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1 **Combined U-Th/He and $^{40}\text{Ar}/^{39}\text{Ar}$ Geochronology of Post-shield**

2 **Lavas from the Mauna Kea and Kohala volcanoes, Hawaii**

3
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Abstract

30
31 Late Quaternary, post-shield lavas from the Mauna Kea and Kohala volcanoes on the
32 Big Island of Hawaii have been dated using the $^{40}\text{Ar}/^{39}\text{Ar}$ and U-Th/He methods. The
33 objective of the study is to compare the recently demonstrated U-Th/He age method,
34 which uses basaltic olivine phenocrysts, with $^{40}\text{Ar}/^{39}\text{Ar}$ ages measured on groundmass
35 from the same samples. As a corollary, the age data also increase the precision of the
36 chronology of volcanism on the Big Island. For the U-Th/He ages, U, Th and He
37 concentrations and isotopes were measured to account for U-series disequilibrium and
38 initial He. Single analyses U-Th/He ages for Hamakua lavas from Mauna Kea are
39 87 ± 40 ka to 119 ± 23 ka (2σ uncertainties), which are in general equal to or younger
40 than $^{40}\text{Ar}/^{39}\text{Ar}$ ages. Basalt from the Polulu sequence on Kohala gives a U-Th/He age
41 of 354 ± 54 ka and a $^{40}\text{Ar}/^{39}\text{Ar}$ age of 450 ± 40 ka. All of the U-Th/He ages, and all but
42 one spurious $^{40}\text{Ar}/^{39}\text{Ar}$ ages conform to the previously proposed stratigraphy and
43 published ^{14}C and K-Ar ages. The ages also compare favorably to U-Th whole rock-
44 olivine ages calculated from ^{238}U - ^{230}Th disequilibria. The U-Th/He and $^{40}\text{Ar}/^{39}\text{Ar}$
45 results agree best where there is a relatively large amount of radiogenic ^{40}Ar ($>10\%$),
46 and where the $^{40}\text{Ar}/^{36}\text{Ar}$ intercept calculated from the Ar isochron diagram is close to
47 the atmospheric value. In two cases, it is not clear why U-Th/He and $^{40}\text{Ar}/^{39}\text{Ar}$ ages do
48 not agree within uncertainty. U-Th/He and $^{40}\text{Ar}/^{39}\text{Ar}$ results diverge the most on a low-
49 K transitional tholeiitic basalt with abundant olivine. For the most alkalic basalts with
50 negligible olivine phenocrysts, U-Th/He ages were unattainable while $^{40}\text{Ar}/^{39}\text{Ar}$ results
51 provide good precision even on ages as low as 19 ± 4 ka. Hence, the strengths and
52 weaknesses of the U-Th/He and $^{40}\text{Ar}/^{39}\text{Ar}$ methods are complimentary for basalts with

53 ages of order 100-500 ka.

54 **1. Introduction**

55 Hawaiian lavas are used extensively to probe the chemical composition of the
56 Hawaiian mantle plume (DIXON et al., 1997; FEIGENSON et al., 1983; FREY et al., 1991;
57 FREY et al., 1990; WEST et al., 1988). Multiple chemical components (HART et al.,
58 1992; RODEN et al., 1994), radial (DEPAOLO et al., 2001) and asymmetric
59 (ABOUCHAMI et al., 2005) zonation of the mantle plume source have been invoked to
60 explain the chemical heterogeneity found in the volcanoes. The ability to characterize
61 the temporal evolution of the volcanoes, and to tie the lava geochemistry to the
62 structure of the mantle plume, is critically dependent on accurate dating of the lavas.
63 Dating has proven to be challenging when using $^{40}\text{Ar}/^{39}\text{Ar}$ technique, because they are
64 young (< 750 ka) and have low concentrations of potassium (COUSENS et al., 2003;
65 SHARP and RENNE, 2005). In this study U-Th/He measurements of phenocrystic
66 olivine (ACIEGO et al., 2007) and $^{40}\text{Ar}/^{39}\text{Ar}$ measurements of groundmass (e.g, SHARP
67 and RENNE, 2005) are applied to Late Quaternary lava flows from Hawaii to further test
68 the U-Th/He method on basalts and to improve the detailed geochronology of the
69 youngest volcanoes. This work is in conjunction with a larger study of the trace
70 element and isotopic compositions of the post-shield stage lavas of the Big Island,
71 including samples from Hualalai (HANANO et al., in review). Detailed geochronology is
72 required in order to accurately compare temporal compositional variations of historical
73 lavas from Hualalai with those from the Mauna Kea and Kohala volcanoes.
74
75 Hawaiian lavas are challenging targets for the U-Th/He method because they typically
76 have a large component of trapped helium and low concentrations of uranium and

77 thorium. For this study, we focus on transitional tholeiitic to alkalic lavas, which are
78 largely degassed, and have higher K contents, theoretically allowing high precision
79 $^{40}\text{Ar}/^{39}\text{Ar}$ measurements. Future work will test the U-Th/He method on shield stage
80 tholeiitic basalts, which are traditionally more difficult to date using $^{40}\text{Ar}/^{39}\text{Ar}$ and may
81 have lower U and Th concentrations, but are older and have abundant olivine.

82

83 We present data on post shield lavas from the older Kohala and younger Mauna Kea
84 volcanoes, which constitute the northwest section of the Island of Hawaii (Figure 1).
85 On the Kohala volcano, the northern-most on the island, the volcanic units are
86 classified into two groups: the Polulu Volcanic member, containing the transitional
87 tholeiitic to alkali basalts, and the overlying Hawi Volcanic member, the evolved
88 alkalic cap lavas which range in composition from hawaiitic to trachytic. The Kohala
89 volcano entered the post-shield alkalic stage at about 400 to 500 ka (WOLF et al.,
90 1997). On the Mauna Kea volcano, the lower, transitional basalts are grouped into the
91 Hamakua Volcanic member and the upper, evolved alkalic cap lavas are named the
92 Laupahoehoe Volcanic member (STEARNS and MACDONALD, 1946). Mauna Kea
93 entered the post-shield alkalic stage at about 100 ka (WOLF et al., 1997). For this work,
94 we sampled both sequences of basalts, but found only the Polulu and Hamakua basalts
95 had high enough abundance of phenocrystic olivine for U-Th/He work.

96

97

98 **2. Methods**

99 *2.1 Sample collection and descriptions*

100 Samples were collected from lava flows on the flanks of the Mauna Kea and Kohala
101 volcanoes (Figure 1), exact locations and elevations are summarized in Table 1. The
102 collection points were road and gulch cuts, where the samples could be collected from
103 more than one meter below the original flow surface to minimize cosmogenic ^3He and
104 ^4He production, and more than 1m above the base of the flow which should minimize
105 quenching effects (e.g. glassy groundmass) on the $^{40}\text{Ar}/^{39}\text{Ar}$ ages. At the collection
106 points, sampled lava flows had no direct overlying units and were less than 50m thick.
107 The samples are fresh with some occasional minor alteration of the groundmass and,
108 where present, the olivine is unweathered and free of oxidation. The olivine grains have
109 abundant melt inclusions of glass mixed with micro-crystalline plagioclase (Figure 2)
110 as well as minor inclusions of Fe-Ti-oxides and phosphates (likely apatite or
111 fluoroapatite). Major element compositions were measured by XRF on a Philips
112 PW2400 spectrometer at UC Berkeley and the results are summarized in Table 1.

113

114 The samples of Hamakua lava from Mauna Kea have abundant olivine and pyroxene
115 phenocrysts, and varying plagioclase phenocryst contents. Microprobe analyses
116 indicate olivine compositions in the range Fo_{75} to Fo_{82} with no zonation in composition
117 across grains. Olivine observed in thin sections show a lack of textural indicators of
118 xenocrystic populations such as resorption rims or sieve cores. In general, all of the
119 samples from Mauna Kea have transitional chemical compositions (Table 1), with
120 more alkalic compositions corresponding to lower olivine abundance. Sample AMK7 is

121 from a flow originating near the summit, where glacial moraines provide additional
122 stratigraphic context. The samples from Kohala are all alkalic basalt and only one
123 sample contained olivine phenocrysts (AKA5, Fo₇₃). The ages of the flows overlying
124 sample AKA5 have been measured multiple times using K/Ar analyses; the measured
125 ages range from 135 to 149 ka (MCDUGALL, 1969). Figure 3 shows the stratigraphic
126 relationship between the collected samples and the nearest age markers.

127

128 *2.2 Sample Preparation and Analysis*

129 *2.2a U-Th/He.*

130 Rock samples containing olivine were crushed to pea size, a split taken for whole rock
131 powdering, and the remainder sieved, and re-crushed. Olivine grains in the size range
132 850 to 1000 μm were magnetically separated and handpicked. After picking, the
133 olivine separates, approximately 1 g of material, were air abraded to remove the effects
134 of alpha implantation from the decay of groundmass uranium on the helium
135 concentration or alpha ejection loss from the phenocrysts (ACIEGO et al., 2007;
136 BLACKBURN et al., 2007; MIN et al., 2006). Several attempts were made to separate
137 enough microphenocrysts from samples AMK3, AMK11, AKA2, and AKA7, but the
138 amount of material was not sufficient for helium and U-Th/He analysis. In order to
139 remove enough material by abrasion for microphenocrysts on the order of 100 μm in
140 diameter, more than 70% of the mass must be removed, thereby requiring more than 2
141 grams of olivine grains to start, an amount unattainable with the 5 kg sample sizes
142 collected.

143

144 After abrading, the olivine grains were cleaned, air dried, and loaded into a magnetic
145 mortar and pestle for crushing. The crushing in vacuo releases the trapped (initial)
146 helium component leaving the radiogenic and cosmogenic components. Release of the
147 trapped component was optimized to minimize the effects of overcrushing or
148 undercrushing the samples. Overcrushing can result in the release of radiogenic ^4He
149 (e.g. HILTON et al., 1999) while undercrushing can result in trapped ^4He remaining,
150 between 0.2 and 10% (e.g. WILLIAMS et al., 2005; KURZ et al., 1996). Samples were
151 crushed using 300 beats in 5 minutes, then sieved to remove the remaining pieces
152 larger than 100 μm , which may have magmatic helium remaining.

153

154 The <100 μm size fraction was loaded into platinum packets; powder weights ranged
155 from 0.37 to 0.82 g. The total possible contribution of U and Th from the Pt foil was
156 less than 24 pg. The Pt foil packets were loaded into a resistance furnace designed for
157 low abundance U-Th/He work. Gas release was measured at three temperatures: a
158 300°C extraction step to remove any adsorbed gases, a 1500°C step to melt and release
159 the cosmogenic ^3He and the radiogenic ^4He , and a third 1600°C step to check that gases
160 were fully released. In all cases we found that the gas concentrations released at the
161 300°C and 1600°C steps were at blank level, therefore numbers reported in Table 2 are
162 the blank subtracted 1500°C step. Extracted gases were purified on a series of getters
163 and the helium concentrated by absorption on a charcoal trap prior and release directly
164 into the mass spectrometer prior to measurement. Helium abundance and isotopic
165 measurements were conducted on a VG5400 at Lawrence Berkeley National
166 Laboratory equipped with a Faraday cup and an electron multiplier operating in pulse

167 counting mode. Abundance measurements were calibrated using an aliquot of air and a
168 reference sample of helium of known isotopic composition: $R = 2.4 R_a$ where R_a is the
169 helium isotopic composition of air (${}^3\text{He}/{}^4\text{He} = 1.39 \times 10^{-6}$). The detection limit for ${}^3\text{He}$
170 on the multiplier is 5×10^{-11} nmol; in theory the same detection limit for ${}^4\text{He}$ although
171 the blanks are significantly higher. Blanks were run prior to each sample for both
172 crushing and heating, and varied between 1.3 and 4.0×10^{-6} nmol ${}^4\text{He}$ for the crushers
173 and 1.0 and 3.6×10^{-6} nmol ${}^4\text{He}$ for the furnace, ${}^3\text{He}$ blanks were at the detection limits.

174

175 After total gas extraction, the samples packets are retrieved and the fused sample
176 removed from the foil. The samples were dissolved in an $\text{HNO}_3\text{-HF-HClO}_4$ acid
177 solution. Dissolutions are checked for completeness and formation of any fluorides by
178 centrifuging, and treating dark solids with $\text{HNO}_3\text{-HF-HClO}_4$ again, white solids
179 (fluorides) with an HCl-Boric acid solution. Aliquots of the solutions were spiked with
180 ${}^{229}\text{Th}$ and ${}^{233}\text{U}$, unspiked aliquots were analyzed for ${}^{234}\text{U}/{}^{238}\text{U}$ and ${}^{230}\text{Th}/{}^{232}\text{Th}$.

181 Isolation of U and Th was accomplished using Tru-Spec® column resin following
182 established procedures (LUO et al., 1997). U and Th isotopic and concentration
183 measurements were made at the Woods Hole Oceanographic Institution. U and Th
184 concentration analyses were done by isotope dilution on a ThermoFinnegan Element 2
185 ICP-MS operating in pulse counting mode. Samples were introduced to the mass
186 spectrometer via a CETAC Aridus desolvator. Background counts were evaluated by
187 peak scanning between masses 227 and 240. Standard NBS960 was measured in
188 between every sample to correct for mass fractionation using the natural ${}^{238}\text{U}/{}^{235}\text{U}$
189 ratio. Samples were measured in triplicate, and the uncertainty in the concentrations,

190 0.75-1%, reflects the external reproducibility of the repeat measurements. U and Th
191 isotopic compositions were measured on a ThermoFinnegan Neptune MC-ICP-MS.
192 Thorium and uranium isotopic compositions were measured statically with ^{232}Th , ^{238}U ,
193 and ^{235}U in Faraday cups and ^{230}Th and ^{234}U in the SEM. Thorium measurements were
194 made with the RPQ filter on, resulting in 85% transmission, abundance sensitivity of
195 50ppb over 2 amu, and tail corrections of ^{232}Th on ^{230}Th of $\sim 0.3\%$. Sample
196 measurements were bracketed with measurements of UCSC ThA which was used to
197 correct for mass bias and SEM/Faraday gain of the $^{232}\text{Th}/^{230}\text{Th}$. Sample measurements
198 for uranium were corrected for mass bias using an internal normalization, the natural
199 $^{238}\text{U}/^{235}\text{U}$ ratio, and bracketed with NBS U10 measurements to determine SEM/Faraday
200 gain. WHOI 's analytical protocols for measuring U and Th isotopes and
201 concentrations are detailed in (BALL et al. 2007 and SIMS et al., 2008a). Accuracy of
202 the spike compositions, and thereby the concentration measurements, and isotopic
203 measurements were monitored by the measurement of rock standard TML, which is
204 well known to have an [$^{230}\text{Th}/^{238}\text{U}$] activity ratio of one (see. e.g. SIMS et al., 2008a).
205 The TML powders dissolved and spiked at the same time as the olivine samples had a
206 [$^{230}\text{Th}/^{238}\text{U}$] activity ratio of 1.01, which is within the analytical uncertainties of the
207 measurements. Uranium isotopic compositions for all samples were found to be within
208 error of equilibrium, $^{234}\text{U}/^{238}\text{U}$ activity ratios were 1 ± 0.01 . The analytical techniques
209 used for U-Th/He dating are identical to those found in (ACIEGO et al., 2007).
210
211
212 *2.2b $^{40}\text{Ar}/^{39}\text{Ar}$.*

213 Lava rock chunks were crushed into fine chips. Phenocrysts were removed using
214 conventional Frantz magnetic separation. Groundmass grains (300-500 microns) that
215 showed no sign of alteration were further hand-picked and leached in diluted (2N) HF
216 for one minute and then thoroughly rinsed with distilled water in an ultrasonic cleaner.
217 One irradiation of 15 minutes duration was performed in the Cd-shielded (to minimize
218 undesirable nuclear interference reactions) CLICIT facility of the TRIGA reactor at
219 Oregon State University. Samples were irradiated in aluminum discs along with the
220 Alder Creek sanidine standard, for which an age of 1.193 Ma is adopted (NOMADE et
221 al., 2005). $^{40}\text{Ar}/^{39}\text{Ar}$ analyses were performed at the Berkeley Geochronology Center
222 using a CO_2 laser. The gas was purified in a stainless steel extraction line using two C-
223 50 getters and a cryogenic condensation trap. Ar isotopes were measured in static mode
224 using a MAP 215-50 mass spectrometer with a Balzers electron multiplier mostly using
225 10 cycles of peak-hopping. A more complete description of the mass spectrometer and
226 extraction line is given in (RENNE et al., 1998). Blank measurements were generally
227 obtained before and after every three sample runs. The correction factors for interfering
228 isotopes correspond to the weighted mean of 10 years of measurements of K-Fe and
229 CaSi_2 glasses and CaF_2 fluorite in the OSTR reactor: $(^{39}\text{Ar}/^{37}\text{Ar})_{\text{Ca}} = (7.60 \pm 0.09) \times 10^{-4}$;
230 $(^{36}\text{Ar}/^{37}\text{Ar})_{\text{Ca}} = (2.70 \pm 0.02) \times 10^{-4}$; and $(^{40}\text{Ar}/^{39}\text{Ar})_{\text{K}} = (7.30 \pm 0.90) \times 10^{-4}$. Ages were
231 calculated using the decay constants of (STEIGER and JAGER, 1977). J- and mass
232 discrimination values range from 0.0000680 ± 0.0000003 (0.43%) to $0.0000701 \pm$
233 0.0000001 (0.19%) and from 1.00634 ± 0.00216 to 1.00682 ± 0.00242 per dalton
234 (atomic mass unit), respectively. Our criteria for the determination of age plateaus are:
235 (1) to include at least 70% of ^{39}Ar ; and (2) to be distributed over a minimum of 3

236 consecutive steps agreeing at 95% confidence level and satisfying a probability of fit of
237 at least 0.05. Plateau ages are given at the 2σ level and are calculated using the mean of
238 all the plateau steps, each weighted by the inverse variance of their individual
239 analytical error, and assuming that the initial $^{40}\text{Ar}/^{36}\text{Ar}$ ratio is that of air (295.5 by
240 convention (STEIGER and JAGER, 1977)). A more recent determination of atmospheric
241 $^{40}\text{Ar}/^{36}\text{Ar}$ (LEE et al., 2006) yields indistinguishable ages because this value is also used
242 to determine mass discrimination and the effects almost entirely cancel out. Integrated
243 ages (2σ) are calculated using the total gas released for each Ar isotope. Data were also
244 cast in inverse isochron diagrams, and in cases where the $^{40}\text{Ar}/^{36}\text{Ar}$ intercept ratio is
245 statistically higher than the atmospheric value, the inverse isochron age is used. Inverse
246 isochrons include the maximum number of consecutive steps with a probability of fit \geq
247 0.05. Complete descriptions of the analytical procedure are given in (SHARP and
248 RENNE, 2005) and (NOMADE et al., 2005). Detailed $^{40}\text{Ar}/^{39}\text{Ar}$ results are shown in
249 Appendix 1 and summarized in Table 2.

250

251 **3. Results**

252 *3.1 U, Th, and He concentrations and isotopic compositions*

253 The U, Th and He concentrations are compiled in Table 2. The ^4He concentrations in
254 olivine are in the range 0.64 to 4.5×10^{-5} nmol/g; three of the samples have
255 concentrations lower than 1.8×10^{-5} nmol/g. The low concentrations limit the accuracy
256 with which ^4He concentration can be measured, and if the U and Th concentrations of
257 these samples are representative of Hawaiian olivine, it means that the lower limit age
258 for which the U-Th/He method can be useful for Hawaiian basalt geochronology is
259 about 50 ka using our measurement techniques. Only one sample (AMK7) had
260 cosmogenic ^3He (4.59×10^{-9} nmol/g) after crushing. Gas released during crushing has a
261 helium R/Ra (helium isotopic composition normalized to air) of 7.5, while the gas
262 released in melting has an R/Ra of > 150 for sample AMK7. All other samples also
263 had crush-release helium compositions between 6.7 and 12 R/Ra, and corresponding
264 concentrations of ^3He released in heating below detection limits.

265

266 Hawaiian lavas reported in the literature have a large range of trapped helium
267 concentrations, from 2.2 to 1560×10^{-5} nmol/g (KURZ et al., 2004). Even the lower
268 limit of this range is comparable to the amount of radiogenic helium measured in our
269 samples. The post-shield lavas, however, are apparently more thoroughly degassed, as
270 indicated by the low amounts of helium released in crushing. The crushing step
271 yielded small amounts of helium that we assume that if there was any trapped helium
272 remaining in the sample at the heating stage, it was minor compared to the amount of
273 helium released in heating. In the worst case, and 10% of the gas remained after

274 crushing (e.g. KURZ et al., 1996), all of the samples would in fact have much younger
275 ages, which would make the U-Th/He ages less consistent with the $^{40}\text{Ar}/^{39}\text{Ar}$ ages.
276 However, if the crushing and sieving procedures done were as adequate as those in
277 (WILLIAMS et al., 2005), and less than 1% of the trapped component remained than the
278 difference in age would be insignificant.

279

280 The olivine U and Th concentrations are compared to those of the whole rocks in Table
281 1. According to trace element partitioning studies, olivine should contain virtually no
282 U and Th (<0.05 ppb) if it forms in equilibrium with typical basalt liquid (BEATTIE,
283 1993). The olivine U and Th concentrations are far higher than expected based on
284 published distribution coefficients; instead of a concentration ratio between olivine and
285 whole rock of ca. 10^{-5} , the measured ratios are about 0.01 to 0.1. The olivine U and Th
286 concentrations are much more variable than those of the whole rocks, and as shown in
287 the duplicate measurements of AMK12; olivine samples from the same lava flow have
288 variable U and Th concentrations. This variability demonstrates that the U and Th are
289 likely held in inclusions, and therefore the necessity to measure He concentration and
290 U and Th concentrations on the same olivine fraction. The Th/U ratios of the olivine
291 samples are typically lower than those of the whole rocks. The relatively large
292 differences between olivine Th/U and whole rock Th/U indicates that mineral
293 inclusions rather than melt inclusions play the largest role in determining the U-Th
294 concentrations in olivine. Our observations of oxides, phosphates and plagioclase
295 inclusions within the grains is consistent with this hypothesis; PEATE et al. (1996)
296 observed highly variable Th/U ratios in mineral separates of magnetite and plagioclase.

297

298 3.2 $^{40}\text{Ar}/^{39}\text{Ar}$ ages

299 We obtained two plateau ages and one isochron age from Kohala (190 ± 20 to 450 ± 40
300 ka) and four plateau ages and one isochron age from Mauna Kea (19 ± 4 to 239 ± 84 ka).
301 Associated MSWD and P range from 0.25 to 0.94 and from 0.51 to 0.99 respectively
302 (Figure 4 a,b and Table 3). Associated errors are reported as 2 sigma uncertainties
303 within the text.

304

305 Sample AMK7 yielded a well-defined plateau age of 123 ± 5 ka. For this sample, the
306 percentage of radiogenic $^{40}\text{Ar}^*$ is relatively high (11 - 17%), The isochron age (116
307 ± 14 ka) agrees very well with the plateau age and yielded a $^{40}\text{Ar}/^{36}\text{Ar}$ intercept value of
308 298 ± 4 indistinguishable from the argon atmospheric ratio. For samples AMK3,
309 AMK11, AMK13, AKA5, and AKA7 the fraction of radiogenic $^{40}\text{Ar}^*$ (less than 10%)
310 and their K concentration (i.e. ranging from 0.3 to 1.2%) are significantly lower than
311 for sample AMK7 which limits the age precision, although the estimated initial
312 $^{40}\text{Ar}/^{36}\text{Ar}$ are within 1% of the air value which lends confidence to the plateau age
313 estimates.

314

315 Samples AKA2 and AMK12 yielded plateau ages according to our definition of a
316 plateau, but the $^{40}\text{Ar}/^{36}\text{Ar}$ intercept values (313 ± 22 and 305 ± 6 ; 2σ) are higher than the
317 atmospheric value and their age spectra follow a slight saddle-shaped pattern. These
318 features suggest the presence of excess $^{40}\text{Ar}^*$. For these samples, we use the isochron
319 age calculation, which should provide a better estimate of the crystallization age

320 (SHARP and RENNE, 2005). Additionally, AMK12 exhibits a strong tilde-shaped age
321 spectrum. This additional shape suggests that this sample underwent ^{39}Ar and ^{37}Ar
322 redistribution during the neutron irradiation (JOURDAN et al., 2007; ONSTOTT et al.,
323 1995). If this is the case, the plateau and isochron calculation cannot be confidently
324 used to define the age of the sample. Furthermore, the fraction of radiogenic $^{40}\text{Ar}^*$ is 6
325 – 10% lower and the estimated initial $^{40}\text{Ar}/^{36}\text{Ar}$ is $6 \pm 7\%$ higher than the air value.
326 The low percentage of $^{40}\text{Ar}^*$ in comparison to the uncertainty in the initial $^{40}\text{Ar}/^{36}\text{Ar}$
327 makes the AMK12 age the least reliable of those obtained.

328

329 Most of the samples show increasing age and Ca/K over the last 10-20% of the
330 spectrum, at high temperature. These steps also depart from the isochron mixing lines,
331 arguing for a distinct excess $^{40}\text{Ar}^*$ reservoirs included in refractory Ca-rich phases (i.e.
332 interstitial pyroxene). These steps were not included in the plateau and isochron age
333 calculation. Overall, all but one (AMK12) of the $^{40}\text{Ar}/^{39}\text{Ar}$ ages obtained in this study
334 are in agreement with their inferred stratigraphic ages given by previous K/Ar and ^{14}C
335 dates (Figure 3). Furthermore, the precision of these new $^{40}\text{Ar}/^{39}\text{Ar}$ ages far surpasses
336 the precision obtained by K/Ar dating on similar lavas which have uncertainties of 10-
337 30% as shown in Figure 3.

338

339 *3.3 U-Th/He Ages*

340 Ages were calculated using the measured U, Th and He concentrations and isotopic
341 compositions coupled with the U-Th/He age equation given in (FARLEY et al., 2002)
342 and (ACIEGO et al., 2003). For samples with U-series out of radioactive equilibrium, a

343 correction factor must be applied to take into account variations from secular
344 equilibrium. For samples older than 20 ka and less than 1 Ma, this departure from
345 equilibrium will be dominated by the Th-U fractionation. Therefore, the correction is
346 based on the estimate of the initial U-Th disequilibrium (initial $^{230}\text{Th}/^{238}\text{U}$ activity =
347 D_{230}) at the time of helium closure. D_{230} can be calculated using either the
348 concentrations of U and Th within the whole rock and olivine separates or by using the
349 $^{230}\text{Th}/^{238}\text{U}$ of the olivine (see FARLEY et al. (2002) and ACIEGO et al. (2007)for
350 discussion). In this case we report the ages calculated directly from the olivine; use of
351 the whole rock-olivine concentrations change the ages by -10% for samples AMK12
352 and AMK13 and by +10% for sample AMK7. Calculated ages are shown in Table 2
353 and range from 354 ± 54 to 87 ± 40 ka (2 sigma uncertainties). In general, relatively
354 small amounts of radiogenic ^4He limit the precision of the calculated ages from ± 30 to
355 ± 50 ka. Unlike the $^{40}\text{Ar}/^{39}\text{Ar}$ ages, there is no additional information (plateau quality,
356 isochron fits, estimate of initial $^{40}\text{Ar}/^{39}\text{Ar}$) with which to assess the quality of the age
357 determinations. However, for one sample, AMK12, we have duplicate ages of 87 ± 40
358 and 91 ± 36 ka, which are identical. Unfortunately, that sample has the lowest quality
359 $^{40}\text{Ar}/^{39}\text{Ar}$ age determination of the samples we measured.

360

361 *3.4 $^3\text{He}_C$ Age*

362 AMK7 is unique because it was collected from a narrow, shallow gully where it was
363 not possible to collect a sample completely shielded from cosmic radiation exposure.
364 However, although exposed, there was significant cosmic ray shielding due to
365 obstruction of the gully face and the opposite gully wall. We calculated a minimum

366 exposure age of the sample based on the ^3He concentration and a production rate. An
367 average equatorial, sea level production rate for ^3He in olivine, $103 \text{ atoms g}^{-1}\text{yr}^{-1}$, was
368 scaled for latitude and elevation to $415 \text{ atoms g}^{-1}\text{yr}^{-1}$ (DUNAI, 2001) and again to
369 account for 50% azimuthal shielding and a surface dip angle of 90° (DUNNE et al.,
370 1999) resulting in a production rate of $101 \text{ atoms g}^{-1}\text{yr}^{-1}$. The calculated age is $\sim 28 \text{ ka}$.
371 This age is not the age of the bottom of the gully, but is an integrated age based on the
372 increasing exposure of the rock as the gully was cut. Given that the production rate of
373 cosmogenic He is negligible more than 10cm away from the exposed surface, we can
374 infer that the gully was close to 1 m deep at least 25 ka, which is consistent with an
375 eruption age of 120 ka.

376

377

378 **4. Discussion**

379 *4.1 Comparison of $^{40}\text{Ar}/^{39}\text{Ar}$, U-Th/He, and U-series ages*

380 As discussed earlier, the Mauna Kea summit lavas have the best age constraints
381 because of the broad glacial moraine coverage. Sample AMK7 must be older than 15
382 ka, because it is overlain by the Makaanaka glacial moraine. It is also likely to be older
383 than the 100ka age of the overlying Laupahoehoe Volcanics. The new U-Th/He age of
384 119 ± 26 ka and $^{40}\text{Ar}/^{39}\text{Ar}$ plateau age of 123 ± 5 ka confirm this hypothesis. The $^3\text{He}_C$
385 age of ~ 28 ka suggests a slow incision rate that is consistent with the aridity of this
386 region of the Mauna Kea volcano – both the sample collection point and the drainage
387 area for the gully are east of the coastal wet areas on the west coast of the island
388 (EHLMANN et al., 2005). Again, we underscore that for this sample, the $^{40}\text{Ar}/^{39}\text{Ar}$ age
389 and the U-Th/He age agree extremely well.

390

391 For sample AMK12 we have the poorest agreement between the U-Th/He results
392 (89 ± 28 ka) and the $^{40}\text{Ar}/^{39}\text{Ar}$ result (239 ± 84 ka). As noted above, the Ar results for
393 this sample are not likely to be as reliable as those of the other samples due to the
394 combined effects of low $^{40}\text{Ar}^*$, [\$^{39}\text{Ar}\$ and \$^{37}\text{Ar}\$ recoil redistribution](#) and an uncertain
395 initial $^{40}\text{Ar}/^{36}\text{Ar}$. While there are both large vertical and lateral stratigraphic distances
396 between the collected samples and the closest previously dated samples, all of the
397 available ages in the region where this sample was collected are between 70 and 150 ka
398 (WOLF et al., 1997). Hence we infer that in this instance the duplicated U-Th/He age
399 may be accurate whereas the $^{40}\text{Ar}/^{39}\text{Ar}$ age is spuriously old, possibly beyond estimated
400 uncertainty. The fact that this sample is the most tholeiitic in composition (lowest

401 alkalinity, Table 1) highlights the crux of this work: $^{40}\text{Ar}/^{39}\text{Ar}$ is a powerful dating tool
402 even for young samples, but tholeiites require an alternate dating method, such as U-
403 Th/He.

404

405 The minimum age of sample AKA5 from Kohala is constrained by the ~137 ka K-Ar
406 age for a unit (MCDUGALL, 1969) located 200m stratigraphically higher. The
407 sampled flow is also within the Polulu Volcanic series (Figure 1), which has a
408 documented age range of 250-500 ka based on several previous K-Ar analyses
409 (MCDUGALL, 1969; MCDUGALL and SWANSON, 1972). The calculated U-Th/He age
410 of 354 ± 54 and the $^{40}\text{Ar}/^{39}\text{Ar}$ age of 450 ± 40 ka, therefore, are both broadly compatible
411 with the previous data although statistically distinguishable.

412

413 Our U-series results provide some additional perspective on the reliability of the U-
414 Th/He and $^{40}\text{Ar}/^{39}\text{Ar}$ ages. The U and Th isotopic composition of the olivine and whole
415 rock samples are plotted on a ^{230}Th - ^{238}U activity diagram (ALLEGRE and CONDOMINES,
416 1976) in Figure 5. Model 'isochron' ages, with errors based solely on the analytical
417 errors, can be calculated (Table 1) from each whole rock – olivine pair. The calculated
418 isochron age would be the eruption age if the olivine and whole rock had identical
419 initial $^{230}\text{Th}/^{232}\text{Th}$ and had remained undisturbed since eruption. There are few olivine
420 U-Th isochron data available in the literature with which to compare these results, so
421 we are not certain how well the requirement of identical initial $^{230}\text{Th}/^{232}\text{Th}$ is likely to
422 be met. One possibility is that the olivine grains did not have an identical initial
423 $^{230}\text{Th}/^{232}\text{Th}$ to the host lava because they are xenocrystic rather than phenocrystic.

424 Based on the petrographic analysis, this is unlikely, but can not be completely ruled
425 out. In previous work on olivine in basalts (SIMS et al., 2007) it was found that internal
426 U-series isochrons that include olivine separates form linear trends, but more accurate
427 and precise age results are generated by removing the olivine from the age calculation,
428 likely because the olivine grains were xenocrystic or antecrystic.

429

430 The whole rock and olivine samples from AKA5 both lie on the equiline and have
431 U/Th ratios that differ only slightly. The U-Th data for AKA5 do not define an age,
432 but are consistent with the age being in the 350 – 450 Ka range as determined by the
433 other methods. Sample AMK7 has an OL-WR U-Th age of 163 ± 9 ka, which is
434 somewhat older than our new $^{40}\text{Ar}/^{39}\text{Ar}$ and U-Th/He ages of ca 120 ka. This
435 difference could be an indication that the olivine in this sample is partly xenocrystic,
436 which could skew the U-Th age to older values but not the U-Th/He age. Sample
437 AMK13 also has a well-defined OL-WR U-Th age of 102 ± 11 ka, which is
438 indistinguishable from the U-Th/He age (111 ± 24 ka) and slightly younger than the
439 $^{40}\text{Ar}/^{39}\text{Ar}$ (143 ± 22 ka) age. The two olivine analyses from sample AMK12 give two
440 distinct ages of 20 ± 9 ka and 50 ± 10 ka. The older age is closer to the U-Th/He age
441 but much younger than the $^{40}\text{Ar}/^{39}\text{Ar}$ age for this sample. This may be further evidence
442 that the $^{40}\text{Ar}/^{39}\text{Ar}$ age for AMK12 is too old.

443

444 With the exception of sample AMK7, all of the samples have systematically younger
445 U-Th/He ages than $^{40}\text{Ar}/^{39}\text{Ar}$ ages. As discussed above, for AMK12, the most likely
446 cause of this discrepancy is $^{40}\text{Ar}^*$ excess and/or ^{39}Ar and ^{37}Ar recoil. However, the

447 reason for the age difference is much less clear for the two other samples. AKA5 and
448 AMK13 both yielded $^{40}\text{Ar}/^{36}\text{Ar}$ intercepts of atmospheric composition on the isochron
449 plot, thereby suggesting that no excess $^{40}\text{Ar}^*$ component is present in these samples.
450 There are two possible reasons for the U-Th/He to produce erroneously low ages. The
451 first possibility is diffusive loss, where the higher diffusivity of helium than argon in
452 the crystallized lava flow would result in the observed difference in age. However, as
453 (HART, 1984) has shown, olivine in lava flows with thicknesses less than 50m cool too
454 rapidly for helium loss to occur. Similarly, heat from overlying lavas would dissipate
455 too rapidly for the samples to lose helium, at least at these collection locations where
456 the thicknesses of overlying lavas is 0-10m. The second possibility is a systematic error
457 in the estimation of U-series disequilibria (ACIEGO et al., 2007). For the young samples
458 (<300 ka), there is a general agreement (within 10%) between initial $^{230}\text{Th}/^{238}\text{U}$ (D_{230})
459 disequilibrium calculated using the olivine and the initial value calculated using a Th-U
460 fractionation model for crystals and melts (e.g. (FARLEY et al., 2002)). Therefore, for
461 these samples we are confident in the errors in the U-Th/He ages due to U-series
462 disequilibria. However, sample AKA5 does have an age that falls in the range of
463 maximum possible error due to uncertainty in D_{230} , between 300ka and 1 Ma (FARLEY
464 et al., 2002; ACIEGO et al., 2007), which could result in uncertainties up to 12%. In this
465 case, using the Th-U fractionation model provides a best estimate for the D_{230} , which
466 lowers the error to 2-5%, well below the difference between the U-Th/He age and the
467 $^{40}\text{Ar}/^{39}\text{Ar}$ age.

468

469 *4.2 Implications for future U-Th-He work*

470 A potentially important overall observation is that the U-Th/He ages are consistently
471 either equal to or younger than the $^{40}\text{Ar}/^{39}\text{Ar}$ (Figure 6, uncertainties shown are 1-
472 sigma). The sample with the highest percentage of radiogenic $^{40}\text{Ar}^*$ is the one that has
473 the best agreement. While the U-Th/He ages are relatively imprecise, this method may
474 yield accurate results as demonstrated by result on sample AMK7 and these results are
475 encouraging. In any case, as noted above, the uncertainty in the $^{40}\text{Ar}/^{39}\text{Ar}$ ages may be
476 as high as that for U-Th/He when the samples have low percentages of radiogenic ^{40}Ar .
477 Thus, our results suggest that the U-Th/He chronometer may be a valuable additional
478 tool for dating young mafic volcanic rocks. Substantially more work will be needed,
479 however, before we can be confident about the generality of our conclusions. If the
480 reliability of olivine U-Th/He technique can be demonstrated, we may be able to
481 develop additional insight about which aspects of the Ar data are indicators of
482 unreliable ages by comparing $^{40}\text{Ar}/^{39}\text{Ar}$ and U-Th/He ages.

483

484 This work reinforces the conclusions (ACIEGO et al., 2007) that U-Th/He dating can be
485 usefully applied to dating basalts in the age range of 50 – 500 ka, and provides further
486 evidence about the reliability of the U-Th/He method by comparison to the $^{40}\text{Ar}/^{39}\text{Ar}$
487 ages on the same samples. The $^{40}\text{Ar}/^{39}\text{Ar}$ ages determined here also strengthen the
488 possibility of using $^{40}\text{Ar}/^{39}\text{Ar}$ to measure ages of increasingly younger alkalic basalts,
489 down to the range of radiocarbon dating, although the comparisons with U-Th/He ages
490 suggest that careful attention must be paid to the percentage of radiogenic ^{40}Ar
491 measured and the pattern defined by the age spectrum (i.e. sample AMK12 having 6-
492 10% of $^{40}\text{Ar}^*$ and a tilde-shaped age spectrum).

493

494 The U-Th/He dating method using olivine of course requires that the samples contain
495 olivine phenocrysts. As noted above and shown by the fewer U-Th/He dates, samples
496 with ~1% microphenocrysts have inadequate olivine for U-Th/He dating. This could be
497 considered a disadvantage in that K-Ar and $^{40}\text{Ar}/^{39}\text{Ar}$ ages can be determined on
498 groundmass and hence are more widely applicable. On the other hand, volcanic
499 groundmass phases may be more susceptible to cryptic alteration that can affect the age
500 determination. Olivine phenocrysts that are useful for U-Th/He dating are also large
501 enough that alteration can be assessed optically. Even in samples where there is some
502 olivine alteration it may be possible to isolate unaltered olivine. However, the
503 possibility of incomplete separation of the helium reservoirs within the olivine remains
504 an issue that has to be carefully considered. Incomplete release of trapped helium
505 would result in *older* calculated U-Th/He ages. And, while over-crushing could release
506 radiogenic and cosmogenic helium, a significant amount of *in situ* produced helium
507 (>1%) is not likely to be released unless longer crushing times and greater crushing
508 force is used (MOREIRA and MADUREIRA, 2005). Ultimately, more comparison between
509 U-Th/He and $^{40}\text{Ar}/^{39}\text{Ar}$ ages are desirable to fully assess the validity of the former
510 technique, and a particularly interesting comparison will be for submarine lavas, where
511 it is well known that there are issues with incomplete degassing of Ar (DALRYMPLE and
512 MOORE, 1968).

513

514 A stringent test of the U-Th/He method will come in applying it to a wider range of
515 lava compositions. Other ocean island lavas, (e.g. the Azores, Canary, Comores

516 Islands, Samoa) have similar U and Th concentrations to the alkalic and transitional
517 lavas measured in this work, between 0.6 to 7 ppm U (BOURDON et al., 1998;
518 BOURDON et al., 2005; CHABAUX and ALLEGRE, 1994; CLAUDE-IVANAJ et al., 1998;
519 CLAUDE-IVANAJ et al., 2001; SIMS and HART, 2006; SIMS et al., 1995; 1999; 2008b).
520 For these lavas, assuming U and Th distribution coefficients of order 0.01, the
521 radiogenic helium production will allow U-Th/He ages to be measured in the same age
522 range as in this work. Other ocean island (e.g. Galapagos, Iceland) and mid-ocean ridge
523 basalts (e.g- EPR) basalts have lower U and Th concentrations, between 0.01 and 0.6
524 ppm (HEMOND et al., 1988; KOKFELT et al., 2003; KOKFELT et al., 2005; STRACKE et al.
525 2003; LUNDSTROM et al., 1999; SIMS et al., 2002; 2003). Therefore, even given optimal
526 analysis conditions of large sample sizes and low blanks, the U-Th/He method will be
527 limited to an older age range, greater than 300 ka. At these low concentrations, the
528 measurement of the $^{230}\text{Th}/^{238}\text{U}$ disequilibria within the olivine will be especially
529 difficult. However, the measurement may be unnecessary if the Th-U fractionation
530 model is valid. Even older samples (> 1 Ma) have the advantage of $^{230}\text{Th}/^{238}\text{U}$ activity
531 ratios close enough to one for multiple half-lives to make the $^{230}\text{Th}/^{238}\text{U}$ disequilibria
532 irrelevant in calculating the U-Th/He age (FARLEY et al., 2002).

533

534 One additional complication for future use of the U-Th/He method on OIBs and
535 MORBs is the likely higher initial helium concentration. If samples have both a high
536 initial helium concentration and cosmogenic helium, distinguishing between the
537 radiogenic and initial components of ^4He will be difficult, leading to large errors in the
538 age. However, one advantage of submarine samples is that they lack cosmogenic He.

539

540 *4.3 Implications for Hawaiian plume dynamics*

541 One application of these new ages is to interpret the spatial-temporal evolution of the
542 volcanoes, and in particular, their relationship to the plume source. One way to do this
543 is to map the source of the lava flow, the individual vents, relative to the location of
544 maximum melt supply at the time of eruption (DEPAOLO et al., 2001). The petrology
545 and geochemistry of the lavas can then provide information about the section of the
546 plume it is sampling: the source material via radiogenic isotopes and melting dynamics
547 via U-series isotopes (see e.g. SIMS et al., 1999). However, the combination of
548 geochemical and spatial evidence depends on having reliable ages, which provide the
549 basis for this paleo-mapping. The geochronology of the Big Island has largely been
550 constrained by K-Ar and ^{14}C ages; the sheer number of ages per stratigraphic unit (20-
551 25; WOLFE and MORRIS, 1996) provide a “brute force” basis for the age ranges
552 assigned because the standard deviation of the mean for all of the ages is relatively low
553 (< 10%). But, individual K-Ar ages have poor errors – as much as 50%, therefore
554 reconstruction of the vent locations could be in error by as much as 40 km, the radius
555 of the melting region of the plume. Finer scale analysis of the plume structure and
556 temporal evolution requires more accurate, precise ages, such as those in this work.
557 Figure 7 illustrates the paleo-mapping for these samples, where the position of the
558 vents relative to the center of the plume, the position of maximum melt supply, is
559 determined based on a Pacific plate motion of N30W at 9 cm/yr. Based on these paleo-
560 locations, the sampled post-shield building lavas of Kohala and Mauna Kea erupted in
561 a front 80 – 100 km away from the melt supply maximum and 0 – 15 km away from

562 the center of the plume. Therefore, variations in petrology and geochemistry must be
563 related to magma chamber processes, such as residence time and magma interaction, or
564 temporal variations in the composition of the plume. Both issues which are addressed
565 in (HANANO et al., in review).

566

567 **5. Conclusions**

568 The U-Th/He method, applied to olivine phenocrysts in four postshield basalt lavas
569 from the Mauna Kea and Kohala volcanoes in Hawaii yield ages for lavas in the age
570 range 90 to 350 ka. The uncertainty in the ages is estimated to be the larger of $\pm 10\%$ or
571 20 ky at the 1-sigma level, although duplicate measurements on one lava agree to
572 within a few percent. The age determinations are consistent with previous geologic
573 mapping and geochronological data from the island of Hawaii. Olivine-whole rock U-
574 Th ages measured on the same samples also agree reasonably well with the U-Th/He
575 ages; and the observed discrepancies could have petrological significance. $^{40}\text{Ar}/^{39}\text{Ar}$
576 ages measured on groundmass from the same four samples yield identical ages in one
577 case, slightly older ages in two cases, a much older (2x) age in one case. The degree of
578 agreement between the $^{40}\text{Ar}/^{39}\text{Ar}$ ages and the U-Th/He ages; the best agreement is for
579 the sample with the largest percent radiogenic $^{40}\text{Ar}^*$ and identical plateau and isochron
580 ages, the worst agreement is for the tholeiitic sample with a clearly identified perturbed
581 age spectrum, low $^{40}\text{Ar}^*$ and higher-than-atmospheric $^{40}\text{Ar}/^{39}\text{Ar}$ trapped component.
582 For the two intermediate cases, it is not clear yet why we observe some age
583 discrepancy between the two methods and further calibration work is needed. Samples
584 with insufficient olivine for U-Th/He dating yielded robust $^{40}\text{Ar}/^{39}\text{Ar}$ ages, indicating
585 the advantages of $^{40}\text{Ar}/^{39}\text{Ar}$ technique for groundmass samples.

586

587 The results presented here are encouraging regarding the applicability of U-Th/He
588 geochronology using olivine phenocrysts in subaerially-erupted ocean island basalts.

589 The analytical uncertainty in the U-Th/He ages depends on the He content and age of

590 olivine, and limits the usefulness of the method for samples like those measured here to
591 ages that are greater than about 50 ka. The data from this study and that of (ACIEGO et
592 al., 2007) show the method to be useful for lavas in the age range from 50 to 500 ka,
593 and that the U-Th/He ages can complement $^{40}\text{Ar}/^{39}\text{Ar}$ ages. Further work needs to be
594 done to evaluate other circumstances where the method can complement existing
595 techniques, such as for both subaerial and submarine Quaternary shield stage tholeiitic
596 basalts.

597

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Figure 1.

612 General geologic map of the Kohala and Mauna Kea Volcanoes on the Big Island of

613 Hawaii, inset indicates location relative to the chain of Hawaiian islands. Most samples

614 were collected from lavas that could be traced back to a specific vent; those vents are

615 noted on the map.

616

617

618

Figure 2.

619 XRF backscatter image of micro-inclusion within sample AMK12 showing plagioclase
620 crystallization nucleating at contact with the surrounding olivine grain. Also present:
621 quenched melt, Ti-Fe oxides and phosphates.

622

623

624

Figure 3.

625

626 Simplified stratigraphic columns indicating relationship of samples to nearest age
627 marker and U-Th/He ages. (a) Sample AMK7 constrained by the overlying Makanaka
628 moraine deposited after the last glacial maximum (PORTER, 1986), C-14 age of material
629 recovered from Laupahoehoe lavas (WOLF et al., 1997), and overlying lava flow dated
630 by K-Ar (WOLF et al., 1997). (b), (c), and (d) Samples constrained only by K-Ar ages
631 (MCDUGALL, 1969; WOLF and MORRIS, 1996; WOLF et al., 1997).

632

Figure 4. $^{40}\text{Ar}/^{39}\text{Ar}$ Release Patterns

(a) Age spectra: $^{40}\text{Ar}/^{39}\text{Ar}$ apparent age and related Ca/K ratio spectra of the groundmass separates versus the cumulative percentage of ^{39}Ar released. Errors on plateau (>70% ^{39}Ar released) ages are quoted at 2σ and do not include systematic errors (i.e. uncertainties on the age of the monitor and on the decay constant). MSWD and probability (P) are indicated. Ages in bold represent the most reliable ages for each sample. * = age likely containing excess Ar; the isochron calculation technique has been used for these samples.

(b) Isochron plots: Inverse isochron plot of $^{36}\text{Ar}/^{40}\text{Ar}$ vs. $^{39}\text{Ar}/^{40}\text{Ar}$ of two samples having $^{40}\text{Ar}/^{36}\text{Ar}$ intercept values higher than the atmospheric ratio. Isochron ages are given at 2σ .

Figure 5.

Activity diagram for the Kohala and Mauna Kea basalts' whole rock powders (WR) and olivine (OL) pairs.

Figure 6.

Comparison of measured $^{40}\text{Ar}/^{39}\text{Ar}$ ages with U-Th/He ages; $^{40}\text{Ar}/^{39}\text{Ar}$ ages are plateau ages except for AMK12 which is an isochron age. Error bars are 1-sigma, and initial Ar composition is noted. Argon plateau ages are older than U-Th/He ages in all samples.

Figure 7.

Reconstructed vent locations of the sampled lavas relative to the Hawaiian plume based on the $^{40}\text{Ar}/^{39}\text{Ar}$ and U-Th/He ages, and Pacific plate motion of N30W at 9 cm/yr.

Table 1.

Description of samples.

Table 2.

U-Th concentrations and $^{230}\text{Th}/^{232}\text{Th}$ compositions of Hawaiian whole rock powders and olivine separates are noted in ppb and square brackets denote activity ratios. Errors in concentration and isotopic composition based on the external reproducibility of the standard TML run at the same time as the samples. Helium concentrations and isotopic compositions are in nmol and R/Ra, the $^3\text{He}/^4\text{He}$ ratio in the sample normalized to air. D values are the calculated distribution coefficients based on the measured whole rock-olivine U and Th concentrations, Experimental* D values are the experimentally determined values from BEATTIE (1993).

**uncertainties in whole rock U, Th concentrations and isotopic compositions are 1% (2-sigma), olivine U, Th concentration and isotopic composition uncertainties are 1.5% (2-sigma)

Table 3. $^{40}\text{Ar}/^{39}\text{Ar}$, U-Th/He and ^3He ages.

Indicates integrated, plateau, isochron $^{40}\text{Ar}/^{39}\text{Ar}$ ages, U-Th/He crystallization and ^3He exposure ages for Kohala and Mauna Kea samples. MSWD for plateau and isochron, percentage of ^{39}Ar degassed used in the plateau calculation, number of analysis included in the isochron, and $^{40}\text{Ar}/^{36}\text{Ar}$ intercept are indicated. Analytical uncertainties on the ages and $^{40}\text{Ar}/^{36}\text{Ar}$ intercept are quoted at 2 sigma (2σ). Bold data indicates the accepted $^{40}\text{Ar}/^{39}\text{Ar}$ ages for a given sample.

Appendix 1.

Ar data summary for the Kohala and Mauna Kea samples. Relative Argon abundances are given in nanoamperes (nA) of amplified beam current. Values are corrected for mass discrimination, blanks, and radioactive decay. Errors in parentheses (1σ) are for the smallest significant digits when not otherwise mentioned. $^{40}\text{Ar}^*$ = radiogenic argon. Age is based on comparison with the Alder Creek sanidine monitor (1.194 Ma; Nomade et al., 2005) and on the decay constant of Steiger and Jäger (1977). J- and discrimination values are provided. Laser beam power (W) is provided for step-heated samples. The correction factors for interfering isotopes correspond to the weighted mean of 10 years of measurements of K-Fe and CaSi_2 glasses and CaF_2 fluorite in the OSTR reactor: $(^{39}\text{Ar}/^{37}\text{Ar})_{\text{Ca}} = (7.60 \pm 0.09) \times 10^{-4}$; $(^{36}\text{Ar}/^{37}\text{Ar})_{\text{Ca}} = (2.70 \pm 0.02) \times 10^{-4}$; and $(^{40}\text{Ar}/^{39}\text{Ar})_{\text{K}} = (7.30 \pm 0.90) \times 10^{-4}$.

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