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Here we document a detailed analytical characterisation of zircon M127, a homogeneous 12.7 carat gemstone from Ratnapura, Sri Lanka. Zircon M127 has TIMS-determined mean U-Pb radiogenic isotopic ratios of $0.084743 \pm 0.000027$ for $^{206}$Pb/$^{238}$U and $0.67676 \pm 0.00023$ for $^{207}$Pb/$^{235}$U (weighted means, 2s uncertainties). Its $^{206}$Pb/$^{238}$U age of $524.36 \pm 0.16$ Ma (95% confidence uncertainty) is concordant within the uncertainties of decay constants. The $\delta^{18}$O value (determined by laser fluorination) is $8.26 \pm 0.06\%o$ VSMOW (2s), and the mean $^{176}$Hf/$^{177}$Hf ratio (determined by solution ICP-MS) is $0.282396 \pm 0.000004$ (2s). The SIMS-determined $\delta^{7}$Li value is $-0.6 \pm 0.9\%o$ (2s), with a mean mass fraction of $1.0 \pm 0.1 \mu$g g$^{-1}$ Li (2s). Zircon M127 contains $\sim 923 \mu$g g$^{-1}$ U. The moderate degree of radiation damage corresponds well with the time-integrated self-irradiation dose of $1.82 \times 10^{18}$ alpha events per gram. This observation, and the (U-Th)/He age of $426 \pm 7$ Ma (2s), which is typical of unheated Sri Lankan zircon, enable us to exclude any thermal treatment. Zircon M127 is proposed as a reference material for the determination of zircon U-Pb ages by means of SIMS analysis in combination with hafnium and stable-isotope (oxygen and potentially also lithium) analysis.
The development of the SIMS (secondary ion mass spectrometry) technique opened the world of high spatial resolution determination of U-Pb ages from polished zircon surfaces (e.g., Andersen and Hinthorne 1972, Compston et al. 1984, Williams 1998). The SIMS technique is a comparative method and the determination of accurate isotopic ratios in zircon requires calibration of the unknown results against the results of a well-characterised calibrant material measured under identical conditions during the same measurement session. In principle either a natural or synthetic reference material can be used as such a calibrant; unfortunately well-performing synthetic reference materials for the U-Pb analysis of zircon are not available. This is due to two main reasons. First, no synthesis technique is currently available that allows one to produce homogeneous Pb-bearing zircon crystals, owing to the exceptionally low partition coefficient of Pb in zircon. Second, glass reference materials cannot be used due to their different sputtering behaviour under the oxygen ion beam, compared with that of the zircon crystals to be analysed. Consequently, high-quality natural zircon reference materials must be used, and are therefore increasingly sought.

Any potentially suitable reference zircon needs to be characterised thoroughly to provide users with the best currently obtainable values, with regards to precision and accuracy, for the isotope ratios and concentrations of the calibrant material. Also, a well-performing reference zircon needs to be exceptionally homogeneous (Pidgeon 1997, Kennedy 2000), because each individual chip, and micro-areas within chips, of the reference material must have the chemical, physical and isotopic properties of the entire, previously characterised, bulk material. Another important characteristic that determines the quality of a reference material is its reliability.
during analysis. In U-Pb geochronology, higher count rates – in particular for the radiogenic Pb isotope, which is always lower in abundance than the parent U isotope – lead to better Poisson statistics. This may allow the operator to shorten the analysis time and/or to decrease the spot diameter without loss of precision. A suitable SIMS U-Pb reference should therefore contain sufficiently high concentrations of U and radiogenic Pb (whereas the concentration of non-radiogenic Pb should be negligible). However, higher levels of radiogenic Pb correlate with elevated radiation damage and this leads to an analytical difficulty, as the secondary ion sputtering behaviour under the ion beam may be different for the unknowns and reference if the structural states are sufficiently different. In searching for suitable U-Pb zircon references it is therefore advantageous to find gem-quality specimens that have some, but not too much radiation damage. Zircon M127 fulfils all of these important prerequisites. Zircon M127 contains notably higher levels of U and radiogenic Pb isotopes than most other SIMS references, resulting in a somewhat higher level of radiation damage. The degree of damage is however similar to that of zircon M257, whose damage still does not cause any identifiable matrix effects under the ion beam, which might bias the SIMS results (Nasdala et al. 2008).

The characterisation of zircon M127 began in 2006–2008, as this specimen was among a batch of fifteen gemstones that were checked for their potential suitability as future SIMS references. After this early round of analyses, zircon M257 was preferentially selected as the first zircon to be characterised as a SIMS U-Pb reference because of its particularly large size (Nasdala et al. 2008), whereas zircon M127, in spite of its excellent characteristics, was not fully characterised because it was much smaller than M257. Although small amounts of M127 were used in several research projects (e.g., Mattinson et al. 2007, Váczi et al. 2009, Mattinson 2010, Guenthner et al. 2013, Lenz and Nasdala 2015), a major fraction is still available. It has subsequently been discovered that M127 is exceptionally useful for SIMS U-Pb geochronology, especially for studies that combine U-Pb with Hf and stable-isotope determination [i.e., SIMS O-isotope and/or LA-ICP-MS (laser ablation-inductively coupled plasma-mass spectrometry)] using the same sample mounts. In this contribution we therefore summarise both early and recent measurement results characterising zircon M127.
General characterisation of zircon M127

General description and preparation

Zircon M127 was a brownish oval shaped gem (Figure 1a). The cut and faceted stone weighed 2.54 g (12.7 carats) and its longest dimension was 14.5 mm. The specimen came from the collection of the Institut für Edelsteinforschung Idar-Oberstein, Germany. It was originally found in a secondary placer deposit in the Ratnapura area, which belongs to the Highland Complex, Sri Lanka (Munasinghe and Dissanayake 1981, Kröner et al. 1994, Zoysa 2014).

Prior to preparation, thorough homogeneity checks of the original gemstone specimen were done by (i) inspection in white-light illumination under a high-resolution optical microscope and in an optical CL (cathodoluminescence) microscope, and by (ii) multiple laser spectroscopy analyses placed on different faces and under different sample orientations. The stone was found to be internally flawless, free of noticeable fractures, and uniform in colour. Neither inclusions nor growth zoning or other heterogeneities were observed. The initial Raman and laser-induced PL (photoluminescence) analyses indicated that zircon M127 has a uniform, moderate degree of radiation damage (Figure 2a) and uniform REE (rare earth element) emission patterns (Figure 2b).

The stone was cut in half along its longest dimension, and halves were cut into ~2.5 mm thick slices (Figure 1b), using a 200 µm, diamond-trimmed tungsten wire. Slices were then crushed in a steel-cylinder apparatus. The central slice pictured in Figure 1b (maximum dimension ~8 mm) was cut again very close to one of the large side planes, i.e., parallel to the initial cut direction. This was done to produce one particularly large, doubly polished thin section for homogeneity tests.
Chemical composition and homogeneity

Contents of major elements were measured by means of electron probe micro-analysis (EPMA). Multiple spots (n = 63) were measured across two large slices of M127, including the ~ 8 mm thin section. These wavelength-dispersive X-ray analyses were done using the Jeol 8900 RL EPMA system at Geowissenschaftliches Zentrum, Universität Göttingen, Germany. The accelerating voltage was 20 kV and the beam current was set at 80 nA. The electron beam was focused to a spot area 2 μm in diameter. Analyses of zircon M127 were done in the same session as those of zircon M257, hence experimental details correspond to the detailed description in Nasdala et al. (2008). Counting times varied between 15 s peak and 2 × 5 s backgrounds for main elements (Zr–Kα, Si–Kα) up to 300 s peak and 2 × 150 s backgrounds for actinides (U–Mα and Th–Mα). Synthetic calibrant materials were used, including ZrSiO₄ (for Zr and Si), HfSiO₄ (for Hf), YAG (yttrium-aluminium garnet, for Y), Yb-glass (for Yb), apatite (for P), wollastonite (for Ca), Al₂O₃ (for Al), haematite (for Fe), ThSiO₄ (for Th), and UO₂ (for U). Data were processed using the CITZAF routine in the JEOL software, which is based on the φ(pZ) method (Armstrong 1991, 1995).

Prior to EPMA analysis, detailed imaging was performed using the large, 8 mm thin section as well as seven smaller chips embedded in epoxy sample mounts and polished. First, slices and chips were inspected under a research-grade optical microscope. Neither inclusions nor growth or other zoning were found; uniform interference colours were observed in cross-polarised light. Second, back-scattered electrons (BSE) and panchromatic CL imaging was done using the EPMA. All BSE and CL images revealed uniform grey scale intensity, and neither primary growth zoning nor other internal heterogeneities were observed. In view of the particularly high sensitivity of the CL technique to reveal internal textures (for a summary see Corfu et al. 2003), the apparent absence of any heterogeneity in all CL images obtained supports the internal homogeneity of zircon M127.
Trace element analysis of zircon M127 by means of LA-ICP-MS was done in 2006 by Jan Košler (deceased), University of Bergen, Norway, as part of the characterisation of zircon M257. Analyses were done using a ThermoFinnigan Element2 single-collector double-focusing magnetic-sector system coupled to a NewWave/Merchantek UP213 laser. Analytical details and data reduction are described elsewhere (Nasdala et al. 2008). In addition, a fairly large amount of M127 (∼ 300 mg) was subjected to a solution ICP-MS analysis at GEUS (Nationale Geologiske Undersøgelser for Danmark og Grønland), Copenhagen, Denmark. This was done in 2008, to determine reliably the U mass fraction of M127. Zircon M127 was then used to re-calibrate the U data obtained by EPMA for zircon M257 (for discussion and analytical details see Nasdala et al. 2008).

The EPMA and LA-ICP-MS results are presented in Table 1. Zircon M127 contains rather moderate levels of non-formula elements; only the hafnium-oxide content (1.50% m/m) is in the per cent range. The uranium mass fraction was determined by solution ICP-MS at ∼ 923 µg g⁻¹ U (Nasdala et al. 2008). This value is considerably higher than in most other zircon reference materials [with the exception of M257 whose U mass fraction (∼ 840 µg g⁻¹) is at a similar level].

The Th/U weight ratio of zircon M127 is ∼ 0.45. This ratio is not unusual for Sri Lankan zircon (Murakami et al. 1991, Nasdala et al. 2004); however it should be noted that the value of ∼ 0.45 is much higher than 1/9, which was postulated as typical “Sri Lankan” actinide-concentration ratio by Holland and Gottfried (1955). The homogeneity of the chemical composition, and in particular of the U and Th mass fractions, was checked by two line scans across a large polished chip of zircon M127. Results obtained along a ∼ 7.8 mm long traverse are plotted in Figure 3. No lateral chemical variations exceeding the measurement uncertainties were found.

The REE mass fractions found (Table 1) are typical of igneous zircon (Figure 4; compare Hoskin and Schaltegger 2003). Also, the REE pattern is typical of unaltered zircon from igneous crustal rocks (Hoskin and Schaltegger 2003). First, the HREE (heavy rare earth elements) are enriched relative to the middle and light REE (Figure 4). Second, a positive Ce anomaly (Ce/Ce* = 15.4) and a negative europium anomaly (Eu/Eu* = 0.13) were observed. The former (i.e., HREE

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enrichment) was also evident in the PL spectrum (Figure 2b) where strong emissions were observed for the tri-valent Dy and Tm, and to a lesser extent Sm (Gaft et al. 2000, Lenz et al. 2015).

Finally, it is important to note that elemental species such as Ca, P and Fe that are typically excluded or low in primary zircon but are often absorbed extensively during alteration events (Smith et al. 1991, Geisler et al. 2003, Pérez-Soba et al. 2007, Nasdala et al. 2010), have low mass fractions in zircon M127 (Table 1). This observation shows that M127 has not undergone intense weathering or other post-crystallisation chemical alteration.

Structural state and radiation damage

Natural zircon may have appreciable structural damage created by alpha-decay of U and Th and their unstable daughter nuclei. The accumulation of this damage leads to systematic changes of physical parameters, which in turn can be used to estimate the degree of radiation damage sustained by an unknown zircon sample. We have therefore applied a number of measurement techniques for characterising the structural state of M127 (by determining specific gravity, unit-cell parameters, Raman spectral parameters and sharpness of electron-diffraction maxima). This was done to check the degree to which the observed deviation from the ideal structural state corresponds, within the well-known “Sri Lankan trend”, to the calculated self-irradiation. Obvious mismatches, in turn, could unravel structural recovery as for instance caused by post-growth alteration or synthetic thermal gem enhancement.

The self-irradiation causing the radiation damage is quantified commonly by the alpha dose \(D_\alpha\), i.e., the time-integrated number of alpha-decay events experienced by the zircon since primary growth (more precisely, since the time of closure of the U-Pb system). This parameter is calculated according to (Murakami et al. 1991, Nasdala et al. 2001)
\[ D_\alpha = \frac{6 \cdot c_{Th} \cdot N_A}{10^6 \cdot M_{232}} \left( e^{\lambda_{232} \cdot t} - 1 \right) + \frac{7 \cdot c_U \cdot 0.0072 \cdot N_A \cdot \left( e^{\lambda_{235} \cdot t} - 1 \right)}{10^6 \cdot M_{235}} + \frac{8 \cdot c_U \cdot 0.9928 \cdot N_A \cdot \left( e^{\lambda_{238} \cdot t} - 1 \right)}{10^6 \cdot M_{238}} \]

(1)

where \( c_U \) and \( c_{Th} \) = U and Th mass fractions in µg g\(^{-1}\) (Table 1); \( N_A \) = Avogadro's Number; \( M_{238} \), \( M_{235} \), and \( M_{232} \) = molecular weights of isotopes; \( \lambda_{238} \), \( \lambda_{235} \), and \( \lambda_{232} \) = decay constants of \( ^{238}\text{U}, ^{235}\text{U}, \) and \( ^{232}\text{Th} \); and \( t \) = zircon U-Pb age (here 524.3 Ma; see below). It should be noted that the common use of the U-Pb age in calculating \( D_\alpha \) adds slight uncertainty, as this practice is based on the – often incorrect – assumption that the closure of structural recovery by annealing corresponds exactly with U-Pb system closure. For natural zircon, \( D_\alpha \) actually is not a “measure” of the structural damage present in a given natural zircon sample. Rather, \( D_\alpha \) quantifies the total self-irradiation since the time of closure of the U-Pb system. How much of the damage has been accumulated or has been thermally annealed, depends on the thermal history of the sample. Nasdala et al. (2001, 2004) and Palenik et al. (2003) found that the Ratnapura, Sri Lanka, gem-zircon suite is not as radiation-damaged as expected given the U and Th mass fractions and ages; only about 55% of the damage was retained. This is assigned to the particular geological history of the area (involving slow cooling prior to Ordovician uplift, and perhaps a re-heating event; Nasdala et al. 2004). In summary, radiation damage cannot be compared among samples of different origin by the simple evaluation of their calculated \( D_\alpha \) values. However, internal comparison is straightforward with Sri Lankan zircon gems, because – provided no gem enhancement involving elevated temperatures was applied – they all show uniform trends of parameter changes depending on \( D_\alpha \) (Nasdala et al. 2004). The calculated value of \( D_\alpha = 1.82 \times 10^{18} \) events per gram for zircon M127 is rather moderate. It is well below the first percolation point (for Sri Lanka zircon determined at \( \sim 3.5 \times 10^{18} \alpha \) g\(^{-1}\); Salje et al. 1999) that is characterised by the beginning of three-dimensional inter-connection of amorphous clusters in the crystalline matrix. The nearly complete transformation of Sri Lankan zircon to the amorphous state requires self-irradiation doses of \( \sim 10^{19} \alpha \) g\(^{-1}\) (Salje et al. 1999, Nasdala et al. 2002). The calculated alpha dose of \( 1.82 \times 10^{18} \) events hence suggests a notable but still moderate level of radioactive self-irradiation.

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The specific gravity of zircon M127 was determined at 4.625 ± 0.005 g cm\(^{-3}\). This was done prior to preparation, by repeated weighing of the original stone in air and in distilled water. The gravity of M127 is ~2% lower than that of crystalline zircon (> 4.7 g cm\(^{-3}\)) but well above than that of metamict zircon (< 4 g cm\(^{-3}\); e.g., Sahama 1981, Nasdala et al. 2002), indicating a notable, but still moderate degree of self-irradiation induced volume expansion. As zircon M127 has a “regular” chemical composition with generally low amounts of non-formula elements, and as it is free of inclusions, any chemical or textural effects on the sample’s specific gravity must be very minor. The moderately decreased specific gravity is therefore assigned solely to the moderate degree of radiation damage present.

Unit-cell dimensions of zircon M127 were determined from four 50–100 µm sized chips, by obtaining ten frames each on a Bruker Nonius KappaCCD single-crystal X-ray diffractometer. Analyses were done with Mo–K\(_\alpha\) radiation. The distance between chip and detector was 35 mm. The measurement time per frame (width 2\(^\circ\)) was 3 s. The dimensions of the tetragonal unit-cell were determined at \(a_0 = 6.6408(4)\) Å and \(c_0 = 6.0367(5)\) Å, resulting in a unit-cell volume of \(V = 266.22(4)\) Å\(^3\). Compared with well-crystallised zircon, zircon M127 is characterised by expansion of the unit cell of ~2%. First, this corresponds very well with the ~2% decrease of the specific gravity (cf. above). Second, the ~2% cell expansion characterises the material as moderately radiation-damaged (Holland and Gottfried 1955, Murakami et al. 1991). Note that the observed expansion of parameters \(a_0\) and \(c_0\) is rather “homogeneous” (Holland and Gottfried 1955; compare also figure 6A in Nasdala et al. 2004). Partially annealed zircon, in contrast, is typically characterised by an apparent \(a_0 - c_0\) mismatch, caused by preferential recovery of the original expansion along the crystallographic \(a\) axis (Weber 1990, 1993).

The degree of radiation damage was further estimated from the full width at half maximum (FWHM) of the main Raman band of zircon \([\nu_3(SiO_4)\) vibrational mode; Nasdala et al. 1995, 2001]. Spectra were obtained with He-Ne 632.8 nm excitation (8 mW at the sample surface) by
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Isotopic composition of zircon M127

U-Pb age determination by TIMS

The U-Pb age of zircon M127 was determined by thermal ionisation mass spectrometry (TIMS) analysis in the geochronology laboratories of the University of California at Santa Barbara, USA, and University of Oslo, Norway. Results of chemical abrasion (CA) TIMS analyses of one 1.437 mg chip at the University of California have already been published elsewhere (Mattinson 2010).

At the University of Oslo, one 0.646 mg chip of M127 was subjected to isotope dilution (ID) TIMS analysis. The chip was cleaned with HNO₃, H₂O and acetone, then weighed on a microbalance, and transferred to a Krogh-type bomb (see Krogh 1973), to which a measured amount of ²⁰⁹Pb – ²⁰⁸Pb – ²³⁵U spike was added. The spike was evaporated before adding a mixture of HF and HNO₃ (12:1). The bomb was placed in an oven for 5 days at 190 °C. The solution was then evaporated to dryness and re-dissolved in 3 mol l⁻¹ HCl overnight at 190 °C. The solution was then split with a pipette onto four columns with anion exchange resin to

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separate out Pb and U. The latter were then loaded on zone-refined-Re filaments together with Si-gel and H$_3$PO$_4$ and measured in a MAT262 mass spectrometer. The main Pb and U peaks were measured on multiple faraday cups in static mode, and the $^{207}$Pb/$^{204}$Pb ratio was measured by peak-jumping on an electron multiplier. Other details of the procedure can be found in Corfu (2004). The spike composition was calibrated against the synthetic ET100 solution (Condon et al. 2008) provided by the EARTHTIME initiative (http://www.earth-time.org).

The $^{238}$U decay constant of Jaffey et al. (1971; 1.55125 × 10$^{-10}$ a$^{-1}$) and the re-revised $^{235}$U decay constant of Mattinson (2010; 9.8571 × 10$^{-10}$ a$^{-1}$), which is marginally higher than the revised value of Schoene et al. (2006; 9.8569 × 10$^{-10}$ a$^{-1}$), were used for converting isotopic ratios into ages and for plotting the results. Age calculations and the preparation of the Concordia plot were done using the Isoplot program (Ludwig 2003). Note that isotopic ratios are reported as measured (i.e., without Th fractionation correction). The uncertainties reported include quadratically propagated measurement errors, a minimum reproducibility term of ± 0.06% amu$^{-1}$ on the fractionation corrections for Pb and U (based on regular runs of standard solutions), blank corrections of 2 ± 1 pg Pb and 0.1 ± 0.05 pg U, and uncertainties of 2% on the $^{206}$Pb/$^{204}$Pb and 1% on the $^{207}$Pb/$^{204}$Pb compositions of the blank.

The U-Pb results obtained in the University of Oslo laboratory are summarised in Table 2, and a Concordia plot including both these results and the results of Mattinson (2010) is presented in Figure 7. From all twenty TIMS analyses, a weighted mean $^{206}$Pb/$^{238}$U ratio for zircon M127 of 0.084743 ± 0.000027 (2s) was calculated, which corresponds to a $^{206}$Pb/$^{238}$U age of 524.36 ± 0.16 Ma (uncertainty quoted at the 95% confidence level). The weighted mean $^{207}$Pb/$^{235}$U isotopic ratio was calculated at 0.67676 ± 0.00023 (2s). The mean Concordia age calculated is 524.37 ± 0.46 Ma (95% confidence; errors of decay constants included). The U-Pb system of zircon M127 is well concordant within the uncertainties of the decay constants (for further discussion see Schoene et al. 2006, Schmitz and Schoene 2007, Mattinson 2010). We
recommend the above mean isotopic ratios and $^{206}\text{Pb} / ^{238}\text{U}$ age to be used when applying zircon M127 as reference material in SIMS U-Pb dating.

**Hafnium isotope determination by solution ICP-MS**

Hafnium mass fractions, and $^{176}\text{Hf} / ^{177}\text{Hf}$ and $^{176}\text{Lu} / ^{177}\text{Hf}$ ratios, were measured by solution isotope-dilution analysis. Three multiple collector (MC) ICP-MS systems at different laboratories were used: the Radiogenic Isotope and Geochronology Laboratory (RIGL) at Washington State University (WSU), Pullman; the Institute of Geology and Geophysics, Chinese Academy of Sciences (IGGCAS), Beijing; and the joint Köln-Bonn laboratory, Germany. In order to check for the internal homogeneity of Hf isotopes within zircon M127, each of the three laboratories was provided with five small samples (1–3 mg per sample consisting of one, two, or three individual chips). The five samples were recovered from the five main slices of the original gemstone (see Figure 1b; one sample per slice).

In the RIGL facility, each fragment was separately dissolved in a mixture of concentrated HF and HNO$_3$ in pre-cleaned 0.5 ml Savillex® perfluoroalkoxy screw-top vials. The vials were loosely capped and placed in Teflon® liner in a large-capacity Parr dissolution bomb to which ~ 5 ml of concentrated HF were added to provide vapour pressure within the bomb. The sealed bomb was placed in an oven at 250 °C for 48 hours. Following dissolution, the contents of each bomb were transferred to at 7 ml Savillex® beaker, dried, and fluxed overnight in a mixture of 2.5 mol l$^{-1}$ HCl/H$_3$BO$_3$ to convert to chlorides and minimise production of fluoride species. Samples were again dried and redissolved in 1 ml 1 mol l$^{-1}$ HCl plus 0.1 mol l$^{-1}$ HF (precisely weighed). From this solution an aliquot for analysis was removed and added to a clean 7 ml Savillex® beaker to which a zircon-specific mixed $^{176}\text{Lu} - ^{180}\text{Hf}$ tracer solution was added. Further details of sample preparation, column chemistry, and Hf and Lu mass spectrometry are described in Vervoort et al. (2004) and Goodge and Vervoort (2006). The purified Hf of samples (and calibrants) were run as 50 ng ml$^{-1}$ solutions and introduced to the RIGL ThermoFinnigan Neptune MC-ICP-MS using an Aridus desolvating nebuliser. Mass fractionation was internally corrected using
$^{179}\text{Hf} /^{177}\text{Hf} = 0.7325$ and all sample analyses were normalised using the Hf isotope reference material JMC 475 (Patchett and Tatsumoto 1980; with a presently accepted value of $^{176}\text{Hf} /^{177}\text{Hf} = 0.282160$). Five analyses of JMC 475 were interspersed between samples and yielded a mean $^{176}\text{Hf} /^{177}\text{Hf} = 0.282132 \pm 0.000004$ (2s; $n = 5$).

For analyses at the IGGCAS, single chips of zircon M127 were weighed and dissolved in HF-HNO$_3$ in 3 ml vials placed in high-pressure bombs at 210 °C for one week. Afterwards the samples were dried down and re-dissolved in 3 mol l$^{-1}$ HCl. Each sample solution was then split. About 20% was spiked with a mixed $^{176}\text{Lu}$ and $^{180}\text{Hf}$ tracer for determining the Lu and Hf mass fractions. The spike solution used was calibrated beforehand against a standard solution made from pure metals (Yang et al. 2010). It was tested on several international calibrants (BCR-2, W-2) and yielded results corresponding to reference values (Münker et al. 2001). The remaining $\sim$80% of each initial sample solution was used to determine the Hf isotopic composition. The chemical purification procedure described in Nebel-Jacobsen et al. (2005) and Morel et al. (2008) was applied. Isotope measurements were performed on a Thermo Scientific Fisher Neptune MC-ICP-MS following the protocol listed in Yang et al. (2010). Instrumental mass bias was corrected offline using the exponential law and $^{179}\text{Hf} /^{177}\text{Hf} = 0.7325$. For possible interferences of $^{176}\text{Yb}$ and $^{176}\text{Lu}$ on $^{176}\text{Hf}$, isotopes $^{173}\text{Yb}$ and $^{175}\text{Lu}$ were monitored and corrected for, applying $^{176}\text{Lu} /^{175}\text{Lu} = 0.02655$ and $^{176}\text{Yb} /^{173}\text{Yb} = 0.79631$ (Vervoort et al. 2004). The $^{176}\text{Hf} /^{177}\text{Hf}$ results for M127 are reported relative to a value of 0.282160 for JMC 475.

The procedure in the Köln-Bonn laboratory was broadly similar. Lutetium and Hf isotope measurements were performed using a Thermo Neptune MC-ICP-MS and a procedure after Nebel-Jacobsen et al. (2005), employing a mixed $^{176}\text{Lu}-^{180}\text{Hf}$ tracer specifically designed for zircon. Results for $^{176}\text{Hf} /^{177}\text{Hf}$ are given relative to a value of 0.282160 for the Münster AMEs standard solution that is isotopically indistinguishable from JMC 475.

All results are presented in Table 3 and Figure 8. They indicate that zircon M127 is fairly uniform in Hf mass fraction and $^{176}\text{Hf} /^{177}\text{Hf}$ ratio. Considering all eighteen individual analyses (but

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without consideration of re-analyses which are merely reported for completeness), the
unweighted mean $^{176}\text{Hf}/^{177}\text{Hf}$ ratio is determined at $0.282396 \pm 0.000004$ (2s), which we
suggest be used as a "working value". The $^{176}\text{Hf}/^{177}\text{Hf}$ ratios determined in Köln may seem to be
slightly higher compared with data produced in the two other laboratories. Differences among
the three laboratories are, however, within measurement uncertainties and hence cannot be
considered systematic.

**Hafnium isotope determination by LA-ICP-MS**

To check the internal homogeneity of the $^{176}\text{Hf}/^{177}\text{Hf}$ ratio, 130 individual LA-ICP-MS analyses
were performed using the Neptune MC-ICP-MS system of the Institute of Geology and
Geophysics, Chinese Academy of Sciences, Beijing. The JMC 475 standard solution with 200 ng
g$^{-1}$ Hf was used for evaluating the reproducibility and accuracy of the instrument prior to laser
ablation analyses. During laser ablation determinations of Hf isotopes, the isobaric interference
of $^{176}\text{Lu}$ on $^{176}\text{Hf}$ was relatively small compared with the interference of $^{176}\text{Yb}$ on $^{176}\text{Hf}$, which
must be carefully corrected since the contribution of $^{176}\text{Yb}$ to $^{176}\text{Hf}$ can profoundly affect the
accuracy of the measured $^{176}\text{Hf}/^{177}\text{Hf}$ ratio (e.g., Fisher et al. 2011). Here, the mean $^{173}\text{Yb}/^{171}\text{Yb}$
ratio of all individual spot analyses was used to calculate the fractionation coefficient ($\beta_{\text{Yb}}$), and
then to calculate the contribution of $^{176}\text{Yb}$ to $^{176}\text{Hf}$ using an isotopic ratio of $^{176}\text{Yb}/^{172}\text{Yb} = 0.5887$
(Wu et al. 2006). Reference zircon 91500 was used for 'external' correction. During the
measurement sessions, the $^{176}\text{Hf}/^{177}\text{Hf}$ value of 91500 was $0.282294 \pm 0.000039$ (2s, $n = 436$),
which was adjusted to $0.282305$ (correction of 0.000011), a reference value recommended for
91500 (Wu et al. 2006). During data acquisition, analyses of GJ-1 as an unknown yielded a
weighted $^{176}\text{Hf}/^{177}\text{Hf}$ ratio of $0.282018 \pm 0.000048$ (2s, $n = 116$), within error the recommended
value (Morel et al. 2008). Data are not presented in detail here; they are available online (Table
S1; see link given at the end of the paper). Analyses did not yield any significant variations
within and among chips, underlining the homogeneity of zircon M127; the mean $^{176}\text{Hf}/^{177}\text{Hf}$
value of all 130 analyses was $0.282392 \pm 0.000031$ (2s).
Hafnium isotope determination – discussion

One puzzling observation related to the solution isotope-dilution results (Table 4) is the diversity of Lu mass fractions (and, connected with that, of $^{176}\text{Lu}/^{177}\text{Hf}$ ratios). Fairly uniform Lu mass fractions were measured in Köln-Bonn whereas the results of the other two laboratories show a comparably large scatter. These differences could point to sample heterogeneity. Recall, however, that all three laboratories received corresponding fragments recovered from the same areas in the five gemstone slices, so samples may be assumed to represent a nearly identical set. This implies that sample heterogeneity is most likely not the cause of scattered Lu mass fractions, because otherwise similar scatters would be expected to have been found by all three laboratories. Possible other reasons for Lu-data variations include difficulties relating spiking sub- aliquots some time period after dissolution (as done in the RIGL facility and at the IGGCAS). This practice may have resulted in differing behaviour of Hf and Lu during the chemical treatment. For instance, it may be speculated that REE-fluoride complexes might have formed that prevented the spike and sample from equilibrating completely with respect to Lu. We therefore cannot be sure the variations in $^{176}\text{Lu}/^{177}\text{Hf}$ observed by the RIGL and IGGCAS laboratories are real. Rather we assess the Köln-Bonn results as reliable, which suggest the true $^{176}\text{Lu}/^{177}\text{Hf}$ is homogeneous. Thus we suggest the Köln-Bonn result of $^{176}\text{Lu}/^{177}\text{Hf} = 0.0006146 \pm 0.0000130$ (2s) be used as a “working value” in Hf-Lu studies. It is however clear that the possibility of interlaboratory variations needs to be resolved in future measurement campaigns.

Oxygen isotope determination by laser fluorination

Sixteen fragments of zircon M127 (masses in the range 1.83–3.98 mg) were analysed for oxygen-isotope ratios at the University of Wisconsin, Madison. Analyses were done in four separate sessions. The routine procedure at UW is to treat zircons in cold HF for 8 hours before laser fluorination and analysis for oxygen isotope ratios. This procedure removed non-zircon impurities and also dissolved intra-crystalline domains that were severely radiation damaged and thus susceptible to post crystallisation alteration of $\delta^{18}\text{O}$ (Valley et al. 2005, 2015); it has
been shown not to affect $\delta^{18}$O of undamaged zircon (Valley 2003). Two chips that were HF-treated for the December 10, 2015, session lost 14% m/m, which is more than expected for such clean fragments. The fragments were then pre-treated at room temperature in a BrF$_5$ atmosphere for 12 hours and each was heated in the presence of fresh BrF$_5$, using an infrared laser ($\lambda = 10.6$ µm). The evolved O$_2$ gas was cryogenically purified, converted to CO$_2$, and analysed by dual-inlet gas-source mass spectrometer as described by Valley et al. (1995).

Garnet reference material UWG-2 (Valley et al. 1995) was analysed four to six times in each analysis session of M127. The $\delta^{18}$O values obtained for M127 were normalised to the recommended value of 5.80‰ VSMOW (Vienna Standard Mean Ocean Water) for UWG-2 (Valley et al. 1995). Nine analyses of untreated fragments yielded a value of $\delta^{18}$O = 8.26 ± 0.06‰ VSMOW for zircon M127 (Table 4), whereas seven analyses of HF-treated fragments yielded a value of $\delta^{18}$O = 8.49 ± 0.05‰ VSMOW (2s; unweighted means). These data are systematically higher for the HF-treated zircon grains by an average of 0.23‰.

The difference obtained by laser fluorination analysis of M127 between untreated and HF-treated zircon is small and comparable to the measurement precision obtained by in situ SIMS analysis. We interpret the HF-treated value as most reliably reflecting the original composition of the M127 zircon. If the HF-treatment has selectively removed zircon domains that are slightly higher in $\delta^{18}$O, then the untreated value for $\delta^{18}$O best represents the bulk composition of the M127 crystal. Any higher $\delta^{18}$O domains will be included in a SIMS analysis and if present, the homogeneity of SIMS analyses suggests that higher $\delta^{18}$O domains are sub-micrometre and randomly distributed. Thus the untreated value $\delta^{18}$O = 8.26 ± 0.06‰ VSMOW (2s) is recommended for calibration of SIMS analysis with the caveat that there are currently no estimates for the effect of radiation damage on the instrument bias (or instrumental mass fractionation). Although, it is usually recommended to use a zircon with less radiation damage for determination of oxygen isotope ratios, such as KIM-5 (Valley 2003), M127 might be a better reference for samples with comparable amounts of damage.
Oxygen isotope determination by SIMS

To check the homogeneity of the oxygen-isotope ratio, SIMS analyses were done at Institute of Geology and Geophysics, Chinese Academy of Sciences, Beijing, and at the Helmholtz-Zentrum Potsdam (GFZ). In Beijing, a total of thirty point analyses placed in six chips originating from five slices of the original stone were performed using a Cameca IMS 1280. The primary Cs\(^+\) beam was accelerated at 10 kV, with a current of 1.2 nA. The spot size was about 20 \(\mu\)m in diameter (10 \(\mu\)m beam diameter plus 10 \(\mu\)m raster). Low-energy electron flooding was used to compensate for sample charging during analysis. Secondary ions were extracted with a -10 kV potential. Oxygen isotopes were measured in multi-collection mode with two off-axis Faraday cups. Each analysis consisted of twenty cycles per 4 s counting time. The instrumental mass fractionation factor was corrected using the Penglai zircon reference (with \(\delta^{18}\)O = 5.3‰ VSMOW; see Li et al. 2010). Measurement repeatability was better than 0.4‰ (2s) for \(\delta^{18}\)O. Calibrated against the Penglai reference, the data set yielded a mean of \(\delta^{18}\)O = 8.34 ± 0.34‰ VSMOW (2s).

The Potsdam Cameca IMS 1280-HR performed a total of forty-one point analyses on a total of five chips. These five chips were recovered from the five main slices of the original gemstone (see Figure 1b; one chip per slice). These analyses employed a \(\sim\)2.5 nA, mass filtered \(^{133}\)Cs\(^+\) beam with a Gaussian density distribution, which was focussed to a \(\sim\)5 \(\mu\)m diameter on the polished sample surface. The total impact energy of the Cs\(^+\) ions was 20 keV. Each analysis was preceded by 20 \(\mu\)m x 20 \(\mu\)m rastered pre-sputter for 60 s. The total pressure in the sample chamber was \(1 \times 10^{-7}\) Pa at the time of analysis. Charge compensation was achieved with low-energy, normal-incidence electron flooding using a current of \(\sim\)1 \(\mu\)A. All analyses were conducted in duplicate in sequence with an offset between the two analytical locations of 50 \(\mu\)m, to assess the short-term stability of our technique. In order to suppress within-run drift in the isotope ratio, a 10 \(\mu\)m x 10 \(\mu\)m raster was employed during data collection; thus assuring a crater geometry with a flat bottom. Secondary ions were accelerated by a -10 kV potential applied to the sample holder. The mass spectrometer was operated in static multi-collection.
mode, with the \(^{16}\text{O}^-\) being collected in the \(L2'\) Faraday cup and the \(^{18}\text{O}^-\) signal being collected in the \(H2'\) Faraday cup; the amplifier system employed thermally stabilised \(10^{10} \text{\Omega}\) and \(10^{11} \text{\Omega}\) amplifiers, respectively. The mass resolution of the instrument was determined to be \(M/\Delta M = 1670\) on the \(^{16}\text{O}\) mass station and \(M/\Delta M = 1970\) on the \(^{18}\text{O}\) mass station. This is effectively full transmission for the tool and fully suffices to eliminate both the \(^{16}\text{O}^1\text{H}_2\) and the \(^{16}\text{O}^2\text{H}\) isobaric interferences from the \(^{18}\text{O}\) mass station. A single analysis consisted of twenty integrations of 4 s each. Using white light profilometry, we determined the test portion mass for our analyses was equivalent to ca. 150 picograms. Data for M127 were corrected for instrumental mass fractionation using the 91500 reference (with \(\delta^{18}\text{O} = 9.86\text{‰ VSMOW};\) see Wiedenbeck et al. 2004). Analysis on M127 yielded a mean of \(\delta^{18}\text{O} = 7.96 \pm 0.20\text{‰ VSMOW}\) (2\(\sigma\)) for forty-one determinations on the five fragments, meaning this data set is consistent with being a homogeneous population.

Data are available online (see supporting information Table S2). No systematic variations or “outliers” were found among the thirty analyses in the Beijing laboratory. Among the forty-one analyses in Potsdam, two measurements in a single area gave results 0.25‰ lighter than the remainder of the population, and suspiciously these were the first data acquired during our measurement series. Discarding these two, seemingly discrepant, values yielded a measurement repeatability of \(\pm 0.16\text{‰ } (2\sigma)\) on \(n = 39\) determinations, indicating that M127 is at least as homogeneous as the 91500 zircon, which we presume to be homogeneous in its \(\delta^{18}\text{O}\) content. We therefore conclude that M127 is well suited for use as a reference material in SIMS studies that correlate U-Pb age and \(\delta^{18}\text{O}\) of unknown zircon samples (see Cavosie et al. 2005, Valley et al. 2005).

**Oxygen isotope determination – discussion**

Unlike the homogeneous Sri Lanka zircon references CZ3 (Cavosie et al. 2011) and M257 (Nasdala et al. 2008), the \(\delta^{18}\text{O}\) value determined for zircon M127 is not remarkably high and hence does not fall outside the range typical of igneous zircon (Figure 9). Unlike CZ3 and M257,
the oxygen-isotope ratio of M127 does not provide a clear indication of the formation environment (whereas the particularly high $\delta^{18}O$ values of some Sri Lanka zircon samples are presumed to indicate formation in a marble or Ca-silicate metamorphic rock; Cavosie et al. 2011). We can therefore merely speculate that the remarkable homogeneity (i.e., absence of all signs of “igneous” growth zoning) of zircon M127 might be explained by (i) metamorphic formation that however did not result in a very high $\delta^{18}O$ value, or (ii) igneous formation that must have resulted in the formation in particularly thick growth zones. As zircon M127 was derived from a gem gravel, with the original host rock being unknown, the formation environment remains uncertain.

**Lithium isotope determination by SIMS**

Lithium-isotope determinations were performed using the Cameca IMS-1280 HR at the Institute of Geology and Geophysics, Chinese Academy of Sciences, Beijing. A detailed description of the procedure followed can be found in Li et al. (2011) and therefore only a brief description is given here. The primary O’ ion beam was accelerated at −13 kV, with a beam current of 28 nA. The beam was focused to an elliptical spot ca. 20–30 µm in size. Positive secondary ions were extracted with a 10 kV potential. A single ion-counting electron multiplier (EM) was used as the detection device. The mass resolution was 1300 (at 10% peak height). Each measurement consisted of eighty cycles, with a total measurement time of 20 min. The M257 zircon reference (Li = 0.86 µg g$^{-1}$ and $\delta^{7}Li = 2.1 \pm 1.0‰$; see Li et al. 2011) was used as calibrant material.

A total of fifty-five analyses were performed on six fragments of M127 that were embedded in a polished epoxy sample mount. The mean Li mass fraction of zircon M127 is $1.0 \pm 0.1 \mu g \, g^{-1}$ (2s), and the mean $\delta^{7}Li$ value was determined at $-0.6 \pm 0.9‰$ (2s). This can initially be used as a working value for M127, though there is an urgent need for further characterisation of this system using other, independent analytical methods (i.e., solution ICP-MS). In view of the low $\delta^{7}Li$ variability observed in the present study, M127 may be a useful reference material in SIMS.
studies that use Li as a petrogenetic tracer (Ushikubo et al. 2008, Li et al. 2011, Bouvier et al. 2012).

(U-Th)/He analysis

The retention of the radiogenic \(^4\)He was evaluated by (U-Th)/He analyses at the University of Arizona, Tucson. Our experimental details are described elsewhere (Nasdala et al. 2004, Reiners 2005), and our results are presented in Table 5. Analyses on chips M127zA and M127zB were done in 2006, in the course of the characterisation of zircon M257; the other chips were analysed in 2015. All five analyses yielded a weighted mean He age of 426 ± 7 Ma (2s) for zircon M127. This value is well within the range of typical helium ages for unheated zircon samples from Sri Lanka (Hurley 1954, Nasdala et al. 2004). It also confirms that zircon M127 has never been subjected to heat treatment, as this would have caused near total loss of radiogenic helium (Reiners et al. 2005).

Conclusions

Zircon M127 is a suitable reference material for the SIMS U-Pb analysis of unknown zircon samples, especially in studies that subject the same sample mounts to the additional analysis of oxygen (SIMS), lithium (SIMS), or hafnium isotopes (LA-ICP-MS). Zircon M127 is homogeneous in terms of chemical and isotopic composition and has a remarkably concordant U-Pb isotopic system. Therefore M127 is suitable for calibration, quality control and method validation purposes for analytical techniques working at the low- to sub-nanogram sampling scale, and its broad acceptance by the SIMS community would greatly improve the metrological traceability between facilities. A particularly advantageous property of M127 is the fairly high uranium mass fraction of \(\sim 923 \mu g \, g^{-1}\) (and correspondingly high levels of radiogenic Pb), which will result in exceptionally high count rates and good counting statistics for SIMS analyses.
The material will in future be used and distributed by the Beijing SHRIMP Centre, Institute of Geology, Chinese Academy of Geological Sciences (contact: liudunyi@bjshrimp.cn). Zircon M127 is reserved as SIMS reference. In particular, it will not be provided for only U-Pb geochronology in LA-ICP-MS laboratories. Despite this decision, it will of course be possible, and worthwhile, to subject re-polished SIMS mounts containing M127 to subsequent analyses by means of LA-ICP-MS, including Hf-isotope or LASS (laser ablation split stream method for combined U-Pb and Hf analyses; Fisher et al. 2014). This decision was made to reduce the consumption of zircon M127 to a minimum and ensure that the material will be available for SIMS work for several more years.

Acknowledgements

This article is dedicated to Jan Košler (University of Bergen, Norway), who passed away in 2014 at the age of 49. Jan was involved in the initial characterisation of zircon M127.

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Supporting information

The following supporting information is available online:

Table S1. Results of a total of 130 Hf isotope determinations by LA-ICP-MS (six measurement sessions 2009–2013) at the Chinese Academy of Sciences, Beijing

Table S2. Results of O isotope determinations by SIMS at the Chinese Academy of Sciences, Beijing, and the Helmholtz-Zentrum Potsdam (GFZ)

This material is available as part of the online article from: http://onlinelibrary.wiley.com/doi/10.1111/ggr.11111/abstract (This link will take you to the article abstract).

Figure captions

Figure 1. White-light photographs of zircon M127, a clear, dark brownish, cut and faceted gemstone originating from Ratnapura, Sri Lanka. (a) Original stone as purchased. (b) One half after cutting into five slices.

Figure 2. Spectroscopy of zircon M127. (a) Raman spectra in the SiO$_4$ stretching range, showing M127 in comparison with three zircon reference materials ranging from very well crystallised (Penglai; spectrum this work; for sample description see Li et al. 2010), mildly radiation damaged (91500, data from Wiedenbeck et al. 2004) to strongly radiation damaged (OR1, data from Nasdala et al. 2004). (b) The PL spectrum (473 nm excitation) of M127 is dominated by groups of narrow, REE-related emissions. For assignment see Gaft et al. (2000) and Lenz and Nasdala (2015).
Figure 3. Results of Raman microprobe and EPMA line scans across an ~ 8 mm long traverse. Lateral variations of observed Raman and concentration parameters were below the measurement uncertainties (visualised by grey bars).

Figure 4. Plot of chondrite-normalised REE mass fractions determined by LA-ICP-MS (University of Bergen). The grey area visualises REE mass fraction ranges obtained for unaltered igneous zircon (graphically extracted from figures 4 and 5 in Hoskin and Schaltegger 2003).

Figure 5. Plots of unit-cell parameter $c_0$ and the width of the main Raman band at $\sim 1000$ cm$^{-1}$ against the time-integrated alpha dose, for zircon samples from Sri Lanka. Literature data for unit-cell constants from Holland (1954), Murakami et al. (1991) and Nasdala et al. (2004). Literature data for Raman FWHMs from Zhang et al. (2000) and Nasdala et al. (2001, 2004). Data for zircon M257 from Nasdala et al. (2008). Note that parameters of zircon M127 plot well within the “Sri Lankan” trends (visualised by dashed arrows).

Figure 6. Electron diffraction in the TEM. (a) Photograph of the diffraction pattern of zircon M127 observed along [001]. (b) Plot of the intensity distribution along the line marked in a with arrows (solid graph), in comparison with an analogous intensity distribution pattern obtained from highly crystalline zircon from Ban Lung, Ratanakiri, Cambodia (dotted graph; for sample description see Lenz and Nasdala 2015). Graphs are shown with vertical offset for clarity.

Figure 7. Concordia plot of U-Pb data obtained at the University of California at Santa Barbara (CA-TIMS; sixteen plateau steps, data from Mattinson 2010) and at the University of Oslo (ID-TIMS, four individual analyses). All ellipses represent 2s uncertainties. The uncertainty of the mean age quoted includes uncertainties of decay-constants.

Figure 8. Results of Hf-isotope determinations (solution-ICP-MS) done at University of Cologne (5 individual analyses), Washington State University, Pullmann, WA (7), and the Chinese Academy of Sciences, Beijing (6). Error bars represent 2s uncertainties.

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Figure 9. Plot of $\delta^{18}$O data for zircon M127 (laser fluorination; University of Wisconsin, Madison) and data for references KIM-5 (Valley 2003), M257 (Nasdala et al. 2008) and CZ3 (Cavosie et al. 2011) in comparison with those for igneous zircon from localities worldwide (Valley et al. 2005).

Table 1.
Chemical composition of zircon M127 (EPMA, University of Göttingen; LA-ICP-MS, University of Bergen)

<table>
<thead>
<tr>
<th>Oxide</th>
<th>Mass fraction [% m/m]**</th>
<th>Element</th>
<th>Isotope measured</th>
<th>Mass fraction [µg g$^{-1}$]**</th>
<th>Element</th>
<th>Isotope measured</th>
<th>Mass fraction [µg g$^{-1}$]**</th>
</tr>
</thead>
<tbody>
<tr>
<td>SiO$_2$</td>
<td>32.6 ± 0.2</td>
<td>Y</td>
<td>89</td>
<td>785 ± 55</td>
<td>Dy</td>
<td>163</td>
<td>67.1 ± 2.0</td>
</tr>
<tr>
<td>P$_2$O$_5$</td>
<td>0.057 ± 0.006</td>
<td>Nb</td>
<td>93</td>
<td>1.61 ± 0.45</td>
<td>Ho</td>
<td>165</td>
<td>25.7 ± 1.6</td>
</tr>
<tr>
<td>Y$_2$O$_3$</td>
<td>0.109 ± 0.012</td>
<td>La</td>
<td>139</td>
<td>0.35 ± 0.17</td>
<td>Er</td>
<td>166</td>
<td>121 ± 9</td>
</tr>
<tr>
<td>ZrO$_2$</td>
<td>65.9 ± 0.3</td>
<td>Ce</td>
<td>140</td>
<td>16.9 ± 0.8</td>
<td>Tm</td>
<td>169</td>
<td>28.0 ± 2.1</td>
</tr>
<tr>
<td>Yb$_2$O$_3$</td>
<td>0.028 ± 0.012</td>
<td>Pr</td>
<td>141</td>
<td>0.18 ± 0.07</td>
<td>Yb</td>
<td>172</td>
<td>286 ± 19</td>
</tr>
<tr>
<td>HfO$_2$</td>
<td>1.50 ± 0.02</td>
<td>Nd</td>
<td>146</td>
<td>1.95 ± 0.56</td>
<td>Lu</td>
<td>175</td>
<td>53.7 ± 1.8</td>
</tr>
<tr>
<td>ThO$_2$</td>
<td>0.046 ± 0.006</td>
<td>Sm</td>
<td>147</td>
<td>3.95 ± 0.65</td>
<td>Hf</td>
<td>178</td>
<td>12400 ± 500</td>
</tr>
<tr>
<td>UO$_2$</td>
<td>0.099 ± 0.008</td>
<td>Eu</td>
<td>153</td>
<td>0.39 ± 0.13</td>
<td>Pb</td>
<td>206</td>
<td>78.4 ± 4.2</td>
</tr>
<tr>
<td>Total</td>
<td>100.4 ± 0.4</td>
<td>Gd</td>
<td>157</td>
<td>15.6 ± 1.5</td>
<td>Th</td>
<td>232</td>
<td>413 ± 17</td>
</tr>
</tbody>
</table>

* Al$_2$O$_3$, CaO and FeO were not detected or averages were below 0.005% m/m.
** Quoted uncertainties are 2s.
*** The U Mass fraction was determined by solution ICP-MS at ~ 923 µg g$^{-1}$ (Nasdala et al. 2008), which is suggested to be used as the “working value.”

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<table>
<thead>
<tr>
<th>Analysis number</th>
<th>Pb (µg g⁻¹)</th>
<th>U (µg g⁻¹)</th>
<th>Th /U</th>
<th>Total common Pb [pg]</th>
<th>207Pb/235U</th>
<th>2s (abs.)</th>
<th>207Pb/235U age [Ma]</th>
<th>206Pb/238U</th>
<th>2s (abs.)</th>
<th>206Pb/238U age [Ma]</th>
<th>ρ (err. corr.)</th>
<th>207Pb/206Pb</th>
<th>2s (abs.)</th>
<th>Discordance [%]</th>
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<td>153/5 3A</td>
<td>72</td>
<td>817</td>
<td>0.4</td>
<td>7.5</td>
<td>0.677</td>
<td>0.002</td>
<td>524.88 ± 1.22</td>
<td>0.0848</td>
<td>0.00</td>
<td>525.04 ± 1.38</td>
<td>0.98</td>
<td>0.0579</td>
<td>0.00</td>
<td>-0.2</td>
</tr>
<tr>
<td>153/5 3B</td>
<td>72</td>
<td>816</td>
<td>0.4</td>
<td>8.6</td>
<td>0.676</td>
<td>0.002</td>
<td>524.44 ± 1.64</td>
<td>0.0847</td>
<td>0.00</td>
<td>524.70 ± 1.86</td>
<td>0.98</td>
<td>0.0578</td>
<td>0.00</td>
<td>-0.3</td>
</tr>
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<td>153/5 3C</td>
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<td>816</td>
<td>0.4</td>
<td>9.8</td>
<td>0.678</td>
<td>0.002</td>
<td>525.34 ± 1.66</td>
<td>0.0849</td>
<td>0.00</td>
<td>525.64 ± 1.88</td>
<td>0.97</td>
<td>0.0579</td>
<td>0.00</td>
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<td>817</td>
<td>0.4</td>
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<td>0.677</td>
<td>0.002</td>
<td>525.02 ± 1.71</td>
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<td>0.00</td>
<td>525.16 ± 1.90</td>
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<td>0.0579</td>
<td>0.00</td>
<td>-0.2</td>
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</table>

ID-TIMS means:
- Pb: 0.677 ± 69
- U: 0.0848 ± 71
- 207Pb/206Pb: 0.0579 ± 12

One 646 µg chip was dissolved (split after dissolution).

Table 2.
Results of ID-TIMS U-Pb analyses of M127 (University of Oslo)

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Table 3.
Results of Hf-isotope analyses by solution ICP-MS

<table>
<thead>
<tr>
<th>Laboratory</th>
<th>Analysis number</th>
<th>Lu [µg g⁻¹]</th>
<th>Hf [µg g⁻¹]</th>
<th>176Lu/177Hf</th>
<th>176Hf/177Hf *</th>
<th>2s (abs.)</th>
<th>176Hf/177Hf **</th>
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<tr>
<td>Academy of Sciences, Beijing</td>
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<td>M127 #5B re-analysis</td>
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<td>M127 #6Y</td>
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<td>0.282388</td>
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</table>

Mean observed 176Hf/177Hf ratio of six*** analyses: 0.282392 ± 0.000006 (2s)

| Washington State University, Pullman | 11-chip 3     | 64.5        | 12616       | 0.00073     | 0.282397     | 0.000006  | 0.282390      |
|                                      | 11-chip 3 re-analysis |          |             |             | 0.282402     |           |               |
|                                      | Chip 12        | 103.1       | 12590       | 0.00116     | 0.282401     | 0.000009  | 0.282389      |
|                                      | Chip 12 re-analysis |          |             |             | 0.282395     |           |               |
|                                      | Chip 13        | 48.8        | 12529       | 0.00055     | 0.282394     | 0.000008  | 0.282389      |
|                                      | 14-chip 1      | 78.7        | 12578       | 0.00089     | 0.282395     | 0.000008  | 0.282386      |
|                                      | 14-chip 2      | 68.7        | 12604       | 0.00077     | 0.282388     | 0.000006  | 0.282380      |
|                                      | 15-chip 1      | 69.6        | 12539       | 0.00079     | 0.282385     | 0.000007  | 0.282377      |
|                                      | 15-chip 2      | 11.5        | 12546       | 0.00013     | 0.282394     | 0.000005  | 0.282393      |

Mean observed 176Hf/177Hf ratio of seven*** analyses: 0.282393 ± 0.000010 (2s)

| Universität zu Köln | M127-1 (# 6)   | 55.8        | 12970       | 0.0006102   | 0.282406     | 0.000004  | 0.282400      |
|                     | M127-2 (# 7)   | 56.8        | 12956       | 0.0006223   | 0.282407     | 0.000005  | 0.282401      |
|                     | M127-3 (# 8)   | 56.1        | 12847       | 0.0006198   | 0.282403     | 0.000004  | 0.282397      |
|                     | M127-4 (# 9)   | 54.9        | 12840       | 0.0006066   | 0.282401     | 0.000004  | 0.282395      |
|                     | M127-5 (# 10)  | 54.7        | 12631       | 0.0006142   | 0.282407     | 0.000004  | 0.282401      |

Mean observed 176Hf/177Hf ratio of five analyses: 0.282405 ± 0.000005 (2s)

Results of replicate analyses are quoted but were not considered in the calculation of means.

Note that additional results of Hf isotope determinations, done by LA-ICP-MS at the Chinese Academy of Sciences, Beijing, are listed in Table S1.

* M127 data are corrected to analyses of JMC 475.
** Age-corrected 176Hf/177Hf ratio (524 Ma).
*** In the calculation of means, results of re-analyses were not considered.
Table 4.
Oxygen isotope determinations by laser fluorination (University of Wisconsin at Madison)

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<tr>
<th>Analysis No.</th>
<th>Sample/reference name</th>
<th>Material analysed</th>
<th>Mass [mg]</th>
<th>δ^{18}O Raw [‰ VSMOW]*</th>
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<td>Garnet reference</td>
<td>2.31</td>
<td>5.76</td>
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(Table 4 continued)

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<tr>
<th>Analysis No.</th>
<th>Sample/reference name</th>
<th>Material analysed</th>
<th>Mass [mg]</th>
<th>δ¹⁸O [‰ VSMOW]*</th>
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<td>M127 Zircon (HF-treated)</td>
<td>1.83</td>
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Summary:

Zircon M127 (HF-treated; seven individual analyses): Average δ¹⁸O = 8.49 ± 0.05 (2s)
Zircon M127 (non HF-treated; nine individual analyses): Average δ¹⁸O = 8.26 ± 0.06 (2s)

Note that additional results of O isotope determinations, done by SIMS at the Chinese Academy of Sciences, Beijing, and the Helmholtz-Zentrum Potsdam (GFZ), are listed in Table S2.

* M127 data are corrected to the respective UWG-2 reference analyses.

Table 5.

(U+Th)/He age of M127 (University of Arizona at Tucson)

<table>
<thead>
<tr>
<th>Sample name</th>
<th>⁴He [pmol]</th>
<th>U [ng]</th>
<th>Th [ng]</th>
<th>Age (Ma)</th>
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</thead>
<tbody>
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<td>M127zA</td>
<td>11.6 ± 0.2</td>
<td>4.41 ± 0.16</td>
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<td>M127zB</td>
<td>24.2 ± 0.6</td>
<td>9.24 ± 0.34</td>
<td>4.26 ± 0.14</td>
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<td>15B346</td>
<td>0.809 ± 0.010</td>
<td>0.303 ± 0.008</td>
<td>0.133 ± 0.004</td>
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<td>2.05 ± 0.06</td>
<td>0.772 ± 0.024</td>
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<td>15B348</td>
<td>0.227 ± 0.006</td>
<td>0.088 ± 0.002</td>
<td>0.041 ± 0.002</td>
<td>417 ± 18</td>
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</table>

Weighted mean age of five analyses: 426 ± 7 Ma (2s)

All analytical uncertainties are 2s.

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