

24 during production was responsible for emission levels higher than those from petrol. After
25 wind-generated electricity was incorporated, the emissions were significantly reduced below
26 the levels of those from petrol under both SMR and AE scenarios. However, with the
27 incorporation of wind-generated electricity, the production of hydrogen, particularly from
28 electrolysis, is more environmentally friendly than the SMR process.

29

30 *Keywords:*

31 Hydrogen fuel

32 Life cycle assessment

33 Environmental feasibility

34

35 1. Introduction

36

37 There is a growing necessity for an alternate energy carrier to replace the ever decreasing,
38 and high emissions generating, supplies of fossil fuels. This is particularly notable in the
39 transport sector, where the overwhelming majority of vehicles operate on petroleum products
40 [1]. Considering the enormous environmental, and economic impact of the transport industry,
41 the introduction of alternative fuels will be key to a sustainable transport sector [2].

42 With petrol as the most common vehicle fuel, the Western Australian transport sector
43 generates approximately 14% of the state's total greenhouse gas (GHG) emissions. This is
44 attributable to the heavy reliance on passenger vehicles for most West Australians, coupled
45 with the sparsely populated landscape and large distances between population centers [3]. In
46 2007, approximately 78.9% of the total vehicle fleet was registered as using unleaded petrol
47 and 85.9% of these vehicles were classed as passenger vehicles [4]. With ownership of
48 private vehicles in Australia on the increase (up 13.1% from 2004 to 2009) [4], transportation
49 is a major factor in the ever increasing demand for fossil fuels [5], in turn having a significant
50 effect on the Western Australian environment [6].

51 With the overwhelming majority of Western Australia's vehicles operating on petrol,
52 environmentally damaging emissions are constantly being introduced into the atmosphere,
53 resulting in the per capita GHG emissions for Western Australia being significantly higher
54 than for other Australia states [3]. These passenger vehicles are also the primary emitters of
55 nitrogen oxides (NO_x) and volatile organic compounds (VOCs) causing photochemical smog
56 and negative health impacts [7].

57 Considering the growing atmospheric pollution and the current energy crisis, studies
58 have been conducted in Australia that assess the environmental feasibility of alternative
59 transport fuels such as liquefied petroleum gas (LPG), liquefied natural gas (LNG), bio-diesel

60 and ethanol. While the use of these fuels reduces GHG emissions, they can have other
61 environmental impacts during the combustion stage. For example, ethanol was a potentially
62 renewable fuel with reduced carbon monoxide (CO) emissions compared to petrol, but the
63 NO_x emissions resulting from combustion were significantly higher than those from
64 petroleum products [8].

65 Alternative fuels may produce relatively less GHGs than conventional fuel during
66 combustion, but more emissions are produced during the production process. For example, a
67 study in 2011 by Biswas *et al.* [9] found that biodiesel production and combustion from
68 canola is not “carbon neutral”, as GHGs are emitted from production of farm inputs and
69 during crop growth. Similarly, LNG has been considered one of the safest and cleanest fossil
70 fuels [10, 11, 12, 13, 14, 15] in comparison with other fossil fuels such as coal and oil in
71 terms of NO_x, sulphur dioxide (SO₂) and carbon dioxide (CO₂) emissions, but the production
72 and liquefaction of LNG is energy intensive and not free of environmental impacts.
73 Therefore, a life cycle assessment (LCA) that takes into account emissions from all stages
74 needs to be conducted to assess the environmental impacts of alternative fuels and to identify
75 the most polluting processes for applying mitigation strategies.

76 Many alternative fuels have been studied over the years; however, the fuel which appears
77 to be a more promising alternative is hydrogen due to its clean burning characteristics and
78 limitless supply. Although research into hydrogen fuel is limited in Australia, a 2003
79 Australian study identified a number of hydrogen feed stocks suitable for mass production in
80 Australia. These feed stocks included coal, fuel oils, industrial chemical by-products, coal,
81 coal seam methane and natural gas [16].

82 One of numerous foreign studies into hydrogen as an automotive fuel, a life cycle
83 emissions study for hydrogen fuel production found that hydrogen could potentially be
84 produced with comparatively less emissions than petrol [17]. Similarly, a 2005 Canadian

85 study [18] found that the life cycle emissions from hydrogen could also be comparable to
86 those of petrol when producing hydrogen from natural gas feed stocks. Other studies have
87 assessed the viability of hydrogen from alternative production sources and processes [19, 20,
88 21].

89 Western Australia possesses abundant fossil fuel resources, particularly coal and natural
90 gas. Black coal accounts for around 49% of total fossil fuel resources within the state, with
91 natural gas accounting for around 40% and growing as more sources are identified [22]. This
92 makes reforming of natural gas, or steam methane reforming (SMR), an attractive option for
93 Western Australia due to its availability in large reserves. While there are available resources
94 to produce environmentally friendly hydrogen fuel in Western Australia, the upstream
95 activities, such as feedstock production, processing and storage stages, can have adverse
96 environmental impacts because of the state's fossil fuel dependent electricity mix and
97 scattered settlements [8, 22].

98 This study aims to assess the life cycle environmental feasibility of using hydrogen as an
99 automotive fuel in Western Australia through two commonly used hydrogen production
100 process (SMR and electrolysis). This study utilizes the functional unit VKT (vehicle
101 kilometer travelled) in order to assess the well-to-wheel emissions of vehicles per kilometer
102 of travel, so that there is a common unit of measure between the petrol and hydrogen results.

103 Firstly, the paper discusses the methodology for carrying out the life cycle environmental
104 feasibility study of hydrogen fuel in Western Australia. Secondly, the life cycle
105 environmental impact of hydrogen fuel has been compared with that of petrol and the
106 “hotspot” – the inputs causing the most pollution – is identified. Finally, appropriate
107 mitigation strategies have been considered for reducing the life cycle environmental impacts
108 of hydrogen fuel use in passenger transport.

109

110 **2. Methodology**

111

112 LCAs model the complex interactions between a product and the environment throughout all
113 phases of the product's life. The methodology for this LCA of hydrogen as an automotive
114 fuel has followed the guidelines set out by ISO14040–14043 [23]. The LCA methodology
115 consists of four steps

116 i. goal and scope

117 ii. life cycle inventory (LCI) analysis, which provides information on the input data
118 (chemicals and energy) used to determine the life cycle emissions during each life
119 cycle phase

120 iii. impact assessment, which evaluates the environmental impacts of the emissions of
121 each life cycle phase and classifies impacts into environmental impact categories
122 (e.g. global warming)

123 iv. Interpretation, which evaluates the LCA model by identifying significant issues
124 based on the results of LCI and LCA, considering completeness and consistency
125 and making conclusions and recommendations (as presented in the results and
126 discussions section of this paper).

127

128 **2.1. Goal and scope definition**

129

130 The goal of this life cycle study is to evaluate the environmental feasibility of hydrogen as an
131 automotive fuel in Western Australia. The study also provides a reasonable comparison of the
132 life cycle environmental impacts of hydrogen compared to petrol as a vehicle fuel. For the
133 purposes of this comparative study, the functional unit used is VKT. This allows the
134 identification and comparison of life cycle impacts between hydrogen and petrol vehicles.

135 Road tests using hydrogen fuel in a Volkswagen Polo (1.4 liter engine) in 2011 gave a
136 maximum speed of 125 km/h and an estimated consumption of 1 kg of hydrogen per 100 km
137 at an average speed of 90 km/h [24]. Therefore, the average consumption of hydrogen (0.01
138 kg hydrogen/VKT) [24] was used as a functional unit. The same model vehicle with the same
139 engine size consumes 0.059 liters of regular unleaded petrol per kilometer [25]. Using the
140 density of BP unleaded petrol (730 kg/m^3 [26]), the fuel consumption by mass was found to
141 be 0.043 kg/VKT, where VKT is the functional unit for petrol. It should be noted that
142 Volkswagen Polo cars are sold in Australia [27], which justifies their use in this case study.

143 The life cycle environmental impacts of the use of 0.01 kg hydrogen have been
144 compared with 0.043 kg of petrol for driving a passenger car for 1 km.

145 This LCA study considers the well-to-wheel approach, which means that it takes into
146 account all stages from resource extraction to eventual fuel consumption.

147 Three system scenarios have been assessed within this LCA. The first is the LCA of
148 hydrogen as an automotive fuel when the hydrogen is produced by SMR. The second
149 scenario will assess the LCA of hydrogen when the hydrogen is sourced from alkaline
150 electrolysis (AE). Finally, the third scenario is the LCA of petrol for comparison.

151 The determination of impacts associated with the modification of the existing
152 Volkswagen engine into a petrol-H₂ engine was beyond the scope of this research.

153

154 **2.2. Life cycle inventory analysis**

155

156 LCI is the collection of data that describes the inputs required for each stage of the well-to-
157 wheel life cycle. The purpose of these inventories is to provide the basis for an assessment of
158 the environmental impacts of running a vehicle on hydrogen compared to running a vehicle

159 on conventional petrol. Figure 1 presents the life cycle pathways for SMR and AE to produce
160 the same amount of hydrogen required to drive a passenger vehicle for 1 km.

161 *2.2.1. Steam methane reforming scenario*

162

163 The SMR scenario includes seven life cycle stages of well-to-wheel (or production to
164 combustion), which are as follows:

- 165 1 Natural gas extraction and distribution: this phase takes into account the energy
166 and resources required to extract and distribute the gas.
- 167 2 SMR: this phase takes into account the natural gas, steam and electricity required
168 for the process. The SMR process is assumed to occur at 20 bar.
- 169 3 Compression of the hydrogen into large transport trailers: SMR produces
170 hydrogen gas at pressures of around 20 bar; however, large-scale CP-12 hydrogen
171 delivery trucks have 12 storage tubes which operate at 165 bar [28]. Therefore a
172 compressor is used to increase the pressure of the hydrogen to 165 bar for travel
173 and delivery.
- 174 4 The distribution of hydrogen gas by tanker truck: the CP-12 hydrogen delivery
175 trailers weigh 42.5 tons and are typically pulled by large diesel trucks. The mean
176 delivery distance was also calculated based on Western Australia. BP locations
177 and the average distance were found to be 233 km. This phase takes into account
178 delivery distance and diesel consumption by a tanker truck.
- 179 5 The compression of the hydrogen into medium-term storage tanks at the fuelling
180 station: mid-term storage tanks at fuelling stations contain hydrogen at 300 bar to
181 allow for faster refueling of vehicle tanks [29]. This means that the hydrogen
182 must again be compressed from 165 bar in the delivery tanker tubes to 300 bar
183 using an electrical compressor. The energy required to pump petrol into a fuel

184 tank was not considered as it is negligible when compared with the energy
185 required to compress hydrogen into a vehicle tank.

186 6 The compression of the hydrogen into smaller vehicle fuel tanks: from the 300
187 bar storage cylinders at the fuelling station, the hydrogen gas needs to be
188 compressed to 350 bar inside the vehicle fuel tank [30, 31]. Again, this process is
189 performed by an electrical compressor.

190 7 Hydrogen used by vehicle: the emissions associated with hydrogen combustion
191 have been sourced from Wallnera *et al.* [32].

192 Table 1 details the inputs and quantities required for production, delivery and
193 combustion of 0.01 kg of hydrogen gas produced through SMR.

194

195 2.2.2: Alkaline electrolysis scenario

196

197 LCA for the AE scenario includes five life cycle stages of well-to-wheel analysis, which are
198 as follows:

199 1 Electrolysis process: this phase takes into account the water, electricity and
200 electrolytes used during the electrolytic process (Table 2). The process used as a basis
201 for this research operated at 8.14 bar [31].

202 2 Compression of the hydrogen into large transport trailers: compression into the
203 transport trailer requires more energy when the hydrogen is produced by AE as the
204 hydrogen gas is produced at a lower pressure than during SMR. This phase takes into
205 account the electricity required to compress the hydrogen from 8.14 bar to 165 bar for
206 transport.

207 3 The distribution of hydrogen gas by tanker truck: the distribution method is identical
208 to when hydrogen is produced by SMR.

209 4 The compression of the hydrogen into medium-term storage tanks at the fuelling
210 station: as with the SMR scenario, the electricity required to compress the hydrogen
211 from 300 bar to 350 bar is taken into account in this phase.

212 5 Hydrogen use by vehicle: this is same as for SMR.

213 A separate inventory for petrol has not been developed as the software used has the
214 emission values of petrol production and use.

215

216 **2.3 Life cycle impact assessment**

217

218 The environmental impacts associated with the production and use (combustion) of hydrogen
219 includes two steps. Firstly, the energy and material flow data provided in the LCI were input
220 to *Simapro 7.24* software [33] to calculate the environmental impacts of the production and
221 use of hydrogen fuel. Secondly, the program categorized the emissions for all impact
222 categories and then converted them to equivalent environmental impacts, including global
223 warming, photochemical oxidation, eutrophication, carcinogens, land use, water use, solid
224 waste, embodied energy and mineral depletion impacts.

225 Step 1: The input and output data in the LCI were input to the *Simapro* software to
226 calculate the emissions for different environmental impact categories due to the use of
227 hydrogen and petrol per VKT. The input/output data of the LCI were linked to relevant
228 libraries in *Simapro*. The LCA Library is a database of energy consumption, emissions and
229 materials data for the production of one unit of an input (e.g. electricity, diesel).

230 This study utilized the Australian LCA libraries [34] developed by RMIT University for
231 Australian conditions to calculate the emissions associated with the production and use of
232 inputs. The library for the Western Australian electricity generation mix was used to calculate

233 the environmental impacts associated with the use of electricity for hydrogen production,
234 storage and compression [34].

235 Step 2: *Simapro* software calculated the environmental impacts once the inputs and
236 outputs were linked to the relevant libraries. The program sorted the relevant emissions for
237 particular impacts, and then converted them to an equivalent amount of environmental
238 impacts. The Australian Environmental Impact calculation method, developed locally [34],
239 was used to assess the environmental impacts of the use of hydrogen and petrol for VKT.

240

241 **3. Results and discussions**

242

243 **3.1. Comparison of environmental performance of hydrogen with petrol**

244

245 The comparative environmental performance of three scenarios has been carried out. The first
246 scenario is the life cycle of hydrogen when the hydrogen is produced by SMR. The second
247 scenario is for hydrogen produced by AE. The last scenario is the life cycle of petrol.

248 Contributions to global warming, photochemical smog and eutrophication have been
249 found to be the predominant environmental impacts in these three scenarios (Figure 2). While
250 hydrogen is a cleaner burning fuel than petrol, the AE scenario produces more life cycle
251 global warming and eutrophication impacts than the latter in the petrol scenario. This is
252 mainly due to the emissions of CO₂ (causing global warming) and NO_x (nitrogen oxides
253 causing eutrophication) from electricity and diesel consumption during upstream activities
254 (alkaline electrolysis, compression for distribution and storage, and transportation) being
255 higher than those for petrol.

256 The life cycle global warming impacts due to the use of hydrogen produced in the AE
257 scenario are 2.3 times greater than those of petrol. Walwijk *et al.* [35] also found that CO₂-e

258 emissions from electrolytic hydrogen production and use would be higher (approximately 1.6
259 times) than those from petrol. There are similar results in terms of emissions for
260 eutrophication. Figure 2 indicates that PO_4^- -e eutrophication emissions from the AE scenario
261 are significantly greater than for petrol. However, in terms of photochemical oxidation
262 emissions, the results are quite different. Both the hydrogen scenarios produce less SO_x and
263 NO_x (C_2H_2 -e emissions) throughout the life cycle from a photochemical perspective.

264 The SMR scenario produces slightly lower environmental impacts than the petrol scenario.
265 About 4%, 91% and 23% of the global warming, photochemical smog and eutrophication
266 impacts, respectively, can be avoided due to the replacement of petrol with hydrogen fuel
267 produced under the SMR scenario. In addition, hydrogen production from the SMR scenario
268 is less harmful to the environment than the from AE scenario in its global warming,
269 photochemical smog and eutrophication impacts, because electricity consumption in the AE
270 scenario is about 6.7 times higher than that in the SMR process (Tables 1 and 2).

271 The life cycle emissions from the AE scenario were found to be significantly higher than
272 for the SMR scenario across every environmental impact category. This is likely attributable
273 to the large quantities of coal (37%) and natural gas (60%) in the Western Australian energy
274 mix required to produce the electricity for electrolysis; however, this will be examined in
275 more detail in the following section. Further investigation has been carried out to determine
276 the inputs or processes causing the most environmental impacts (hotspots) so that the
277 appropriate mitigation strategies can be considered for making hydrogen fuel
278 environmentally competitive with petrol.

279

280

281 *3.2 Breakdown of environmental impacts of the use of hydrogen produced by steam*
282 *methane reforming*

283

284 In order to find the hotspots, the percentage distribution of global warming, photochemical,
285 and eutrophication impacts in terms of inputs for the SMR and AE scenarios have been
286 determined (Table 3).

287 Greenhouse Gas Emissions: The majority (88.64%) of GHGs are generated by SMR, the
288 generation of electricity and the production of steam. The SMR process itself produces the
289 largest amount of CO₂-e (44.9% of the total emissions).

290 The generation of electricity, in particular from coal and natural gas, produces the second
291 largest amount of CO₂-e. This life cycle phase accounts for 29.6% of the total emissions due
292 to the heavy reliance on fossil fuels as the primary source of fuel for generating electricity.
293 The production of steam is also a carbon intensive process, accounting for 15.5% of the total
294 emissions.

295 The results of the SMR model in a 2007 Canadian study (0.3602 kg CO₂-e per VKT) are
296 similar to those in the current study (0.252 kg CO₂-e per VKT) [36]. The difference in
297 emission output is likely attributable to the technical efficiency improvement during this
298 period. The average hydrogen fuel consumption during 2006–12 was 0.0227 kg/VKT, while
299 the present study considered the latest consumption figure in 2011 (0.01 kg/VKT). The
300 emissions breakdown clearly indicates that for GHG emissions to be reduced, improvements
301 need to be made to the aforementioned CO₂-e intensive life cycle phases.

302 Photochemical Smog Emissions: The major life cycle phases contributing to
303 photochemical emissions are also the production of steam, the steam reforming operation and
304 electricity generation. Together, these three life cycle phases represent 63% of the total C₂H₂-
305 e emissions due to significant levels of NO_x and VOCs released into the atmosphere. The
306 second largest contribution is from tailpipe emissions (30%), mainly NO_x.

307 Eutrophication Emissions: Eutrophication emissions are produced primarily from the
308 production of steam, the production of electricity and from the steam reforming process. In
309 total, these processes account for 86.83% of the total of eutrophication emissions. Producing
310 the steam required for reforming emits 0.016 g of PO₄^{-e} per VKT while the generation of
311 electricity for the steam reforming and compression processes produces 0.0385 g of PO₄^{-e}
312 per VKT.

313

314 ***3.3 Breakdown of environmental impacts of the use of hydrogen produced by alkaline*** 315 ***electrolysis***

316

317 Table 3 also shows the breakdown of global warming, photochemical, and eutrophication
318 impacts that would result from the production and use of hydrogen fuel generated by AE.
319 Greenhouse Gas Emissions: The overwhelming majority of life cycle GHGs emitted during
320 the alkaline electrolysis scenario are attributable to the generation of electricity. Table 1
321 shows that 93.1% of the total GHG emissions are generated from the electricity supply, of
322 which 78.3% of the CO₂-e comes from electricity generation from coal and 14.8% comes
323 from electricity generation from natural gas. AE is very energy intensive, requiring 62.7 kWh
324 per kilogram of hydrogen production which equates to 0.63 kWh per VKT. Although AE
325 itself is virtually emission free, generating the required electricity is currently very carbon
326 intensive.

327 Photochemical Oxidation Emissions: Table 3 clearly shows that electricity generation
328 from coal and gas accounts for 73.4% of total C₂H₂-e emissions; however, vehicle tailpipe
329 emissions are also significant. Tailpipe emissions account for 16.5% of the total C₂H₂-e
330 emissions and this is attributable to the combustion of hydrogen within the vehicle engine.
331 NO_x, as well as fugitive hydrocarbon emissions, are also emitted during electricity generation
332 and contribute to the development of photochemical smog.

333 Eutrophication Emissions: The majority of the emissions (about 93.9%) causing
334 eutrophication are generated during the production of electricity from coal and natural gas,
335 with these processes contributing 84.6% and 9.3% respectively. The first compression stage
336 of hydrogen gas is somewhat significant with a 2.5% contribution. Producing the electricity
337 required for electrolysis emits 0.3 g of PO₄^{-e} per VKT while the compression processes
338 produces 0.008 g of PO₄^{-e} per VKT.

339

340 ***3.4 Mitigation and reduction of emissions using wind***

341

342 The previous sections identified electricity generation as a major source of global warming,
343 photochemical oxidation and eutrophication emissions for both the SMR and AE scenarios. It
344 is clear from the breakdowns of the life cycle emissions that reducing the carbon intensity of
345 electricity production would have the greatest environmental benefit and would significantly
346 reduce total emissions in each impact category.

347 The implementation of wind-generated electricity for hydrogen production has the
348 potential to substantially reduce the emissions across all impact categories in every life cycle
349 phase, excluding for the vehicle use phase as the only input is hydrogen gas.

350 Wind power is a promising technology in Australia with a potential to generate
351 renewable and virtually emissions-free electricity. As of 2009, Western Australia's wind

352 energy capacity was 202.7 MW which represents a significant investment [37] and currently
353 Western Australia has 42 operating wind farms [38,39].

354 Wind technology is poised to be a potential solution to reducing emissions during
355 hydrogen production by greatly reducing reliance on coal and gas. The potential benefits are
356 greatest for the AE scenario as the only life cycle phase which relies directly on fossil fuels is
357 the transportation of hydrogen by diesel truck.

358 The emissions from the SMR scenario will also benefit from lower emission levels;
359 however, there is still a reliance on fossil fuels, particularly natural gas, during the extraction
360 and steam reforming processes. This means that although emissions from electricity
361 production will be reduced, there is still potential for significant environmental impacts
362 resulting from the use of fossil fuels.

363 The efficacy of wind electricity needs to be assessed for both the SMR and AE scenarios
364 before any conclusions can be made regarding the net environmental effects. Figure 3 shows
365 that the environmental impacts can be significantly reduced due to the use of wind energy in
366 the production, delivery and storage of hydrogen fuel. This is because the substitution of coal
367 and natural gas powered electricity with wind generated electricity for production and storage
368 purposes have significantly reduced the emissions of CO₂, NO_x and O₃, which cause global
369 warming, eutrophication and photochemical smog impacts, respectively. About 31%, 19%
370 and 35% of the total global warming, photochemical smog and eutrophication impacts can be
371 reduced by using wind electricity in the SMR scenario. In the AE scenario, global warming
372 and eutrophication impacts have been almost completely eliminated (by 99%) with the use of
373 wind energy in the life cycle of hydrogen fuel.

374 The replacement of grid electricity with wind electricity could make hydrogen fuel
375 environmentally competitive with petrol from the global warming, photochemical smog and
376 eutrophication impacts perspectives. Although the SMR scenario using grid electricity (coal

377 and natural gas mix) produced less environmental impacts than petrol, a further reduction in
378 environmental impacts is possible when grid electricity is replaced with wind-generated
379 electricity. About 37%, 91% and 64% of the total global warming, photochemical smog and
380 eutrophication impacts can be reduced by replacing petrol with hydrogen fuel under the SMR
381 scenario with wind-generated electricity. The AE scenario has significant potential to reduce
382 global warming (97%), photochemical smog (96%) and eutrophication (98%) impacts due to
383 replacement of petrol with hydrogen fuel.

384 Therefore, the use of wind-generated electricity in the hydrogen fuel cycle not only
385 reduces overall environmental impacts in hydrogen fuel production but also makes the
386 hydrogen fuel environmentally friendlier than petrol. When grid electricity was used for
387 hydrogen production, the SMR scenario appeared to be more environmentally friendly than
388 the AE scenario. Interestingly, if wind is only source of electricity used in hydrogen
389 production, then the AE scenario becomes much more environmentally friendly than the
390 SMR scenario.

391

392 **4 Conclusions and recommendations**

393

394 LCA has been demonstrated as an effective tool for modeling and quantifying the
395 environmental impacts from the use of hydrogen as an automotive fuel. Global warming,
396 photochemical smog and eutrophication have been found to be the predominant
397 environmental impacts associated with the use of hydrogen fuel produced from both SMR
398 and AE. The initial results of the models found that the SMR scenario emitted 0.252 kg of
399 CO₂-e, 0.000079 kg of C₂H₂-e and 0.00012 kg of PO₄⁻-e per VKT. The AE scenario was
400 found to emit 0.67 kg CO₂-e, 0.000139 kg of C₂H₂-e and 0.000322 kg of PO₄⁻-e per VKT.

401 In order to determine the feasibility of hydrogen as an automotive fuel, the life cycle
402 impacts were compared to those of petrol. When grid electricity is used in the hydrogen fuel
403 life cycle, the use of hydrogen fuel was found to be environmentally friendlier than petrol
404 from global warming, photochemical oxidation and eutrophication perspectives under the
405 SMR scenario. Except for the photochemical smog impact, the AE scenario produces higher
406 global warming and eutrophication impacts than petrol. The global warming and
407 eutrophication impacts associated with the production and use of petrol have been found to be
408 2.3 and 1.8 times lower than hydrogen fuel produced from the AE scenario, respectively. For
409 both the SMR and AE scenarios, electricity was a major source of emissions; however, the
410 AE model required nearly seven times the electricity of the SMR model, hence the greater
411 environmental impacts. Natural gas was also a major source of emissions, particularly in the
412 SMR model, as it was required in large quantities during the SMR process.

413 In order to mitigate the environmental impacts further, the LCAs were reworked so as to
414 incorporate electricity from wind turbines to reduce the reliance on coal and gas. The results
415 from the wind hydrogen models revealed significant improvements in all impact categories
416 and emissions reduction below the levels of petrol.

417 However, the situation is different when electricity generated by wind is incorporated
418 into the LCA analysis. The incorporation of wind-generated electricity into the SMR model
419 reduced the global warming impact ($\text{CO}_2\text{-e}$), photochemical smog ($\text{C}_2\text{H}_2\text{-e}$) and
420 eutrophication ($\text{PO}_4\text{-e}$) emissions by 31%, 19% and 35.0% respectively. More impressively,
421 the $\text{CO}_2\text{-e}$, $\text{C}_2\text{H}_2\text{-e}$ and $\text{PO}_4\text{-e}$ emissions from the AE model were reduced by 99%, 84% and
422 99% respectively. Also, hydrogen production can be environmentally feasible compared to
423 petrol under the AE and SMR scenarios when the electricity is generated by wind.

424 The results of this study could be improved by widening the scope to include
425 consideration of economic factors. The study has indicated that, from an environmental

426 perspective, both hydrogen models can be made feasible by incorporating wind-generated
427 electricity. However, the capital costs of wind-generated electricity have not been considered,
428 nor the prices of grid electricity. For instance, a preliminary review of capital costs found that
429 South West Interconnected System (SWIS) connected wind farms commissioned in Western
430 Australia after 2000 cost, on average, \$2.22 million/MW of output [40]. The cost of natural
431 gas and water could also be incorporated into the models to provide an improved
432 environmental-economic analysis, particularly for the SMR model.

433 This study also assumed that for the wind scenario, the electricity needed for
434 compressing the hydrogen gas was sourced from wind generation. Given that the models
435 employed centralized hydrogen production, where hydrogen gas was transported from a
436 production facility to fuelling stations within the metropolitan area, it is inaccurate to assume
437 that the electricity used at the fuelling station would be sourced from wind turbines. A more
438 accurate emissions model could be developed if the electricity required for compressing the
439 hydrogen was sourced from SWIS.

440 The study could also include alternative hydrogen storage systems, such as cryogenic
441 liquid hydrogen tanks or hydride systems, as opposed to compressed hydrogen tanks, which
442 may require less energy during refueling.

443

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446

447

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537

538 Table 2- Life cycle inventory for 0.01 kg of hydrogen required for 1 km of travel using AE

539 Table 3 – Breakdown of three major impacts in terms of inputs for two hydrogen production

540 processes

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542

543 **Table 1 Life cycle inventory for 0.01 kg of hydrogen required for 1 km of travel using**544 **SMR**

Inputs	Amount	Unit	Reference
Extraction of natural gas			
Electricity	2.22E-02	kWh	[35]
SMR of natural gas			
Electricity	6.56E-02	kWh	[17]
Natural gas	3.92E-02	kg	
Steam	1.88E-01	kg	
Compression for distribution			
Electricity	2.27E-02	kWh	[39]
Distribution to fuelling station			
Diesel Fuel	0.20	L	[41]
Compression for storage at fuelling station			
Electricity	2.23E-03	kWh	[39]

Compression for storage on board vehicle

Electricity	5.37E-04	kWh	[39]
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Vehicle usage

NO _x emissions	2.20E-05	kg	[30, 32]
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CO ₂ emission	8.19E-04	kg	
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549 **Table 2 Life cycle inventory for 0.01 kg of hydrogen required for 1 km of travel using**550 **AE**

Input	Amount	Unit	Reference
AE process			
Electricity	0.63	kWh	[31]
Potassium hydroxide (KOH)	7.05E-05	kg	
Water	0.11	kg	
Compression for distribution			
Electricity			
	4.00E-02	kWh	[39]
Distribution to fuelling station			
Diesel fuel			
	0.20	L	[41]
Compression for storage at fuelling station			
Electricity			
	2.23E-03	kWh	[39]

Compression for storage on board vehicle

Electricity

5.37E-04

kWh

[39]

Vehicle usage

Hydrogen

1.00E-2

kg

[30, 32]

551 Table 3 Breakdown of three major impacts in terms of inputs for two hydrogen production processes
 552

	Global warming impact		Photochemical smog		Eutrophication	
	kg CO ₂ -e/VKT	%	kg C ₂ H ₂ -e/VKT	%	kg PO ₄ -e/VKT	%
SMR process						
Electricity supply	7.47E-02	29.6%	1.27E-05	17.3%	3.85E-05	32.0%
Steam reforming operation	1.13E-01	44.9%	1.17E-05	16.0%	1.68E-05	14.0%
Steam production from natural gas	3.92E-02	15.5%	2.2E-05	30.1%	4.91E-05	40.8%
Natural gas extraction for steam reforming	1.30E-02	5.2%	2.56E-06	3.5%	5.94E-06	4.9%
Compression of hydrogen for tanker delivery	9.52E-03	3.8%	1.72E-06	2.4%	5.10E-06	4.2%
Compression of hydrogen at the Fuelling station	2.12E-03	0.8%	3.8E-07	0.5%	1.13E-06	0.9%
Hydrogen distribution via tanker truck	2.27E-04	0.1%	6.58E-08	0.1%	1.20E-07	0.1%
Compression of hydrogen for vehicle tank	5.05E-04	0.2%	8.78E-08	0.1%	2.65E-07	0.2%
Vehicular emission	0.00E+00	0.0%	2.2E-05	30.1%	3.28E-06	2.7%

	Total	2.52E-01	100.0%	7.31E-05	100.0%	1.20E-04	100.0%
AE process							
Electricity supply		6.23E-01	93.09%	1.02E-04	73.37%	3.03E-04	93.86%
Compression of hydrogen for tanker delivery		3.24E-02	4.84%	8.32E-06	5.98%	7.96E-06	2.47%
Electrolysis of water		8.04E-03	1.20%	4.87E-06	3.50%	6.61E-06	2.05%
Compression of hydrogen for storage		4.35E-03	0.65%	7.52E-07	0.54%	1.74E-06	0.54%
Compression of hydrogen for vehicle tank		1.07E-03	0.16%	1.11E-07	0.08%	2.58E-07	0.08%
Hydrogen distribution via tanker truck		2.68E-04	0.04%	6.96E-08	0.05%	9.67E-08	0.03%
Production of KOH		1.34E-04	0.02%	2.78E-08	0.02%	3.22E-08	0.01%
Vehicular emissions		0.00E+00	0.00%	2.29E-05	16.46%	3.10E-06	0.96%
	Total	6.70E-01	100.00%	1.39E-04	100.00%	3.22E-04	100.00%

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555 **List of figures**

556

557 Figure 1 Simplified block diagram for hydrogen fuel life cycle models

558

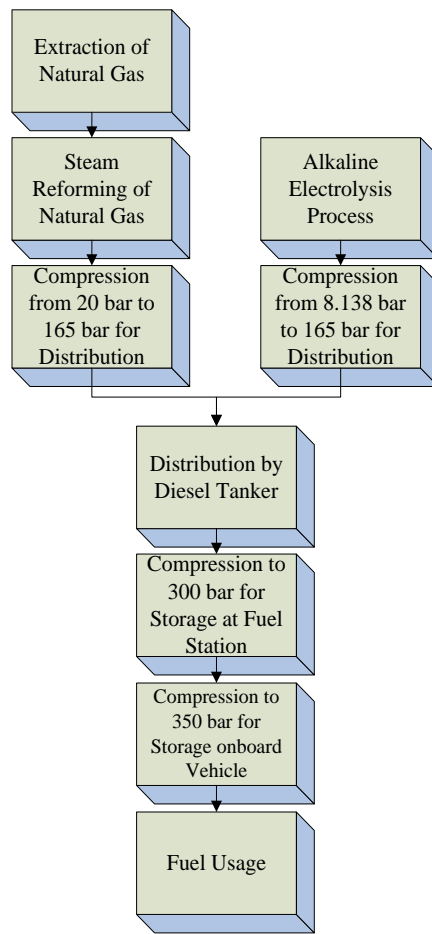
559 Figure 2 Hydrogen models compared to conventional petrol model on an environmental

560 impact basis

561 Figure 3 Implication of mitigation strategies

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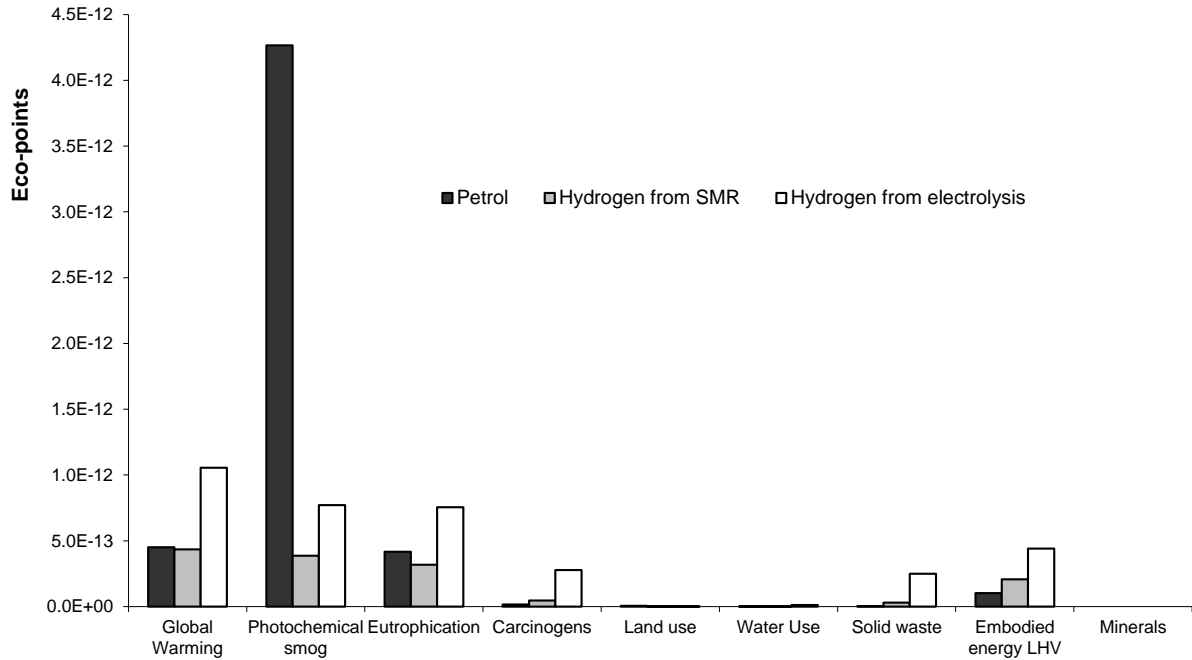
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Figure 1 Simplified block diagram for hydrogen fuel life cycle models

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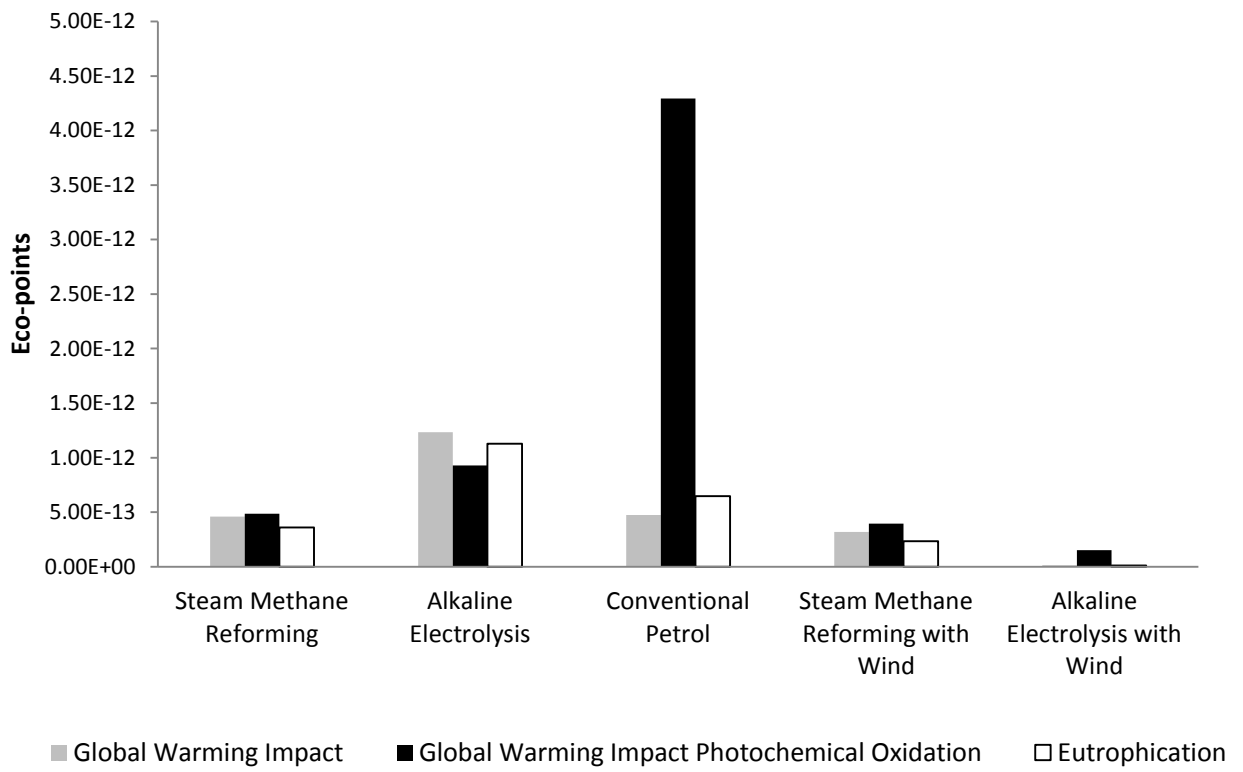
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572 Figure 2 Hydrogen models compared to conventional petrol model on an environmental
573 impact basis

574 Note: eco-points represent the relative importance of environmental impacts assigned by
575 industry and society.

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579 Figure 3 Implication of mitigation strategies

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