Annealing radiation damage and the recovery of cathodoluminescence

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ABSTRACT

The structural recovery upon heat-treatment of a highly metamict, actinide-rich zircon (U \approx 6000 ppm) has been studied in detail by a complex of techniques including X-ray powder diffraction, Raman spectroscopy, SHRIMP ion probe, electron microprobe, transmission electron microscopy and cathodoluminescence analysis. The structural regeneration of the amorphous starting material depends on random nucleation. It starts between 800 and 950°C when amorphous ZrSiO₄ decomposes to form crystalline ZrO₂ and amorphous SiO₂. At around 1100°C, well crystallised ZrSiO₄ grows at the expense of the oxides. U has been retained in the newly grown zircon whereas Pb was evaporated during the heat-treatment. This process is in marked opposition to the reconstitution of moderately metamict minerals, which experience a gradual recovery controlled by the epitaxial growth at the crystalline-amorphous boundaries. Both of these recovery processes are not the direct inverse of metamictisation. The structural regeneration was found to be connected with a significant increase in the emission of CL. In all cases (annealing heavily damaged zircon and moderately damaged zircon and monazite) we observe that the final, well crystallised annealing products emit more intense CL than their radiation-damaged starting minerals, even though having almost identical elemental composition. Our observations are taken as evidence that the CL is not only determined by the chemical composition of the sample but is also strongly controlled by structural parameters such as crystallinity or the presence of defect centres.

Keywords: radiation damage; thermal annealing; cathodoluminescence; Raman spectroscopy; X-ray powder diffraction; SHRIMP ion probe analysis; transmission electron microscopy

1. Introduction

Self-irradiation over geologic timescales due to the radioactive decay of incorporated actinide elements may transform initially crystalline minerals into an amorphous state that is called "metamict". Even though the study of metamict minerals has a long history (e.g., Hurley and Fairbairn, 1953; Holland and Gottfried, 1955), the process of radiation damage generation and retention in mineral structures has not yet been

fully clarified. One of the main goals of such studies was to understand the causes of different properties of radiationdamaged minerals, such as their generally enhanced chemical reactivity and decreased physical resistance, when compared with their crystalline analogues (e.g., Ewing, 1994; Lumpkin, 2001). In the past, most such research was initiated by geochronologists because understanding the structural causes of the strongly enhanced susceptibility of metamict minerals to secondary Pb loss (Silver and Deutsch, 1963; Nasdala et al., 1998) is crucial for the interpretation of U-Th-Pb dating results. The interest in studying radiation damage phenomena has increased in the past two decades, stimulated inter alia by the necessity to find durable host materials for the safe long-term storage of radioactive waste such as the highly radioactive and poisonous Pu from disarmed nuclear weapons (e.g., Ewing, 1999; Wang et al., 1999).

Natural actinide-bearing minerals such as zircon (e.g., Murakami et al., 1991; Weber et al., 1994) and monazite (Meldrum et al. 1998) may cover the full range from little damaged by radiation to entirely metamict. The real structure of radiation-damaged minerals has been investigated with a wide variety of analytical techniques (Ewing, 1994). Additional information is obtained by studying their behaviour when treated with high temperature, high pressure or in hydrothermal experiments (e.g., Pidgeon et al., 1966; Mursic et al., 1992; Weber et al., 1994; Ellsworth et al., 1994). Annealing experiments were initially done to eliminate the radiation damage step-by-step and, thus, to retrace the formation of metamict structures back to their crystalline origin. We have learned, however, that none of the different recovery processes (for example, long-term annealing over geologic timescales, natural recrystallisation processes, or heat-treatment in the laboratory) is the direct inverse of metamictisation. Annealing experiments are, nevertheless, a powerful tool for the study of metamict structures, particularly in view of the fact that the damage recovery must be strongly controlled by the actual microstructure of the radiationdamaged mineral.

It has already been described in several annealing studies that moderately metamict samples, which have a domain structure consisting of amorphous and little to heavily damaged crystalline nano-areas (Murakami et al., 1991), undergo a gradual recovery process upon annealing (e.g., Colombo et al., 1999a; Zhang et al., 2000b). Their structural recovery is a complicated process that involves, for example, random nucleation in the amorphous phase, epitaxial growth of the crystalline domains in favour of their surrounding remnant amorphous areas, recombination of Frenkel defect pairs in the crystalline phase through point defect diffusion (e.g., Weber et al., 1994), structural recovery as a result of He diffusion and loss (cf. Seydoux-Guillaume et al., 2002a), and the escape of hydrous species (Nasdala et al., 2001b). Even though these processes may have different relationships depending on temperature, duration of heating and other conditions, the heating-induced gradual recovery of moderately metamict minerals is most probably strongly controlled by the epitaxial growth at crystalline-amorphous boundaries.

Highly amorphised minerals, however, react differently to the heat-treatment. Because crystalline remnants are lacking, their epitaxial growth is impossible and the structural recovery must depend on the nucleus formation. In addition, it has already been observed in several studies that the recovery of highly metamict zircon involves an intermediate stage of ZrO_2 formation (e.g., Weber, 1993; Weber et al., 1994; McLaren et al., 1994; Capitani et al., 2000; Begg et al., 2000; Zhang et al., 2000b, c). However, no comparative documentation of this process comprising long-range and short-range methods has been documented so far. In the present paper we report a detailed investigation of the structural regeneration of a highly metamict zircon upon heat-treatment.

The products of our and previous annealing experiments were also subjected to a CL (cathodoluminescence) investigation. It is still controversial as to which features of the chemical composition and/or structural state affect the generation of luminescence when irradiating minerals with an electron beam. Cathodoluminescence images of minerals allow one to recognise internal zones and micro-areas of different chemical composition (e.g., Vavra, 1990; Paterson et al., 1992; Hanchar and Miller, 1993). It has also been demonstrated that the CL correlates closely with the pattern of heterogeneous metamictisation (Nasdala et al., 2001a). It has, however, not been clarified so far whether (1) the latter is only an apparent correlation, caused by the dependence of metamictisation on the chemical zoning that controls the CL, or (2) the structural radiation damage has causal effects on the CL emission. This question can be answered by investigating samples having the same chemical composition but different degrees of radiation damage. Annealing products seem to be an ideal case for this and, therefore, we have studied them by means of CL imaging and spectral CL.

2. Experimental

In this paper we mainly investigate the annealing of a highly metamict but, nevertheless, gemstone-quality zircon from Sri Lanka (sample number N17). Results of CL investigations for three other samples that have been annealed and characterised in previous studies (zircons K1 and K2, Nasdala et al., 2001b; monazite Moacir; Seydoux-Guillaume et al., 2002a) are additionally reported.

Zircon N17 is a clear, dark green, cut stone of 4.6 ct (0.92 g) weight and about $7.5 \times 11 \times 5$ mm size. Slices between 200 µm and 1 mm in size were used for heating experiments in air. We did not do stepwise heating, rather we started always with fragments of the untreated zircon. The slices were placed in a 1 ml Pt crucible and heated up to the desired annealing temperature (500, 800, 950, 1100, 1250, 1400 and 1500°C) at a rate of \approx 30 degrees per minute. At the end of the 150 h run, the furnace was switched off. After about 1 h the temperature had decreased to ca. 300-500 °C and then the furnace door was slightly opened, thus again increasing the cooling speed. The crucible was taken out of the furnace after another 30 min (then at \approx 100-150°C) and cooled down to room temperature within a few minutes.

Cathodoluminescence images were taken using an OXFORD INSTRUMENTS CL System interfaced to a JEOL 8900 RL electron microprobe. The accelerating voltage was 20 kV and the beam current was 50 nA. To avoid effects of varying experimental conditions on the observed CL intensity, samples to be compared were brought simultaneously in the vacuum chamber and were measured under identical conditions, i.e. equally sized areas were scanned for the same time and images were obtained with constant signal amplification. Conclusions about absolute CL intensities are still limited, however, relative comparison is possible between simultaneously obtained images. Because chemical compositions have been determined in previous studies (Nasdala et al., 2001b; Seydoux-Guillaume et al., 2002a), only two fragments zircon N17 (untreated and annealed at 1400°C) were subjected to multiple electron microprobe analyses, to check again for potential compositional changes upon annealing

Cathodoluminescence spectra of samples N17 and Moacir were obtained using a JEOL 840A scanning electron microscope operated at 35 kV and 50 nA. The luminescence signal was led into an external optical spectrometer by placing an OPEA parabolic collector in the beam path. Signal analysis was done by means of two different external systems. Spectra in the range 200-500 nm were obtained using an INSTRUMENTS S. A. H10UV spectrometer equipped with GaAs cathode-based photomultiplier. The spectral resolution was 8 nm. Spectra in the range 300-900 nm were obtained using a JOBIN YVON TRIAX triple monochromator equipped with liquid nitrogen-cooled CCD (charge-coupled device) detector, with a spectral resolution of better than 4 nm. All spectra were normalised to the quantum efficiency of the respective detector.

U, Th and Pb contents and U-Pb isotopic ratios for the zircons were determined by means of the SHRIMP (Sensitive High mass Resolution Ion MicroProbe) II at the Department of Applied Physics, Curtin University of Technology, Perth. The general SHRIMP technique has been described in more detail elsewhere (e.g., Compston et al., 1984; Vavra et al., 1996; Nelson, 1997) and the performance characteristics of the Curtin SHRIMP II was documented by De Laeter and Kennedy (1998). The SHRIMP was operated at 10 kV and with a mass resolution (M/ Δ M) better than 5000, and the sensitivity for lead isotopes was in the range 14 to 20 counts per second per ppm. Data were calibrated using the Curtin University CZ3 standard zircon (Pidgeon et al., 1994). The ²⁰⁴Pb method was employed for the correction for common Pb. For further experimental details see Nasdala et al. (1999b).

Raman microprobe measurements were done using a JOBIN YVON LabRam-HR (high resolution; this long-geometry version of the LabRam system has, due to its focal length of 850 mm, a threefold improved spectral resolution). The Raman system was equipped with OLYMPUS optical microscope (100× objective; numerical aperture 0.9). Spectra were excited with the He-Ne 632.8 nm line (3 mW at the sample surface). The wavenumber accuracy was ± 0.5 cm⁻¹ and the spectral resolution was about 0.4 cm⁻¹. The degree of metamictisation of zircons was estimated from the FWHM of the B_{1g} Raman band at about 1000 cm⁻¹ (Nasdala et al., 1995). Such quantitative estimation of the radiation damage is not possible in the case of monazite, because chemical variations may greatly affect the Raman spectrum of this mineral. We have, however, used the FWHM of the monazite A_{1g} mode at ≈ 974 cm⁻¹ as a relative measure of the recovery of the short-range order upon annealing (cf. Seydoux-Guillaume et al., 2002a). For further experimental details, correction of measured FWHMs for the apparatus function and errors in the estimation of metamictisation from Raman data see Nasdala et al. (2001a).

X-ray powder data for zircon N17 and its annealing products were collected on a PHILIPS PW 3050 based X'PERT MPD diffractometer in Debye-Scherrer geometry equipped with an inhouse built capillary spinner and a RAYTECH PSD (position sensitive detector). Zircon fragments were ground in an agate mortar and pestle and filled into a Linde-glass capillary (Hilgenberg, no. 14) with an inner diameter of 0.28 mm. For the monochromatic, parallel-beam Cu K α X-ray source ($\lambda = 1.54056$ Å) a ceramic-type X-ray copper tube (45 kV, 40 mA) with a combination of a multilayer X-ray mirror and a channel-cut Ge crystal (PHILIPS) was used. A soller slit of 1.15 ° axial

divergence was inserted on the primary side to reduce the peak asymmetry. The area of the primary X-ray beam irradiating the glass capillary was limited to $0.7 \times 6.0 \text{ mm}^2$, thus resulting in an illuminated sample volume of roughly 0.4 mm³ per sample. The goniometer radius of the secondary side was 343.8 mm, which corresponds to an angular resolution of 6 mm/° 20 and a physical detector resolution of 0.011 °20. Measurements were performed at 24(2)°C in a continuous mode, using a twofold data collection, each set over the range 10-100 $^{\circ}2\theta$ with a step interval of 0.011 °20, a scan time of 11.6 h, and an angular PSD opening of 4 °20. This was equivalent to a summed counting time of 3600 sec/step. The determination of cell parameters and FWHMs was done using the PC-Rietveld Plus package of Fischer et al. (1993). Standard deviations were corrected according to Berar and Lelann (1991). The background was set manually and the absorption correction was calculated with the values for cylindrical samples according to Dwiggins (1975). For the peakshape simulation, the pseudo-Voigt function was used. Angular dependencies of FWHMs were detemined using the formula of Cagliotti et al. (1959). Semiguantitative results were calculated as described by Hill and Howard (1987). For the Rietveld refinements the atomic coordinates and displacement factors given by Mursic et al. (1992) and Howard et al. (1988) were used. Ionic X-ray scattering factors were taken from Cromer and Waber (1974; for Zr^{4+} and Si^{4+}) and Hovestreydt (1983; for O^{2-}).

TEM (transmission electron microscope) investigations were done by means of a PHILIPS CM200 system equipped with EDAX X-ray analyser and GATAN imaging filter. Samples were first grind and then prepared by conventional Ar ion beam thinning and subsequently carbon coated. The electron microscope was operated at a voltage of 200 kV. For more experimental details see Wirth et al. (2001a, b).

3. General characterisation of samples

A general characterisation of the chemical compositions of the four mineral samples under investigation is presented in Table 1 and structural parameters (powder diffraction and Raman data) are listed in Table 2. Note the comparably high actinide content of all four samples.

Zircon N17 has an X-ray powder diffraction pattern (Fig. 1A) that lacks any Bragg peaks and the Raman spectrum (Fig. 1B) shows a pattern of extremely broad bands, which is typical of an amorphous material (cf. Zhang et al., 2000a). Correspondingly, HREM (high resolution electron microscopic) image and electron diffraction pattern did not show any indication of remnant lattice order (Fig. 2A). In Fourier-filtered HREM images, however, a pattern of slightly misoriented domains can be seen (see small inset in Fig 2A), which indicates that the sample is not perfectly amorphous but there exists some residual short-range order. A thin (25 µm) platelet of zircon N17 showed very low interference colours (not extending grey I) in cross-polarised light (cf. Chakoumakos et al., 1987). The density of this zircon exceptionally low with $\sigma = 3.86$ g/cm³ (cf. Koechlin, 1903; Sahama, 1980). All of these observations suggest that zircon N17 is highly metamict.

Four SHRIMP analyses (Table 3; Fig. 3) gave a concordant, weighted mean 207 Pb/ 206 Pb age of 551 ± 7 Ma (95 % confidence) for zircon N17. This Cambrian age is consistent with other ages

reported for Sri Lankan zircons (Kröner et al., 1987, Hölzl et al., 1991). From actinide content and age, a time integrated selfirradiation dose of 10.6×10^{18} α -events per gram is calculated. Generally incomplete damage retention has already been reported for zircons from the Sri Lankan Highland Southwestern Complex (Nasdala et al., 2001a), suggesting that the high α -fluence above, which was calculated using the zircon U-Th-Pb age, overestimates by about two times the α -dose that has effectively caused the presently observe radiation damage. This explains that the zircon is heavily damaged but still not fully amorphised. The Pb loss in zircon N 17 is, in spite of the high self-irradiation dose and structural radiation damage, still insignificant.

The X-ray diffraction patterns of samples K1 and K2 showed significantly broadened Bragg peaks when compared with well crystallised zircon. The notably expanded unit cell (compare Murakami et al, 1991) and the moderate Raman band shift and broadening (compare and Nasdala et al., 1995, and Zhang et al., 2000a) of K1 characterise this zircon as moderately metamict (see Table 2). Zircon K2 has a lower degree of unit cell expansion and Raman band broadening and, therefore, represents a low degree of metamictisation. Zircons K1 and K2 have concordant $^{206}\text{Pb}/^{238}\text{U}$ ages of 324 \pm 4 Ma and of 552 \pm 5 Ma, respectively. We have calculated α -fluences of 3.1 \times 10¹⁸ α/g (zircon K1) and 4.3×10^{18} α/g (zircon K2), respectively. Zircon K2 must have experienced extensive structural recovery through long-term annealing, because it is notably less radiation-damaged than zircon K1 (see Table 2) even though it has experienced a higher self-irradiation dose. In addition, both of these zircons are less metamict than would correspond to their U and Th contents and ages (cf. Nasdala et al., 2001a).

Monazite Moacir has a low to moderate degree of radiation damage. Its X-ray powder pattern actually consisted of two different crystalline monazites, which were interpreted by Seydoux-Guillaume et al. (2002a) to represent the two major phases of a mosaic structure. The age of Moacir monazite was determined at 474 ± 1 Ma (Seydoux-Guillaume et al., 2002b). The observed moderate radiation damage of Moacir is in clear contradiction to the fact that this monazite has experienced a selfirradiation dose of 24.3×10^{18} α/g since the time of closure of its U-Th-Pb system, which is high enough to displace all lattice atoms several times. Consequently, a major portion of the radiation damage must have been annealed. The simultaneous observation of U-Pb concordance and incomplete retention of the radiation damage in all samples underlines again that long-term, low-T annealing is not necessarily connected with the loss of radiogenic Pb (cf. Nasdala et al., 2001a).

4. Annealing radiation damage

4.1. General remarks

Corresponding with previous findings of a gradual regeneration of moderately metamict minerals upon heattreatment (e.g., Colombo et al., 1999a; Zhang et al., 2000b), the three little to moderately metamict samples K1, K2 (Nasdala et al., 2001b) and Moacir (Seydoux-Guillaume et al., 2002a) were found to recover gradually with temperature. There is, however, a clear difference in the temperature-dependencies of the annealing processes of the two minerals species. Most of the structural recovery of monazite Moacir has occurred between 500 and 800°C and its reorganisation is complete at 900°C. In contrast, zircons K1 and K2 were found to be still notably radiation-damaged at 800°C and relatively well crystallised at 1200°C (compare Fig. 4C). This corresponds with the observations of Capitani et al. (2000) who found that a significant portion of the structural recovery of moderately metamict zircons occurs well above 1000°C. Zhang et al. (2000b) demonstrated that the recovery process of little to moderately metamict zircons starts below 500°C and is not yet completed at 1300°C. The difference in the annealing temperatures of monazite and zircon has already been discussed, for example, by Meldrum et al. (1998).

Zircon N17, however, reacted differently to the heattreatment. Its structural recovery was not gradual but involved an intermediate stage of $ZrSiO_4$ decomposition and fully recrystallised zircon was observed after annealing at 1250°C and above. Results are listed in Table 3, X-ray diffraction patterns and Raman spectra are shown in Fig. 1 and TEM images and electron diffraction patterns are compiled in Fig. 2.

4.2. Annealing up to 800°C

After annealing zircon N17 at 500°C, both the X-ray powder pattern and the Raman spectrum are virtually identical with those of the untreated zircon and have the typical appearance of a fully non-crystalline compound. Patterns of the 800°C sample are similar, suggesting that the sample is still mainly amorphous. There are, however, additional weak, broad bands of tetragonal ZrO₂ and ZrSiO₄ in the Raman spectrum (Fig. 1B). Correspondingly, three extremely broad and diffuse haloes around d-spacings 2.95, 1.82 and 1.54 Å in the X-ray pattern, which correspond to the three most intense diffraction peaks [(101), (112) and (103) + (211)] of tetragonal ZrO₂, indicate some initial three-dimensional periodicity over several unit translations. We take this as evidence for initial nucleation at 800°C.

Note again that our observations, characterising sample N17 as still mainly amorphous at 800°C, are in apparent contradiction to the annealing of moderately metamict zircon. After heating at 800°C, zircon K1 was, although still being moderately disordered, notably less radiation-damaged than before the treatment (see Raman data in Fig. 4C). We explain the different "onset temperatures" of the recovery by the reflection that the low-temperature epitaxial growth of crystalline domains is impossible in the highly metamict zircon N17 because of the lack of crystalline $ZrSiO_4$ remnants. Therefore, its structural reconstitution depends on random nucleation, which does not start significantly below 800°C.

4.3. Annealing at 950°C: an intermediate stage

After annealing at 950°C, the amorphous zirconium silicate has experienced dramatic structural changes. The sample consists of a multitude of randomly oriented, round crystals (some ten nm in size) embedded in an amorphous matrix (Fig. 2B and C). The crystalline fraction is dominated by monoclinic ZrO_2 exhibiting strong peak broadening in the X-ray diffraction pattern (Fig. 1A). The refined unit cell parameters (Table 2) deviate significantly from literature data for monoclinic ZrO_2 , which s is probably

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mainly due to the poor crystallinity. Furthermore, relatively sharp peaks of (relatively well ordered) tetragonal ZrO₂ are resolved in X-ray and Raman patterns. The formation of tetragonal ZrO₂ normally occurs only well above 1100°C, however, it is also known that metastable, tetragonal ZrO₂ is preferentially formed at very small crystallite sizes (Garvie, 1965; Mitsuhashi et al., 1974). It must also be cautiously considered that impurities may potentially stabilise the tetragonal ZrO₂ at lower temperatures.

It seems a bit surprising that only the tetragonal ZrO_2 is observed in the Raman spectrum whereas the monoclinic ZrO_2 is not (compare the spectrum of the 950°C sample in Fig. 1B with that of the monoclinic ZrO_2 reference). We suspect that the Raman signal of the poorly crystallised monoclinic ZrO_2 is obscured by the much more intense signal of the relatively well crystallised tetragonal ZrO_2 . This conclusion is supported by the observation of dramatic intensity losses (about two orders of magnitude) of Raman bands with decreasing crystallinity (for zircon documented by Nasdala et al., 2001a).

None of the further known cubic, orthorhombic, or rhombohedral ZrO_2 polymorphs (cf. Howard et al., 1990) were found by consecutive Rietveld refinement trials. The minor presence of $ZrSiO_4$ is indicated by low-intensity zircon Raman bands (Fig. 1B) and a small deviation from the background noise at d-spacing 3.30 Å in the X-ray diffractogram, corresponding to the main zircon (200) peak. Calculation of cell parameters, however, was impossible.

A crystalline SiO₂ polymorph or another SiO₄-bearing phase is neither observed in the X-ray diffraction pattern nor in the Raman spectrum. Energy dispersive element analyses and dark field images obtained in the TEM, however, showed that Si is concentrated the amorphous phase whereas Zr was only detected in the round nanocrystals (compare Figs. 2B and C). The amorphous ZrSiO₄ has obviously decomposed to form crystalline ZrO₂ and amorphous SiO₂. Capitani et al. (2000) explained this by considering that the ionic bond character of ZrO₂ strongly favours the crystalline state whereas bonding of SiO₂ is predominantly covalent.

The decomposition of zircon into oxides has different effects to non-formula elements. U is preferentially incorporated in the ZrO₂ phase whereas Pb is concentrated in the SiO₂ phase. This can be seen from the behaviour of the 950°C sample when analysed in the SHRIMP. The impact of the oxygen beam causes enhanced sputtering of amorphous SiO₂ whereas the crystalline ZrO₂ is more resistant. Too much Pb is thus released and we obtain, as an analytical artefact, apparently increased Pb concentrations (Table 3). Consequently, U-Pb isotopic ratios determined with the normal calibration equations show strong reverse discordance (Fig. 3; compare McLaren et al., 1994). It will be worthwhile to subject intermediate annealing products to precise chemical analyses, in order to check whether the entire Pb content of the original zircon has been completely transferred into the SiO₂ phase or Pb has already partially escaped.

4.4. Annealing at 1100°C

At 1100 °C the two ZrO_2 polymorphs have become minor constituents with decreased crystallinity (see FWHMs Table 2) and $ZrSiO_4$ is the main phase. Zircon Bragg peaks are also significantly broadened (Fig. 1A). The hypothesis that this broadening may be due to moderate radiation damage and, thus, indicate incomplete damage recovery at this stage is disproved by the facts that (1) this zircon is newly grown and (2) its unit cell parameters are identical within their errors with those of the fully reorganised zircon (Table 2). Instead, we explain the broadening of Bragg peaks by the very small crystal size of the newly grown ZrSiO₄, which is probably on the order of ≤ 10 nm.

This interpretation is supported by the Raman results. In both the 950°C and the 1100°C Raman spectrum, the frequency of the $\nu_3({\rm SiO}_4)$ band shows a notable shift by about 2 $\text{cm}^{\text{-1}}$ toward higher wavenumbers, when compared with crystalline zircon (Table 2). Recall that the loss of crystallinity upon metamictisation causes vibrational frequencies to shift toward *lower* wavenumbers (Nasdala et al., 1995). For instance, $v_3(SiO_4)$ may decrease from ≈ 1008 cm⁻¹ (Dawson et al., 1971; Hoskin and Rodgers, 1996) to $< 990 \text{ cm}^{-1}$. When annealing moderately metamict zircon, the gradual structural recovery is connected with gradual recovery of the band frequencies and $v_3(SiO_4)$ finally increases back to the initial value of $\approx 1008 \text{ cm}^{-1}$ (see Zhang et al., 2001b). It has, however, never been reported before that vibrational frequencies of intermediate annealing products can surpass those of well crystallised zircon. The now observed $v_3(SiO_4)$ frequency of 1011 cm⁻¹ is, therefore, most remarkable. We explain the slight shift toward higher vibrational energies by the compressive strain in small ZrSiO₄ particles, which increases with decreasing particle size. This effect has already been reported for other materials, e.g., by Scamarcio et al. (1992) and Hwang et al. (1996).

4.5. Annealing at 1200°C and above

All three samples that were heat-treated at 1250°C and above consist of crystalline ZrSiO₄ with a crystal size on the order of some hundred nm (Fig. 2D). Because the scale factors of the Rietveld refinements are very similar (Table 2), we may assume that almost all previous non-crystalline parts of the sample are recrystallised at 1250°C, in favour of the remnant ZrO₂. Small differences can only be observed for the FWHM, which is, however, very close to the theoretical resolution of our X-ray measurements (0.047 °2 θ). The cell parameters of the recrystallised zircon are slightly increased in comparison to synthetic ZrSiO₄ (see Table 3). We explain this difference by the non-ideal chemistry of the zircon .

Analogously, the Raman spectra at 1250°C and above show only the bands of well crystallised zircon and differences between the three annealing products do not extend beyond the analytical uncertainties. Note that the Raman parameters of the annealing product show only minor variations from those of pure, synthetic zircon. In particular the $v_3(SiO_4)$ FWHM deviates by less than $+1 \text{ cm}^{-1}$ from pure ZrSiO_4 (Table 2). It is obvious that the mere presence of some $10^3 \mbox{ ppm U}$ at the ${}^{[8]}\mbox{Zr}^{4+}$ sites has only minor effects on the Raman spectrum. In contrast to natural monazites, where internal PO₄ vibrations may be strongly broadened depending on potentially extensive chemical variations (Podor, 1995; Nasdala et al., 1999a), Raman band broadening in natural zircons is mainly due to structural effects such as radiation damage and chemical effects can mostly be excluded. This confirms again that the Raman band broadening in zircon can be used to estimate the degree of structural radiation damage.

Multiple electron microprobe analyses on the untreated, highly metamict sample and the recrystallised zircon (Table 1) demonstrate that the heat-treatment did not cause major changes of the chemical composition, except for the escape of Pb. Considering this observation and the fact that the elemental composition of zircons is relatively easily altered in hydrothermal experiments (Pidgeon et al. 1966; Rizvanova et al., 2000; Geisler et al., 2001), we suppose that both the lack of a suitable transporting medium (e.g., chemically "aggressive" fluids, etc.) and the lack of diffusion pathways may have prevented notable migration of elements. The degree of Pb loss, however, is obvious from the SHRIMP results. Calculated Pb/U ratios are close to zero, with huge errors greatly extending the values. Obviously the heat-treatment has fully reset the U-Pb system of the zircon.

This reset may also account for the occurrence of tiny SiO_2 inclusions in the recrystallised zircon (Fig. 2E). Due to the Pb escape, the initial equilibrium between large (Zr, U, Pb) and small (Si) cations is disturbed and the sample has now a slightly non-stoichiometric composition. Obviously the excess Si was not incorporated in the newly grown zircon but has formed a separate silica phase.

4.6. Discussion

The reconstitution of a highly metamict zircon involves an intermediate stage of decomposition into oxides. An energetic explanation for this phase separation was given by Ellsworth et al. (1994) who discussed that the difference in enthalpy between highly radiation-damaged and crystalline zircon is much greater than between metamict zircon and a mixture of tetragonal ZrO₂ and glassy SiO₂. Also, Ellsworth et al. (1994) stated that the separation into oxides would be favoured if the metamict starting material had a non-uniform structure comprising ZrO₂- and SiO₂rich domains, because ZrO2-rich regions would recrystallise more easily upon heat-treatment. Such structure was described, for example, by Vance and Anderson (1972) who explained the anomalous optical absorption of metamict zircon by the presence of tiny crystalline ZrO₂ particles in the radiation-damaged ZrSiO₄. Decomposition into oxides, however, does not seem to depend on the hypothetical presence of ZrO₂ domains in the starting material, because it was also observed when annealing highly radiation-damaged zircons without this feature (Capitani et al., 2000; this study).

We found the formation of well crystallised $ZrSiO_4$ to be nearly completed at 1250°C in our experiments. We have also observed at 1100°C that the oxide content is lowered and the content of zircon nanocrystals is greatly increased with respect to the 950°C sample, so the ZrSiO₄ formation must have started even below 1100°C. Consequently, our results suggest that the temperature interval in which the intermediate oxide stage is realised extends between > 800°C / < 950°C and roughly 1100°C. This is in agreement with the results of Begg et al. (2000) who reported that amorphous, Pu-substituted zircon had decomposed into oxides at 1000°C whereas at 1200°C zircon had fully recrystallised.

Our results, however, do not correspond with the results of McLaren et al. (1994). These authors observed rounded monoclinic ZrO_2 crystallites in a glassy silica matrix after annealing a highly metamict zircon for 5 h at 1250°C. A detailed

explanation for these different results cannot be given. One may suspect that stepwise annealing and annealing of untreated zircon may give different results. It seems to be most likely that ZrSiO₄ formation after phase separation at lower temperatures requires higher energies than ZrSiO₄ recrystallisation directly from amorphous zircon (e.g., Begg et al., 2000). When heating amorphous zircon to 1250°C, it is most probably quickly driven through the temperature range of oxide nucleation and, therefore, ZrSiO₄ nucleation in the amorphous zircon will become more important. However, this potentially different behaviour cannot be applied to explain the different results of McLaren et al. (1994) and the present work, because untreated fragments were heated in both studies. There is certainly some doubt about the accuracy of the temperature calibration (i.e. whether or not their and our 1250°C experiments were really done at exactly the same temperature). Also, one might suspect that extending the annealing times from 5 h (McLaren et al.) to 150 h (his study) may have allowed for slow recrystallisational processes. Even though annealing experiments have often been done for only < 2 h (e.g., Woodhead et al., 1991; Biagini et al., 1997; Colombo et al., 1999a), it is well known that structural parameters show notable changes when extending the annealing time (Ellsworth et al., 1994; Colombo et al., 1999b; Zhang et al., 2000b). This suggests that annealing times of only a few hours may result in incomplete reaction.

Another apparent contradiction is found when comparing our results with the results of Capitani et al. (2000). These authors found zirconia grains in an amorphous silica matrix after heat-treating a highly metamict, natural zircon at 1400 K (1127°C) but did not report zircon at this temperature. Their sample, however, was richer in actinides (UO₂ 0.86 wt-%) and also even more radiation-damaged than the zircon studied here, which may have caused slightly different annealing behaviours. The different experimental conditions may also have affected the results: Capitani annealed their zircon for 16 h and in an N₂ atmosphere whereas in the present work the experiments lasted 150 h and were done in an open furnace (air).

5. Cathodoluminescence results

Our and previous heating experiments (Nasdala et al., 2001b; Seydoux-Guillaume et al., 2002a) have produced small series of mineral samples with almost identical chemical compositions but different structural states. Apparently complete Pb escape during the high-temperature treatment is the only major chemical difference observed in the present study. It should, however, be cautiously considered that many trace elements, which potentially may also cause or suppress luminescence, were below the detection limit of the electron microprobe. In spite of this uncertainty, comparison of crystalline annealing products and their radiation-damaged starting materials allows us to investigate the dependence of CL on metamictisation without having to consider major chemical effects.

To check for general changes of the total CL intensity, CL images were obtained for fragments of the same sample annealed at different temperatures (Fig. 4C). We found that the total CL signal of the annealed fragments is always higher in intensity when compared with the image of the respective untreated, radiation-damaged sample. This observation suggests that the CL

emission is generally suppressed in radiation-damaged structures. The CL recovery seems to be almost completed at 800°C in Moacir monazite (Seydoux et al., 2002a) whereas the CL of zircons K1 and K2 shows significant changes above this temperature (Fig. 4C). This difference in CL recovery concurs with the different annealing behaviours of the two minerals discussed above.

More precise conclusions, however, require the spectral analysis of the CL emission. This was done for several chips of samples N17 (and, in addition, also for monazite Moacir). For the interpretation of spectra, it is necessary to reconsider briefly the origin of CL. The CL emission commonly observed in crustal zircon crystals comprises two different types. The first type is an "intrinsic" broad-band emission centred either in the blue or yellow region of the electromagnetic spectrum. Its origin is still controversial. Most probably it is defect-related, for example to point defects or local charge-imbalance due to impurities (e.g., Ohnenstetter et al., 1991; Kempe et al., 2000) or electronic defects related to SiO₄ tetrahedra (Cesbron et al., 1993; 1995). The second type are comparably narrow REE (rare earth element) emission peaks (e.g., 4f electronic transitions of trivalent REE). These emissions are often observed in clusters of bands, then pointing to a variety of REE site symmetries, and may either be superimposed on the broad-band emission or a flat background. Among the REEs, emission bands of Dy³⁺ are a particularly typical feature of zircon (Mariano, 1978; 1988; 1989; Remond et al., 1992; Hanchar and Rudnick, 1995). Trivalent Tb, Gd³⁺ and Y ³⁺ may also be CL-activated elements in this mineral (e.g., Ohnenstetter et al., 1991; Yang et al., 1992; Blanc et al., 2000).

Fig. 5 shows the CL spectra obtained from zircon N17. After structural reorganisation through annealing at 1400°C, the integrated CL intensity has increased by about two orders of magnitude. No significant yellow broad-band emission is detected in the 1400°C spectrum and the blue broad-band intrinsic emission is comparably low in intensity and is centred in the ultraviolet. The spectrum is dominated by narrow emissions related to Gd³⁺, Dy³⁺ and Tb³⁺. The fact that these bands are extremely low in intensity in the 25°C and 1100°C spectrum, even though the same amount of REEs is present, may hypothetically be explained in two different ways. Firstly, it may be assumed these REEs are not CL-activated elements when in the metamict zircon, which could, for example, be due to ionisation effects caused by the self-irradiation. Secondly, it is possible that the CL is "quenched" in the damaged structure, for example by the highly disturbed site symmetries of the REEs or electronic defects in their neighbouring range. As Pb loss from the 1400°C sample was observed, it cannot be ruled out that traces of radiogenic Pb may have some suppressing influence on the CL signal. Even the presence of He (implanted α -particles) in the low-luminescent, metamict zircon, which has certainly also escaped at 1400°C, should be considered. Dehydration of solids may also increase their CL (e.g., Gutzov and Peneva, 1995), however, this effect is of minor importance in the studied zircons because of their comparably low content of hydrous species (Nasdala et al., 2001b).

For comparison, CL spectra of monazite Moacir were also obtained. All spectra (Fig. 6) have a similar pattern that is dominated by five Nd³⁺ bands in the range 860-895 nm. The CL intensity of the annealed sample is clearly enhanced, when

compared with the untreated, radiation-damaged sample. It is well known that minor CL intensity variations may be affected by the crystal orientation of zircon (Cesbron et al., 1995) and apatite (Barbarand and Pagel, 2001). The intensity gain observed here, however, surpasses potential orientation effects. The intensity gain is only about $3 \times$ for the Nd³⁺ bands but $10 \times$ for the 330 nm lattice emission and must be due to the reconstitution of the monazite structure. Note the clear correlation between the CL intensity and the recovery of short-range order, in Fig. 6 demonstrated with the decreasing FWHM of internal PO₄ vibrations.

Poller et al. (2001) have proposed that "U (and possibly Hf) and related radiation damage" may suppress the CL intensity. Our results disprove the hypothesis that the presence of U is the main reason for decreased CL emission of natural zircons. Recall that the CL intensity of zircon N17 is greatly increased after annealing at 1400°C, even though the U content is about the same. We suspect that the anti-correlation of CL intensity and U content observed by Poller et al. (2001) is most probably a link between the radiation damage caused by U and the CL suppression due to radiation damage. In particularly actinide-rich minerals such as natural monazites, however, high levels of U⁴⁺ may play a role in quenching the CL emission.

Consideration of the crystallinity-dependence of CL may perhaps also prove helpful for the interpretation of apparently contradicting BSE (backscattered electrons) and CL images. BSE images reveal, greatly simplified, contrasts in average atomic number of a phase. The element primarily responsible for these BSE variations in crustal zircons is Hf, with U having a secondary effect (Hanchar and Miller, 1993). It is well known from numerous BSE/CL imaging studies (e.g., Vavra et al., 1990; Paterson et al., 1992; Hanchar and Miller, 1993; Koschek, 1993; Hanchar and Rudnick, 1995; Rémond et al., 1995) that both techniques reveal mostly the same internal features, however, usually the bright areas in CL are dark in BSE and vise versa (see Fig. 4A and B). Observations of positively correlated BSE and CL in patchily recrystallised zircons (see Fig. 2c-d in Kempe et al., 2000) or differing BSE and CL patterns can be explained if we consider that processes causing heterogeneous structural changes may have different effects on the CL and BSE patterns.

The results of our CL study give strong evidence that the CL is not only controlled by the presence of REEs and other chemical factors but has also great structural dependence. This is consistent with previous findings that the CL emission greatly increases when amorphous solids crystallise (e.g., Gutzov et al., 1998). Cathodoluminescence generation is, however, probably even more complicated than has previously been thought. In our study, we have observed that the CL is notably suppressed in significantly radiation-damaged samples and recovers upon annealing. The opposite case, however, is also possible. Partial reorganisation of highly radiation-damaged zircon under the impact of an electron beam is not connected with CL recovery but, in contrast, the CL emission is even more suppressed in the better crystallised area (Fig. 4 D and E). It is also well known that radiation damage in minerals may result in enhanced CL emission (for example, see Owen, 1988; Meunier et al., 1990). Correct assignment of radioactivity-induced CL changes will, therefore, require a much more detailed investigation of CL spectra obtained from well characterised, natural and synthetic mineral samples.

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Table 1 General chemic analyses	al compo	sition of	the inves	tigated m	inerals as	determine	d by mea	ns of elec	tron micr	oprobe ar	ialysis. Va	ılues (in w	t.%) repre	sent mea	ns of mul	tiple point	
Zircon	n0. ^(a)	ZrO	² F	IfO_2	ThO_2	UO_2	PbO	Ce ₂ (0 ₃ Y	${}^{1}_{2}O_{3}$	Yb_2O_3	K_2O	SiO_2	Al_2	0 ₃ F	$^{2}O_{5}$	Total
$K1^{(b)}$	9	65.	1 0	.70	0.73	0.14	0.02	0.0	9	0.30	0.05	n.d.	32.2	0.0	02 (0.05	99.4
K2 ^(b)	12	66.(0	.07	0.08	0.24	0.02	0.0	1 (0.04	0.03	0.02	31.8	0.0)5 (0.03	99.4
N17 (25°C)	9	65.(0	.95	0.04	0.75	0.06	n.c	I. (0.04	n.d.	n.d.	32.8	0.0	02 (0.05	100.7
N17 (1400°C)	8		3 1	.94	0.04	0.72	n.d.	n.č	l. ().04	n.d.	n.d.	32.8	0.0)2 (.05	100.0
Monazite Moacir ^(c)	no. ^(a) 30	ThO ₂ 6.92	UO ₂ 0.13	PbO 0.16	Ce ₂ O ₃ 30.6	La ₂ O ₃ 14.5	Pr ₂ O ₃ 3.14	Nd ₂ O ₃ 10.2	Sm ₂ O ₃ 2.05	Gd ₂ O ₃ 0.94	Dy ₂ O ₃ 0.11	Er ₂ O ₃ 0.05	$ m Y_2O_3$ 0.71	CaO 0.44	P ₂ O ₅ 27.8	SiO ₂ 1.42	Total 99.2
a = number of f	oint anal	yses.															

b = data from Nasdala et al. (2001b).c = data from Seydoux-Guillaume et al. (accepted).

n.d. = not detected or calculated mean oxide content < 0.01 wt.%.

Sample			X-ray data						Raman dat	.a ^(a)		
(annealing temperature)	Phase	C ^(b) [wt.%]	a_0 $[Å]$	b_0 [Å]	c_0 [Å]	β [°]	V $[Å^3]$	FWHM [°20]	v [cm ⁻¹]	b_{meas} [cm ⁻¹]	s [cm ⁻¹]	b_{corr} $[ext{cm}^{-1}]$
K1 ^(c)			6.6625(53)	I	6.0103(48)	I	266.79(64)	n.d.	998.7	20.2	2.0	20.0 ± 1.5
K2 ^(c)			6.6070(52)	I	6.0103(47)	I	262.36(62)	n.d.	1003.4	11.1	2.0	10.7 ± 2.0
Moacir ^(d)	phase A phase B	n.d. n.d.	6.823(1) 6.783(5)	7.026(1) 7.014(6)	6.499(1) 6.489(5)	103.79(1) 103.69(9)	302.60(9) 299.98(36)	0.091(1) 0.425(6)	972.2	15.6	3.9	14.6 ± 1.0
N17 (25°C)	4		÷	Ī	*	I	÷	÷	- ;-	+	I	÷
N17 (500°C)			÷	I	+	I	*	-;	+	+	I	-;
N17 (800°C)	$ZrSiO_4$	< <u>~</u> 1	÷	I	- : -	I	÷	-;	1007.5	17.3	0.4	17.3 ± 1.0
	ZrO ₂ (tet.)	~~ 	-}	I	+	I	+	÷	n.d.	n.d.		
N17 (950°C)	$ZrSiO_4$	~	•}=	I	+	I	+	I	1011.0	5.9	0.4	5.9 ± 0.5
	ZrO ₂ (tet.)	40(5)	3.601(1)	I	5.199(1)	Ι	67.42(5)	0.25(1)	n.d.	n.d.		
	ZrO_2 (mon.)	60(5)	5.145(3)	5.201(3)	5.300(3)	98.32(6)	140.33(26)	0.49(3)	n.d.	n.d.		
N17 (1100°C)	$ZrSiO_4$	70(5)	6.606(1)	Ι	5.983(1)	Ι	261.09(12)	0.19(1)	1010.9	5.6	0.4	5.6 ± 0.5
	ZrO ₂ (tet.)	20(5)	3.603(1)	I	5.198(3)	I	67.48(8)	0.31(4)	n.d.	n.d.		
	ZrO ₂ (mon.)	10(5)	5.146(6)	5.201(6)	5.301(6)	98.3(1)	140.39(52)	0.60(5)	n.d.	n.d.		
N17 (1250°C)	$ZrSiO_4$		6.6057(2)	I	5.9829(2)	I	261.07(1)	0.067(2)	1007.9	3.1	0.4	3.0 ± 0.5
N17 (1400°C)	$ZrSiO_4$		6.6058(2)	I	5.9827(2)	Ι	261.06(1)	0.060(2)	1008.1	2.9	0.4	2.8 ± 0.5
N17 (1500°C)	$ZrSiO_4$		6.6060(2)	I	5.9826(2)	Ι	261.08(1)	0.057(2)	1007.7	2.7	0.4	2.6 ± 0.5
Reference: ZrSi($\mathbf{)}_4^{(\mathbf{e},\mathbf{f})}$		6.604	I	5.979	I	260.76	I	1008.3	2.0	0.4	1.9 ± 0.5
Reference: ZrO ₂	(tetragonal) ^(e)		3.592-3.640	I	5.152-5.270	Ι	66.47-69.83	I	I	I		
Reference: ZrO ₂	(monoclinic) ^(e)		5.146	5.212	5.313	99.22	140.66	I	I	I		

a = v is the Raman shift of the main tetrahedral vibration [zircon, $v_3(SiO_4)$; monazite, $v_1(SiO_4)$]. b is the FWHM of this Raman band. Measured FWHMs were corrected for the apparatus function (s) according to Irmer (1985).

b = relative abundance of the respective phase, estimated from the X-ray results.

c = data from Nasdala et al. (2001b).

d = data from Seydoux-Guillaume et al. (accepted).

e = references for unit cell parameters of synthetic standards: zircon, ICDD-PDF (International Centre for Diffraction Data, Powder Diffraction File) 6-266; tetragonal ZrO₂, ICSD (Inorganic Crystal Structure Database); monoclinic ZrO₂, Hill and Cranswick (1994).

f = reference Raman data were obtained from a synthetic pure zircon crystal grown by J. M. H. (flux synthesis as described in Hanchar et al., 2001). n.d. = not determined.

 $\dot{\tau}$ = phase not detected or X-ray peaks insufficient for the calculation of unit cell parameters.

Table 2

Zircon	Th (ppm)	U (ppm)	Th/U	Pb (ppm)	$^{206}\mathrm{Pb}/^{238}\mathrm{U}$	Age ^{206/₂₃₈ (Ma)}	²⁰⁷ Pb/ ²³⁵ U	Age ²⁰⁷ / ₂₃₅ (Ma)	²⁰⁷ Pb/ ²⁰⁶ Pb	Age ^{207/} 206 (Ma)
N17 (25°C)	351	5601	0.06	475	0.08950 ± 0.00115	552 ± 7	0.7235 ± 0.0101	553 ± 6	0.05853 ± 0.00022	549 ± 8
N17 (25°C)	303	5437	0.06	460	0.08881 ± 0.00138	548 ± 7	0.7168 ± 0.0099	549 ± 6	0.05843 ± 0.00022	546 ± 8
N17 (25°C)	348	5599	0.06	456	0.08772 ± 0.00111	542 ± 7	0.7116 ± 0.00984	546 ± 6	0.05872 ± 0.00024	557 ± 9
N17 (25°C)	374	5637	0.07	420	0.08399 ± 0.00102	520 ± 6	0.6813 ± 0.00952	527 ± 6	0.05845 ± 0.00034	566 ± 12
N17 (950°C)	309	5585	0.06	1298	0.28098 ± 0.00288	1596 ± 14	2.2618 ± 0.0268	1200 ± 8	0.05834 ± 0.00029	543 ± 11
N17 (950°C)	364	5403	0.07	895	0.20023 ± 0.00197	1177 ± 11	1.6225 ± 0.0291	979 ± 11	0.05877 ± 0.00081	559 ± 30
N17 (950°C)	347	5284	0.07	1125	0.25567 ± 0.00262	1468 ± 13	2.0563 ± 0.0245	1134 ± 8	0.05832 ± 0.00028	542 ± 10
N17 (1400°C)	382	5753	0.07	÷	÷	- ;-	-;	÷	- ;	÷
$\dot{\tau} = no U-Pb isoto$	ope ratios	are giver	1 for the	1400°C sa	mple. The observed Pb co	ount rate was ins	ignificant, resulting in a	pparently mean	ungless isotope ratios wit	h huge errors.

Table 3 SHRIMP ion probe data for zircon N17 (corrected for common Pb). Isotope values are given with 1 σ errors



Fig. 1. X-ray powder diffraction patterns (A) and Raman spectra (B) for fragments of zircon N17 annealed for 150 h at different temperatures (M = monoclinic ZrO_2 ; T = tetragonal ZrO_2 ; Z = $ZrSiO_4$). Spectra and diffraction patterns are shown in comparison with those of the monoclinic ZrO_2 standard described by Hill and Cranswick (1994). *The assignment of the zircon Raman band at 356 cm⁻¹ is still controversial. This E_g mode has been interpreted as internal SiO₄ vibration (antisymmetric bending) by Dawson et al. (1971) and external lattice vibration (rotational) by Kolesov et al. (2001).



Fig. 2. TEM results for sample N17. (A) HREM image of the untreated, highly metamict zircon. The small inset at the right is the Fourier-filtered version of a detail of the HREM image (dashed). Note the diffuse, halo-like ring in the electron diffraction pattern which is typical of amorphous solids. (B) Bright field image of N17 annealed at 950°C. A multitude of tiny, rounded ZrO₂ crystals having random crystallographic orientation (note the "polycrystalline" electron diffraction pattern) is embedded in an amorphous matrix. (C) HREM image, showing a detail enlargement of the boundary between one ZrO₂ crystal and the surrounding SiO₂-rich matrix. (D) Bright field image, obtained after annealing at 1400°C. The zircon is polycrystalline with a grain size of several hundred nm and contains SiO₂-rich inclusions. (E) HREM image of the boundary between crystalline zircon and amorphous inclusion.



Fig. 3. Concordia plot for SHRIMP U-Pb data obtained from the untreated zircon N17 (marked "25°C") and annealed fragments. Sizes of square symbols extend the errors. Strongly reversely discordant isotope ratios and ages at 950°C result from an ion microprobe analytical artifact. Nearly complete U-Pb resetting was observed at 1400°C, which corresponds to an age close to zero. Axes are shown on a logarithmic scale for better clarity of the picture.



Fig. 4. Correlation between CL and radiation damage. (A) Heterogeneously metamict zircon crystal from the Adirondack Mountains, N. Y. State, BSE (backscattered electrons) image. (B) Same crystal, CL image. Note that the least intense CL emission is observed in the most radiation-damaged micro-areas and vice versa. (C) CL images (widths 100 µm) of zircon K1 from Ontario, Canada, and its annealing products. Diamonds, Raman band broadening; circles, CL intensity measured on an arbitrary grayscale. Note the clear intensity gain with temperature, which corresponds to CL recovery through structural recovery. (D) BSE image of a highly metamict zircon from Rogaland, Norway, after electron microprobe analysis. The impact of the focused electron beam has caused local structural recovery in the analysis spot area, as revealed by the greatly decreased band broadening in the Raman spectrum (left). (E) CL image of the same sample. Note that the spot area emits less CL, even though it is not as radiation-damaged as the surrounding zircon.



Fig. 5. CL spectra for the untreated zircon N17 and two annealing products. Spectra were obtained under the same experimental conditions. Note that the spectrum obtained from the recrystallised zircon has about 100 times higher intensity, when compared with its highly metamict analogue.



Fig. 6. CL spectra (A) and Raman spectra (B) obtained from monazite Moacir. The two ranges of the CL spectra (240-440 and 550-900 nm) were obtained with two different detectors and intensities can, therefore, not be directly compared. To provide a rough estimate, the ultraviolet range was intensity-expanded for about 10 times to make changes better recognisable.