HETEROCYCLES, Vol. 71, No. 9, 2007, pp. 2049 - 2054. © The Japan Institute of Heterocyclic Chemistry Received, 27th April, 2007, Accepted, 11th June, 2007, Published online, 12th June, 2007. COM-07-11092

SYNTHESIS OF NEW HETEROCYCLIC SYSTEM: 1(5)*H*-1,5-DIAZACYCL[3.3.2]AZINE-2,4-DIONE

Anton V. Dolzhenko, Anna V. Dolzhenko, and Wai-Keung Chui*

Department of Pharmacy, Faculty of Science, National University of Singapore, 18 Science Drive 4, Singapore 117543, Singapore E-mail: phacwk@nus.edu.sg

unknown heterocyclic Abstract – A synthesis of hitherto system 1(5)*H*-1,5-diazacycl[3.3.2]azine-2,4-dione (5) presented. The is cyclocondensation of 2,6-diaminopyridine (1) with maleic anhydride yielded (5-amino-2-oxo-2,3-dihydroimidazo[1,2-a]pyridin-3-yl)acetic acid (2), which was converted to methyl (5-amino-2-oxo-2,3-dihydroimidazo[1,2-*a*]pyridin-3-yl) acetate hydrochloride (3). The ester (3) upon treatment with sodium carbonate underwent solution intramolecular cyclization to 2a,3-dihydro-1(5)H-1,5-diazacycl[3.3.2]azine-2,4-dione (4), which was oxidized in alkaline condition to provide 5.

INTRODUCTION

Cyclazines are common subjects in theoretical and experimental studies.¹ The diazacycl[3.3.2]azines, which contain the imidazo[1,5-*a*]pyridine pharmacore, can be categorized into three types of systems (**I-III**) (Figure). To date only one report² on the synthesis of 1,3-diazacycl[3.3.2]azines (**I**) has appeared. The chemistry of 1,4-diazacycl[3.3.2]azines (**II**) has been further developed,³ but there is no information on the 1,5-diazacycl[3.3.2]azine heterocyclic system (**III**) available.



As a continuation to our program on the reactions of maleic anhydride with binucleophiles,⁴ particularly in the synthesis of new heterocyclic systems,⁵ we describe herein a method for the preparation of

1,5-diazacycl[3.3.2]azine system (III) derivative, namely 1(5)*H*-1,5-diazacycl[3.3.2]azine-2,4-dione (1,5,8b-triazaacenaphthylen-2,4-dione).

RESULTS AND DISCUSSION

of anhydride The reaction 2-aminopyridines with maleic has been shown afford to (2-oxo-2,3-dihydroimidazo[1,2-a]pyridin-3-yl)acetic acid.⁶ Using a modification of this method we prepared amino acid (2) from 2,6-diaminopyridine (1) and maleic anhydride (Scheme 1). The tautomeric form **2A** solely was observed in the NMR spectra (DMSO- d_6 solution). However, H-3 signal in ¹H NMR spectrum disappeared upon addition of D_2O to DMSO- d_6 solution of 2 that indicated keto-enol tautomerism in the compound (2). The ¹H NMR spectrum of 2 in D_2O contained no H-3 signal, but the signals of the methylenic group appeared as AB system (instead of AMX system in DMSO- d_6). This type of splitting confirmed diastereotopicity of the methylenic group protons that corresponded to the form 2A. The signal of C-3 in ¹³C NMR spectrum appeared as an equally populated triplet at 62.6 ppm (${}^{2}J_{CD} = 22.9$ Hz), which was observed in D_2O solution, also supported the existence of the form 2A.

Attempts of the direct thermal intramolecular cyclization of **2** were not successful. Therefore, the acid (**2**) was treated with thionyl chloride in methanol to afford **3**. The tautomerism, similar to the compound (**2**), was observed for the ester (**3**). Interestingly, N-1 atom was found to be protonated in the compound (**3**) as indicated by cross-peaks between NH (12.89 ppm) and H-8 (6.40 ppm) signals in 2D NOESY experiment.



Scheme 1

The ester (3) upon treatment with sodium carbonate solution in mild condition underwent intramolecular ring closure and provided the tricyclic system 2a,3-dihydro-1(5)H-1,5-diazacycl[3.3.2]azine-2,4-dione (4) (Scheme 2). The compound (4) was readily enolized in alkaline condition and oxidized in aqueous solution by standing exposed to air overnight. After treatment of the resulting product with acetic acid, 1(5)H-1,5-diazacycl[3.3.2]azine-2,4-dione (5) was isolated and characterized. Annular tautomerism (forms

5A and **5B**) was found to be possible along with keto-enol tautomerism. However, we were not able to distinguish between these forms using the spectral data and X-ray crystallographic study might be required.



Scheme 2

In conclusion, an efficient and simple synthesis of new cyclazine type heterocyclic system viz. 1(5)*H*-1,5-diazacycl[3.3.2]azine-2,4-dione was successfully developed. Further investigation of the 1(5)*H*-1,5-diazacycl[3.3.2]azine chemistry is in the progress.

EXPERIMENTAL

General Methods. Melting points (uncorrected) were determined on a Gallenkamp melting point apparatus. IR spectra were performed on a Shimadzu IRPrestige-21 spectrophotometer in KBr discs. ¹H and ¹³C NMR spectra were recorded on a Bruker DPX-300 spectrometer, using DMSO- d_6 or D₂O as solvents and TMS or TSP as internal references. The assignments were made using the reported data for (2-oxo-2,3-dihydroimidazo[1,2-*a*]pyridin-3-yl)acetic acids,⁶ DEPT135, 2D COSY, NOESY and HMQC spectral data.

(5-Amino-2-oxo-2,3-dihydroimidazo[1,2-a]pyridin-3-yl)acetic acid (2).

The freshly recrystallized from toluene 2,6-diaminopyridine (1, 2.18 g, 20 mmol) was dissolved in EtOAc (40 mL) and a solution of maleic anhydride (1.96 g, 20 mmol) in EtOAc (20 mL) was slowly added with stirring at 0°C (ice bath). After stirring at 0°C for 30 min, ice bath was removed and stirring was continued for 12 h at rt. The precipitate was filtered, washed with EtOAc and refluxed in EtOH for 30 min. After cooling, the product (2) was filtered, washed with EtOH and recrystallized from water to give sandy crystals. Yield 2.82 g (68%); mp > 360°C (water).

IR (KBr, v, cm⁻¹): 3550, 3466, 3369, 3102, 1727, 1682, 1634, 1591, 1551, 1407.

¹H NMR (300 MHz, DMSO- d_6 / TMS): δ 2.73 (1H, dd, ${}^{3}J_{AX} = 4.5$ Hz, ${}^{2}J_{MX} = 17.7$ Hz, H_X), 3.14 (1H, dd, ${}^{3}J_{AM} = 3.8$ Hz, ${}^{2}J_{MX} = 17.7$ Hz, H_M), 4.68 (1H, dd, ${}^{2}J_{AM} = 3.8$ Hz, ${}^{2}J_{AX} = 4.5$ Hz, H-3), 6.16 (1H, d, ${}^{3}J = 8.3$ Hz, H-8), 6.19 (1H, d, ${}^{3}J = 8.3$ Hz, H-6), 7.54 (1H, t, ${}^{3}J = 8.3$ Hz, H-7), 8.30 (3H, br. s, NH₃⁺).

¹³C NMR (75 MHz, DMSO-*d*₆ / TMS): δ 34.4 (CH₂), 58.7 (C-3), 94.7 (C-8), 99.4 (C-6), 143.0 (C-7), 151.5 (C-5), 158.1 (C-8a), 171.5 (C-2), 178.4 (COO⁻).

¹H NMR (300 MHz, D₂O / TSP): δ 3.06 (1H, d, ²*J*_{AB} = 17.7 Hz, H_A), 3.33 (1H, d, ²*J*_{AB} = 17.7 Hz, H_B), 6.48 (1H, d, ³*J* = 7.9 Hz, H-8), 6.63 (1H, d, ³*J* = 8.7 Hz, H-6), 7.79 (1H, t, ³*J* = 8.3 Hz, H-7).

¹³C NMR (75 MHz, D₂O / TSP): δ 37.4 (CH₂), 62.6 (t, ¹*J*_{CD} = 22.9 Hz, C-3), 97.7 (C-8), 108.1 (C-6), 147.5 (C-7), 152.0 (C-5), 154.7 (C-8a), 177.6 (C-2), 178.3 (COO⁻).

Anal. Calcd for C₉H₉N₃O₃: C, 52.17; H, 4.38; N, 20.28. Found: C, 51.85; H, 4.72; N, 20.28.

Methyl (5-amino-2-oxo-2,3-dihydroimidazo[1,2-a]pyridin-3-yl)acetate hydrochloride (3).

To a suspension of acid 2 (2.07 g, 10 mmol) in MeOH (30 mL), thionyl chloride (1.46 mL, 20 mmol) was added slowly with stirring at 0°C (ice bath). After stirring at 0°C for 30 min, ice bath was removed and stirring was continued for 4 h at rt. The solvent and excess of thionyl chloride were evaporated under reduced pressure and residue was recrystallized from EtOH to give sandy crystalline product (3). Yield 2.45 g (95%); mp > 360°C (EtOH).

IR (KBr, v, cm⁻¹): 3372, 3075, 1757, 1730, 1675, 1646, 1586, 1410.

¹H NMR (300 MHz, DMSO- d_6 / TMS): δ 3.17 (1H, dd, ³ J_{AX} = 2.7 Hz, ² J_{MX} = 18.1 Hz, H_X), 3.52 (3H, s, COO*Me*), 3.79 (1H, dd, ³ J_{AM} = 4.9 Hz, ² J_{MX} = 18.1 Hz, H_M), 5.27 (1H, dd, ³ J_{AM} = 4.9 Hz, ³ J_{AX} = 2.7 Hz, H-3), 6.40 (1H, d, ³J = 7.9 Hz, H-8), 6.66 (1H, d, ³J = 8.7 Hz, H-6), 7.81 (1H, t, ³J = 8.1 Hz, H-7), 8.82 (2H, s, NH₂), 12.89 (1H, br. s, NH).

¹³C NMR (75 MHz, DMSO-*d*₆ / TMS): δ 30.1 (CH₂), 52.0 (COO*Me*), 57.7 (C-3), 92.7 (C-8), 104.7 (C-6), 144.3 (C-7), 148.5 (C-5), 151.7 (C-8a), 168.6 (COOMe), 170.1 (C-2).

¹H NMR (300 MHz, D₂O / TSP): δ 3.37 (1H, d, ²*J*_{AB} = 17.7 Hz, H_A), 3.63 (3H, s, COO*Me*), 3.68 (1H, d, ²*J*_{AB} = 17.7 Hz, H_B), 6.51 (1H, d, ³*J* = 7.9 Hz, H-8), 6.62 (1H, d, ³*J* = 8.7 Hz, H-6), 7.81 (1H, t, ³*J* = 8.3 Hz, H-7).

¹³C NMR (75 MHz, D₂O / TSP): δ 33.2 (CH₂), 55.7 (COOMe), 60.9 (t, ${}^{1}J_{CD}$ = 22.6 Hz, C-3), 98.7 (C-8), 107.8 (C-6), 148.0 (C-7), 154.2, 154.8 (C-5 and C-8a), 173.6 (COOMe), 178.2 (C-2).

Anal. Calcd for C₁₀H₁₂N₃O₃Cl: C, 46.61; H, 4.69; N, 16.31. Found: C, 46.38; H, 4.87; N, 16.09.

1(5)H-1,5-Diazacycl[3.3.2]azine-2,4-dione (5).

Salt 3 (1.29 g, 5 mmol) was dissolved in 10% aqueous Na_2CO_3 (5.5 mL) was stirred at rt. After 30 min, the precipitated bright yellow compound 4 was filtered and washed with cold water. Without further

purification, compound **4** was dissolved in 3% aqueous Na₂CO₃ (20 mL) and stirred overnight in an open flask. To the formed suspension of sodium salt of **5**, AcOH (5 mL) was added and the orange precipitate was filtered and washed with cold water. The purification of the product (**5**) can be performed using recrystallization from DMSO or, more conveniently, by reprecipitation from the solution in aqueous 10% aqueous Na₂CO₃ using AcOH. Yield 0.77 g (82%); mp > 360°C (DMSO).

IR (KBr, v, cm⁻¹): 3448, 3059, 1735, 1699, 1667, 1617, 1605, 1485, 1456, 1393.

¹H NMR (300 MHz, DMSO- d_6 / TMS): δ 6.69 (1H, d, ³J = 7.9 Hz), 6.86 (1H, d, ³J = 8.7 Hz), 6.99 (1H, s), 7.91 (1H, t, ³J = 8.3 Hz), 13.01 (1H, br. s. NH).

Anal. Calcd for C₁₅H₁₂N₄O₃: C, 57.76; H, 2.69; N, 22.45. Found: C, 57.99; H, 2.82; N, 22.78.

ACKNOWLEDGEMENTS

This work is supported by the Academic Research Fund, National University of Singapore.

REFERENCES

- A. Taurins, 'The Chemistry of Heterocyclic Compounds: The Chemistry of Cyclazines,' Vol. 30, ed. by A. Weissberger and E. C. Taylor, John Wiley, New York, 1977, pp. 245-270; K. Matsumoto, T. Uchida, and J. Yamauchi, *Yuki Gosei Kagaku Kyokaishi*, 1977, **35**, 739; W. Flitsch and U. Kraemer, 'Advances in Heterocyclic Chemistry: Cyclazines and Related *N*-Bridged Annulenes,' Vol. 22, ed. by A. R. Katritzky and A. J. Boulton, Academic Press, New York, 1978, pp. 321-365; D. Leaver, *Pure Appl. Chem.*, 1986, **58**, 143; Y. Tominaga, Y. Shiroshita, and A. Hosomi, *Heterocycles*, 1988, **27**, 2251; Y. Matsuda, *Yakugaku Zasshi*, 2001, **121**, 971; Y. Tominaga, 'Science of Synthesis: Product Class 7: Cyclazines,' Vol. 17, ed. by S. M. Weinreb, Georg Thieme Verlag, Stuttgart - New York, 2004, pp. 1025-1079; Y. Tominaga and K. Ueda, *J. Heterocycl. Chem.*, 2005, **42**, 337.
- 2. J. M. Chezal, E. Moreau, G. Delmas, A. Gueiffier, Y. Blache, G. Grassy, C. Lartigue, O. Chavignon, and J. C. Teulade, *J. Org. Chem.*, 2001, **66**, 6576.
- K. Kurata, H. Awaya, Y. Tominaga, Y. Matsuda, and G. Kobayashi, Yakugaku Zasshi, 1978, 98, 631;
 M. Takatani, Y. Shibouta, K. Tomimatsu, and T. Kawamoto, WO Pat. 9602542, 1996 (Chem. Abstr., 125, 33679);
 M. Wakimasu and T. Ikemoto, JP Pat. 09249666, 1997 (Chem. Abstr., 127, 293222);
 T. Kawamoto, Y. Shibouta, M. Takatani, and M. Noda, EP Pat. 826686, 1998 (Chem. Abstr., 128, 204886);
 S. Imoto, M. Yoshioka, and T. Kashihara, WO Pat. 9829136, 1998 (Chem. Abstr., 129, 113515);
 T. Kawamoto, K. Tomimatsu, T. Ikemoto, H. Abe, K. Hamamura, and M. Takatani, *Tetrahedron Lett.*, 2000, 41, 3447;
 T. Ikemoto, T. Kawamoto, K. Tomimatsu, M. Takatani, and M. Wakimasu, Tetrahedron, 2000, 56, 7915;
 J. S. Chun, E. Ishikawa, M. Kobayashi, and T. Kawamoto, JP Pat. 2002201193, 2002 (Chem. Abstr., 137, 93751);
 K. Kamiyama, N. Kanzaki, A. Hasuoka, M. Mochizuki, and T. Kawamoto, JP Pat. 2002371042, 2002 (Chem. Abstr., 138, 2002)

55742).

- 4. A. V. Dolzhenko, W. K. Chui, and A. V. Dolzhenko, Heterocycles, 2006, 68, 821.
- 5. A. V. Dolzhenko and W. K. Chui, *Heterocycles*, 2004, 63, 2623.
- A. V. Dolzhenko, N. V. Kolotova, V. O. Kozminykh, W. K. Chui, P. W. S. Heng, and V. N. Khrustalev, *Heterocycles*, 2004, 63, 55; M. E. Baumann, H. Bosshard, W. Breitenstein, G. Rihs, and T. Winkler, *Helv. Chim. Acta*, 1984, 67, 1897.