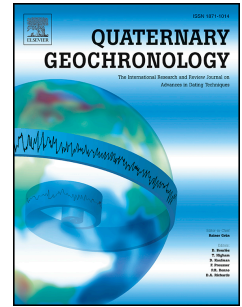


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1 **Exploring the dating of “dirty” speleothems and cave sinters using radiocarbon dating of**
2 **preserved organic matter**

3

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15 **Abstract**

16

17 Speleothems and other carbonate deposits such as tufa containing high proportions
18 of detrital material can be difficult to chemically date due to detrital thorium levels causing
19 a high level of error in conventional U-Th disequilibrium dating. Here we investigate the use
20 of an alternative technique centring on radiocarbon dating of organic matter preserved
21 within the detrital fraction. Non-acid soluble humic, particulate and detritally absorbed
22 organic matter was recovered from eight samples from a flowstone sinter formed within a
23 roman aqueduct at Trento in Italy with a maximum age of 100 CE (1850 cal yr BP), and two
24 repeat samples from a dripstone formed within the 20th Century on a wire fence at Lilly-Pilly
25 Cave, Buchan Caves Reserve in Victoria, Australia. In the aqueduct samples the median
26 calibrated ¹⁴C ages ranged from 2232 to 2889 cal yr BP, with 95.4% probability age range in
27 the youngest and oldest samples of 2153 - 2337 and 2342 – 3449 cal yr BP respectively. The
28 median age of the more modern dripstone was 336 cal yr BP, with a 95.4% probability age
29 range of 148 to 486 cal yr BP. These results provide very approximate ball-park estimates of
30 the age of the sample, but are consistently too old when compared to the known maximum
31 ages of formation. It is hypothesised that this offset is due to a combination of the nature of
32 the organic carbon transported from the source organic matter pools, and reworking of
33 both modern and old organic carbon by in situ microbial communities.

34

35 **Keywords**

36 Speleothems; carbonate sinter; dating; radiocarbon; organic matter;

37

38 1. Introduction

39

40 One of the strengths in the use of carbonates such as speleothems as palaeoclimatic
41 archives lies in their amenability to precise dating. The carbonate in speleothems is most
42 frequently dated by uranium - thorium disequilibrium series (U-Th), which works well back
43 to 500 ka (Dorale and Richards 2003; Dorale et al., 2004; Hellstrom et al., 2006; Hellstrom &
44 Pickering 2015). Uranium – lead dating (U-Pb) extended our capability of obtaining ages
45 from terrestrial carbonates well beyond this, from 1 Ma to at least 8 Ma (Woodhead et al.,
46 2006). Radiocarbon has been relatively under-used as a dating technique in speleothems,
47 principally because radiocarbon dating of the calcite requires corrections for issues such as
48 the dead carbon proportion (dcp), meaning that most of the studies to date have focused on
49 younger material where detection of the bomb spike allows for more accurate modelling
50 (Genty & Massault 1997; Blyth et al., 2007; Hodge et al., 2011; Hua et al., 2012).
51 Radiocarbon analysis in older speleothems has been focused on samples where the analyses
52 could be paired with high precision U-Th dating for radiocarbon calibration (e.g. Beck et al.,
53 2001; Hoffman et al., 2010; Southon et al., 2012) and investigation of temporal dcp
54 variations (e.g., Rudzka et al., 2011; Griffiths et al., 2012; Noronha et al., 2014). These
55 analyses of radiocarbon in speleothems have focused on the carbonate component derived
56 from dissolved carbon dioxide from the breakdown of soil and vadose zone organic matter,
57 and/or from dissolved calcium carbonate bedrock (Genty & Massault, 1999; Genty et al.,
58 2001; Hua et al., 2012).

59 “Dirty” carbonates, those with significant detrital components incorporated in the
60 matrix, are problematic from a dating point of view, as the detrital content has a tendency
61 to lead to high detrital thorium levels which affects conventional chemical dating (Labonne

62 et al., 2002; Hellstrom 2006; Hua et al., 2012; Meyer et al., 2012). Although many samples
63 are amenable to correction (e.g. Hellstrom 2006), and others allow for radiocarbon dating of
64 the carbonate (e.g. Labonne et al., 2002), for some, it can become impossible to secure a
65 usable dating frame-work. This effectively renders these samples defunct for many forms of
66 palaeoenvironmental work, leading to the collected material being under-studied (Meyer et
67 al., 2012). However, “dirty” material often also contains a relatively high organic content
68 (Blyth et al., 2008; Blyth et al., 2015), thus opening up the possibility of using radiocarbon
69 dating of the preserved organic matter as a partial solution to this problem. To date, we are
70 aware of only limited radiocarbon studies that utilised the organic matter preserved in
71 speleothems, probably because the proportion of TOC has been historically hard to measure
72 in this context and is generally very low (0.01 – 0.3% of total carbon; Blyth et al., 2013; Li et
73 al., 2014; Quiers et al. 2015; Blyth et al., 2016). However, radiocarbon dating of organic
74 compounds in moonmilk flowstones from high alpine settings, consisting of fibrous calcite
75 crystals and with a large organic matter component (Borsato et al., 2000; Blyth & Frisia,
76 2008), have yielded ages consistent with those of crystalline speleothems at lower altitudes
77 in the same area, which allowed reconstruction of long-term Holocene climate (Frisia &
78 Borsato, 2010). Radiocarbon dating on calcareous tufa flowstones from the same alpine
79 region, which consist of a mixture of biomediated calcite crystals, detrital components and
80 organic matter, yielded similar ages, although inversions were common (Borsato et al.,
81 2007). Separately, dates from Lascaux cave in France have been published, based on calcite
82 gours, which were observed to have a mineral structure reminiscent of moonmilk (Genty et
83 al., 2011). Four organic matter dates were obtained from this material, based on organic
84 residue that was observed to have a filamentous structure similar to that from fungal and
85 algal remains found in carbonate tufas (Genty et al., 2011). These dates were found to be

86 older by approximately 500 years than those from charcoal from comparable layers, and to
87 have a disturbed chronology (Genty et al., 2011). These results suggest that “dirty”
88 conventional speleothems (such as stalagmites and flowstones) known to contain organic
89 matter could potentially be dated with the radiocarbon method, but that sources and
90 chronological disturbances need careful consideration. In this study we consider two
91 different speleothem types from very different contexts to assess whether the issue of
92 radiocarbon dating of organic matter returning older than expected ages is likely to be a
93 widespread problem, or site-specific.

94

95 **2. Materials and methods**

96

97 *2.1 Samples*

98

99 The principal sample in this study is a flowstone sinter formed inside a roman
100 aqueduct at Trento in the Italian Dolomites (henceforth referred to by the sample code T-
101 AQ). The use of a sample from a man-made structure allows us to place a maximum date of
102 formation independent of any chemical techniques, in this case of circa 100 CE, or 1850 cal
103 yr BP, the aqueduct being believed to have been built during the last quarter of the 1st
104 century (75-100 CE) and utilised until the fall of the Roman Empire and the Early Middle
105 Ages, around 500 CE (Bassi, 2004, 2007). The sample is taken from a section of the aqueduct
106 that was unearthed in the late 1990s, with studies suggesting the water was sourced from
107 groundwater draining Permian to Mesozoic carbonate-evaporite-siliciclastic rocks (Bassi,
108 2004). That the waters feeding the aqueduct were captured from karst springs, rather than
109 surface waters, is supported by the mineral fabrics of the flowstone and $\delta^{13}\text{C}$ data, which

110 are typical for stalagmites/flowstones grown in caves at mid-altitudes, below the conifer
111 vegetation belt in the region (Frisia & Borsato 2010; Borsato et al., 2015)

112 The flowstone, or sinter, which has grown upon a cement mortar layer, is about 400
113 mm wide and up to 105 mm thick. The most common facies is characterized by lamina
114 couplets marked by differences in porosity and fabric, with variations in the percentage of
115 sparite and micrite. The columnar fabric (Frisia, 2015) consists of fans of sparry calcite
116 crystals, commonly showing rounded crystal terminations capped by micrite laminae. Spar
117 crystals may be partially truncated by micrite, show fine impurity-rich lines, and split into
118 crystals arranged in fans (Fig. 1). Crystal tips can be corroded, as documented also in
119 stalagmitic flowstones. Micrite is arranged in laminae capping the spar crystals, and is
120 commonly aggregated in peloids (Frisia, 2015) associated with Fe-oxides, silt and small
121 bivalves (Fig 1.). Both micrite and sparite consist of low-Mg calcite with variable amounts of
122 detrital fraction, mostly associated with the micrite, including quartz, feldspar, dolomite,
123 gypsum, particulate organic matter, clay minerals. EDXS analyses identified the presence of
124 P, up to 5%, in the micrite layers, which is most likely to be associated with dead microbial
125 matter and, possibly, allochthonous bone fragments. The micrite layers show fluorescence when
126 excited by blue light at 365 nm, whereas the sparry calcite show fluorescence when excited
127 by both green (470 nm) and blue light. This may be indicative of the presence, in both the
128 micrite and sparite, of the by-products of organic matter degradation.

129 The T-AQ sinter grew in an artificial tunnel, with a higher discharge (in the order of 5
130 to 20 l/sec) than that commonly observed for cave flowstones. However, as the aqueduct
131 was capped by stone slabs, most of the microflora and fauna living within the structure
132 would be adapted to a dark environment. As is the case for cave speleothems (Blyth et al.,
133 2008; Blyth et al., 2016), the sinter's organic content is likely to reflect a mixture of

134 transported particles and solutes and locally produced organic compounds. The T-AQ sinter
135 can therefore be taken as an analogue for those flowstones and rimstone dams (cave
136 calcareous tufa) which developed in periods of relatively fast flow, in a context amenable to
137 microbial growth.

138 Radiocarbon analyses were performed on eight subsamples collected from the
139 flowstone in a consecutive sequence from top to base (Fig 2.), each weighing between 8 and
140 12 g dry calcite. Repeat calcite subsamples were processed and analysed for levels 2, 3, 5, 6,
141 and 7 to provide better dating control. Subsections were cut using a diamond wire saw, and
142 only clean water as lubricant. After air drying, samples were stored in aluminium foil.

143 To test the method in a more conventional speleothem, two samples of 5 g of dry
144 calcite each were analysed from a dripstone (LP-13) grown in Lilly-Pilly Cave, Buchan Caves
145 Reserve, Murrindal, Victoria, Australia. This dripstone sample formed on a wire fence within
146 the cave, thus enabling an external control to limit its time of formation to the 20th century.
147 The LP-13 calcite contains reddish brown dirt visible by eye, which is commonly seen in
148 speleothems at this site, and is believed to result from flushes of detrital material during
149 periods of high flux (Nicholas White, pers comm.). The dripstone was sampled directly into
150 annealed glass containers to minimise contamination and any exposure to fingerprints or
151 plastic.

153 *2.2 Uranium-thorium dating*

154
155 The T-AQ material forms a challenge for U-Th dating because of its high porosity and
156 visible interspersed detritus, but U-Th analysis was attempted to provide a context for the
157 radiocarbon and to test whether the sample has behaved as a closed system for U. Samples

158 were drilled from the top and base of the specimen (layers 1 and 8 respectively). ~25 mg
159 subsamples were dissolved and spiked with a mixed ^{229}Th - ^{233}U - ^{236}U tracer followed by
160 extraction of the actinide fraction using Eichrom TRU resin. U and Th isotope ratios were
161 measured simultaneously using a Nu Plasma multi-collector ICP mass spectrometer at the
162 University of Melbourne (Hellstrom, 2003; Drysdale et al, 2012).

163

164 *2.3 Extraction method for radiocarbon dating*

165

166 The aim of the extraction method was to maximise recovery of the particulate
167 organic matter present in the speleothems along with as much of the non-acid soluble
168 organic matter as possible, whilst also removing any acid salts that would interfere with safe
169 combustion, minimising contamination levels from exposure to reagents and equipment,
170 and delivering a method that could be undertaken in a standard laboratory within a
171 reasonable time-frame, and be adaptable to a range of carbonates. Early experiments with
172 recovery of organics via filtering the sample were not suited to speleothems without
173 significant clay detritus, as effective recovery of organic matter from the filter proved
174 problematic, and the technique exposed the samples to increased potential contamination
175 sources. A method based on centrifuge recovery was therefore chosen.

176 The solid sectioned speleothem subsamples were cleaned in 3 M HCl (Sigma-Aldrich,
177 ACS grade, in glass) to remove surface contamination, and then washed sequentially in
178 MilliQ water, and by sonication (x 10 mins) in methanol, and then dichloromethane (DCM)
179 to wash out the surface pores of the calcite. The cleaned sections were digested in 3 M HCl
180 in 50 ml pyrex centrifuge tubes which had been pre-annealed at 450 °C. The minimum
181 amount of acid required for full digestion was used; as per Blyth et al. (2006), the acid was

182 cleaned prior to use via liquid-liquid extraction in DCM. Once digestion was complete, the
183 solution was settled, and as much supernatant as possible was removed with a pre-
184 annealed glass pipette without disturbing the particulate sediment. The top of the pipette
185 was plugged with a loose small wad of cleaned glass wool to prevent any contamination
186 being transferred from the pipette bulb.

187 MilliQ water was added to the sample tubes to remove acid salts, and the tubes
188 centrifuged at 1500 rpm for 10 mins to settle the maximum amount of organic sediment,
189 before excess wash water was removed with a pipette. For this material, two washes of 40
190 ml MilliQ water each were found to be adequate. The particulate sediment in each tube was
191 taken up in the last 2 ml of water from the final wash, and transferred to pre-weighed
192 annealed 4 ml vials, the residue was oven-dried at 80-100 °C and weighed. This step is only
193 necessary in samples like these where the physical amount of particulate matter is too large
194 to be suitable for direct transfer to the quartz combustion tube. After weighing, an aliquot
195 of the residue equivalent to 2 – 12 mg dry weight (n.b. this refers to complete dry residue
196 including clay, not carbon content) was transferred in 0.5 ml of MilliQ water to pre-annealed
197 wide bore quartz tubes (6 mm OD x 120 mm), and oven-dried at 100 °C. Quartz tubes were
198 capped with cleaned foil and lab film for transfer from Curtin University to the Australian
199 Nuclear Science and Technology Organisation (ANSTO) for radiocarbon analysis.

200 At ANSTO, the quartz tubes containing sample sediment were combusted and the
201 resultant sample CO₂ was converted to graphite following standard procedures (Hua et al.,
202 2001). The graphite was loaded into an aluminium cathode by rear pressing for ¹⁴C analysis
203 using the ANTARES accelerator mass spectrometry (AMS) facility at ANSTO (Fink et al.,
204 2004).

205

206 2.4 Preparation of procedural blanks

207

208 To estimate the level of carbon contamination or extraneous carbon incorporated
209 during the full process of sample preparation including particulate extraction and AMS
210 procedure, two series of full procedural blanks at various carbon masses were prepared.
211 Blanks followed the full cleaning and extraction procedure using empty glassware, and were
212 spiked at the end of the process with either spectroscopic grade powdered graphite (Lot No.
213 40V) from Union Carbide containing no measurable ^{14}C or lignin alkali powder (Sigma-
214 Aldrich CAS 8068-06-1) having a ^{14}C level of ~125 percent Modern Carbon (pMC). This
215 determination of the ^{14}C content and mass of extraneous carbon is necessary for a proper
216 blank correction for the measured ^{14}C ages of our particulate samples.

217

218 3. Results and discussion

219

220 3.1 U-Th dating

221

222 Table 1 shows the U-Th data for the base (B) and top (T) of the T-AQ sinter. The
223 known historical age range of the flowstone constrains the initial $^{230}\text{Th}/^{232}\text{Th}$ activity ratio to
224 between 0.88 and 0.99, and Th-corrected U-Th ages were calculated using an average value
225 of 0.94 with an uncertainty of 5%. The resulting ages are inverted, but within the
226 uncertainty resulting from initial $^{230}\text{Th}/^{232}\text{Th}$ variability. There is no clear indication of post-
227 depositional diagenetic change, and the results can be explained entirely in terms of closed-
228 system evolution from a reasonable initial $^{230}\text{Th}/^{232}\text{Th}$ value.

229

230 3.2 Radiocarbon

231

232 All the T-AQ and LP samples were of sufficient size to allow radiocarbon dating, with
233 carbon masses ranging from 0.06 – 0.19 mgC per sample. The measured ^{14}C values of these
234 15 samples in pMC are shown in Table 2.

235

236 The 21 procedural blanks (11 spiked with powdered graphite and 10 with lignin
237 alkali) had carbon masses ranging from 0.045 to 0.96 mgC. Their pMC values presented in
238 Fig. 3 show clear mass-dependent trends following hyperbolic curves very well, indicating
239 that a constant level of carbon contamination was incorporated during our sample
240 processing (Donahue et al., 1990). Using the method described by Hua et al. (2004), we have
241 determined the ^{14}C content and mass of extraneous carbon incorporated during our sample
242 processing. These values are 20.72 ± 3.09 pMC and 4.13 ± 0.43 μgC , respectively. We have
243 used these values for blank correction for all measured samples using a two component
244 mixing model (Brown and Southon, 1997; Hua et al., 2004), and the corrected ^{14}C ages of all
245 particulate samples are also presented in Table 2. Radiocarbon ages were converted to
246 calendar ages using OxCal Program v.4.2 (Bronk Ramsey, 2009), and the SHCal13 data set
247 (Hogg et al., 2013) for LP samples or the IntCal13 data set (Reimer et al., 2013) for T-AQ
248 samples. Where replicate speleothem samples have been analysed, the ^{14}C ages agree with
249 each other within 1σ uncertainties. The only exceptions to this are the samples from T-AQ-7,
250 whose ages do not agree within 2σ uncertainties. This greater level of variation in one
251 sample is likely to derive from a particularly high degree of mixed and reworked detrital
252 material in this layer (Fig 2.).

253 All the calibrated ^{14}C ages of particulate samples from both LP-13 and T-AQ are
254 consistently older than their known maximum age of formation by around 300 years for LP-
255 13, and 300 – 1000 years for T-AQ (300 – 700 years, excluding T-AQ-7r, Fig 4.). This is
256 consistent with the 500 year offset found by Genty et al. (2011). The middle samples from T-
257 AQ (layers 2 – 6) also all have very similar dates within error of each other. We hypothesise
258 that the consistent older dates of these samples are due to substantially older organic
259 carbon being transported from the source organic matter pools with slow turnover times,
260 combined with recycling of dead carbon by microbes within the aquifer. A recent study of a
261 karst soil showed that the radiocarbon age of the bulk particulate organic matter spans from
262 modern to over 2000 years BP (Hobley et al., 2014). As leaching from soil organic pools is
263 assumed to be a major source for organic matter in karst groundwaters (Baker et al., 1999),
264 it would be consistent for a similar mixture to be seen in speleothems (Blyth et al., 2008).

265 In the T-AQ samples, the fabrics support this interpretation. The columnar and
266 micrite fabrics observed in T-AQ are commonly observed in other roman aqueduct
267 carbonate sinters (Adolphe, 1973; Guendon & Vaudour, 1986), modern freshwater tufas
268 (Gradzinski, 2010; Frisia & Borsato, 2010), cave flowstones and in some stalagmites (Frisia et
269 al., 2000; Frisia & Borsato, 2010; Frisia, 2015). Similar features described in freshwater
270 laminar tufa carbonates have been interpreted as primary depositional fabrics (Brasier et al.,
271 2011) related to seasonal fluctuations in discharge (Arenas et al., 2010; Brasier et al., 2011)
272 and, subsequently, in particulate content and water supersaturation with respect to calcite
273 (Genty, 1992). Sparry columnar crystal development has been related to high-energy
274 (discharge) periods in tufa (Gradziński, 2010) and to relatively constant drip rate in
275 stalagmites (Frisia, 2015), which prevents colonization by micro-organisms, whilst micrite
276 has been observed to form in low-energy periods (Gradziński, 2010) or under low drip rates

277 (Frisia & Borsato, 2010; Frisia, 2015). As documented for Holocene stalagmites from the
278 same region, organic-rich layers mark autumnal rains and flushing of soil organic
279 compounds and colloidal particulates (Frisia et al., 2000; Scholz et al., 2012). In T-AQ, thus,
280 we interpret fabrics as being mainly controlled by discharge.

281 The presence of peloidal micrite associated with Fe oxides arranged in stromatolite-
282 like layers in the T-AQ specimen suggests the presence of iron oxidising bacteria, similar to
283 *Leptotrix ochracea*, which grow in freshwater environments in periods of low discharge
284 (Pelczar et al., 1982). Thus, the formation of micrite should mark a combination of relatively
285 low flow and a higher abundance of particulate matter (cf. Arenas et al., 2010). Micritic
286 fabrics in freshwater tufa have been linked to the presence of Extra Polymeric Substances
287 (EPS) produced by microbes, which are believed to provide nucleation sites for calcite
288 (Rogerson et al., 2008). It is thus reasonable to infer that low discharge promoted growth of
289 microbial mats or filaments, which trapped and incorporated organic and detrital particles.
290 Small freshwater bivalves appear to have also been part of the community dwelling on the
291 aqueduct floor during low discharge periods. Radiocarbon ages of spring tufas show that the
292 “autochthonous” organic compounds (i.e. those generated in situ by microbial activity)
293 included in the deposit inherit their C from a mixture of atmospheric CO₂, biogenic old C and
294 old dissolved carbonates (Özkul et al., 2010). Similarly, in the Roman aqueduct, organic
295 compounds generated by the activity of microorganisms at the interface between substrate
296 and water may have resulted in the incorporation in the sinter of old C from dissolved
297 carbonate rocks, particulates or shells. Old C can also be associated with trapped particles
298 such as charcoal, bones or soil micro aggregates, which are known to retain soil organic
299 carbon (Hobley et al., 2013) and are to be expected in waters captured from a natural spring
300 or stream. Thus, both the transported and in situ organic matter deposited in T-AQ would

301 be expected to contain a mixture of older carbon and contemporaneous terrestrial carbon,
302 as is seen in tufa. A similar argument can be made for the cave dripstone deposit LP-13,
303 which contained visible amounts of transported detritus, but also contains glycerol dialkyl
304 glycerol tetraethers (Blyth, unpublished data), which have been argued in speleothems to
305 be principally derived from microbial populations living in-situ in the cave environment
306 (Blyth et al., 2014; Baker et al., 2016).

307

308 **4. Potential for future application**

309

310 The results from this experiment suggest that radiocarbon dating of particulate
311 organic matter in “dirty” speleothems (as well as other carbonates such as tufa) provides an
312 age which is not overly distorted by contamination, but likely to be offset by 300-1000 years
313 older than the true age of carbonate formation. The T-AQ samples (excluding T-AQ 7) define
314 a growth period of around 400 years, which corresponds in length to the known available
315 time-period of formation, although this is primarily driven by the younger age of T-AQ-1,
316 with the older samples being well within error of each other. The precise degree of age
317 offset in any particular sample will be controlled by a number of factors including the ages
318 and leachability of the organic matter pools at the source, type of detrital input, the flow-
319 rate and length of water transport pathways, and microbial communities present in the
320 growth area at the time of deposition. Analysis of mineral fabrics in a sample prior to dating
321 may help resolve some of these issues, for example via identification of micrite layers
322 associated with low flow regimes and microbial activity.

323 This experiment focused on only one of the available organic fractions that can be
324 recovered from speleothems, sinters and non-photoc tufas – the humic non-acid soluble and

325 particulate material. It is possible that other organic matter fractions preserved may provide
326 more accurate dates due to more rapid deposition; for example *n*-alkanes preserved in a
327 speleothem have previously been shown to respond rapidly to changes in the overlying
328 land-use and vegetation where other parts of the organic matter pool did not (Blyth et al.,
329 2007). Recovery of molecules connected to specific sources such as plant derived lignins is
330 also possible (Blyth & Watson, 2009). Aside from the issue of sample chronology, the
331 application of radiocarbon dating to a range of organic matter fractions, might provide
332 crucial information about the ages and transport lags associated with different parts of the
333 organic matter signal in underground archives, and so advance our understanding of related
334 palaeo-proxies.

335

336 **Conclusion**

337

338 These results indicate that radiocarbon dating of the organic matter component in
339 speleothems has potential for use in palaeoenvironmental sciences, but that careful choice
340 of sample, prior analysis of mineral fabric, and isolation of the most appropriate organic
341 matter fraction will be required. In terms of use in dating “dirty” speleothems, dating of
342 particulate organic matter has been shown to provide an approximate indication of broad
343 age, which may be of value as a sample screening technique and can complement what is
344 possible by U-Th for such samples. However, a time-series dating strategy looks unlikely to
345 be achieved with this technique alone, with feasibility depending on the origin and relative
346 age of the particulate organic matter at the time of deposition. Further research is now
347 required to establish the extent to which dating of other organic matter classes preserved in
348 speleothems and tufa can contribute to chronological studies.

349

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358

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541

542 **Table and figure captions**

543

544 **Table 1.** U-Th data for the top (Aq.T) and base (Aq.B) of the T-AQ flowstone sinter545 **Table 2** - Radiocarbon results of particulate samples extracted from LP and T-AQ

546 speleothems. Note: Calendar age conversion was performed using OxCal Program v.4.2

547 (Bronk Ramsey, 2009), and the SHCal13 data set (Hogg et al., 2013) for LP samples or the

548 IntCal13 Data set (Reimer et al., 2013) for T-AQ samples.

549 **Figure 1.** T-AQ thin section showing sparry columnar calcite crystals capped by micrite

550 peloids. The micrite layer incorporates small bivalves. Note that crystal splitting occurs

551 toward the top of columnar crystals, most probably coinciding with the input of organic

552 compounds. These likely facilitated the occurrence of micrite. The bottom edge of the photo

553 is 7 mm across.

554 **Figure 2.** Subsection of the T-AQ flowstone sample showing the eight groups of layers

555 sampled. Note the clear detrital layers associated with level 7.

556 **Figure 3.** Measured pMC vs carbon mass for lignin alkaline (A) and powdered graphite (B).557 Lines denote hyperbolic curves in the form of $pMC = a + b/\text{Mass}$ fitted to the experimental

558 data (solid dots), with a and b being constant and subscripts la and g designating for lignin

559 alkali and graphite, respectively. Error bars are 1σ .

560 **Figure 4.** The T-AQ age data. Black dots are median calibrated ^{14}C ages, with error bars
561 showing the 95.4% probability age ranges. Black crosses are the top and basal U-Th dates,
562 which show an age inversion. The green/grey box shows the archaeological age of formation
563 of the flowstone.
564

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Sample	U(ngg^{-1})	$[\text{}^{230}\text{Th}/\text{}^{238}\text{U}]^a$	$[\text{}^{234}\text{U}/\text{}^{238}\text{U}]^a$	$[\text{}^{232}\text{Th}/\text{}^{238}\text{U}]$	$[\text{}^{230}\text{Th}/\text{}^{232}\text{Th}]$	Age(ka BP) ^b	$[\text{}^{234}\text{U}/\text{}^{238}\text{U}]_i^c$
Aq.T	95	0.1242(34)	1.3740(28)	0.1075(23)	1.16	1.9(0.6)	1.3761(29)
Aq.B	144	0.0946(63)	1.3724(39)	0.08175(26)	1.16	1.4(0.6)	1.3740(40)

^a Activity ratios determined after Hellstrom (2003) and Drysdale et al (2012)

^b Age in kyr before 1950 AD corrected for initial ^{230}Th using eqn. 1 of Hellstrom (2006), the decay constants of Cheng et al (2013) and $[\text{}^{230}\text{Th}/\text{}^{232}\text{Th}]_i$ of 0.94 ± 0.05

^c Initial $[\text{}^{234}\text{U}/\text{}^{238}\text{U}]$ calculated using corrected age

2- σ uncertainties in brackets are of the last two significant figures presented

Table 1. U-Th data for the top (Aq.T) and base (Aq.B) of the T-AQ flowstone sinter

Sample ID	Lab ID	$\delta^{13}\text{C}$ (‰)	Sample Mass (mgC)		Measured pMC		Blank-corrected pMC		Conv. ^{14}C Age (BP)		Calibrated ^{14}C Age (cal years BP)				
			Mean	1σ	Mean	1σ	Mean	1σ	Mean	1σ	68.2% probability		95.4% probability		Median
LP13 r1	OZR099	-20.7	0.1037	0.0011	92.98	0.35	95.86	0.48	340	40					
LP13	OZR100	-20.7	0.0879	0.0010	93.38	0.36	96.80	0.53	261	44					
LP13 mean									304	55	446	158	486	148	336
T-AQ-1	OZR101	-24.5	0.1759	0.0100	74.48	0.32	75.74	0.35	2232	38	2322	2160	2337	2153	2232
T-AQ-2	OZR102	-23.6	0.1859	0.0100	72.47	0.57	73.63	0.59	2460	64					
T-AQ-2r	OZR103	-24.5	0.1191	0.0013	70.95	0.41	72.69	0.46	2562	51					
T-AQ-2 mean									2522	71	2745	2492	2753	2380	2587
T-AQ-3	OZR104	-24.5	0.1443	0.0015	72.58	0.35	74.06	0.39	2412	43					
T-AQ-3r	OZR105	-24.5	0.0822	0.0009	70.76	0.35	73.28	0.46	2498	51					
T-AQ-3 mean									2447	60	2697	2365	2714	2357	2532
T-AQ-4	OZR106	-25.4	0.1283	0.0014	72.26	0.34	73.92	0.39	2428	43	2680	2360	2704	2353	2481
T-AQ-5	OZR107	-23.0	0.1659	0.0100	72.04	0.41	73.32	0.44	2493	48					
T-AQ-5r	OZR108	-25.3	0.1659	0.0100	71.64	0.27	72.91	0.31	2538	34					
T-AQ-5 mean									2523	30	2737	2514	2745	2492	2599
T-AQ-6	OZR109	-24.3	0.1440	0.0015	73.03	0.32	74.53	0.37	2361	40					
T-AQ-6r	OZR110	-25.3	0.1659	0.0100	72.61	0.33	73.90	0.36	2430	40					
T-AQ-6 mean									2395	48	2651	2350	2701	2340	2444
T-AQ-7	OZR111	-24.5	0.0590	0.0008	69.47	0.59	72.88	0.72	2541	80					
T-AQ-7r	OZR112	-24.5	0.0825	0.0009	67.54	0.47	69.89	0.55	2878	64					
T-AQ-7 mean									2747	232	3211	2508	3449	2342	2889
T-AQ-8	OZR113	-24.5	0.0613	0.0008	70.34	0.42	73.68	0.59	2453	64	2700	2380	2717	2358	2539

Table 2 - Radiocarbon results of particulate samples extracted from LP and T-AQ speleothems

Note: Calendar age conversion was performed using OxCal Program v.4.2 (Bronk Ramsey, 2009), and the SHCal13 data set (Hogg et al., 2013) for LP samples or the IntCal13 Data set (Reimer et al., 2013) for T-AQ samples.

Figure 1.



Figure 2.



Figure 3.

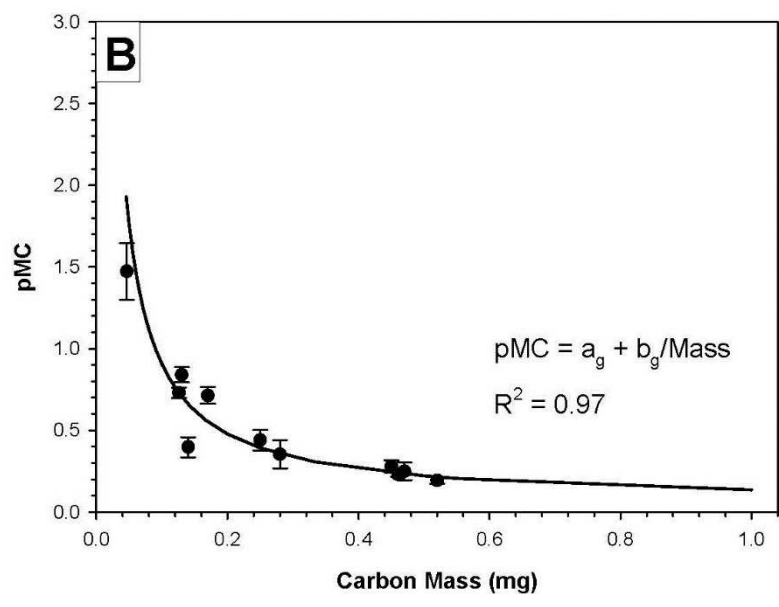
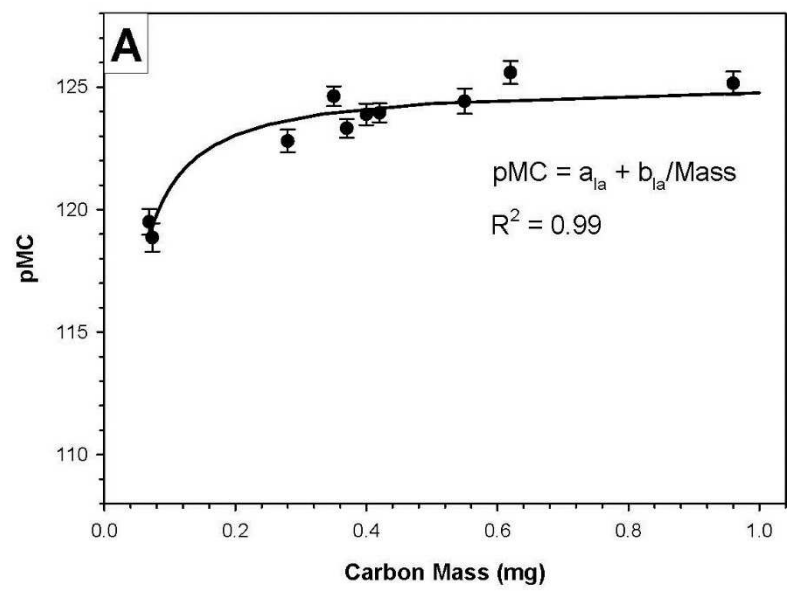
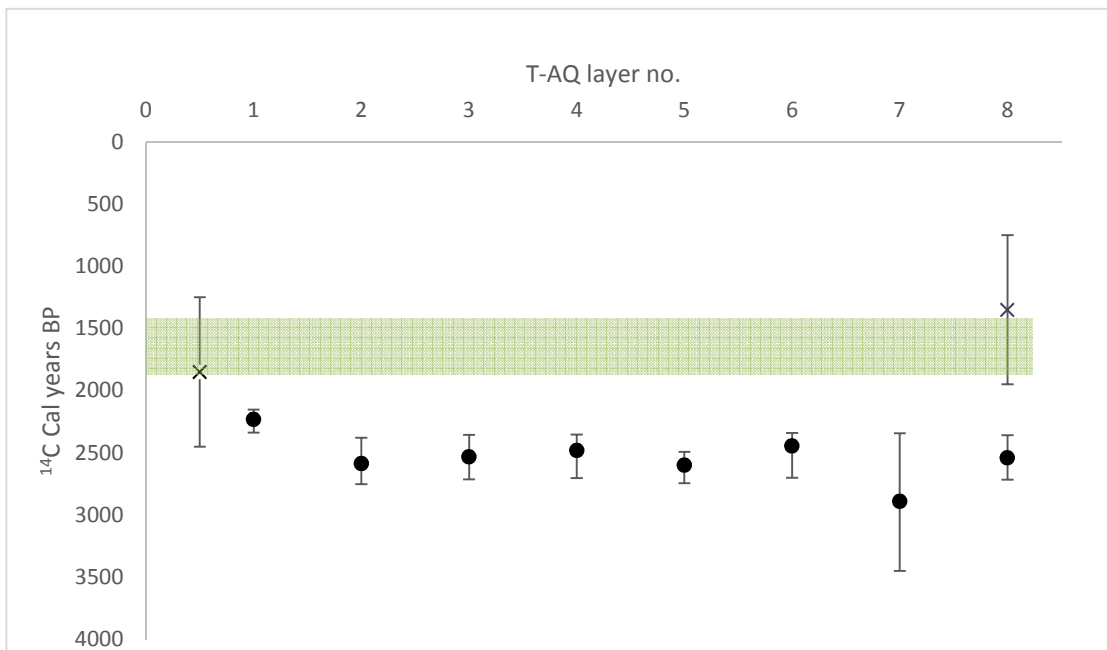


Figure 4.



Highlights

- Speleothems contain particulate and non-acid soluble organic matter amenable to radiocarbon dating
- Dates from the OM are consistently 300 – 1000 years older than the known date of formation
- The offset is hypothesised to derive from transported and reworked organic carbon