

Faculty of Science and Engineering

**Influence of Rock Wettability on Reservoir-Scale CO₂ Geo-
Sequestration**

Emad Abdulhusain Fakher Al-Khdheawi

**This thesis is presented for the Degree of
Doctor of Philosophy
of
Curtin University**

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Declaration of Academic Integrity

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Dedication

*I would like to dedicate my thesis to all those people who are source of my motivation!
To my mother, for your unconditional love, support, and prayers.*

To the memory of my father (Peace be upon him).

To my wife, Doaa, for your unconditional love, support and understanding, and for all the encouragement and humour that kept me going. Without your support, it would have been difficult to complete this thesis. Thank you from the bottom of my heart.

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Abstract

Concerning the underground storage of CO₂ in geological formations, CO₂ flow through porous media and CO₂ storage efficiency are significantly influenced by various parameters. Rock wettability, which is the tendency of the rock surface to be preferentially in contact with one fluid when more than one fluid is present in the same system, is one key parameter. It has been shown that wettability can vary from strongly water-wet to strongly CO₂-wet depending on the chosen reservoir. Moreover, wettability can strongly vary within a selected reservoir. Despite its importance, rock wettability has received little attention in CO₂ storage studies; specifically, rock wettability controls relative permeability, capillary pressures, residual fluid saturations, CO₂-brine interfacial areas, and CO₂ cluster morphologies.

Thus, the main focus of this research is to investigate the direct influence of rock wettability on CO₂ movement through porous media. By performing different multiphase flow reservoir simulations on a hectometre scale, this research investigates the effects of wettability and wettability spatial distribution on the efficiency of CO₂ trapping mechanisms. Further, this research includes a sensitivity analysis of the effect of factors affecting wettability on CO₂ movement and trapping capacities, including formation water salinity, reservoir temperature, and permeability and porosity spatial heterogeneity. This research also compares the efficiency of different CO₂ injection technologies, including injection well configurations (horizontal versus vertical injection wells) and CO₂ injection scenarios (continuous, intermittent injection, and water-alternating-gas injection (WAG)).

The results of this research show that reservoir wettability significantly influences CO₂ storage efficiency. CO₂-wet reservoirs have the highest vertical CO₂ plume migration while water-wet reservoirs have the best CO₂ retention. In addition, less residual CO₂ but more dissolved CO₂ is obtained in a CO₂-wet reservoir. Moreover, the results of this investigation clearly show that wettability heterogeneity significantly accelerates the vertical CO₂ migration distance, CO₂ mobility, and solubility trapping while reducing residual trapping. Further, the results of this research also show that formation water salinity, reservoir temperature, and permeability and porosity heterogeneity all affect CO₂ geo-sequestration efficiency. Lower salinity decreases CO₂ mobility and migration distance, and enhances residual and solubility trapping

significantly. Higher temperatures significantly increase vertical CO₂ migration, mobile CO₂, and dissolution trapping capacities while reducing the residual trapping capacity. Permeability and porosity heterogeneity reduces vertical CO₂ migration and induces significant horizontal migration but shows lower residual and solubility storage capacities. Moreover, this research demonstrates that a smart choice of CO₂ injection technology can significantly enhance CO₂ storage efficiency. Horizontal wells have reduced CO₂ plume migration, CO₂ mobility, and CO₂ solubility trapping and improved CO₂ residual trapping. WAG injection results in more favourable CO₂ storage outcomes when compared with continuous or intermittent injections. Specifically, residual and solubility trapping were significantly improved while vertical CO₂ migration was reduced.

This thesis concludes that water-wet rock and homogeneous wettability improve CO₂ storage capacity and containment security. Further, this work concludes that low porosity and permeability heterogeneity, low formation water salinities, low reservoir temperatures, using horizontal injection wells, and WAG injection all enhance CO₂ geo-sequestration capacity.

Publications by the Author

Journal Articles forming part of thesis as standalone chapters

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5. **Emad A. Al-Khdheawi**, Stephanie Vialle, Ahmed Barifcani, Mohammad Sarmadivaleh, Yihuai Zhang, and Stefan Iglauer. (2018). Impact of salinity on CO₂ containment security in highly heterogeneous reservoirs. *Greenhouse Gases: Science and Technology*, 8(1), 93-105. doi:10.1002/ghg.1723
6. **Emad A. Al-Khdheawi**, Stephanie Vialle, Ahmed Barifcani, Mohammad Sarmadivaleh, and Stefan Iglauer. (2018). Enhancement of CO₂ trapping efficiency in heterogeneous reservoirs by Water Alternating Gas injection. *Greenhouse Gases: Science and Technology*, 8(5), 920-931. doi:10.1002/ghg.1805

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Chapter 1 Introduction

1.1 Background

CO₂ injection into oil reservoirs is an essential technology for enhancing oil recovery (CO₂-EOR). Further, CO₂-EOR is a good option for long-term CO₂ geological sequestration, which is a significant method of mitigating anthropogenic carbon emission (Bachu, 2008). This is done by capturing CO₂ and injecting it into geological sites, such as depleted oil and gas reservoirs, unminable coal seams, and deep saline aquifers (Metz et al., 2005). Owing to their high geo-sequestration capacity and wide geographical spread, saline aquifers are the most preferable CO₂ geo-sequestration formations (Lackner, 2003). Typically, the CO₂ injection depth should be deeper than 800m, where the associated reservoir conditions (i.e., temperature and pressure) are higher than the critical temperature (31.04 °C) and the critical pressure (7.39 MPa) of CO₂. Thus, the injected CO₂ remains in a supercritical state (Pruess et al., 2003). However, the injected CO₂'s density is lower than that of formation water; therefore, injected CO₂ migrates upwards as a result of the buoyancy forces (Flett et al., 2007). This upward migration can be reduced by four trapping mechanisms: structural, residual, dissolution, and mineral trapping. Once the CO₂ is injected into the reservoirs, some of it will be trapped by impermeable seals (i.e., by structural trapping) (Hesse et al., 2008; Iglauer et al., 2015a). In addition, large volumes of the injected CO₂ will be immobilised by the capillary trapping mechanism, which is controlled by capillary forces (Pentland et al., 2011; Iglauer et al., 2011). Some of the supercritical CO₂ will be dissolved in formation water to be trapped as dissolution trapping, which is a function of the CO₂ and water interface area (Lindeberg and Wessel-Berg, 1997; Spycher et al., 2003; Iglauer, 2011). Finally, some of the CO₂ will be trapped by the mineral trapping mechanism, which results from chemical reactions between the reservoir minerals and dissolved CO₂ (Xu et al., 2004; Gaus, 2010). Only the first three trapping mechanisms are treated in this thesis. The capacity of these trapping mechanisms and CO₂ movement in porous media are affected by many caprock and reservoir parameters. Rock wettability is one of these parameters, which, despite its importance, has received little attention in CO₂ geo-sequestration reservoir simulations (Iglauer et al., 2015a). Specifically, rock wettability has a significant

effect on capillary pressure (Anderson, 1987a; Batycky et al., 1981; Heiba et al., 1983; Melrose, 1965; Morrow, 1976), relative permeability (Owens and Archer, 1971; McCaffery and Bennion, 1974; Heiba et al., 1983; Anderson, 1987b; Krevor et al., 2012; Levine et al., 2014), residual CO₂ saturations (Iglauer et al., 2011; Anderson, 1987b; Chaudhary et al., 2013), reservoir fluid displacement mechanisms, and phase distribution in the reservoir (Morrow, 1990). Based on the experimental observations, rock wettability can vary tremendously from strongly CO₂-wet to strongly water-wet. This wettability variation is a result of factors such as reservoir temperature (Broseta et al., 2012; Al-Yaseri et al., 2016; Arif et al., 2016a,b), salinity and ion type (Al-Yaseri et al., 2016; Arif et al., 2016b; Chiquet et al., 2007; Espinoza and Santamarina, 2010; Wang et al., 2012), reservoir pressure (Al-Yaseri et al., 2016; Arif et al., 2016a,b; Chiquet et al., 2007; Broseta et al., 2012), and surface chemistry (Iglauer et al., 2015a,b). Experimental studies have shown that low salinity water flooding changes the rock wettability towards a more water-wet state (Morrow and Buckley, 2011; Myint and Firoozabadi, 2015; Sheng, 2014). Importantly, experimental measurements performed at laboratory scale (mm to cm) clearly investigated that rock wettability affects residual trapping (Iglauer et al., 2011; Iglauer, 2017; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016) and structural trapping (Naylor et al., 2011; Iglauer et al., 2015; Iglauer, 2017).

Moreover, most of the previous studies assumed that the reservoir has a homogeneous wettability (strongly water-wet) state. However, this is unlikely (Iglauer, 2017). Based on experimental investigations at different laboratory scales (ranging from the microscale to the macroscale) rock wettability is distributed heterogeneously (Aspenes et al., 2003; Chaouche et al., 1994; Graue et al., 2002; Masalmeh, 2002; Morrow et al., 1986; Spinler et al., 2002; Standnes and Austad, 2000; Vizika and Duquerroix, 1997). This heterogeneous wettability distribution is a result of various factors, including pore-surface roughness heterogeneity, mineral surface chemical heterogeneity, grain size heterogeneity, permeability heterogeneity and the associated fluid flow, adsorption of organics, reservoir depth, and chemistry variations of the rock surface (Crocker and Marchin, 1988; Drelich and Miller, 1994; Gaydos and Neumann, 1987; Iglauer et al., 2015a,b; Iglauer, 2017; Jafari and Jung, 2016; Laroche et al., 1999; Li, 1996; Lin et al., 1993; Morrow et al., 1986; Saghafi et al., 2014; Van Lingen et al., 1996; Vizika and Duquerroix, 1997). Regardless that it has been demonstrated that this heterogeneous wettability distribution significantly affects phase distribution

during EOR processes and displacement mechanisms (Laroche et al., 1999; Anderson, 1986, 1987a,b; Chang et al., 1997; Morrow, 1990; Bertin et al., 1998; Blunt, 1997; Kiriakidis et al., 1993), its influence on CO₂ storage efficiency has not been investigated previously.

Thus, in this research, we develop various multiphase flow reservoir simulation models on a hectometre-scale to investigate the influence of the rock wettability and wettability heterogeneity on CO₂ storage efficiency and CO₂ movement through porous media. This thesis also includes a sensitivity analysis on effect of other parameters, such as reservoir temperature, formation water salinity, reservoir permeability and porosity heterogeneity, injection well configurations (i.e., vertical and horizontal wells), and the CO₂ injection scheme (i.e., continuous injection, intermittent injection, and water-alternating-gas (WAG) injection), on CO₂ geo-sequestration and CO₂ flow through porous media.

1.2 Research Objectives

The main objective of this research is to investigate the influence of rock wettability on CO₂ storage efficiency and CO₂ movement through porous media. The specific objectives of this research are described below:

- 1- Investigating the influence of rock wettability on CO₂ storage efficiency and CO₂ movement in homogeneous reservoirs by testing the CO₂ trapping capacity and CO₂ plume migration for five wettability conditions (i.e., strongly water-wet, weakly water-wet, intermediate-wet, weakly CO₂-wet, and strongly CO₂-wet) in a homogeneous reservoir.
- 2- Investigating the effect of reservoir wettability on CO₂ trapping capacities and CO₂ mobility in heterogeneous reservoirs (reservoirs with a heterogeneous distribution of permeability and porosity).
- 3- Investigating the impact of heterogeneous wettability distribution (at isothermal and non-isothermal conditions) on vertical CO₂ plume migration and dissolution and residual trapping storage capacities.

4- Investigating the effect of geological parameters affecting rock wettability (i.e. reservoir temperature, brine salinity, and reservoir heterogeneity) on CO₂ geo-sequestration efficiency.

5- Optimizing CO₂ injection technology by investigating the effect of two important technical parameters: the CO₂ injection scheme (i.e., continuous injection, intermittent injection, and water-alternating-gas (WAG) injection) and the injection well configuration (i.e., vertical and horizontal wells) on CO₂ trapping efficiency.

1.3 Thesis Outline and Organisation

This thesis comprises nine chapters, including the introduction (Chapter 1), literature review (Chapter 2), methodology, results, and discussion (chapters 3 – 8), and conclusions, recommendations and outlook for future work (Chapter 9). The structure of the thesis objectives (Chapters 3 – 8) is presented in Figure 1-1. The nine chapters are detailed below:

Chapter 1 — *Introduction* — Presents a brief introduction and background of the thesis, research objectives, and thesis outline and organisation.

Chapter 2 — *Literature Review* — Provides an overview of the previous studies on rock wettability and CO₂ geo-sequestration processes. First, some fundamentals concerning oil-water-rock and CO₂-water-rock systems are presented. Then, we discuss the various existing wettability measurement methods, as well as the key parameters affecting rock wettability. We also give an overview of the different carbon geo-sequestration processes. Finally, we present a comprehensive review of the previous simulation studies regarding the influence of wettability on CO₂ storage capacity.

Chapter 3 — *Influence of CO₂-wettability on CO₂ migration and trapping capacity in deep saline aquifers. Greenhouse Gases: Science and Technology. 7(2): 328-338 (2017); doi:10.1002/ghg.1648* — Investigates the effect of CO₂ wettability on CO₂ plume migration, CO₂ mobility, and the capacity of residual and solubility trapping in homogeneous reservoirs and demonstrates its key importance. In this chapter, multiphase flow reservoir simulations on a homogeneous reservoir have been

performed using five relative permeability and capillary pressure curves that represent five characteristic wettability scenarios from strongly water-wet to strongly CO₂-wet.

Chapter 4 — *Impact of reservoir wettability and heterogeneity on CO₂-plume migration and trapping capacity. International Journal of Greenhouse Gas Control. 58: 142-158 (2017); doi: 10.1016/j.ijggc.2017.01.012* — Investigates the impact of reservoir wettability on storage capacities (i.e., residual trapping and dissolution trapping), mobile CO₂ capacity, and CO₂ plume migration patterns in heterogeneous reservoirs using 3D reservoir simulation models. This chapter also demonstrates the impact of permeability and porosity heterogeneity within the reservoir on CO₂ storage efficiency.

Chapter 5 — *Influence of injection well configuration and rock wettability on CO₂ plume behaviour and CO₂ trapping capacity in heterogeneous reservoirs. Journal of Natural Gas Science and Engineering. 43: 190-206 (2017); doi: 10.1016/j.jngse.2017.03.016* — Investigates the impact of the CO₂ injection well configuration (vertical versus horizontal) and rock wettability on the CO₂ geo-storage efficiency. In this chapter, three-dimensional multiphase flow simulations have been developed in a heterogeneous reservoir using four injection well scenarios: one vertical well, two vertical wells, four vertical wells, and one horizontal well with different wettability conditions.

Chapter 6 — *Effect of wettability heterogeneity and reservoir temperature on CO₂ storage efficiency in deep saline aquifers. International Journal of Greenhouse Gas Control. 68: 216-229 (2018); doi: 10.1016/j.ijggc.2017.11.016* — Investigates the influence of heterogeneous wettability distribution (at isothermal and non-isothermal conditions) and reservoir temperature on the CO₂ geo-storage efficiency in heterogeneous porosity and permeability reservoirs.

Chapter 7 — *Impact of salinity on CO₂ containment security in highly heterogeneous reservoirs. Greenhouse Gases: Science and Technology. 8(1): 93-105 (2018); doi: 10.1002/ghg.1723* — Computationally examines the effect of salinity (in the range 3-20 wt%) on CO₂ geo-sequestration efficiency using a highly heterogeneous 3D reservoir scale model.

Chapter 8 — *Enhancement of CO₂ trapping efficiency in heterogeneous reservoirs by water- alternating gas injection. Greenhouse Gases: Science and Technology.*

00:1–12 (2018); doi: 10.1002/ghg.1805 — Investigates the influence of CO₂ injection scenarios, namely continuous CO₂ injection, intermittent CO₂ injection, and water alternating gas injection on CO₂ trapping efficiency. In this chapter, numerical multiphase flow simulations have been used in a highly heterogeneous, hectometre-sized storage reservoir.

Chapter 9 — *Conclusions, Recommendations and Outlook for Future Work* — Concludes the thesis and provides recommendations for future research.

Note: Chapters 3 – 8 have all been published by the author in peer-reviewed journals (please refers to the section “Publications by the Author” on page VII) and are reproduced in this thesis as individual chapters. Due to the selected format, some details may appear more than once across different chapters and reader may like to read them separately. Appendix B contains the relevant copyright agreements between the author and the respective journals.

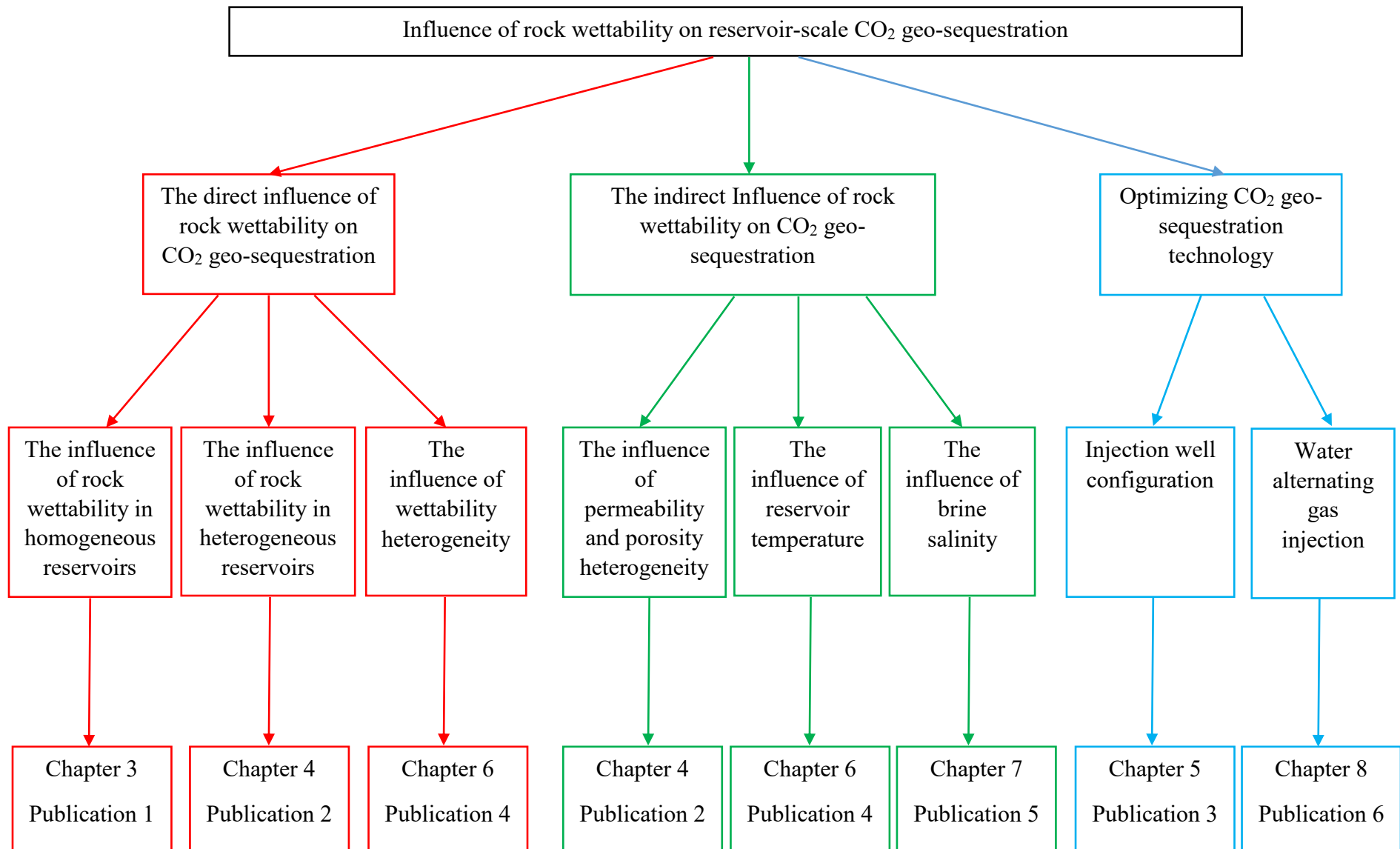


Figure 1-1 Thesis objectives and structural layout

Chapter 2 Literature Review

2.1 Introduction

This chapter provides an overview of previous studies on rock wettability and how it affects CO₂ geo-sequestration processes. First, this chapter presents the fundamentals of wettability in oil-water-rock and CO₂-water-rock systems. Methods of wettability measurements are then discussed in detail, including the contact angle methods, the Amott method, the USBM method, the imbibition method, relative permeability methods, the permeability-residual saturation relationship method, and capillary pressure methods. Among the contact angle methods, which are direct measurement methods, this chapter details the captive bubble method, the tilting plate method, the Wilhelmy balance method, capillary rise at a vertical plate, direct measurement on fibres method, and the capillary tube method. Next, the chapter discusses various chemical and geological factors affecting CO₂/water/rock wettability such as surface chemistry, surface roughness, reservoir pressure, reservoir temperature, brine composition (salinity and ion type), and impurities in the injected CO₂ stream. An overview of the technology of carbon geo-sequestration is then presented; this encompasses CO₂ storage sites (i.e., depleted oil and gas reservoirs, unminable coal seams, and deep saline aquifers) and carbon geo-sequestration trapping mechanisms (i.e., structural trapping, residual trapping, mineral trapping, and solubility trapping). Finally, the chapter is concluded with a comprehensive review of the previous simulation studies that have been carried out on the effect of wettability on CO₂ storage capacity.

2.2 Fundamentals of wettability

Wettability is the tendency of the solid surface to be in contact with a particular fluid relative to other existing fluids in the system (Abdallah et al., 1986; Al-Khdheawi et al., 2017b, c; Iglauer et al., 2015a). Rock wettability strongly controls relative permeability, capillary pressures, the residual saturation of water, and the residual saturation of non-aqueous phase liquids that significantly control multiphase fluid flow through the porous media (Chaudhary et al., 2015; Anderson, 1987a, b; Craig,

1993; Jackson et al., 2005; Morrow, 1990). On a molecular scale, wetting is the result of intermolecular forces and the degree of wetting (or wettability) is determined by the balance between adhesive forces between the liquid and solid and cohesive forces within the liquid.

In oil-water-rock systems, the force vectors are balanced at the point of contact of the oil-water-solid (Figure 2-1). This balance of the forces is represented by Young's equation (Eq.2.1) for the oil-water-solid system, as follows:

$$\sigma_{os} - \sigma_{ws} = \sigma_{ow} \cos \theta \quad (2.1)$$

where σ_{os} is the oil-solid IFT (interfacial tension), σ_{ws} is the water-solid IFT, σ_{ow} is the oil-water IFT and θ is the contact angle.

Depending on the value of the contact angle, the oil-water-rock system can be classified, in petroleum fundamentals, as three wettability scenarios: water-wet, intermediate-wet and oil-wet (Table 2-1). Thus, the oil and water distribution through the porous medium is controlled by the wettability. In water-wet rocks, water fills small pores and covers most of the rock surface while the remaining pore centres are occupied by oil. However, in oil-wet rocks, oil covers the rock surface and fills small pores while water occupies the remaining pore centres (Figure 2-2) (Anderson, 1986b).

Table 2-1 Contact angle values for the different oil-water-rock system wettability scenarios

Wettability scenario	Dake (1978)	Treiber and Ownes (1972)
Water-wet	$0^\circ \leq \theta < 90^\circ$	$0^\circ \leq \theta < 75^\circ$
Intermediate-wet	$\theta = 90^\circ$	$75^\circ \leq \theta < 105^\circ$
Oil-wet	$90^\circ < \theta \leq 180^\circ$	$105^\circ \leq \theta \leq 180^\circ$

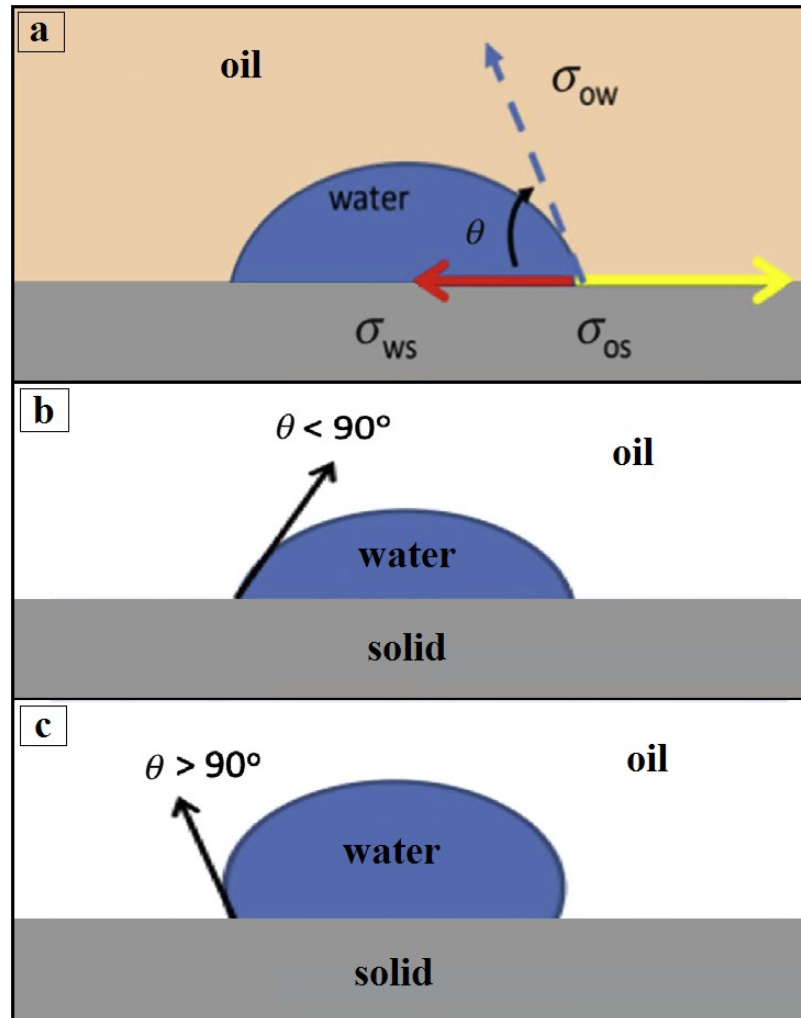


Figure 2-1 An oil-water-solid system showing: (a) interfacial forces interaction (b) contact angle value and the shape of water droplets in a water-wet system and (c) contact angle value and shape of water droplets in the oil-wet system (Wheaton, 2016).

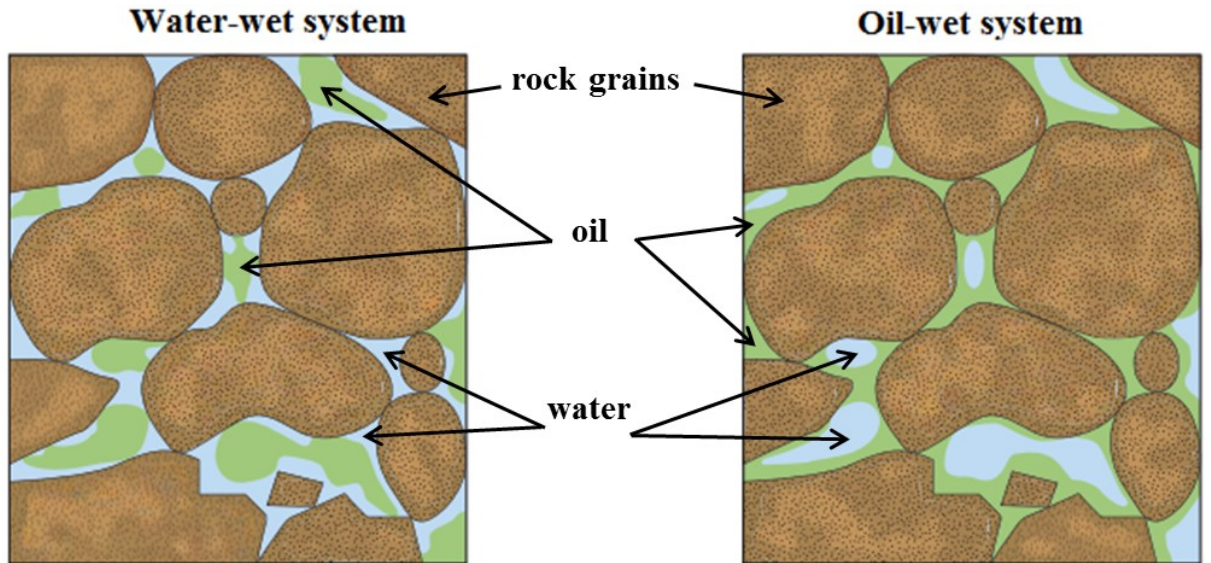


Figure 2-2 Water (blue) and oil (green) distribution in water-wet and oil-wet systems (modified after Abdallah et al., 1986).

In CO₂-water-rock systems, it has been shown that the same wettability classification as the one used for the oil-water-solid system can be used by replacing the oil-wet term by the CO₂-wet term. Thus, based on contact angle measurements, there are different wettability cases for the CO₂-water-rock system ranging from complete wetting of water to complete non-wetting of water (Table 2-2) (Iglauer et al., 2015b).

In the gas (CO₂)-liquid (water)-solid systems, Young's equation (Eq. 2.2) also describes the relationship between surface tension and contact angle, which indicates that contact angle is a function of the forces' balance between the liquid droplet and the solid surface at the line of contact (Figure 2-3). Three surface tension forces result from this liquid-solid surface contact. These are solid-liquid surface tension, liquid-gas surface tension, and solid-gas surface tension, which are used to measure the gas-liquid-rock system wetting state:

$$\sigma_{sg} - \sigma_{sl} = \sigma_{lg} \cos \theta \quad (2.2)$$

where: σ_{sg} , σ_{sl} , and σ_{lg} are the solid-gas, solid-liquid, and liquid-gas interfacial tensions, respectively.

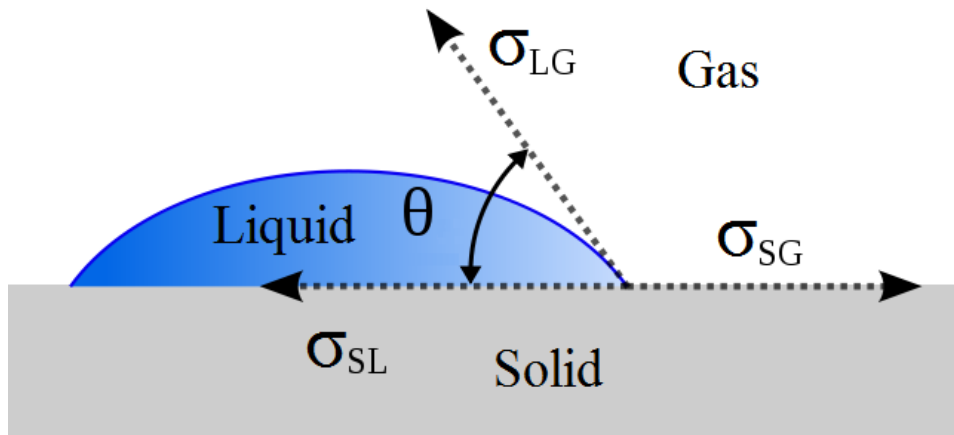


Figure 2-3. Schematic of the gas-liquid-solid system showing the interfacial forces interaction of Young's equation.

Table 2-2 Contact angle values for the different CO₂-water-rock system wettability cases (Iglauer et al., 2015b)

Wettability case	Contact angle
Complete wetting of water	0°
Strongly water-wet	0°-50°
Weakly water-wet	50°-70°
Intermediate-wet	70°-110°
Weakly CO ₂ -wet	110°-130°
Strongly CO ₂ -wet	130°-180°
Complete non-wetting of water	180°

2.3 Wettability measurements

Wettability can be measured by different methods, which can be classified as quantitative methods (i.e., contact angle method, imbibition, and forced displacement (Amott) method, and USBM method) and qualitative methods (imbibition rates, microscope examination, floatation, glass slide, relative permeability-saturation relationships, and reservoir logs) (Anderson, 1986a).

2.3.1 The contact angle method

The contact angle method is considered the most appropriate quantitative method for measuring wettability. The method is also used for measuring the wettability alteration owing to reservoir fluid flow, in addition to investigating the impact of pressure, temperature, and salinity on rock wettability. However, measuring the contact angles of the reservoir cores involves some difficulties. Contact angle measurements do not include rock heterogeneity, rock roughness, and reservoir rock complex geometry. Thus, great care is needed while sampling and cleaning the reservoir rock cores to obtain more representative contact angle measurements. Please refer to Section 2.2 and Tables (2-1) and (2-2) for more details concerning the relationship between contact angle and wettability for both oil-water-rock and CO₂-water-rock systems. Contact angle can be measured using different methods, which include the direct measurement method, the captive bubble method, the tilting plate method, the Wilhelmy balance method, capillary rise at a vertical plate, individual fibre, capillary tube, the capillary penetration method, and the capillary bridge method (Yuan and Lee, 2013).

2.3.1.1 The direct measurement method

Direct measurement of the contact angle is considered as the most common method. It includes measuring the tangent angle at the fluids-mineral contact point on a sessile drop profile using a telescope-goniometer. Bigelow et al. (1946) were the first to create the telescope-goniometer, which is a simple tool for measuring the contact angle of various liquids on polished plates. Then, at the start of the 1960s, an instrument company (Ramé-Hart) manufactured the first commercial goniometer for measuring the contact angle, which consists of a horizontal platform, a micrometre pipette, a source of illumination, and a telescope containing a protractor eyepiece (Figure 2-4). To measure the contact angle using a telescope-goniometer, the sessile drop profile tangent should be aligned at the point of contact with the surface. The protractor should then be read via the eyepiece. Since the 1960s, various changes have been made to this machine to increase its accuracy (e.g., adding a camera to take drop profile pictures to measure the contact angle at any time (Leja and Poling, 1960); increased accuracy of the examination of the contact profile using high magnification (Smithwich, 1988); employing a motor-driven syringe to measure the dynamic contact angle for sessile

drops (Kwok et al., 1996)). Although, using small volumes of substrate and liquid in this method increases the method's simplicity, this reduces the measurement accuracy owing to increasing the risk of impurities .

To measure the contact angle of the CO₂-water-rock system using this method, the rock is immersed in water then its surface is placed in contact with the CO₂ droplet. Typically, the telescope-goniometer is placed in a high-pressure cell. Owing to the ability to control various factors (e.g., size of the droplet, brine salinity, temperature, and pressure), the high-pressure cell can be used to investigate the impact of these factors on contact angle. Experimentally, during the sessile drop method, a droplet of wetting phase fluid (i.e., water) is positioned at the top of the rock surface, which is surrounded by low-density CO₂. In this method, the contact angle is determined through the water.



Figure 2-4 A direct contact angle measurement device (telescope-goniometer) from Yuan and Lee (2013).

2.3.1.2 The captive bubble method

This method is similar to the sessile drop method. However, in the captive bubble method, instead of positioning a drop of water at the top of the rock surface, a bubble of nonwetting phase fluid (e.g., CO₂) is positioned beneath the rock surface, which is surrounded by water. In this method, the contact angle is determined through the CO₂.

2.3.1.3 The tilting plate method

The tilting plate technique, which was introduced by Adam and Jessop (1925) and improved by Fowkes and Harkins (1940), was once the preferred method for measuring the contact angle owing to its accuracy and simplicity. Here, a solid plate is held on one end and the free end is landed towards a water surface in a rotating motion. When the plate's free end is immersed in the water, a meniscus is formed on the plate's sides. Then, the plate is titled until the meniscus becomes parallel with one of the plate's sides. After that, the contact angle is measured, which represents the angle between the horizontal surface and the plate. This procedure is rarely used because it requires considerable skill and the measured values of the contact angle are affected by liquid contamination and subjectivity. In 1940, the accuracy of the tilting plate technique was improved by Fowkes and Harkins (1940) by adding glass barriers for surface cleaning, film balance for detecting liquid impurities, and a microscope for ensuring the edge of intersection between the plate and the water is positioned on the rotation axis.

2.3.1.4 The Wilhelmy balance method

This technique was first used by Wilhelmy (1863) for indirectly measuring the dynamic contact angle. Experimentally, when a thin and smooth plate of solid immerses in liquid, the plate weight will be changed owing to the balance of the interfacial force (wetting force) and buoyancy force. Note that during the wetting measurement, the gravitational force will not be changed. The wetting force (F_w) is represented by the following equation (Eq. 2.3):

$$F_w = \sigma_{lg} p \cos \theta \quad (2.3)$$

where σ_{lg} = surface tension of liquid-gas, P = the plate perimeter, and θ = contact angle.

while the buoyancy force is represented by the following equation (Eq. 2.4):

$$F_b = V\Delta\rho g \quad (2.4)$$

Where:

V = displaced liquid volume.

$\Delta\rho$ = liquid-air density difference.

g = gravity acceleration.

Therefore, the following relationship (Eq. 2-5) represents the total measured force balance (F) during the Wilhelmy balance technique measurements:

$$F = \sigma_{lg} p \cos \theta - V\Delta\rho g \quad (2.5)$$

Thus, if the plate perimeter and the liquid's surface tension are known, the contact angle can be estimated using equation (Eq. 2.5) after measuring F experimentally. In some scenarios, for a zero contact angle with a given solid perimeter, the detected force is directly identified with the liquid's surface tension.

2.3.1.5 The capillary rise at a vertical plate

This method is used to measure the contact angle using the capillary effect by measuring the capillary rise h when a liquid contacts an infinitely wide and vertical plate. Here, the following integrated Laplace equation is used to estimate the contact angle θ (Cain et al., 1983):

$$\sin \theta = 1 - \frac{\Delta\rho g h^2}{2\sigma_{lg}} \quad (2.6)$$

The infinite wide theoretical requirement will be satisfied for a plate width of approximately 2 cm. When such a plate, with a homogeneous surface, is immersed in a moderate surface tension liquid (e.g., water), the contact line will be at the plate's centre. Experimentally, the capillary rise h can be measured using an optical instrument (i.e., cathetometer). This method is suitable for measuring the contact angle at various temperatures (Neumann and Good, 1979). In addition, this method can measure dynamic contact angles by pulling or pushing the plate.

2.3.1.6 Direct measurement on fibres method

The direct measurement of contact angle on fibres was first established by Schwartz and Minor (1959). In this method, a fibre surface should be suspended horizontally in a microscope field. A drop profile will be made from depositing large drops on the fibre surface. The angle of this drop profile, which can be measured using a goniometer eyepiece, represents the contact angle. This method can be used to measure the contact angle hysteresis, advancing contact angle, and receding contact angle for homogeneous and smooth fibre surfaces.

2.3.1.7 Capillary tube method

In this method, the Wilhelmy and capillary rise methods are both applied. As the Wilhelmy method is not limited to flat surfaces only, it can be used to determine the contact angles of different geometries, such as capillaries, plates, wires, rods, and tubes. The perimeter p for the tube geometry in the Wilhelmy method represents the sum of the outer and inner perimeters. Secondly, the capillary rise method can be applied to determine the contact angle of the very narrow vertical capillaries. Here, the meniscus is deemed to have a spherical geometry. The experimental measurements of capillary rise h and the capillary radius r are used to estimate the contact angle:

$$h = \frac{2\sigma_{lg} \cos \theta}{\Delta\rho gr} \quad (2.7)$$

2.3.2 The Amott method

The Amott method is an important quantitative method for measuring the average wettability of reservoir fluids and core samples by combining the imbibition and centrifugal (forced) displacement (Anderson, 1986a). The wettability measurement theory of this method is derived from the fact that the wetting fluid tends to imbibe, spontaneously, into the rock pores and displaces other fluids available in the rock (i.e., nonwetting fluid). The measurement steps in this method are summarised as follows: a) immerse the rock sample in oil for a 20-hour period and measure the volume of water spontaneously displaced (V_{wsp}) by oil imbibition; b) displace the remaining water by oil (using forced displacement) until irreducible water saturation (S_{wi}) is achieved and measure the total displaced water volume by both spontaneous and forced displacement processes (V_{wt}); c) immerse the rock sample in brine for a 20-hour period and measure the volume of the oil spontaneously displaced (V_{osp}) by imbibition of water; d) displace the remaining oil by water (using forced displacement) until irreducible oil saturation (S_{oi}) is achieved and measure the total displaced oil volume by both spontaneous and forced displacement processes (V_{ot}) (Anderson, 1986a). Then, the ratios of displacement by oil (δ_o) and by water (δ_w) can be estimated as follows:

$$\delta_o = \frac{V_{wsp}}{V_{wt}} \quad (2.8)$$

$$\delta_w = \frac{V_{osp}}{V_{ot}} \quad (2.9)$$

Thus, the wettability of core sample is estimated using these ratios as follows:

Water-wet sample: $\delta_w = \text{positive}$ and $\delta_o = 0$

Strongly water-wet: $\delta_w = 1$ and $\delta_o = 0$

Neutral-wet sample: $\delta_w = 0$ and $\delta_o = 0$

Oil-wet sample: $\delta_w = 0$ and $\delta_o = \text{positive}$

Strongly oil-wet: $\delta_w = 0$ and $\delta_o = 1$

2.3.3 United States Bureau of Mines (USBM) method

This method was first established by Donaldson et al. (Donaldson et al., 1969; Donaldson et al., 1980; Donaldson, 1981) for quantitatively measuring the wettability

of the oil-water-rock system. In this method, the required work for displacing a certain fluid by another fluid are compared. Generally, displacing the wetting fluid from the rock sample by nonwetting fluid requires more work than required for displacing the nonwetting fluid, from the same rock sample, by the wetting fluid. Morrow (1970) and Leverett (1941) reported that this required work for fluid displacement is a function of the area below the capillary pressure curve. Specifically, for water-wet rocks, displacing oil by water process has a smaller area below the capillary pressure curve than the process of displacing water by oil. This is due to the fact that most of the water will be spontaneously imbibed inside the water-wet rock and the area below the capillary pressure curve will be small for the process of displacing oil by water. The ratios of these two different areas below the capillary pressure curves are used in the USBM method for calculating the wettability index (W) as follows:

$$W = \log (A_1/A_2) \quad (2.10)$$

where A_1 = the area below the oil-drive curve and A_2 = the area below the brine-drive curve.

Thus, the wettability estimation depends on the wettability index value (W). If the W value is near 1, the rock is water-wet while if the W value is near -1 (i.e., less than zero), the rock is oil-wet. If the W value is near zero, the rock is neutral-wet (Figure 2-5).

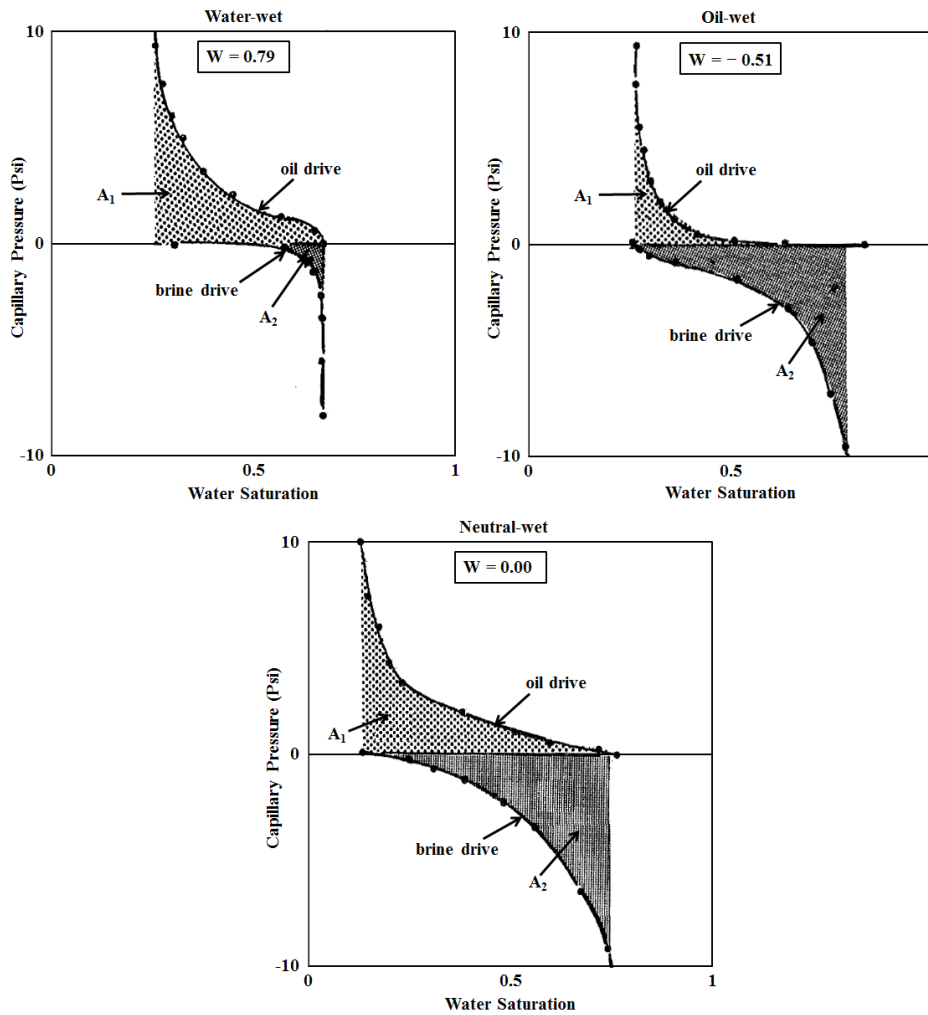


Figure 2-5 Illustration of wettability measurement using the USBM method for three different wettability conditions: water-wet, oil-wet, and neutral-wet. W is the USBM wettability index (modified after (Anderson, 1986a)).

2.3.4 The imbibition method

As a result of its simplicity and rapidity, the imbibition method is a common method for qualitatively measuring wettability. The original method can only measure the wettability at room conditions (temperature and pressure; Bobek et al., 1958). However, the apparatus modified by Kyte et al. (1961) was able to measure the wettability at reservoir conditions. To estimate the wettability using this method, a core sample, at initial water saturation, is immersed in water to simulate an imbibition process and measure the volume of displaced oil, caused by the water imbibition, over time. The rock sample is considered strongly water-wet if the water imbibition process is rapid and weakly water-wet if the water imbibition process is slow. However, for

neutral-wet and oil wet rocks, no water will be imbibed. In this latter case, when no water is imbibed, the core sample should be driven to residual oil saturation and immersed in oil. Then, the volume of displaced water, owing to the oil imbibition, is measured as a function of time. The rock sample is oil-wet if it imbibes oil. The oil imbibition process rapidity will be used to indicate the degree of oil-wetness (i.e., strongly oil-wet rocks will have a rapid oil imbibition process while weakly oil-wet rocks will have slow oil imbibition). If neither water nor oil imbibition occurs, the rock is considered to have neutral wettability. However, if both the water and oil imbibition processes occur, the rock is either mixed-wet or fractional-wet. Importantly, the imbibition rate is not only a function of wettability but it also depends on other factors, such as initial saturation, pore structure, interfacial energy, viscosity, and relative permeability (Anderson, 1987a; Dullien, 1979). Thus, this is considered as the main problem of the imbibition method. This dependency can be decreased by comparing the measured core sample imbibition volumes with a strongly water-wet imbibition measurement (i.e., with a reference measurement).

2.3.5 Relative permeability methods

Relative permeability is an important qualitative method for wettability measurement. Generally, if more than one phase (e.g., gas, water, and oil) flows simultaneously in a system, the relative permeability of a particular phase represents the ratio of the effective phase permeability, which is the permeability of one fluid if more than one fluid is available in the system, to the absolute permeability. Relative permeability for a particular phase has a direct relationship with the phase saturation and the phase relative permeability increases with increasing the phase saturation. For each phase, the relative permeability ranges from zero to 1. Thus, for a given phase saturation, the relative permeability of each phase can be calculated as follows:

$$K_{rg} = \frac{K_g}{K} \quad (2.11)$$

$$K_{rw} = \frac{K_w}{K} \quad (2.12)$$

$$K_{ro} = \frac{K_o}{K} \quad (2.13)$$

Where K_{rg} , K_{rw} , and K_{ro} are the relative permeability of gas, water, and oil, respectively, K_g , K_w , and K_o are the effective permeability of gas, water, and oil, respectively, and K is the absolute permeability.

Thus, if two fluids are flowing simultaneously in a system, Darcy's equation will be modified by using the relative permeability concept to calculate the flow rate for each phase as follows:

$$Q_o = -A \frac{K K_{ro} \Delta P}{\mu_o \Delta L} \quad (2.14)$$

$$Q_w = -A \frac{K K_{rw} \Delta P}{\mu_w \Delta L} \quad (2.15)$$

where Q_o = oil flow rate, Q_w = water flow rate, A = cross sectional area, μ_o = oil viscosity, μ_w = water viscosity, ΔP = pressure drop through the flow system, and ΔL = length of the flow system.

Based on the well-known fact that wettability affects relative permeability curves and the associated residual oil, water, and gas saturation (Owens and Archer, 1971; McCaffery and Bennion, 1974; Anderson, 1987b; Craig, 1993; Batycky et al., 1981), many methods were investigated to qualitatively estimate the rock wettability. Craig (1993) proposed important rules of thumb to distinguish the differences between oil-wet and water-wet rocks, which is considered an important relative permeability method for estimating the rock wettability (Anderson, 1987b). Craig's rules of thumb estimate the rock wettability based on the values of three main relative permeability curve's characteristics: the residual water saturation, the water saturation where the relative permeability of oil and the relative permeability of water are equal, and water relative permeability at floodout (i.e. at maximum water saturation; Table 2-3 and Figure 2-6). Later, in this work, we use Craig's rules of thumb for modelling different wettability cases in our simulations.

Table 2-3 Typical characteristics of water-wet and oil-wet relative permeability curves (Craig, 1993)

Characteristic	Water-wet	Oil-wet
Residual water saturation (S_{wr} , %)	$20 < S_{wr} \leq 25$	$S_{wr} < 10$
Water saturation where relative permeability of oil and relative permeability of water are equal (S_w , %)	$S_w > 50$	$S_w < 50$
Water relative permeability at floodout (K_{rw} , %)	$K_{rw} < 30$	$50 \leq K_{rw} \leq 100$

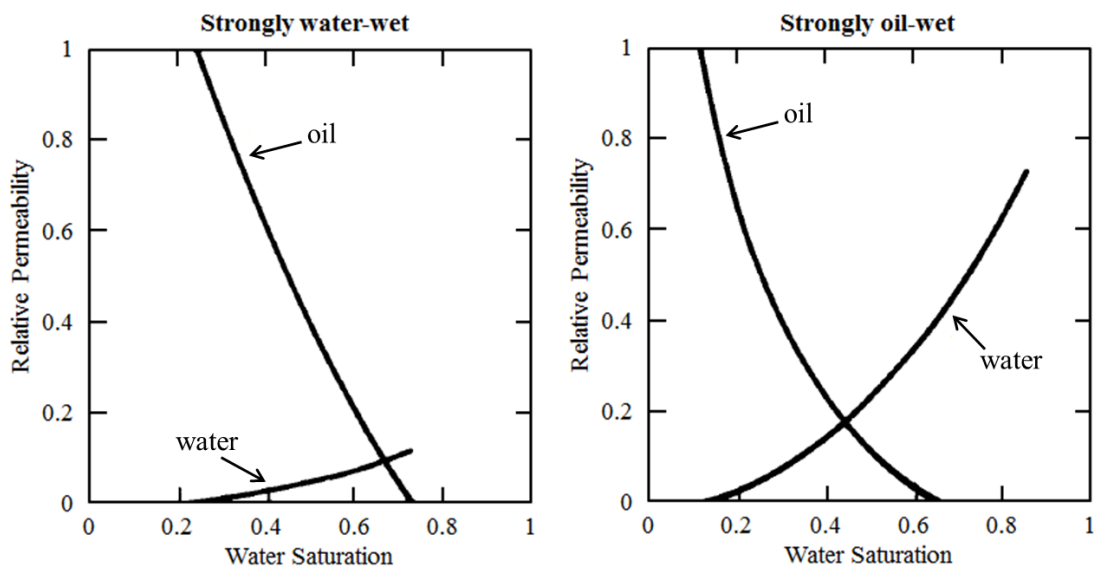


Figure 2-6 Typical relative permeability curves for strongly water-wet and strongly oil-wet rocks (Craig, 1993).

The second method for evaluating rock wettability using relative permeability measurements is presented by Treiber and Owens (1972). In this method, the relative permeability curves of water-oil, oil-gas, and water-gas are compared. Then, the wettability is estimated based on the fact that the strongly wetting phase relative permeability is a function of that same phase saturation only (Craig, 1993; Owens and Archer, 1971). Importantly, for water-wet rock, the oil phase is the preferable wetting phase compared to the gas phase. Thus, in this method, the rock is considered as strongly water-wet when the oil relative permeability in the gas-oil system is found to be a continuation to the water relative permeability in the oil-water system (Owens

and Archer, 1971), as shown in Figure 2-7. However, the rock is not strongly water-wet if the oil relative permeability (in the gas-oil measurement) is significantly different from the water relative permeability (in the oil-water measurement).

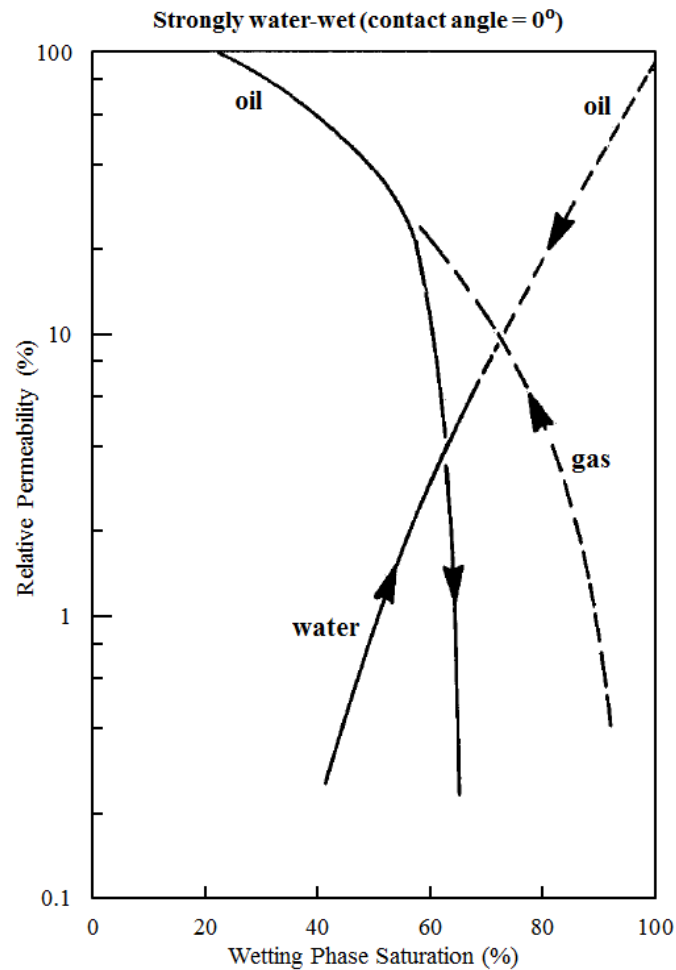


Figure 2-7 Comparison of the oil-gas drainage (dashed lines) and water-oil imbibition (solid lines) relative permeability curves measured in a strongly water-wet rock (Torpedo sandstone). The relative permeability of water in the water-oil test is a continuation of the relative permeability of oil in the oil-gas test. Thus, the rock is strongly water-wet (Owens and Archer, 1971).

2.3.6 The permeability-residual saturation relationship method

Proposed by Raza et al. (1968), this method is used to qualitatively determine rock wettability based on the relationship between air permeability and residual water saturation. The rock wettability is estimated by plotting the air permeability versus the

residual water saturation. The average residual water saturation of oil-wet rock is usually lower than that of the water-wet rock. Thus, if the best-fitting line for the residual water saturation-permeability relationship is found to be approximately vertical and it is extended over a small residual water saturation range, the rock is considered oil-wet. However, if the best-fitting line is found to be inclined with a low vertical gradient and it covers a large residual water saturation range, the rock is water-wet (Figure 2-8).

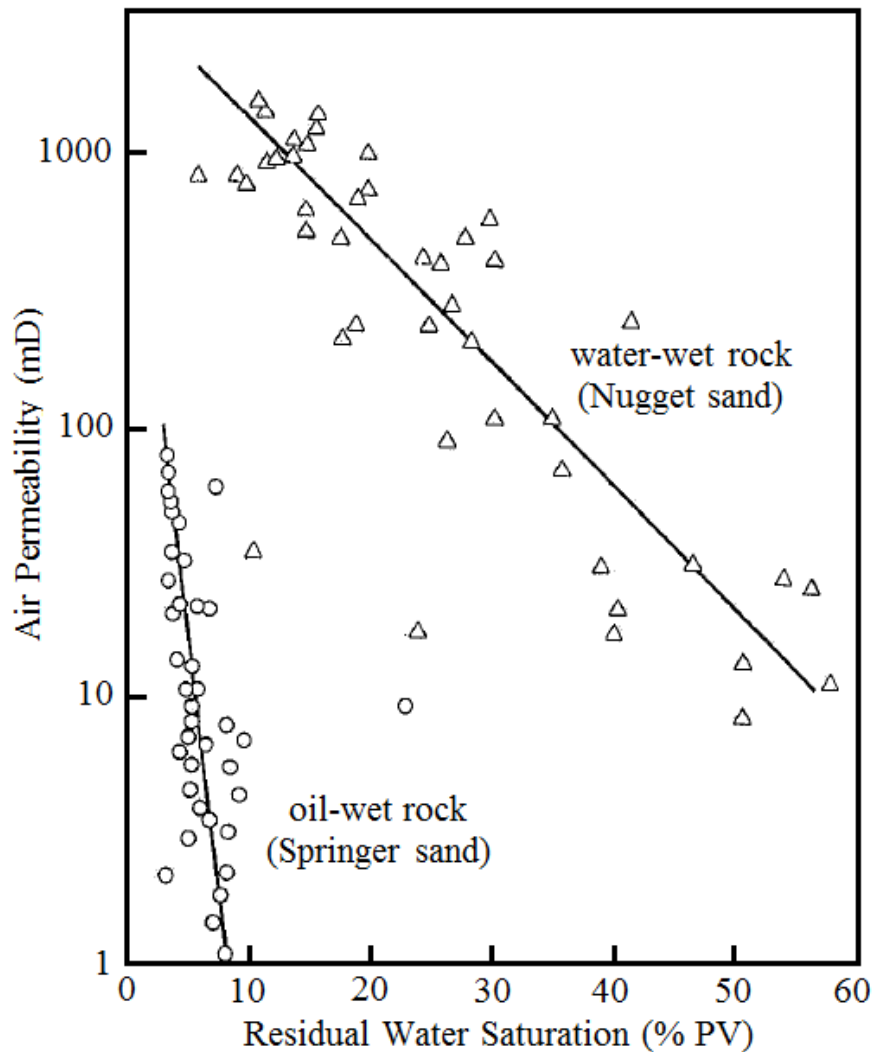


Figure 2-8 Permeability-residual water saturation relationship for water-wet and oil-wet rocks (Raza et al., 1968).

2.3.7 Capillary pressure methods

Capillary pressure (P_c) is the pressure difference between two immiscible fluids coexisting in the same porous medium, which are separated by an interface. Capillary pressure is considered as an important rock property that controls the multiphase flow

in porous media. Specifically, in reservoir engineering fundamentals, capillary pressure (P_c) is defined as the pressure difference between the non-wetting phase and wetting phase as follows:

$$P_c = P_{non-wetting} - P_{wetting} \quad (2.16)$$

In addition, capillary pressure is affected by rock-fluid interfacial tensions, pore size distribution, and rock-wetting characteristics. Thus, the capillary pressure can be calculated using the following relationship:

$$P_c = \frac{2\sigma \cos \theta}{r} \quad (2.17)$$

where σ is the interfacial tension between the wetting and non-wetting fluids, θ is the contact angle, and r is the capillary radius.

The use of capillary pressure curves in estimating rock wettability was first suggested by Calhoun (1951) while the first attempt to use the area below the capillary pressure curve to predict wettability was done by Gatenby and Marsden (1957). They used the areas surrounded by both the imbibition and drainage curves, for negative and positive capillary pressures, to investigate the relationship between capillary pressure and rock wettability. According to the results, they concluded that there is no clear relationship between these capillary pressure areas and rock wettability. However, Donaldson et al. (1969) suggested that only the areas below the water-drive and oil-drive curves should be used when estimating rock wettability.

Another capillary pressure method for estimating the rock wettability is called the method of displacement capillary pressure. In this method, the threshold (or displacement) capillary pressure is used to calculate the apparent contact angle θ_a . Threshold capillary pressure refers to the minimum required capillary pressure for the nonwetting phase to enter a rock that is fully saturated with the wetting phase (Benner et al., 1942; Bartell and Osterhof, 1927). The apparent contact angle can be predicted using the following relationship (Dullien, 1979):

$$P_T = \frac{2\sigma \cos \theta_a}{r_{max}} \quad (2.18)$$

where P_T = Threshold capillary pressure, r_{max} = pore radius by which the nonwetting phase starts to enter the rock, and θ_a = apparent contact angle.

As increasing the pore radius leads to reducing the capillary pressure required for injecting the nonwetting phase, the average radii of the biggest pores of the rock is used to represent r_{max} . There are two limitations of this method. These are the effect of the pore geometry on the measured contact angle and the capability of examining the largest pore wettability only (Anderson, 1986a).

2.4 Factors affecting CO₂/water/rock wettability

CO₂-wettability can differ substantially within the reservoir from strongly CO₂-wet to strongly water-wet (Iglauer, 2017). This CO₂-wettability variation is a result of chemical and geological factors such as surface chemistry, surface roughness, reservoir pressure, reservoir temperature, brine composition (salinity and ion type), and impurities. The influence of these factors on CO₂ wettability is discussed in the sections that follow.

2.4.1 Rock surface chemistry

The chemistry of the rock surface has a significant impact on the CO₂-wettability. Generally, hydrophilic surfaces have a high affinity for water (i.e., water-wet) while hydrophobic surfaces have a low affinity for water (i.e., CO₂- wet) (Iglauer, 2017). Thus, because of their hydrophilicity, the clean rock-forming minerals such as mica, calcite, quartz, and feldspar are water-wet. Moreover, it was shown that wettability is strongly influenced by the degree of surface deprotonation and hydroxylation (Chen et al., 2015b). However, it was reported that owing to their hydrophobicity, oil-wet rocks are intermediate-wet or CO₂-wet (Iglauer et al., 2015b). Importantly, the measured contact angle on the clean surface is significantly lower than the actual contact angle in the reservoir as mineral surfaces in the reservoir are unlikely clean.

In summary, wettability can vary in subsurface formations, at reservoir conditions, from strongly CO₂-wet to strongly water-wet. Thus, in this work, different wettability scenarios (from strongly CO₂-wet to strongly water-wet) have been accurately simulated and the effect of this wettability variation on CO₂ storage capacity and contaminate security has been investigated.

2.4.2 Reservoir pressure

Reservoir pressure, also known as hydrostatic pressure or formation pressure, is the pressure of fluids within the reservoir. In CO₂ geo-sequestration processes, reservoir pressure is usually computed from the hydrostatic pressure gradient of 10 MPa/km (Dake, 2007). Pressure is an important influencing factor on CO₂-wettability (Iglauer, 2017). Previous experimental studies performed on various minerals clearly showed that increasing pressure leads to increased CO₂-wettability (i.e., increasing pressure leads to increasing the advancing and receding contact angle) (Iglauer et al., 2014; Chiquet et al., 2007; Broseta et al., 2012; Farokhpoor et al., 2013). This increase in CO₂-wettability (i.e., contact angle) with increasing pressure is mainly a result of the highly increasing CO₂ density with pressure, which, in turn, leads to enhancing the intermolecular interactions between the CO₂ molecules and minerals (Iglauer et al., 2012).

2.4.3 Reservoir temperature

By reason of the impact of temperature on various factors affecting wettability (e.g., water-CO₂ interfacial energies, fluid dielectric constant, and fluid density), there is an intricate relationship between temperature and contact angle. Thus, previous studies have indicated two different trends of the effect of temperature on water-CO₂-mineral contact angle. Some literature studies observed that the contact angle of quartz substrate in the CO₂-water-mineral system decreases with increased temperature (Yang et al., 2005; Yang et al., 2008a; Bikkina, 2011; Saraji et al., 2013). However, other studies observed that the contact angle of the same substrate (i.e., quartz) increases with temperature (Farokhpoor et al., 2013; Sarmadivaleh et al., 2015). Further, Iglauer et al. (2012) showed that the contact angle decreases with temperature using molecular dynamics simulations. In summary, the effect of temperature on the water-CO₂-mineral system is considered as an open research area and more research work is required to understand its fundamentals (Iglauer, 2017).

2.4.4 Brine composition: salinity and ion type

Brine salinity is widely varied in CO₂ geo-sequestration formations. Previous studies have clearly demonstrated that brine salinity significantly affects CO₂-wettability.

Many previous experimental studies qualitatively investigated that the CO₂-water-mineral contact angle increases with increased salinity for various substrates (Espinoza and Santamarina, 2010; Broseta et al., 2012; Wang et al., 2012; Jung and Wan, 2012; Farokhpour et al., 2013; Saraji et al., 2014; Sarmadivaleh et al., 2015). Mechanistically, the rock surface is charged and this charge is affected by the mineral type and PH value. In brine, the positive ions move toward the negative surface charges; this reduces the surface potential and polarity. This surface potential and polarity reduction leads to reduced hydrophilicity and the contact angle is thus increased (i.e., CO₂-wettability will be increased at higher salinities) (Iglauer, 2017). Further, the dissolved ion type in brine also affects the CO₂-wettability. The ionic strength and the ratios of the charge to volume in the high-valency ions (e.g., Al³⁺, Mg²⁺, SO₄²⁻ and Ca²⁺) are higher than that of the monovalent ions (Cl¹⁻, Na⁺ and K⁺), which results in a more effective shielding to the surface charge, which results in increasing the contact angle (i.e., increasing the CO₂ wettability).

2.4.5 Impurities in injected CO₂ stream

The CO₂ injected in the CO₂ geo-sequestration process could contain impurities such as H₂S, H₂, N₂, O₂, CH₄, CO, SO_x, NO_x, and Ar (Iglauer, 2017). According to the used separation method, the injected CO₂ impurity ranges from 0.05% to 5% by volume (Porter et al., 2015). Moreover, in some petroleum production operations, the produced gas, which could be highly contaminated with H₂S, is reinjected into the reservoir to enhance oil recovery. Thus, it is important to test the impact of these impurities on CO₂-wettability. For instance, previous experimental studies observed that pure CO₂ is more CO₂-wet (i.e., higher contact angle) than the N₂ and CO₂-N₂ mixture (Kaveh et al., 2014; Iglauer, 2017) and less CO₂-wet (i.e., lower contact angle) than H₂S (Broseta et al., 2012).

2.4.6 Surface roughness

It is well established that surface roughness affects the CO₂-water-mineral wettability. Surface roughness affects the CO₂-water-mineral contact angle as represented by the Wenzel equation (Wenzel, 1949):

$$\cos \theta_{rough} = r \cos \theta_{smooth} \quad (2.19)$$

where θ_{rough} : apparent (measured) contact angle, θ_{smooth} : Young contact angle on ideal surface (i.e. totally flat surface), and r : the roughness factor, which represents the ratio of the measured surface area to the ideal surface area; thus, for the ideal surfaces, $r=1$). Typically, surface wetting increases by increasing surface roughness. Thus, in water-wet rock, increasing surface roughness leads to increasing the water wetting state while in CO₂-wet rock, increasing surface roughness results in increasing the CO₂ wetting state. Wang et al. (2013) observed a significant difference in contact angle between smooth (RMS = 5.8 nm) and rough (RMS = 2300 nm) silica surfaces. They found that the contact angle of a smooth silica surface ranged from 31° to 38° while it was only 1° for the rough silica surface, measured at the same pressure (0 to 20 MPa), temperature (323 K), and salinity (1.5 M NaCl).

2.5 Carbon geo-sequestration

Carbon geo-sequestration, also known as geological CO₂ sequestration, is an effective strategy for anthropogenic carbon emissions mitigation. In the carbon geo-sequestration process, CO₂ is captured from large CO₂ sources (e.g., heat supply processes, fossil fuel, power generation, solid fuels burning) and injected, normally as a supercritical fluid, into an accurately selected geological site (Pruess et al., 2003). For many decades, the technology of CO₂ injection into geological formations (e.g., oil and gas reservoirs) has been used for enhanced oil recovery purpose. However, in recent years, this technology has been applied for the purpose of long-term CO₂ geo-sequestration. It is important to mention that the first commercial CO₂ injecting project for CO₂ geo-sequestration and enhanced oil recovery purposes was developed in 2000 in the Weyburn oilfield in Canada (Preston et al., 2005).

There are various geological sites for CO₂ geo-sequestration (Figure 2-9). The CO₂ geo-sequestration site must satisfy some fundamental characteristics such as having high permeability, being overlaid by a caprock seal, being economically feasible, and being environmentally safe (Metz et al., 2005). It was estimated that for a carefully selected and evaluated CO₂ geo-sequestration formation, approximately 99% of the total injected CO₂ can be retained over a long-term storage period (1000 years) and the CO₂ storage period can be extended for millions of years (Metz et al., 2005).

Mainly, CO₂ can be stored in three various storage formations: depleted oil and gas reservoirs, unminable coal seams, and deep saline aquifers. Table 2-4 summarises the estimated storage capacity for the different geological storage sites. These sites are described in more detail in sections that follow.

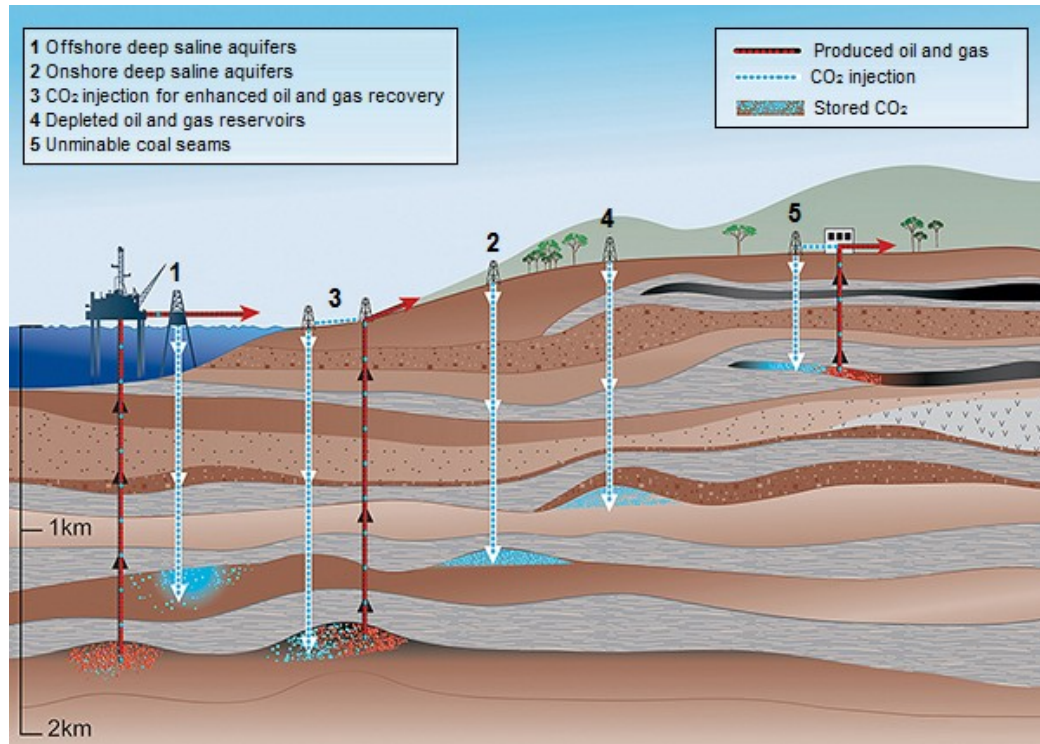


Figure 2-9 An overview of CO₂ injection and geological storage sites (modified after Metz et al., 2005).

Table 2-4 Storage capacity for different storage sites (modified After Metz et al., 2005)

Storage site	Lower estimate for the CO ₂ storage volume (Gt)	Upper estimate for the CO ₂ storage volume (Gt)
Unminable Coal Seams	3-15	200
Depleted Oil and Gas Reservoirs ^a	675	900
Deep Saline Aquifers	1000	Approximately 10000

^aThese lower and upper estimates for the storage volume of depleted oil and gas reservoirs may be increased by 25% after including the undiscovered hydrocarbon fields.

2.5.1 Depleted oil and gas reservoirs

Depleted oil and gas reservoirs are considered as economical carbon geo-sequestration sites owing to various reasons such as the well-understood geology during the oil and gas exploration operations, the wide use of CO₂ injection into oil and gas reservoirs to enhance hydrocarbon recovery, and the availability, in place, of the surface and underground infrastructure, which can be used to inject CO₂ for geo-sequestration purposes (Li et al., 2006). Thus, CO₂ injection into oil and gas reservoirs improves hydrocarbon production and reduces CO₂ emissions. It was estimated that the CO₂ storage capacity in depleted oil reservoirs is approximately 60% of the injected CO₂ (Shaw and Bachu, 2002). However, approximately 40% of the injected CO₂ is produced with oil, which can be reinjected into the oil reservoir (Shaw and Bachu, 2002). CO₂ storage capacity in depleted oil and gas reservoirs is estimated to be approximately 50 Gton worldwide (Firoozabadi and Myint, 2010). The Weyburn CO₂ capture and storage project in Canada and CO₂CRC Otway project in Australia are examples of CO₂ geo-sequestration in depleted oil and gas reservoirs, respectively. In the Weyburn CO₂ capture and storage project, 5000t/day CO₂ is injected in the Weyburn oil field for enhanced oil recovery and CO₂ sequestration purposes (Bachu, 2008). However, in the CO₂CRC Otway project, carbon dioxide is collected from a natural gas field (Buttress field, which contains 20% methane and 80% carbon dioxide) and injected into a small gas field (Naylor gas field), which has an estimated initial gas volume of 5.4 Bscf (Underschultz et al., 2011).

2.5.2 Unminable coal seams

CO₂ sequestration in unminable coal seams is an important technology for reducing CO₂ emissions to the atmosphere (Shi and Durucan, 2005). Recently, CO₂ sequestration in unminable coal seams, particularly in conjunction with enhanced coal-bed methane (ECBM) recovery, has attracted significant attention (Zhang et al., 2016b). This technology is important because of its capability of long-term CO₂

sequestration in addition to producing methane, especially in high-methane-content coal, which assists in reducing the cost of the CO₂ sequestration process (Shi and Durucan, 2005). Physically, CO₂ injected into deep unminable coal formations adsorbs on the coal surface and displaces the existing methane because of its higher affinity to adsorb onto coal compared to methane. This technology was performed in many locations worldwide (e.g., the CO₂ sequestration in deep unminable coal seams and enhanced coalbed methane recovery project in the San Juan Basin (New Mexico) by Burlington Resources) (Stevens et al., 1998).

2.5.3 Deep saline aquifers

Deep saline aquifers are porous rocks layers at depths of over 800 m. They usually contain formation waters with high salinity, which don't have any commercial value (Metz et al., 2005). However, deep saline aquifers are the most preferred CO₂ storage sinks, as they provide the largest CO₂ storage capacity compared to other CO₂ geological sinks (Metz et al., 2005). In deep saline aquifers, for efficiency and safety considerations, CO₂ must be injected at depths deeper than 800 m, where the CO₂ remains in a supercritical phase owing to the associated temperature (higher than critical temperature of CO₂; 31.04°C) and pressure (higher than critical pressure of CO₂; 7.39 MPa) (Holloway and Savage, 1993). Table 2-5 summarises the current and planned projects for CO₂ injection and storage in saline aquifers.

Table 2-5 List of current and planned CO₂ geo-sequestration projects in saline aquifers (Metz et al., 2005)

Name	Location	Type	Injection start	Injection rate (t/day)	Total storage (Mt)
MRCSP (Cincinnati Arch)	USA	Pilot	2009	500	0.001
Frio	USA	Pilot	2004	177	0.0016
SECARB Mississippi	USA	Pilot	2008	160	0.00275
Nagaoka	Japan	Pilot	2003	40	0.01

MRCSP (Michigan Basin)	USA	Pilot	2008	300-600	0.06
Ketzin	Germany	Pilot	2006	100	0.06
MGSC Decatur	USA	Demonstration	2010	1000	1
SECARB Early	USA	Demonstration	2009	2700	1.5
In Salah	Algeria	Commercial	2004	3500	17
Sleipner	Norway	Commercial	1996	2700	20
Snøhvit	Norway	Commercial	2008	2000	23
Gorgon	Australia	Commercial	2014	12300	129

2.6 Carbon geo-sequestration trapping mechanisms

As soon as the supercritical CO₂ is injected into the geological storage sites, most of the injected CO₂ migrates laterally and vertically with a risk of leaking toward the surface because of the buoyancy forces (i.e., the density differences between the formation water and the injected supercritical CO₂). This CO₂ leakage risk can be minimised by various physicochemical trapping mechanisms including structural trapping (Iglauer et al., 2015b), residual trapping (Iglauer et al., 2011), dissolution trapping (Iglauer, 2011), and mineral trapping (Gaus, 2010). These trapping mechanisms have different time scales (Figure 2-10). These four trapping mechanisms are described in detail in the following sections.

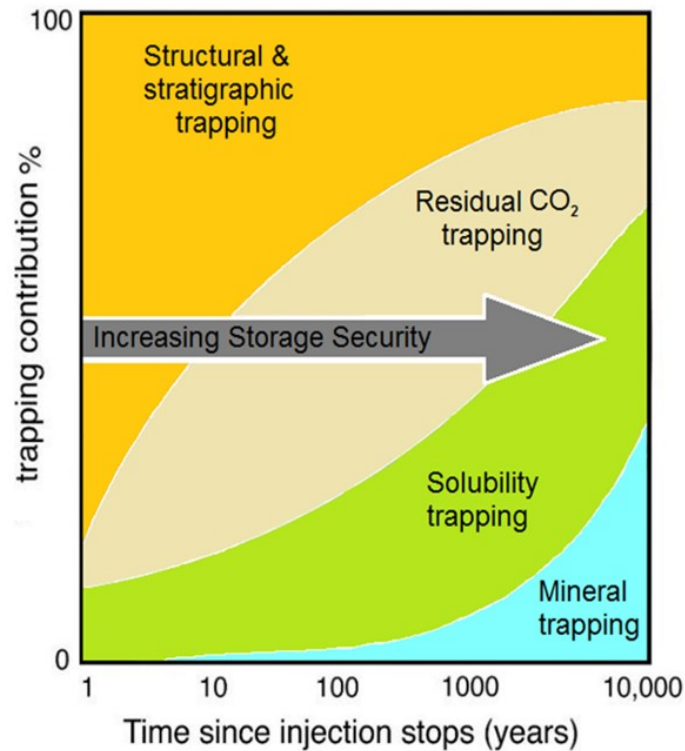


Figure 2-10 The post-injection time scale associated with different trapping mechanisms (Metz et al., 2005).

2.6.1 Structural trapping

Structural/hydrodynamic trapping represents a primary and physical trapping mechanism where the CO₂ is trapped below impermeable (or very low permeability) caprock such as clay, shale, tight carbonates, and mudrock (Iglauer et al., 2015b). CO₂ is often trapped below the caprock owing to the higher capillary forces associated with these very low-permeability seals compared to the buoyancy forces (Hesse et al., 2008). Structural trapping has an important role in any geological storage formation by preventing the CO₂ from migrating upward during the time required to the other storage mechanisms to start to apply (Bachu et al., 1994a).

The efficiency of structural trapping is affected by various parameters, such as interfacial tension, caprock wettability, and caprock properties (Iglauer et al., 2015b; Chiquet et al., 2007; Broseta et al., 2012).

2.6.2 Residual trapping

Residual trapping, also known as capillary trapping, is a major trapping mechanism of CO₂-geosequestration, which can rapidly store large amounts of CO₂ (Juanes et al., 2006; Kumar et al., 2005; Hesse et al., 2008; Pentland et al., 2011). The residual phase is formed when a particular (non-wetting) phase is displaced from the rock pore centre by another immiscible (wetting) phase. Then, a significant amount of CO₂ is trapped in the small pore spaces owing to capillary forces. The capacity of residual trapping is affected by various factors, including characteristics of the residual phase saturation, initial gas saturation, wettability, pore geometry, hysteresis and interfacial tension (Suekane et al., 2008; Iglauer et al., 2011; Chalbaud et al., 2009; Al-Menhali et al., 2015).

2.6.3 Solubility trapping

Solubility/dissolution trapping refers to the dissolution of injected CO₂ into the formation water (Iglauer, 2011). This dissolution of CO₂ in water results in high-density CO₂-enriched brine, which permanently sinks in the formation. The mass transfer in the solubility trapping is a function of the CO₂-water interface area and a larger CO₂-water contact area leads to increasing the amount of CO₂ dissolved in water (Doughty, 2010; Kumar et al., 2005). In addition, solubility trapping is influenced by various factors, such as the reservoir temperature, reservoir pressure, and formation water salinity (Metz et al., 2005). Dissolution trapping is a slow process, ranging from 100 to 1000 years (Emami-Meybodi et al., 2015).

2.6.4 Mineral trapping

Mineral trapping results from the reaction of CO₂ with the reservoir minerals, which is stored in the reservoir in form of minerals (Xu et al., 2005). Mineral trapping is considered an attractive trapping mechanism because it can permanently store the CO₂ (Gaus, 2010). Ortoleva et al. (1998) summarised the chemical reactions associated with the mineral trapping mechanism. First, the weak carbonic acid is produced from dissolving CO₂ in water:



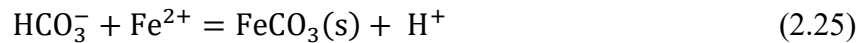
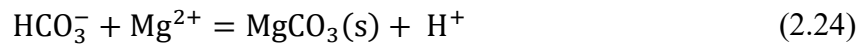
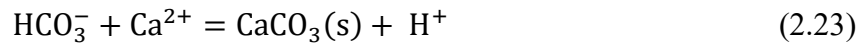
Then, the bicarbonate ion is formed by rapid carbonic acid dissociation:



Increasing acidity leads to inducing the formation minerals dissolution, which results in complexing of the bicarbonate ion with the dissolved cations:



Then, a carbonate mineral is precipitated after the reaction between the species of dissolved bicarbonate and divalent cations. This reaction could result in creating Fe(II), Mg, and Ca carbonates, which represents the mineral trapping (Gunter et al., 1997).



2.7 Previous simulation studies on the effect of wettability on CO₂ storage

Although various experimental studies have demonstrated (at the lab scale ranging from mm to cm) that wettability affects structural and residual trapping capacities (e.g., Iglauer et al., 2011; Iglauer et al., 2015; Espinoza and Santamarina, 2010; Farokhpour et al., 2013), little attention has been placed on the effect of wettability on reservoir-scale processes and associated CO₂ geo-sequestration capacity predictions. Importantly, various parameters (e.g., hysteresis, residual water saturation, residual gas saturation, relative permeability and the capillary pressure) depend on wettability (Anderson, 1987a, b; Craig, 1993; Jackson et al., 2005; Morrow, 1990). This section, thus, introduces a short summary of the previous simulation studies performed on the effect of wettability and wettability dependent parameters on CO₂ geo-sequestration capacity.

First, it is important to mention that the first real research studies about the use of underground formations for storing CO₂ were completed at the beginning of the last decade in the nineteenth century (Holloway and Savage, 1993). After that, a wide

range of studies was conducted to develop the CO₂ geo-sequestration technology. There are different simulation models developed to study the effect of different rock and fluid properties on the CO₂ storage and movements.

Ghanbari et al. (2006) developed a three-dimensional homogeneous model to simulate CO₂ storage in a saline aquifer for 470 years after 30 years of CO₂ injection using the GEM-CMG simulator (Computer Modelling Group). The dimensions of the developed model were (8 km x 8 km x 200 m) and the number of grid blocks was (35 x 35 x 25) in X, Y and Z directions, respectively. The study was aimed at estimating the key parameters that affect the hydrodynamic and solubility trapping mechanisms by evaluating the effect of different parameters including residual gas saturation and residual water saturation. This study compared the ability of each case to increase the capacity of hydrodynamic and solubility trapping mechanisms. The results of this study showed that residual gas saturation increase leads to increasing the amount of immobilised CO₂ while reducing the solubility trapping capacity. In addition, this study concluded that higher residual water saturation results in a higher solubility trapping capacity.

Another simulation of CO₂ injection in deep geological formations was performed by Ukaegbu et al. (2009) using GEM-CMG software. The main focus of this study was on evaluating the effect of various reservoir conditions on the CO₂ distribution between different phases and the effect of gas relative permeability curve hysteresis, residual water, and gas saturation on the amount of CO₂ dissolved in brine. They concluded that hysteresis in the gas relative permeability curve affects the amount of CO₂ dissolved. In addition, they found that increasing the residual gas saturation leads to increasing the amount of immobile CO₂ while increasing irreducible water saturation leads to increasing the amount of CO₂ dissolved.

Ide et al. (2007) used the Eclipse simulator to simulate CO₂ geo-sequestration in deep saline aquifers. They studied the effect of capillary forces and capillary pressure on CO₂ storage capacity. This study concluded that including the capillary pressure in the reservoir simulation model leads to increasing the CO₂ trapping capacity.

Mo and Akervoll (2005) presented a reservoir simulation model using black-oil reservoir software to simulate long-term CO₂ sequestration in a shallow saline aquifer. This study focused on the effect of wettability-dependent parameters (i.e., capillary

pressure, residual gas saturation, and irreducible water saturation) on CO₂ distribution in reservoirs. This study indicated that both capillary pressure and irreducible water saturation have a small effect on CO₂ distribution. However, they concluded that the capillary trapping capacity is highly affected by residual gas saturation.

Doughty (2007) developed a numerical simulation model to simulate a CO₂ geo-sequestration process using the TOUGH2 simulator to study the effect of using hysteretic curves on the residual CO₂ trapping. She found that the use of hysteretic curves has a significant importance in increasing the accuracy of capturing CO₂ plume behaviour.

Kumar et al. (2005) simulated a CO₂ storage process in a saline aquifer using CMG-GEM to study the effect of residual gas saturation, as well as other different parameters (e.g., reservoir permeability, the ratio of vertical to horizontal permeability, aquifer temperature, permeability heterogeneity, mineralisation, and aquifer dip angle) on CO₂ storage. They found that residual gas saturation affects the CO₂ storage process and concluded that increasing residual gas saturation leads to decreasing the gas dissolution in water and gas migration while leading to an increase in the amount of residually trapped CO₂.

Sifuentes et al. (2009) presented a sensitivity analysis study using a reservoir simulation model to study the effect of residual gas saturation in addition to other physical properties (e.g., horizontal permeability and permeability heterogeneity) on the CO₂ storage efficiency in aquifers. The simulations were performed using the compositional simulator (Eclipse software). Sifuentes et al. (2009) showed that residual gas saturation affects the residual trapping capacity and that increasing residual gas saturation leads to increasing the amount of residually trapped CO₂.

Juanes et al. (2006) presented a reservoir simulation model to evaluate the effect of relative permeability hysteresis on the CO₂ storage process using the Eclipse 100 simulator. They found that the use of relative permeability hysteresis curves increases the accuracy of CO₂ geo-sequestration simulation outcomes.

Chalbaud et al. (2007) studied the effect of wettability on the CO₂ storage in aquifers by using micromodel experiments in a laboratory and a pore network model as a theoretical part. They simulated the CO₂ storage in a core scale without involving the actual reservoir heterogeneities.

Krevor et al. (2015) presented a simulation model to study the effect of wettability on the capillary trapping efficiency. They compared only two wettability conditions: water-wet and mixed-wet conditions. They found that the water-wet rocks have a higher capacity of capillary trapping compared to mixed-wet rocks.

Iglauer et al. (2015) presented a comprehensive review of the measurements of CO₂ wettability and its importance in CO₂ geo-sequestration. This study provided an example of the reservoir simulation model to study the effect of wettability CO₂ trapping mechanisms by using the MoRes simulator. They simulated only two wettability scenarios: water-wet and CO₂-wet. They found that wettability affects the amount of mobile and trapped CO₂.

In this thesis, the effect of CO₂-rock-brine wettability on CO₂ storage efficiency has been investigated by performing multiphase flow reservoir simulations on hectometre-scale reservoirs. Here, and for the first time, five wettability conditions (i.e., strongly CO₂ wet, weakly CO₂-wet, intermediate-wet, weakly water-wet, and strongly water-wet) have been simulated by developing five sets of relative permeability and capillary pressure curves (including hysteresis). The wettability effect has been investigated for both homogeneous and heterogeneous reservoirs and for different injection configurations (horizontal and vertical injection well). Further, and again for the first time, the impact of wettability heterogeneity on the effectiveness of CO₂ storage process has been demonstrated. The effect of other important parameters such as reservoir salinity, reservoir temperature, CO₂ injection scenario, injection well configuration on CO₂ geo-sequestration efficiency has also been studied.

Chapter 3 Influence of CO₂-Wettability on CO₂ Migration and Trapping Capacity in Deep Saline Aquifers

3.1 Introduction

CO₂ capture and storage (CCS) is considered an one effective method of mitigating greenhouse gas (GHG) emissions into the atmosphere by collecting CO₂ from large point sources and injecting it into deep geological formations (Pruess et al., 2003; Holloway, 2005). Storage sites, however, need to be screened for storage effectiveness, with prospective targets including unminable coal beds, deep saline aquifers and depleted hydrocarbon reservoirs (Metz et al., 2005; Iglauer, 2012; Juanes et al., 2006; Bachu, 2000; Bickle, 2009; Sakurovs and Lavrencic, 2011). Among all these types of CO₂ storage sites, deep saline aquifers are considered more suitable, because they have the largest CO₂ storage capacity and the widest geographical spread (Lackner, 2003; Bachu, 2000). For safety and efficiency reasons, CO₂ is injected at depths greater than 800 m, so that CO₂ remains in a supercritical (sc) state (Pentland et al., 2011; Hesse et al., 2008; Metz et al., 2005; Pruess et al., 2003). However, scCO₂ - although denser than CO₂ gas - is buoyant compared to formation water and migrates upwards; this migration can be minimized and CO₂ can be prevented from escaping to the atmosphere by four main trapping mechanisms: structural (Hesse et al., 2008; Naylor et al., 2011; Iglauer et al., 2015a), residual (Kumar et al., 2005; Pentland et al., 2011; Iglauer et al., 2011), dissolution (Lindeberg and Wessel-Berg, 1997; Spycher et al., 2003; Iglauer, 2011), and mineral (Bachu et al., 1994b; Xu et al., 2004; Gaus, 2010).

Many factors affect the efficiency and capacity of the main CO₂ trapping mechanisms: temperature (Ofori and Engler, 2011), vertical to horizontal permeability ratio (Basbug et al., 2005), cap rock properties (Iglauer et al., 2015a),¹³ fault-seal behaviour (Bretan et al., 2011; Yielding et al., 2011), or reservoir heterogeneity (Gershenson et al., 2015). One factor, which has received little attention, is the CO₂-wettability of the rock (Iglauer et al., 2015b; Vialle et al., 2016); we show here that CO₂-wettability has a dramatic impact on storage capacities, CO₂ plume migration patterns and CO₂ containment security. It is also important to note that CO₂-rock wettability can vary tremendously. Indeed, water contact angles between 0° (strongly water-wet) and 170°

(strongly CO₂-wet) have been measured, where CO₂-wettability mainly depends on the surface chemistry, and to a lesser extent on temperature, pressure and brine composition (Chiquet et al., 2007; Broseta et al., 2012; Iglauer et al., 2012; McCaughan et al., 2013; Saraji et al., 2013; Iglauer et al., 2014; Iglauer et al., 2015b; Iglauer et al., 2015a; Chen et al., 2015a; Javanbakht et al., 2015; Sarmadivaleh et al., 2015; Al-Yaseri et al., 2016; Arif et al., 2016b).

Wettability, as it has been previously shown in laboratory experiments (at the mm to cm scale), has a significant effect on residual trapping (Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016) and structural trapping (Naylor et al., 2011; Iglauer et al., 2015a). However, despite this laboratory-scale evidence, the effect of wettability on reservoir scale processes and associated storage capacity and containment security predictions has received little attention (Iglauer et al., 2015b) and generally, though the wettability is incorporated in the pilot projects modelling via relative permeability curves and multiphase flow, the values are poorly constrained.

In this chapter, we investigate the influence of CO₂ wettability of rocks on CO₂ plume migration, CO₂ mobility and the capacity of residual and solubility trapping and demonstrate its key importance. For this purpose we performed multiphase flow reservoir simulations on a homogeneous hectometre scale formation using five relative permeability curves (including hysteresis) that represent five characteristic wettability scenarios from strongly water-wet to strongly CO₂-wet.

3.2 Methodology

3.2.1 Numerical model

We built a 3D homogeneous reservoir-scale model (Figure 3-1) using the nonisothermal multicomponent multiphase flow simulator TOUGH2 (Pruess et al., 1999) with the fluid property module ECO2M to model the thermodynamic and thermophysical properties of the H₂O-NaCl-CO₂ mixtures that includes super- and sub-critical conditions, as well as phase changes between liquid and gaseous CO₂ (Pruess, 2011). ECO2M is a tabular EOS and it depends on Altunin's correlations (Altunin, 1975) to compute the molar volumes of CO₂ (including the CO₂ dissolved

in brine). The amount of dissolved CO₂ was used to assess dissolution trapping (Lindeberg and Wessel-Berg, 1997; Spycher et al., 2003; Iglauer, 2011). The aquifer characteristics are summarized in Table 3-1 (Jahangiri and Zhang, 2011; Basbug et al., 2005); the top of the aquifer is at 800 m depth. It is overlaid by a lower permeability unit (10⁻⁶ D compared to 1 D for that of the reservoir) typical of mudstones (Dewhurst et al., 1999; Yang and Aplin, 2007), and is modelled as a water-wet formation, for all wettability scenarios considered in this chapter (including the CO₂-wet models): thus this unit constitutes a barrier preventing the CO₂ from migrating to the surface. Reservoir pressure and temperature were set to 8 MPa (at 800 m depth) and 313 K (40^o C), respectively. The pressure followed the hydrostatic gradient (10 MPa/km) (Dake, 2007), while temperature conditions were isothermal. Dirichlet boundary conditions for pressure (i.e. constant pressure) were assigned on the outer boundary grid cells by applying a large volume multiplier (10000) (Nghiem et al., 2009).

CO₂ was injected into the reservoir at a constant rate of 3.171 kg/sec (100,000 tCO₂/yr) for all modelled scenarios; this is an injection rate similar to that of the Ordos CCS demonstration project in China (Xiuzhang, 2014) and Tomakomai CCS demonstration project in Japan (Tanaka et al., 2014). CO₂ was injected at a depth of 1150m (i.e. near the bottom of the reservoir and at the centre of the model) over a 1 year period (i.e. a total of 100000 tons of CO₂ were injected). Subsequently the CO₂ injection well was shut down and the behaviour of the CO₂ plume was simulated for the following 10 year period (“storage period”). Five different wettability scenarios were analysed, namely strongly water-wet, weakly water-wet, intermediate-wet, weakly CO₂-wet and strongly CO₂-wet with an assumed contact angle (θ) of (0^o, 70^o, 110^o, 130^o, and 170^o), respectively (Iglauer et al., 2015b; Arif et al., 2016b). Note that all wettability states are physically possible and they may prevail in a specific storage reservoir, as mentioned in the introduction.

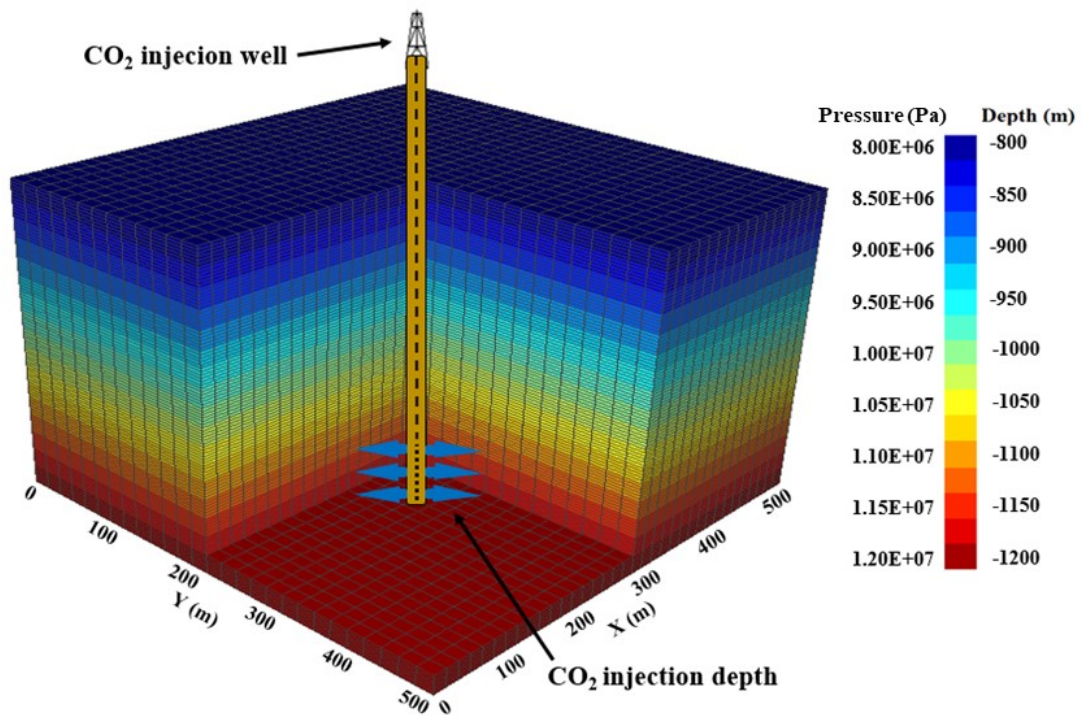


Figure 3-1 A sketch of the 3-D model including location of the injection well, model dimensions, and initial pressure.

Table 3-1 Reservoir model characteristics

Property	Value
Length	500 m
Width	500 m
Thickness	400 m
Cell number	35 x 27 x 100 (94500 cells in total)
Top depth of the reservoir	-800 m
Bottom depth of the reservoir	-1200 m
Reservoir temperature (isothermal)	313 K(40 ⁰ C)
Initial pressure (at depth -800 m)	8 MPa
Initial pressure (at depth -1200 m)	12 MPa
Salinity	15% NaCl by weight
Horizontal permeability	1000 mD

Vertical to horizontal permeability ratio	0.1
Top seal permeability	10^{-3} mD
Porosity	0.25
Initial water saturation	100%

3.2.2 Wettability simulation

CO₂-wettability is an atomistic phenomenon as it is determined by intermolecular forces between CO₂, brine and rock (Liu et al., 2010; Iglauer et al., 2012; McCaughan et al., 2013; Javanbakht et al., 2015; Chen et al., 2015a). Thus, to perform hectometre-scale reservoir simulations, an upscaling mechanism is required; here we directly implement the effect of wettability into the relative permeability and capillary pressure curves (Anderson, 1987b; McCaffery and Bennion, 1974; Levine et al., 2014; Krevor et al., 2012; Anderson, 1987a; Morrow, 1976).

Specifically, we use McCaffery and Bennion's (1974) relative permeability curves to construct the curves for the 5 wettability scenarios investigated in this study and adjusted them based on Craig's criteria (Craig, 1993). This procedure consists in an adjustment of the values of end point saturations, water saturations where CO₂ and water relative permeabilities (k_{rg} and k_{rw}) are equal, and relative permeabilities at water floodout (i.e. the condition when the rock reached its maximum water saturation). According to Craig's criteria, and for a reservoir permeability of 1000 mD, the residual water saturation (S_{wr}) should be less than 15% in strongly CO₂-wet rocks, and should range between 20%-50% in strongly water-wet rocks. In addition, the water saturation where k_{rg} and k_{rw} are equal should be higher than 50% in a strongly water-wet reservoir, while it should be less than 50% in a strongly CO₂-wet reservoir. Moreover, the k_{rw} during the storage period should be less than 30% in the strongly water-wet formation, and it should range between 50%-100% in the strongly CO₂-wet reservoir. All above conditions were applied during construction of the relative permeability curves used in this study, which are displayed in (Figure 3-2). The curves were then fitted with the Van Genuchten-Mualem model (Van Genuchten, 1980; Mualem, 1976) for implementation into the computer code:

$$k_{rw} = \sqrt{S^*} \left\{ 1 - \left(1 - [S^*]^{1/\lambda} \right)^\lambda \right\}^2 \quad \text{if } S_w < S_{ws} \quad (3.1)$$

$$k_{rw} = 1 \quad \text{if } S_w = 1 \quad (3.2)$$

$$k_{rg} = 1 - k_{rw} \quad \text{if } S_{gr} = 0 \quad (3.3)$$

$$k_{rg} = (1 - \hat{S})^2 (1 - \hat{S}^2) \quad \text{if } S_{gr} > 0 \quad (3.4)$$

$$S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}), \quad (3.5)$$

$$\hat{S} = (S_w - S_{wr}) / (1 - S_{wr} - S_{gr})$$

where:

k_{rg} = relative permeability for gas, k_{rw} = relative permeability for water

S_{gr} = residual gas saturation, S_w = water saturation

S_{ws} = saturated (maximum) water saturation (= 1), S_{wr} = residual water saturation.

λ = fitting parameter (pore size distribution index).

Initially, k_{rw} is set to 1 and k_{rg} to 0, which corresponds to full (100%) water saturation. During CO₂ injection (dashed black lines in Figure 3-2), k_{rw} reduces gradually, while k_{rg} increases until it reaches a maximum at the irreducible water saturation (S_{wr}). During the storage period (CO₂ injection has ceased, represented by red lines in Figure 3-2), k_{rg} reduces and k_{rw} increases until residual gas saturation (S_{gr}) is reached. Note that the endpoint saturations (S_{wr} , S_{gr}) depend on wettability (Anderson, 1987b; Craig, 1993; Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016). Moreover, lower water-wettability shifts the k_{rw} curve upwards, and the k_{rw} - k_{rg} cross-over point moves towards the left (i.e. to a lower water saturation value) (Craig, 1993). Furthermore, note that S_{gr} is also a function of the initial CO₂ saturation (Pentland et al., 2011; Krevor et al., 2011; Krevor et al., 2015) and porosity (Iglauer et al., 2011).

Figure 3-3 presents the capillary pressure curves used in this study for the five different wettability scenarios. These curves has been developed by referring on previous studies (Anderson, 1987a; Morrow, 1976), (which look at the wettability-capillary pressure relationship) and by using the Van Genuchten-Mualem model (Van Genuchten, 1980; Mualem, 1976):

$$P_{cap} = P_0 ([S^*]^{-1/\lambda} - 1)^{1-\lambda} \quad (3.6)$$

$$S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}) \quad (3.7)$$

where:

P_{cap} = CO₂-water capillary pressure, P_o = capillary pressure scaling factor,

S_{ws} = maximum (saturated) water saturation, S_{wr} = residual water saturation,

λ = pore size distribution index.

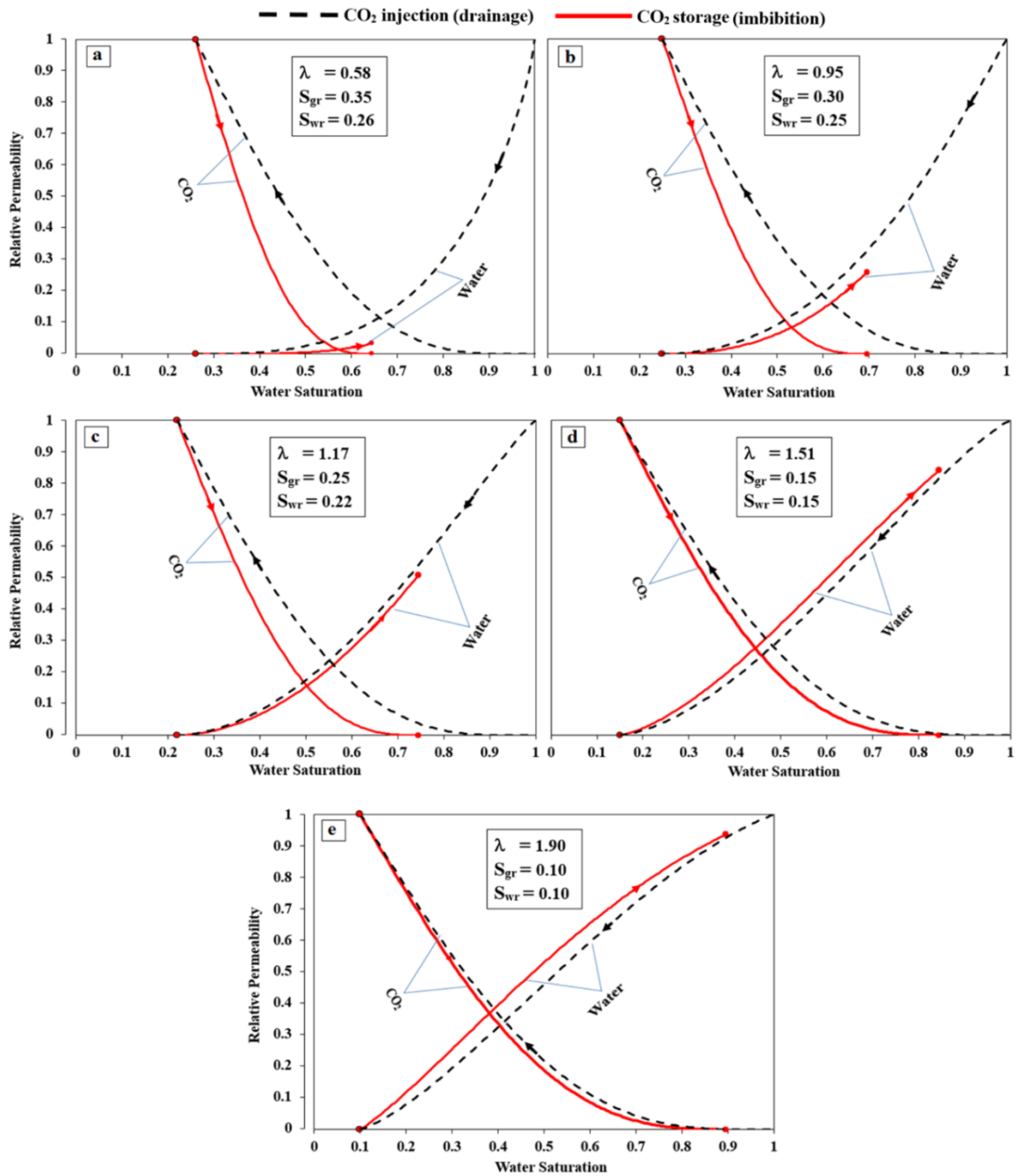


Figure 3-2 Relative permeability curves for the five different rock wettabilities investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. See section 3.2.2 for the construction of these curves.

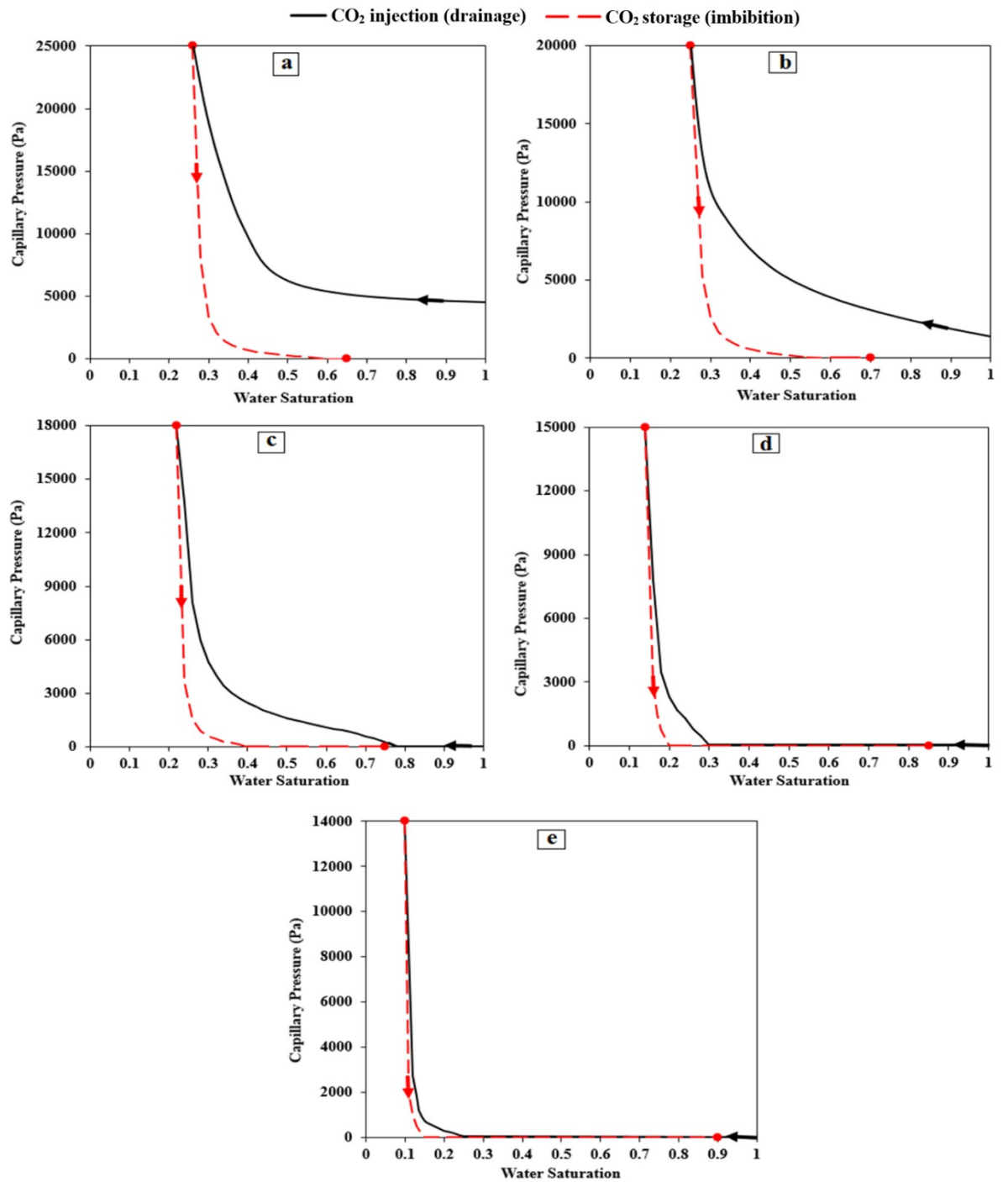


Figure 3-3 Capillary pressure curves for the five different rock wettabilities investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. See section 3.2.2 for the construction of these curves.

3.3 Results and discussion

3.3.1 Influence of wettability on CO₂ plume migration and shape

Clearly wettability has a significant influence on the CO₂ migration pattern. (Figure 3-4) displays 3D views of the CO₂ plume for the five wettability scenarios described above and at various times after the end of the injection. The CO₂ plume moves upwards much more rapidly in the strongly CO₂-wet rock scenario, while it is best retained near the injection well in the strongly water-wet reservoir; this is quantified in Table 3-2, where the depth reached by the (free) CO₂ plume, as well as the total CO₂ vertical migration distance, are reported. Furthermore, rock wettability has a drastic impact on the shape of the CO₂ plume; while the plume is much more compact and “raindrop-like” in a water-wet reservoir, it has a (vertically) elongated “candle-like” shape in the CO₂-wet reservoir (Figure 3-4).

The underlying reason why the CO₂ plume moves upwards much more rapidly in strongly CO₂-wet rock is because of the wettability influence on relative permeability and capillary pressure on dual phase systems. Wettability is also the reason why the spreading patterns are different in CO₂-wet and water-wet reservoirs. Recall that the endpoint saturations (S_{wr} , S_{gr}) depend on wettability (Anderson, 1987b; Craig, 1993; Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016). A CO₂ migration and plume extension are highly dependent on S_{gr} (Metz et al., 2005; Kumar et al., 2005; Doughty, 2010), CO₂ migration rate increases and the plume expands spatially with smaller S_{gr} values. (Figure 3-2), which shows the variation of S_{gr} with wettability, that the lowest S_{gr} (10%) is associated with the strongly CO₂-wet condition, while the highest S_{gr} (35%) is found in the strongly water-wet reservoir. We conclude that CO₂-wettability dramatically affects CO₂ plume migration both in time and space.

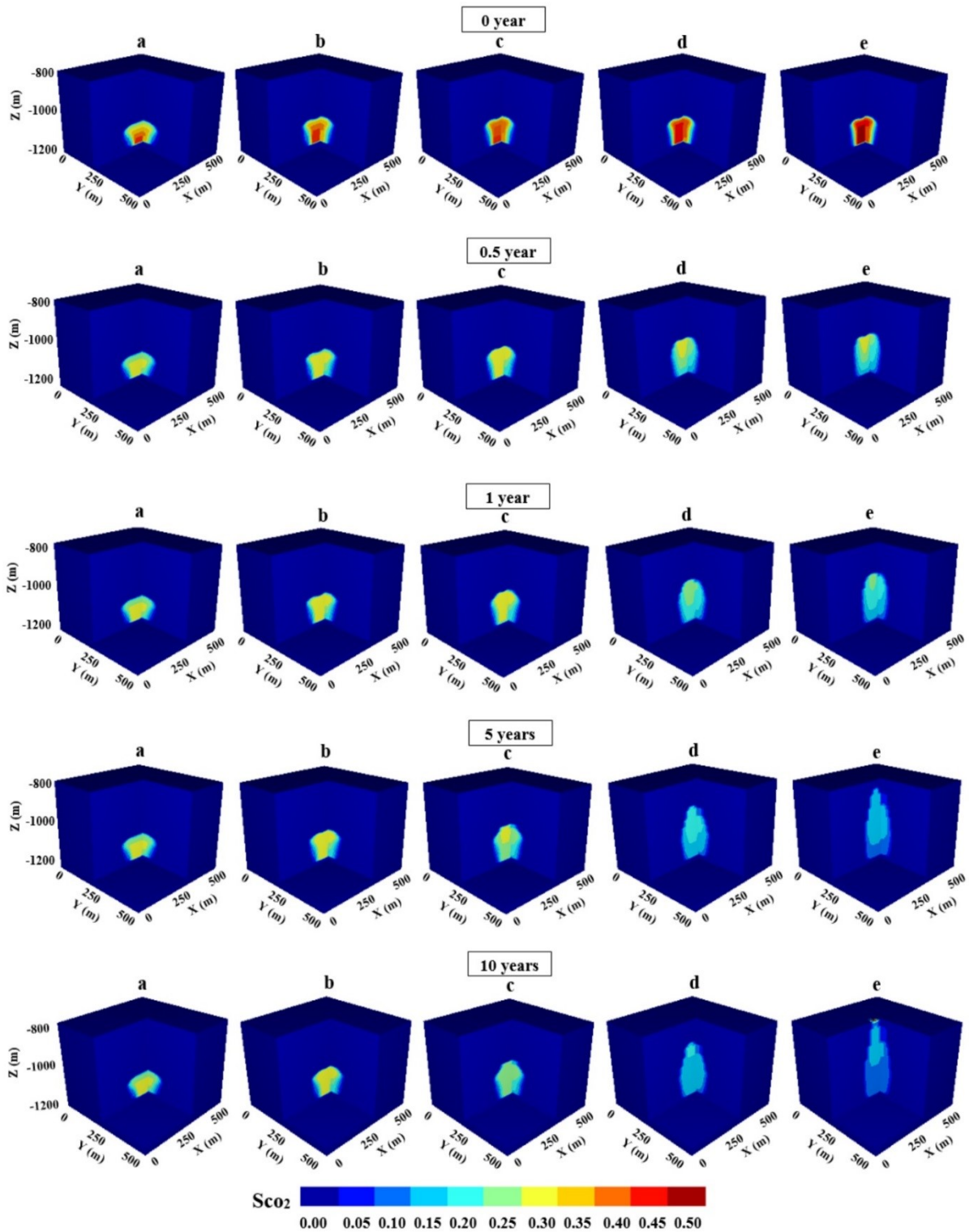


Figure 3-4 3D views of the CO₂ plume as a function of storage time (i.e times since the stop of injection) and wettability: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. Z= model height; X, Y= model length and width.

Table 3-2 Depth^a reached and vertical migration distance of CO₂ plume after the end of the storage period (10 years) for the five different wettability scenarios investigated

Wettability	Depth reached by the free CO₂ plume (m)	Vertical migration distance of the free CO₂ plume (m)^b
Strongly water-wet	-1034	116
Weakly water-wet	-1014	136
Intermediate-wet	-978	172
Weakly CO ₂ -wet	-889	261
Strongly CO ₂ -wet	-800 ^c	350

^aCO₂ injection depth was at (-1150 m).

^bCO₂ plume migration distance measured from the top of the perforated interval.

^cIn the case of the strongly CO₂-wet reservoir, CO₂ reached the top of the model (-800 m) after only 8 years; CO₂ then flowed laterally beneath the caprock.

3.3.2 Influence of wettability on storage mechanisms

We further analyse the impact of wettability on the CO₂ storage mechanisms by distinguishing (at the end of the 10 year storage period) and quantifying the amount of dissolved (in brine) CO₂, mobile CO₂ and residual CO₂. Residual CO₂ is CO₂ that is entrapped in the pore space of the rock by capillary forces (Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016) and that has thus been immobilized.

The percentage of trapped CO₂, either by dissolution or by residual trapping and of mobile (free) CO₂ during the storage period is displayed in (Figure 3-5) for all wettability cases; corresponding percentages are reported in Table 3-3 for the end of the storage period (i.e. after 10 years). Generally for all wettability cases, the percentage of trapped CO₂, either by dissolution trapping or residual trapping, increased over time during the storage period, and consequently the amount of free scCO₂ decreased (Figure 3-5), which is consistent with previous studies (Iglauer et al.,

2015b; Kumar et al., 2005; Suekane et al., 2008; Qi et al., 2009; Gershenzon et al., 2015).

In the case of a strongly water-wet rock, most (99.5%) of the CO₂ is already trapped at the beginning of the storage period, either by dissolution (~18%) or by residual trapping (~81%). This is due to the relatively small amount of CO₂ injected and the high residual CO₂ saturation ($S_{gr} = 35\%$) in the strongly water-wet rock compared with lower residual CO₂ saturation in the less water-wet scenarios (reducing to $S_{gr} = 10\%$ in the strongly CO₂-wet case (see Figure 3-2)); thus the CO₂ plume in the water-wet rock appears stagnant in Figure 3-4 (changes in dissolved CO₂ are insignificant). With increasing CO₂-wettability the amount of residual CO₂ trapping dropped dramatically, from ~80% in case of strongly water-wet rock to ~50% in case of strongly CO₂-wet rock, 10 years after the injection has stopped. This is due to the lower capillary forces in CO₂-wet rock and the resulting lower S_{gr} (Chaudhary et al., 2013; Rahman et al., 2016). Recall that S_{gr} strongly depends on the wettability (Anderson, 1987b; Craig, 1993; Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016), see Figure 3-2. Consequently, residual trapping capacities strongly depend on wettability as wettability strongly impacts S_{gr} and therefore the overall residual CO₂ saturation.

On the contrary, dissolution trapping was more efficient in strongly CO₂-wet rock, with ~18 % for strongly water-wet rock compared to ~29% for strongly CO₂-wet rock after 10 years storage time. It is important to mention here that this difference in dissolution trapping between CO₂-wet and water-wet reservoirs is because the CO₂ plume moves faster through the CO₂-wet reservoir and spreads out more. The more the plume spreads, however, the larger the CO₂-brine interface becomes. And a larger CO₂-brine interface leads to more dissolution trapping (Metz et al., 2005; Kumar et al., 2005; Doughty, 2010).

Thus, our results show that there is a highly significant impact of wettability on the ratio of mobile to residual CO₂, and a significant effect on dissolution trapping.

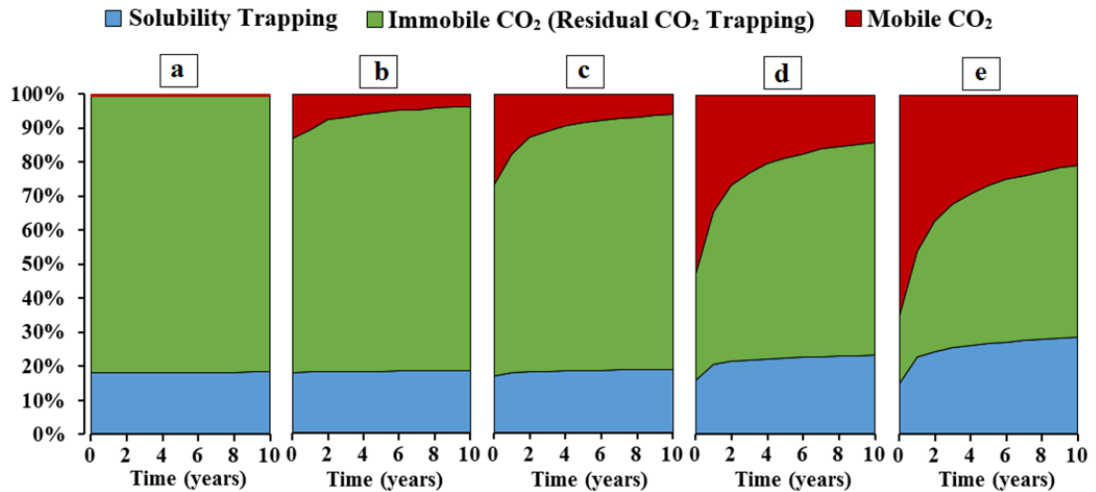


Figure 3-5 Percentage of free and trapped CO₂ for the five different rock wettabilities investigated (dissolution trapping is in blue, residual trapping is in green and mobile CO₂ is in red): a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. Time = 0 is the beginning of the storage period.

Table 3-3 Percentage of free and trapped CO₂ at the end of the storage period (10 years) for the five different wettabilities investigated

Wettability	Mobile CO ₂	Solubility trapped	Residually trapped
	%	CO ₂ %	CO ₂ %
Strongly water-wet	0.5	18.3	81.2
Weakly water-wet	3.9	18.5	77.6
Intermediate-wet	6.0	18.7	75.3
Weakly CO ₂ -wet	13.8	23.2	63.0
Strongly CO ₂ -wet	20.7	28.6	50.7

3.4 Conclusions

CO₂-wettability of rocks can vary tremendously, from strongly water-wet to strongly CO₂-wet (cp. the recent review by (Iglauer et al., 2015b) and the reference list in the introduction of this study). Such wettability variation has been previously shown, in laboratory experiments, to strongly influence residual trapping (Iglauer et al., 2011;

Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016) and structural trapping (Naylor et al., 2011; Iglauer et al., 2015a).

Here, for the first time, we systematically analysed the effect of rock wettability on the CO₂ plume behaviour in an idealized reservoir and computed the amount of mobile and trapped CO₂ (by both residual and solubility trapping mechanisms) at the hectometre-scale via reservoir simulations.

Our simulations clearly indicate that CO₂ is best retained in water-wet rock, while CO₂-wet reservoirs are relatively much more permeable to CO₂. Furthermore, the shape of the CO₂ plume is also strongly affected by wettability, the plume is much more compact in case of water-wet rock, while it is vertically elongated in CO₂-wet rock. Moreover, in our example case study over a 10 year storage period, the amount of residually trapped CO₂ is significantly higher in water-wet rock. On the contrary, dissolution trapping is more effective in CO₂-wet rock.

In summary, wettability significantly changes migration patterns and storage capacities, which is directly relevant to CO₂ geo-storage projects. Moreover, we conclude that strongly water-wet reservoirs are preferable CO₂ sinks due to their higher storage capacities and superior containment security. This result has important implications for designing geoengineering solutions aiming at increasing CO₂ storage especially in situations where an efficient and continuous seal is absent (e.g. the South West Hub project in Western Australia (Stalker et al., 2013)). Recent laboratory experiments indeed show that intermediate-wet and CO₂-wet reservoirs can be treated (e.g. with silica nanofluid (Al-Anssari et al., 2016)) to render them strongly water-wet.

Chapter 4 Impact of Reservoir Wettability and Heterogeneity on CO₂-Plume Migration and Trapping Capacity

4.1 Introduction

Capture and geological storage of Carbon dioxide (CCS) is considered one of the most promising technologies to mitigate greenhouse gas emissions into the atmosphere; CO₂ is captured from large CO₂ stationary sources and injected into deep geological formations (Pruess et al., 2003). There are three main sites for CO₂ storage, namely unminable coal beds, saline aquifers, and depleted hydrocarbon reservoirs (Metz et al., 2005). Among these storage sites, saline aquifers have the largest CO₂ storage capacity so they are considered the best storage sites (Lackner, 2003). However, economic feasibility and particularly safety of these CO₂ sequestration sites need to be insured (Metz et al., 2005; Gershenzon et al., 2015). Typically, CO₂ is injected in a dense supercritical (scCO₂) state, at a minimum aquifer depth of 800 m to maximise storage efficiency (Holloway and Savage, 1993; Pentland et al., 2011; Metz et al., 2005; Pruess et al., 2003). However, scCO₂ has a lower density than that of brine, and it will thus migrate upwards due to the buoyancy forces (Dai et al., 2014; Flett et al., 2007; Hassanzadeh et al., 2009). This CO₂ plume upward migration can be mitigated by four trapping mechanisms which have different timescales (i.e. different time lengths over which the storage mechanisms occurred). The first trapping mechanism is the structural trapping which is represented by the caprock (or seal) (Hesse et al., 2008; Naylor et al., 2011; Iglauer et al., 2015a; Gershenzon et al., 2015). The second trapping mechanism is the residual trapping mechanism (Kumar et al., 2005; Pentland et al., 2011; Iglauer et al., 2011). The third trapping mechanism is the dissolution trapping (Lindeberg and Wessel-Berg, 1997; Pruess and Garcia, 2002; Bachu and Adams, 2003; Spycher et al., 2003; Iglauer, 2011). This is a slow process (on the timescale of 100's to 1000's years) and it has been shown that it can be accelerated by convective dissolution or CO₂- brine mixing at the surface, wellbore, or in the subsurface (Emami-Meybodi et al., 2015). The last trapping mechanism is the mineral trapping (Bachu et al., 1994b; Xu et al., 2004; Gaus, 2010). The injected CO₂ can be permanently stored in the aquifer as the result of chemical reactions between the carbonic acid resulted from CO₂ dissolution in the brine process and the minerals

existing in the aquifer rocks [e.g. (Xu et al., 2003; Xu et al., 2005; Flett et al., 2007)]. In this study, we do not consider the mineral trapping mechanism, as it is a much slower process than the first three [e.g. (Xu et al., 2004; Metz et al., 2005; Gaus, 2010)]. Several studies have shown that geological heterogeneities (i.e. structural or stratigraphic heterogeneity, permeability and porosity distribution or differences in mineralogy) have a significant impact on the behaviour and vertical and horizontal movement of the CO₂ plume (Doughty and Pruess, 2004; Hovorka et al., 2004; Obi and Blunt, 2006; Bryant et al., 2006; Flett et al., 2007; Ide et al., 2007; Saadatpoor et al., 2009; Zhou et al., 2010; Hesse and Woods, 2010; Green and Ennis-King, 2010; Doughty, 2010; Krevor et al., 2015; Li and Benson, 2015) and consequently on the storage efficiency for the different storage mechanisms (Ambrose et al., 2008; Doughty, 2010; Han et al., 2010; Hesse and Woods, 2010; Green and Ennis-King, 2010; Gershenson et al., 2015; Krevor et al., 2016). However, one parameter, reservoir wettability, has received little attention in CO₂ storage modelling, although it is experimentally clear that it strongly affects structural (Iglauer et al., 2015a) and residual trapping (Chaudhary et al., 2013; Rahman et al., 2016; Al-Menhali et al., 2016). In addition, previous simulation works presented by Krevor et al. [2015] (Krevor et al., 2015) and Juanes et al. [2006] (Juanes et al., 2006) showed that wettability and hysteresis in the capillary curves do affect the efficiency of capillary trapping; they compared two wettability scenarios, mixed-wet rocks and water-wet rocks, and concluded that mix-wet rocks have a lower capillary trapping capacity compared to water-wet rocks. Furthermore, it has been shown that rock wettability strongly affects the relative permeability curves (Owens and Archer, 1971; McCaffery and Bennion, 1974; Heiba et al., 1983; Anderson, 1987b; Krevor et al., 2012; Levine et al., 2014), capillary pressure curves (Anderson, 1987a; Batycky et al., 1981; Heiba et al., 1983; Melrose, 1965; Morrow, 1976) and the phase distribution within the rocks pore space (Morrow, 1990) and so, the CO₂-water displacement mechanisms (Morrow, 1990).

These different wettability conditions are the result of various geological and chemical factors including surface chemistry (e.g. organic content (Iglauer et al., 2015a; Iglauer et al., 2015b)), reservoir pressure (pressure increase leads to reduction in water wettability (Al-Yaseri et al., 2016; Arif et al., 2016a; Arif et al., 2016b; Chiquet et al., 2007; Broseta et al., 2012)), reservoir temperature (a higher temperature results in

higher water-wettability on mica or quartz (Broseta et al., 2012; Al-Yaseri et al., 2016; Arif et al., 2016a; Arif et al., 2016b)), salinity and ion type (increasing salinity leads to increasing CO₂ wettability (Al-Yaseri et al., 2016; Arif et al., 2016b; Chiquet et al., 2007; Espinoza and Santamarina, 2010; Wang et al., 2012)). It is thus of key importance to quantify the effect of wettability on storage capacities.

Thus, in this chapter, we developed 3D reservoir simulation models to predict storage capacities (i.e. residual trapping and dissolution trapping), mobile CO₂ capacity and CO₂ plume migration patterns for five different wettability conditions (strongly water-wet, weakly water-wet, intermediate-wet, weakly CO₂-wet, and strongly CO₂-wet). Furthermore, we investigate how reservoir heterogeneity interacts with these processes by comparing a set of simulations using a homogeneous distribution of permeability and porosity values with a set of simulations using heterogeneous distributions.

4.2 Methodology

4.2.1 Reservoir model

We used the nonisothermal multicomponent multiphase flow simulator TOUGH2 (Pruess et al., 1999) for simulating CO₂ migration in a storage reservoir. The thermophysical properties of the H₂O-NaCl-CO₂ mixtures (including super- and sub-critical CO₂ conditions) and phase changes between liquid and gaseous CO₂ were modelled with the tabular equation of state (EOS) ECO2M (Pruess, 2011).

The aquifer dimensions were 400 m × 800 m with a thickness of 85 m, which was equivalent to 24 × 48 × 85 cells (97920 cells in total). The pressure at 1500 m depth (the top of the aquifer) was 15 MPa and a hydrostatic pressure gradient of 10 MPa/km was applied (Dake, 2007) [Dake, 2007], while temperature was kept constant at 333 K (i.e. isothermal conditions were applied). A constant pressure boundary condition (i.e. Dirichlet boundary conditions) was applied by multiplying the volume of the outer boundary cells, on the sides and bottom of the reservoir, by a large factor of 10⁸ (Mo et al., 2005; Nghiem et al., 2009). The aquifer was initially fully saturated with water ($S_{wi} = 100\%$) with a salinity of 15% (by weight). Porosity and permeability were kept constant throughout the reservoir for the set of simulations modelling a homogeneous

scenario. Porosity was set at 0.17. Anisotropy in permeability was introduced by taking a vertical-to-horizontal ratio of 0.1, with a value of 355 mD in the horizontal direction. The top of the reservoir (at 1500 m depth) was overlaid by a seal: it was simulated by reducing the ratio of vertical-to-horizontal permeability from 0.1 in the storage reservoir to 0.0001 at the top boundary of the reservoir in order to create a barrier preventing the CO₂ from leaking out of the reservoir. This model is called 'homogeneous model' in the following paragraphs.

In order to assess the impact on aquifer heterogeneity, we used the porosity and permeability data from the tenth SPE comparative solution project (Christie and Blunt, 2001), (Figure 4-1 and Figure 4-2). The values of porosity and horizontal permeability used in the homogeneous model are average values of these SPE data values. This model is called 'heterogeneous model' in the following paragraphs.

Five different wettability scenarios (see section 4.2.2 for implementation details) were then evaluated both for the homogenous and heterogeneous models.

Furthermore, we tested the impact of CO₂ injection rate by using five different injection rates of 10000 t/year, 15000 t/year, 20000 t/year, 25000 t/year and 30000 t/year; these injection rates were similar to those found in various CO₂ pilot projects: (e.g. 30000 t/year in the CO₂SINK in Ketzin, Germany) (Forster et al., 2006; Prevedel et al., 2009), 7000 ton/ year and 15000 t/year in the Nagaoka pilot project in Japan (Mito et al., 2008; Xue et al., 2006)).

ScCO₂ was injected at a depth of 1570 m at the centre of the model (in the XY plane) (Figure 4-1). The injection period (1 year) was followed by a 10 year shut-off period (representing the storage period).

The CO₂ migration behaviour and the amount of residually and solubility trapped CO₂ were computed as a function of the amount of injected CO₂, storage time, reservoir heterogeneity, and most importantly, reservoir wettability.

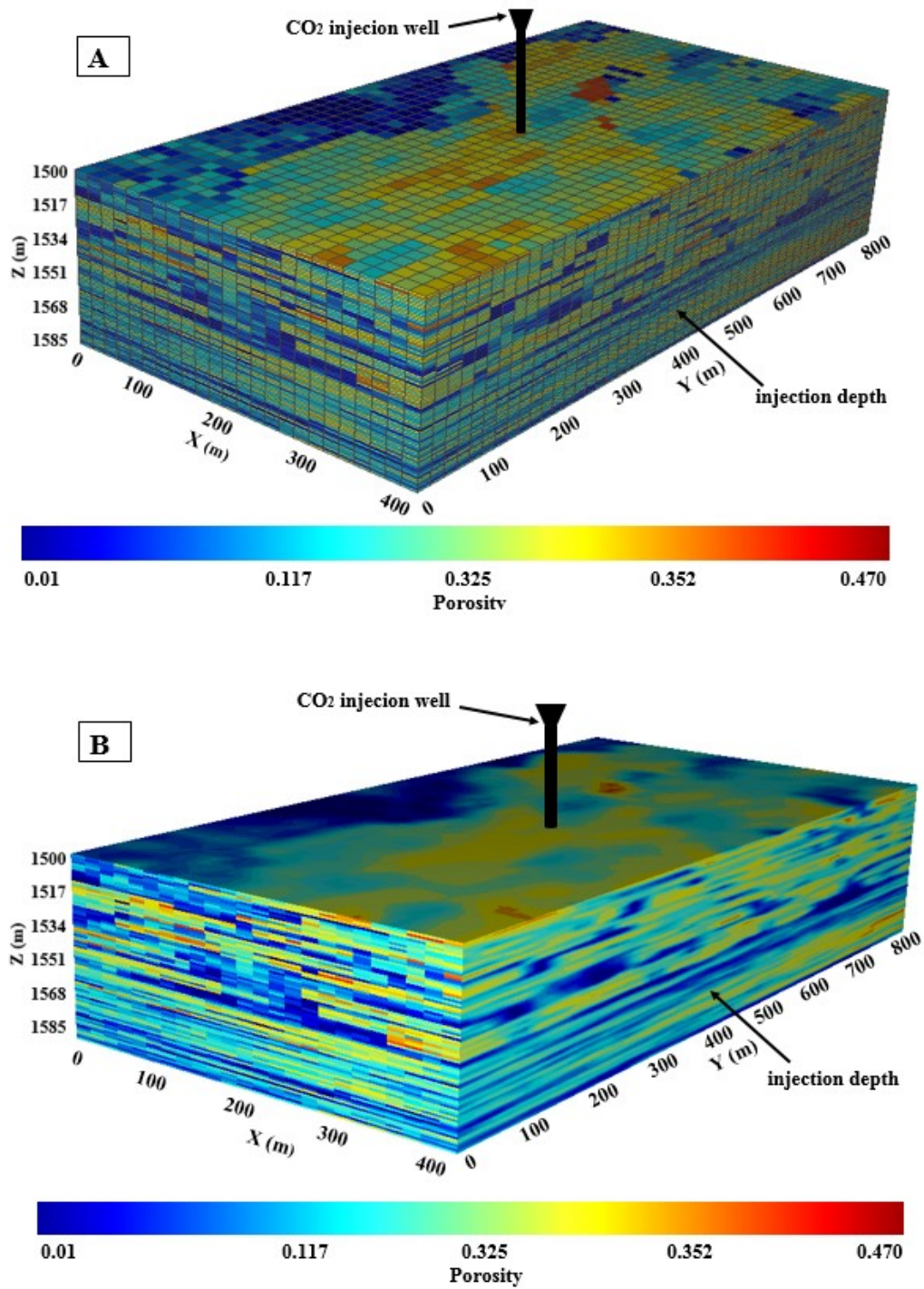


Figure 4-1 Representation of the 3-D model showing the aquifer porosity distribution and including the location of the injection well and model dimensions: A) with grid blocks view; B) compact view without grid blocks.

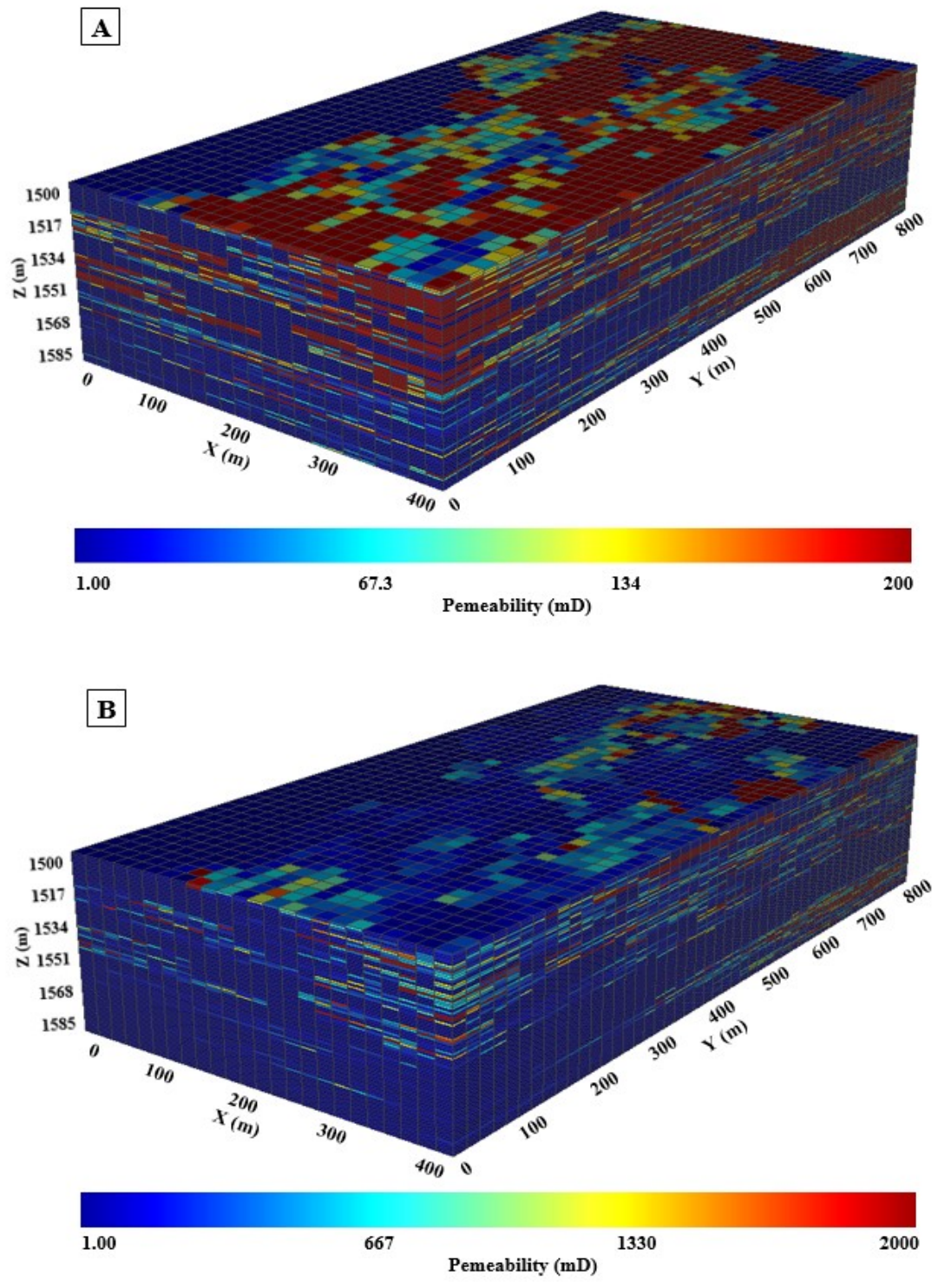


Figure 4-2 Representation of the 3-D model showing the aquifer horizontal permeability distribution for two different maximum permeability scale for permeability visualizing: A) for maximum permeability = 200 mD; B) for maximum permeability= 2000 mD.

4.2.2 Implementation of Reservoir Wettability

In order to simulate the different wettability conditions, we used the corresponding characteristics capillary pressure and relative permeability curves for each wettability condition from strongly water-wet to strongly CO₂-wet. Indeed, it has been shown that the outcome of CO₂ storage simulation results is highly affected by the choice of the characteristics curves (Doughty, 2007; Juanes et al., 2006). In this study, these curves were constructed based on earlier work on relative permeability (Owens and Archer, 1971; McCaffery and Bennion, 1974; Heiba et al., 1983; Anderson, 1987b; Krevor et al., 2012; Levine et al., 2014), capillary pressure (Anderson, 1987a; Batycky et al., 1981; Heiba et al., 1983; Melrose, 1965; Morrow, 1976) and associated residual water- and CO₂-saturations (Anderson, 1987b; Craig, 1993; Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Pentland et al., 2011; Rahman et al., 2016). Importantly, both residual water saturation and residual gas saturation (S_{wr} and S_{gr} , respectively) depend on reservoir wettability (Anderson, 1987a; Anderson, 1987b; Craig, 1993; Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016).

Specifically, for the construction of the relative permeability curves, we used McCaffery and Bennion's data (1974). We used Didger software (Golden Software Inc, 2013, Colorado) to digitize these curves and obtain data for the relative permeability values as a function of water saturation; we adjusted these curves using Craig's rules of thumb (Craig, 1993) (Table 4-1). Hence we took S_{wr} less than 15% for CO₂-wet rocks and more than 25 % for water-wet rocks. Furthermore, the water saturation (S_w) at which the relative permeability of CO₂ (k_{rg}) and that of water (k_{rw}) are equal was taken greater than 50% for strongly water-wet systems and lower than 50% for strongly CO₂-wet systems. Finally, during the storage period, k_{rw} at the maximum water saturation must be less than 30% in the strongly water-wet rocks, while it must be between 50% and 100% in the strongly CO₂-wet rocks. These curves were then fitted using the Van Genuchten-Mualem model described by equations 1 to 5 (Van Genuchten, 1980; Mualem, 1976) and fed into the simulator.

$$k_{rw} = \sqrt{S^*} \left\{ 1 - \left(1 - [S^*]^{1/\lambda} \right)^\lambda \right\}^2 \quad \text{if } S_w < S_{ws} \quad (4.1)$$

$$k_{rw} = 1 \quad \text{if } S_w \geq S_{ws} \quad (4.2)$$

$$k_{rg} = 1 - k_{rw} \quad \text{if } S_{gr} = 0 \quad (4.3)$$

$$k_{rg} = (1 - \hat{S})^2 (1 - \hat{S}^2) \quad \text{if } S_{gr} > 0 \quad (4.4)$$

$$\begin{aligned} S^* &= (S_w - S_{wr}) / (S_{ws} - S_{wr}), \\ \hat{S} &= (S_w - S_{wr}) / (1 - S_{wr} - S_{gr}) \end{aligned} \quad (4.5)$$

with the restriction $0 \leq k_{rw}, k_{rg} \leq 1$

where:

k_{rg} = gas relative permeability, k_{rw} = water relative permeability,

S_{gr} = residual saturation for gas, S_w = water saturation,

S_{ws} = maximum water saturation (= 1), S_{wr} = residual saturation for water,

λ = pore size distribution index (fitting parameter).

Initially the aquifer is fully saturated with brine ($S_w = 1$), thus $k_{rg} = 0$ and $k_{rw} = 1$ (right upper corner in the five plots of Figure 4-3). During CO₂ injection, S_w decreases from 1 to S_{wr} (black lines of Figure 4-3), and k_{rw} decreases steadily until reaching zero at S_{wr} , while k_{rg} increases until its maximum is reached at S_{wr} . During the storage period (which corresponds to water imbibition, represented by the red lines in Figure 4-3), k_{rg} will reduce again gradually to zero (residual CO₂ trapping), and k_{rw} increases again until it reaches its secondary maximum at S_{gr} .

The five capillary pressure curves used in this study (which correspond to the five wettability scenarios) are plotted in Figure 4-4. The curves for the strongly water-wet rock are based on the capillary pressure measurements on Berea sandstone provided by Pini et al. [2012, 2013] (Pini et al., 2012; Pini et al., 2013). The other curves were then adjusted according to previous studies of capillary pressure curves that studied the effect of wettability on the capillary pressure curves (i.e. (Anderson, 1987a; Batycky et al., 1981; Heiba et al., 1983; Melrose, 1965; Morrow, 1976)), as explained above. Table 4-2 presents the data used for the construction of capillary pressure curves for the different wettabilities.

The positive parts of these curves were implemented into the simulations, according to the current capabilities of the software, via the Van Genuchten-Mualem model (Van Genuchten, 1980; Mualem, 1976):

$$P_{cap} = P_0 ([S^*]^{-1/\lambda} - 1)^{1-\lambda} \quad (4.6)$$

with the restriction $0 \leq P_{cap} \leq P_{max}$

$$\text{and, } S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}) \quad (4.7)$$

where:

P_{cap} = CO₂-water capillary pressure, P_o = capillary pressure scaling factor,

S_{ws} = maximum (saturated) water saturation, S_{wr} = residual water saturation,

λ = pore size distribution index.

The Leverett J-function (Leverett, 1941) was then used to scale P_c with the spatially varying porosity and permeability values:

$$J(S_w) = \frac{P_c}{\sigma \cos \theta} \sqrt{\frac{k}{\phi}} \quad (4.8)$$

where:

J = dimensionless capillary pressure, S_w = water saturation, P_c = capillary pressure,

k = permeability, ϕ = porosity, σ = surface tension, θ = contact angle.

Table 4-1 Parameters used for implementing the relative permeability curves into the model

Wettability scenario	CO ₂ injection process			CO ₂ storage process		
	S_{gr}	S_{wr}	λ	S_{gr}	S_{wr}	λ
Strongly water-wet	0	0.26	0.78	0.35	0.26	0.58
Weakly water-wet	0	0.25	1.05	0.30	0.25	0.95
Intermediate-wet	0	0.22	1.22	0.25	0.22	1.17
Weakly CO ₂ -wet	0	0.15	1.41	0.15	0.15	1.51
Strongly CO ₂ -wet	0	0.1	1.7	0.1	0.1	1.9

Table 4-2 Parameters used for implementing the capillary pressure curves for the five different wettability scenarios into the TOUGH2 simulator

Wettability	CO ₂ injection process				CO ₂ storage process			
	S _{wr}	λ	P ₀ [Pa]	P _{max} [Pa]	S _{wr}	λ	P ₀ [Pa]	P _{max} [Pa]
Strongly water-wet	0.259	0.7	1500	25000	0.259	0.51	1000	25000
Weakly water-wet	0.249	0.7	1500	20000	0.249	0.51	1000	20000
Intermediate-wet	0.219	0.7	1500	18000	0.219	0.51	1000	18000
Weakly CO ₂ -wet	0.149	0.7	1500	15000	0.149	0.51	1000	15000
Strongly CO ₂ -wet	0.099	0.7	1500	14000	0.099	0.51	1000	14000

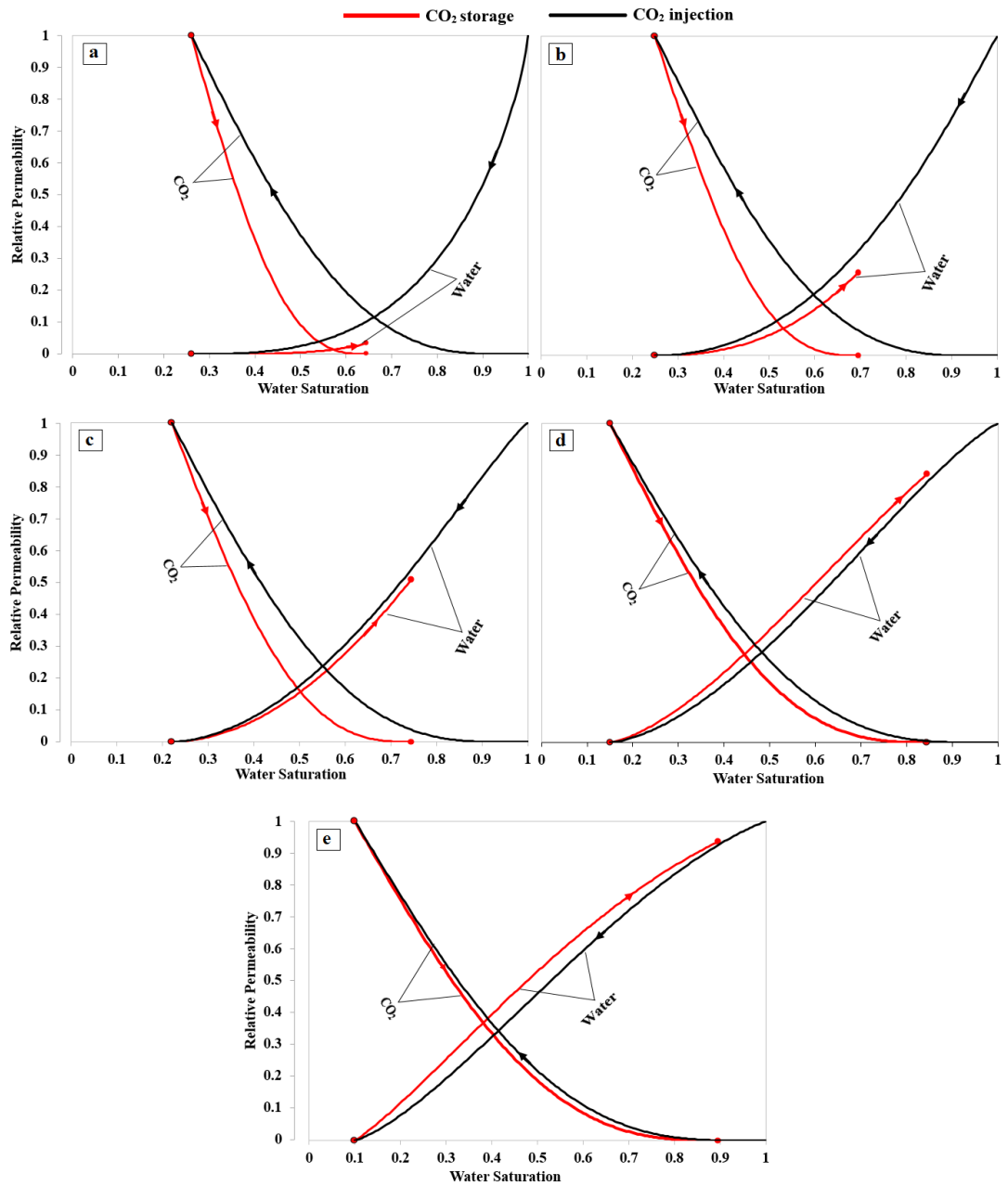


Figure 4-3 Relative permeability curves for the five different rock wettabilities investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet.

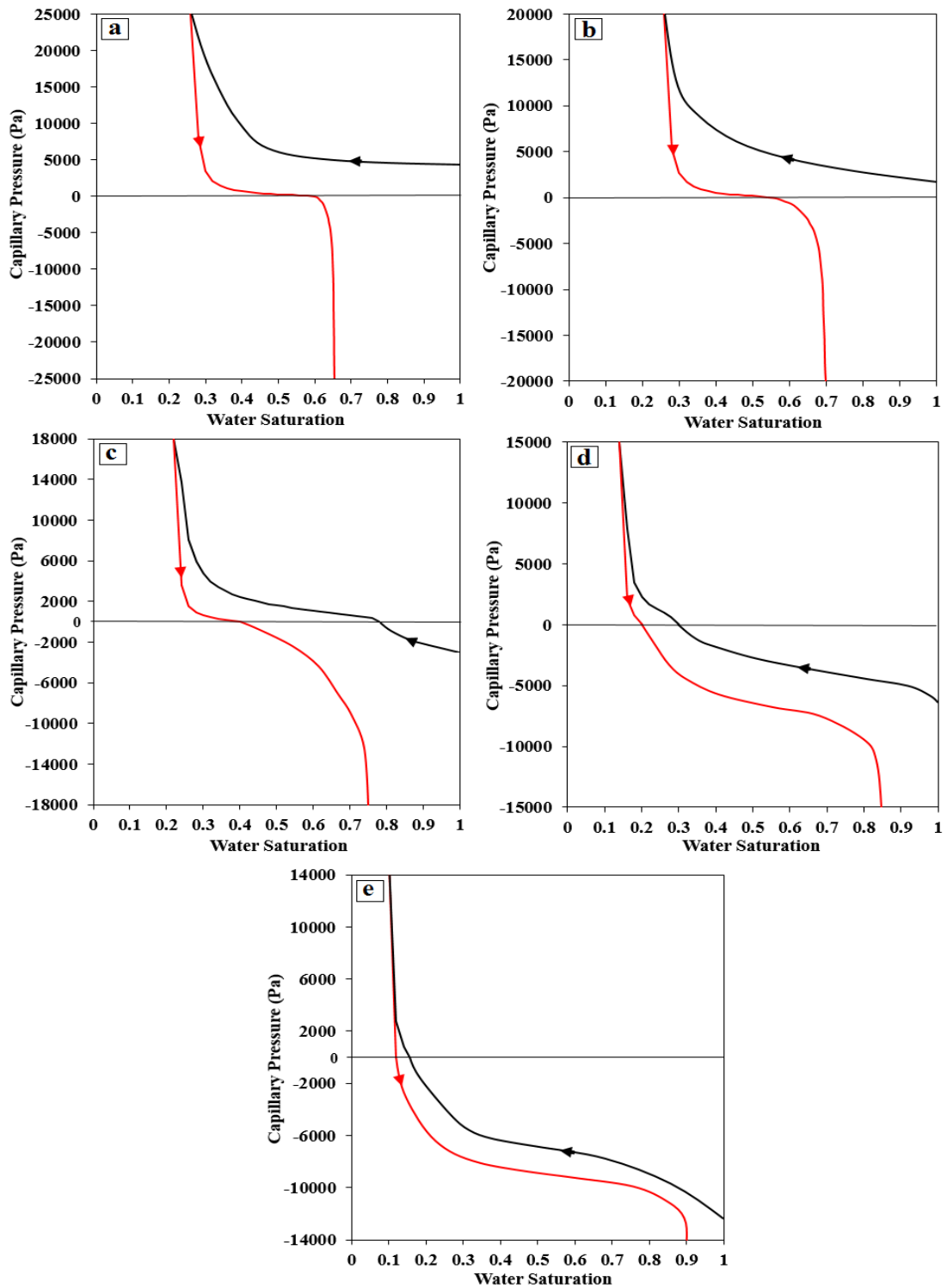


Figure 4-4 Capillary pressure curves for the five different rock wettabilities investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. Only the positive parts of these curves were implemented into the simulations, according to the current software capabilities.

4.3 Results and discussion

4.3.1 Impact of wettability on CO₂ storage in a heterogeneous reservoir

4.3.1.1 Impact of the reservoir wettability on the CO₂-plume migration

It is well established that the CO₂-plume migrates upwards because of the density differences between the injected supercritical CO₂ and the aquifer brine (Dai et al., 2014; Flett et al., 2007; Hassanzadeh et al., 2009).

The set of simulations we have performed shows furthermore that wettability affects this CO₂ plume upward migration (Figure 4-5 to Figure 4-7). CO₂ migrates upward furthest in the strongly CO₂-wet rock (scw), while it is best retained in the strongly water-wet rock (sww) (e.g. the vertical migration distances are scw =32 m and sww =17 m, scw =33 m and sww =18 m, scw =34 m and sww =19 m, scw =34 m and sww =19 m, scw =35 m and sww =20 m, at the end of storage period and after injecting for 1 a period of year 10,000 tCO₂, 15,000 tCO₂, 20,000 tCO₂, 25,000 tCO₂ and 30,000 tCO₂, respectively (Figure 4-8)). The CO₂ plume migration distance for the weakly water-wet, intermediate-wet and weakly CO₂-wet ranges between these two above extreme cases (Figure 4-8). In summary the migration distance increases steadily with increasing the CO₂ wettability of the rocks. Our results are in agreement with previous simulations studies which showed that the CO₂ plume migration distance increases with decreasing the amount of the residual CO₂ saturation (Metz et al., 2005; Kumar et al., 2005; Doughty, 2010).

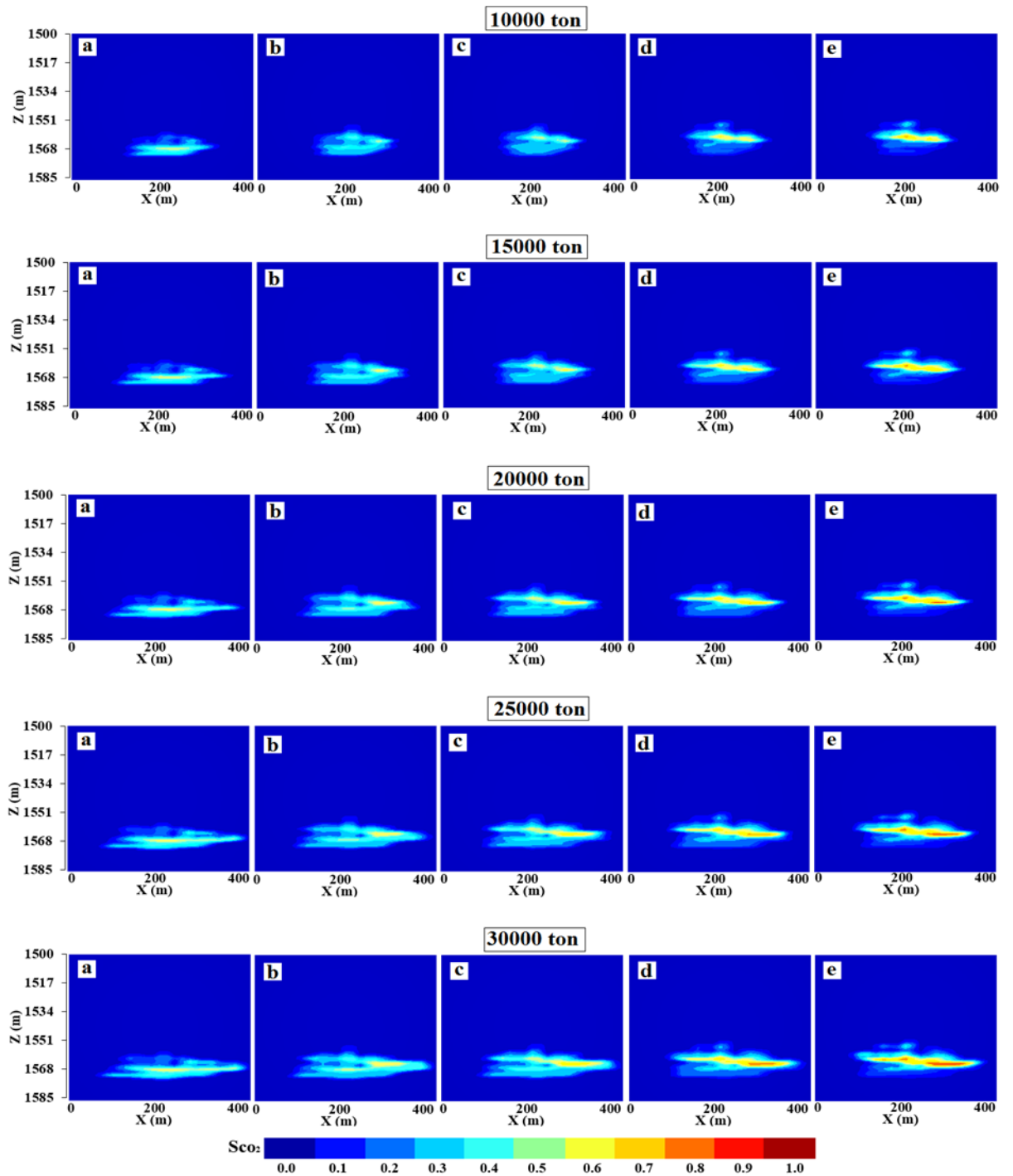


Figure 4-5 2D vertical cross-sections through the middle of the storage heterogeneous aquifer. CO₂ is injected at a depth of 1570 m. The CO₂ plume shape and height in the aquifer are shown for different amount of CO₂ injected after 1 years of storage period for the five different rock wettability cases investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. Z= model height; X= model length.

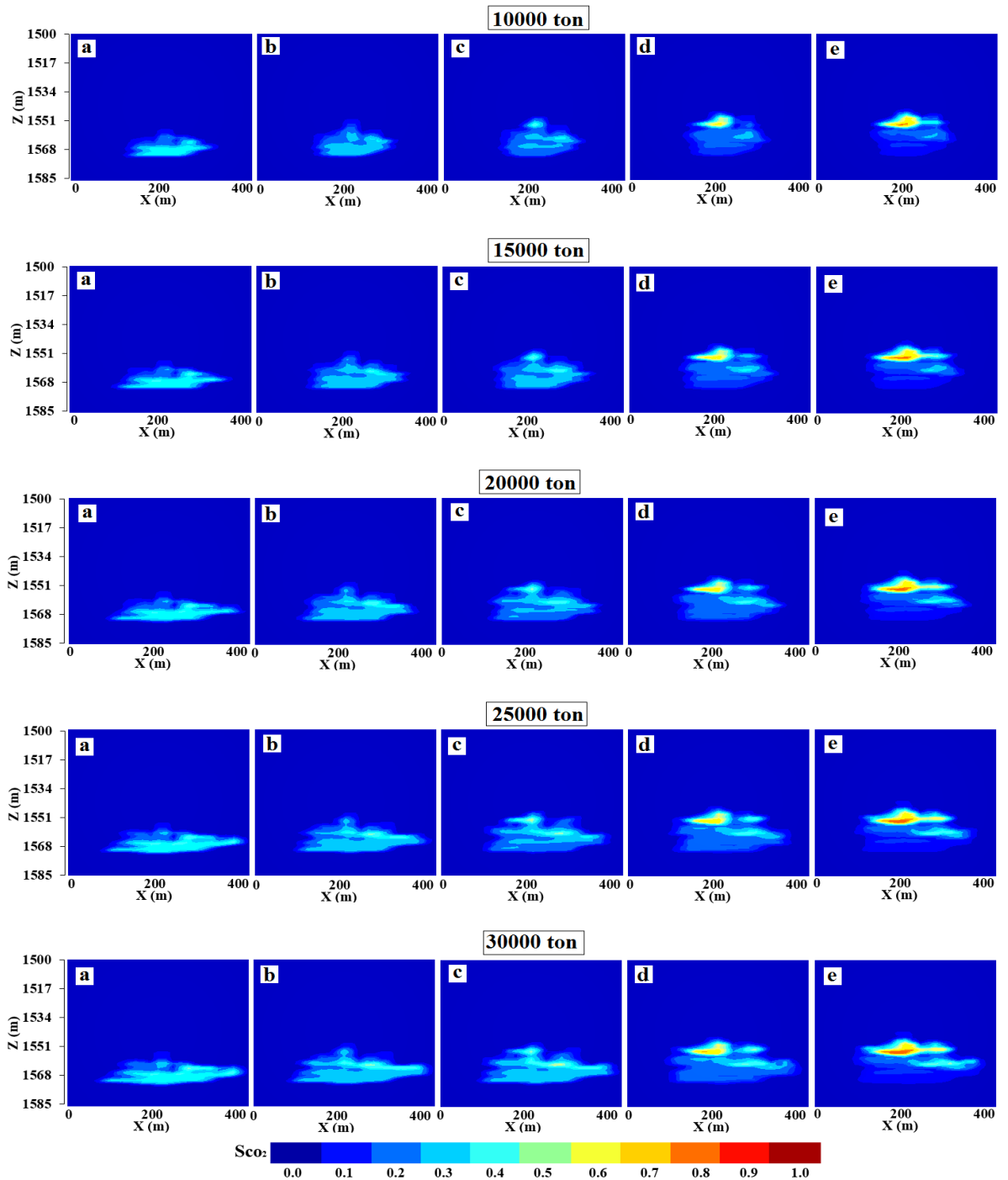


Figure 4-6 2D vertical cross-sections through the middle of the storage heterogeneous aquifer. CO₂ is injected at a depth of 1570 m. The CO₂ plume shape and height in the aquifer are shown for different amount of CO₂ injected after five years of storage period for the five different rock wettability cases investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. Z= model height; X= model length.

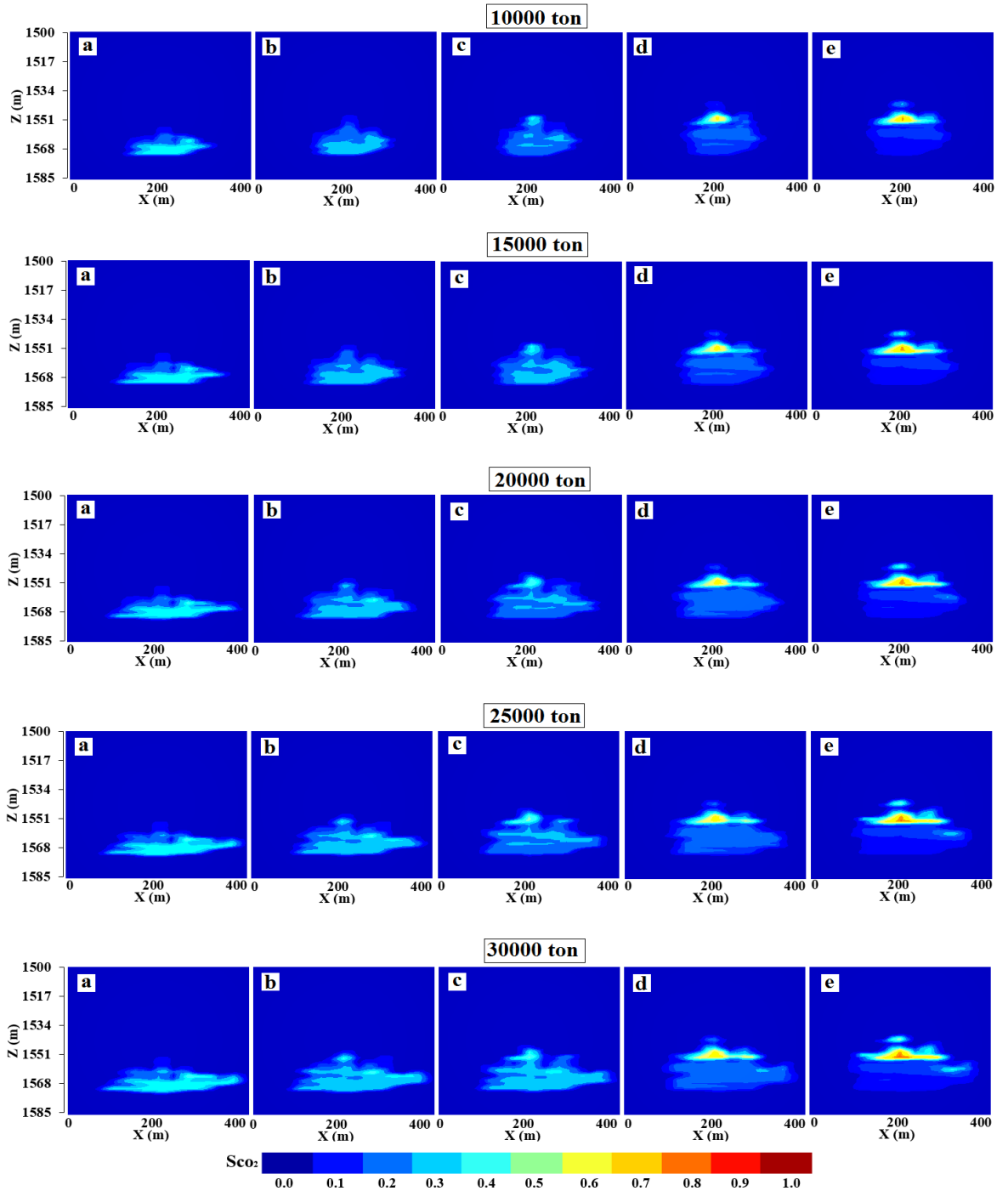


Figure 4-7 2D vertical cross-sections through the middle of the storage heterogeneous aquifer. CO₂ is injected at a depth of 1570 m. The CO₂ plume shape and height in the aquifer are shown for different amount of CO₂ injected at the end of storage period (10 years), for the five different rock wettability cases investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. Z= model height; X= model length.

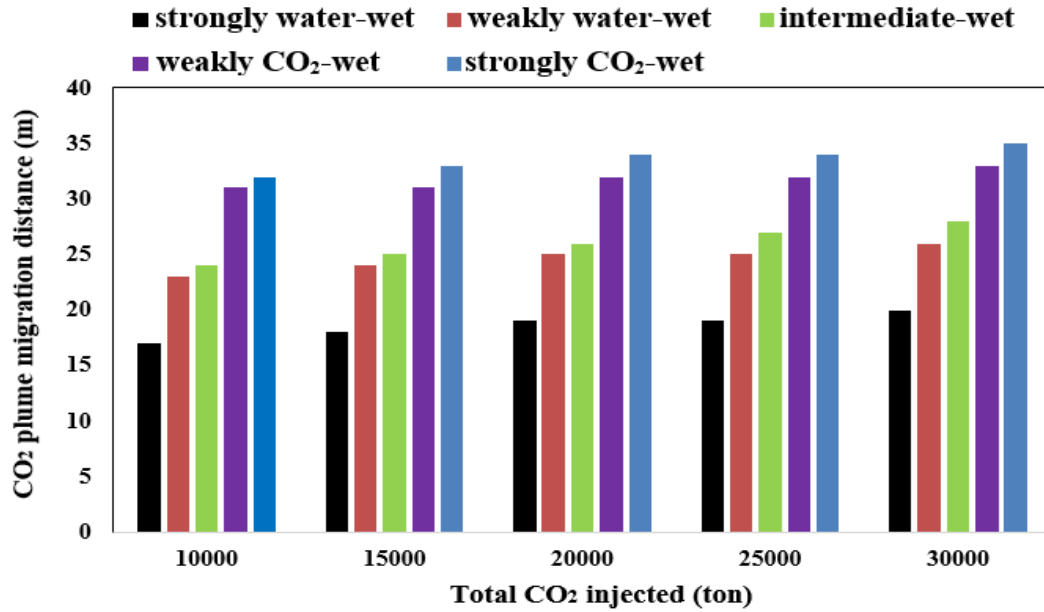


Figure 4-8 CO₂ plume vertical migration distance with different amounts of CO₂ injected, after 10 years of storage period and for the heterogeneous reservoir simulation case.

4.3.1.2 Impact of reservoir wettability on CO₂ trapping capacity

The highest CO₂ mobility (i.e. the amount of free or mobile CO₂) is found for the strongly CO₂-wet reservoir case: mobile CO₂ is 6425 ton, 9876 ton, 13496 ton, 17112 ton and 20761 ton for the five amounts of CO₂ injected, 10,000 tCO₂, 15,000 tCO₂, 20,000 tCO₂, 25,000 tCO₂ and 30,000 tCO₂, respectively (Figure 4-9). For the other wettability conditions, CO₂ mobility reduces with decreasing CO₂ wettability (i.e. when the reservoir becomes more water-wet) (Figure 4-9).

CO₂ trapped by dissolution in the aqueous phase is found to be a function of both reservoir wettability and of the amount of CO₂ injected. From Figure 4-10, it is clear that the amount of dissolved CO₂ increases with increasing CO₂-wettability and the amount of injected CO₂. For example, the amount of dissolved CO₂ at the end of the storage period in the strongly CO₂-wet reservoir (scw) and strongly water-wet reservoir (sww) was computed as follow: scw =2594 ton and sww =1702 ton, scw =3645 ton and sww =2437 ton, scw =4603 ton and sww =3108 ton, scw =5528 ton and sww =3823 ton, scw =6418 ton and sww =4520 ton, after injecting 10,000 tCO₂, 15,000 tCO₂, 20,000 tCO₂, 25,000 tCO₂ and 30,000 tCO₂, respectively (Figure 4-10). Previous simulations studies, which had not incorporated the effect of the wettability

(e.g. (Metz et al., 2005; Kumar et al., 2005; Doughty, 2010)), showed that solubility trapping increases with increasing CO₂ plume migration, which is consistent with our results as this study shows that more CO₂ migration is associated with a more CO₂-wet reservoir.

Residual trapping at the reservoir scale is also strongly influenced by the reservoir wettability. Several studies (Anderson, 1987a; Anderson, 1987b; Craig, 1993; Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Pentland et al., 2011; Rahman et al., 2016) demonstrated that the residual trapping for CO₂ is influenced by residual CO₂ saturation; residual CO₂ saturation is incorporated in our reservoir modelling scenarios via the relative permeability and capillary pressure curves we constructed. Clearly, the results show that residually trapped CO₂ increases with decreasing CO₂ wettability and that the residually trapped CO₂ in the strongly CO₂-wet reservoirs (scw) is less than that in the strongly water-wet (sww) (e.g. scw =981 ton and sww =7154 ton, scw =1479 ton and sww =10595 ton, scw =1901 ton and sww =13654 ton, scw = 2360 ton, and sww =15770 ton, scw =2821 to and sww =18238 ton, at the end of storage period and after injecting 10,000 tCO₂, 15,000 tCO₂, 20,000 tCO₂, 25,000 tCO₂ and 30,000 tCO₂, respectively (Figure 4-11)). These results are consistent with previous simulation studies (e.g. (Krevor et al., 2015; Juanes et al., 2006)) that have shown that mixed-wet rocks have a lower residual trapping capacity when compared to water-wet rocks.

In summary, our results show that reservoir wettability has a strong effect on CO₂ mobility and on residual and dissolution trapping capacities.

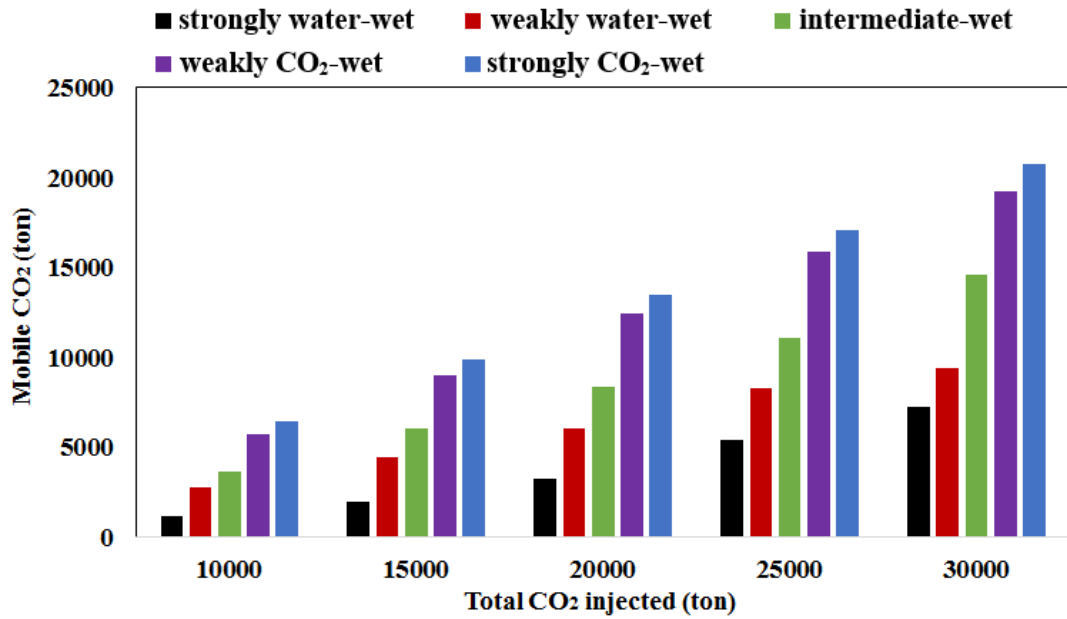


Figure 4-9 Amount of mobile CO₂ as a function of total amount of injected CO₂ and as a function of rock wettability, at the end of storage period (10 years). In all cases, CO₂ was injected for 1 year.

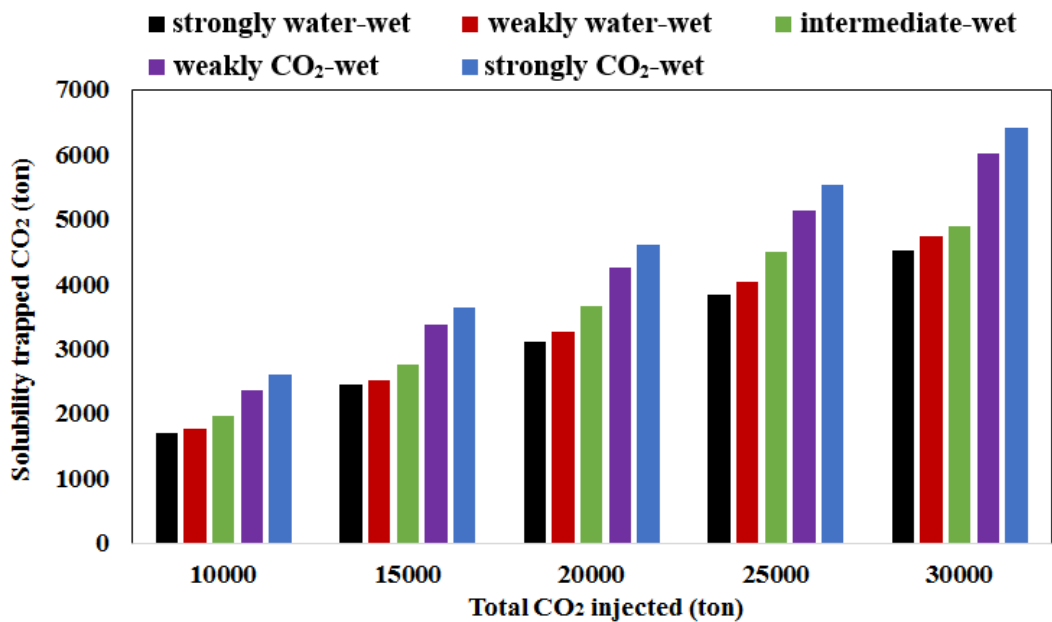


Figure 4-10 Amount of CO₂ trapped by solubility as a function of total amount of injected CO₂ and rock wettability, at the end of storage period (10 years). In all cases, CO₂ was injected for 1 year.

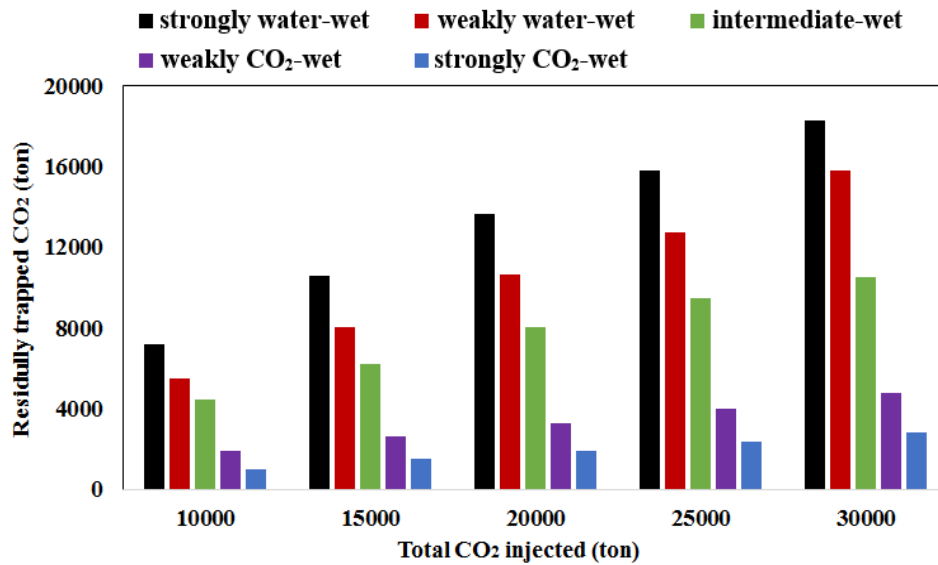


Figure 4-11 Amount of residually trapped CO₂ as a function of total amount of injected CO₂ and rock wettability, at the end of storage period (10 years). In all cases, CO₂ was injected for 1 year.

4.3.2 Impact of heterogeneity on CO₂ storage as a function of rock wettability

Many studies have shown that reservoir heterogeneity has a significant impact on CO₂ plume migration and on trapping capacities (e.g. (Doughty, 2010; Hovorka et al., 2004; Obi and Blunt, 2006; Bryant et al., 2006; Flett et al., 2007; Ide et al., 2007; Saadatpoor et al., 2009; Zhou et al., 2010; Hesse and Woods, 2010; Green and Ennis-King, 2010; Ambrose et al., 2008; Han et al., 2010; Gershenson et al., 2015)). However, the combined effect of heterogeneity and wettability has received little attention. We thus compare the results obtained for the case of a heterogeneous reservoir, presented above, with those obtained for the case of a homogeneous reservoir.

4.3.2.1 Impact of reservoir heterogeneity on CO₂-plume flow behaviour

For all wettability conditions analysed, the CO₂-plume vertical migration distance is larger in the homogeneous reservoir than in the heterogeneous reservoir (Figure 4-12 and Table 4-3). The highest migration distance, obtained for both the homogeneous and heterogeneous cases for the strongly CO₂-wet reservoir, was computed to be 70 m for the homogeneous scenario but only half of that (35 m) for the heterogeneous scenario, at the end of storage period and after injecting 30,000 ton of CO₂ (Table 4-3). Similarly, for the strongly water-wet reservoir case, the highest migration distance was computed to be 64 m for the homogeneous scenario but only 20 m for the

heterogeneous scenario, at the end of storage period and after injecting 30,000 ton of CO₂ (Table 4-3). For all other wettability conditions and for all amounts of CO₂ injected, the CO₂ plume migration distance in the homogeneous model is approximately two times more than that computed for the heterogeneous model (Figure 4-12 and Table 4-3). However, although the reservoir heterogeneity reduces the vertical CO₂ plume migration, it enhanced the lateral movements of the CO₂ in the aquifer, as shown, previously, in Figure 4-5 to Figure 4-7.

In summary, our results show that the reservoir heterogeneity has a significant impact on the CO₂ plume flow behaviour.

Table 4-3 CO₂ plume vertical migration distance with different amount of injected CO₂, for the homogeneous and heterogeneous reservoir scenarios, at the end of the storage period (10 years) and for the five investigated wettability conditions

heterogeneity model	Total injected CO ₂ (ton)	Strongly water -wet	weakly water-wet	Intermediate-wet	weakly CO ₂ -wet	strongly CO ₂ -wet
CO ₂ plume vertical migration distance in heterogeneous model	10000	17	23	24	31	32
	15000	18	24	25	31	33
	20000	19	25	26	32	34
	25000	19	25	27	32	34
	30000	20	26	28	33	35
CO ₂ plume vertical migration distance in homogeneous model	10000	45	50	58	65	70
	15000	53	55	62	67	70
	20000	58	59	64	70	70
	25000	61	63	66	70	70
	30000	64	66	68	70	70

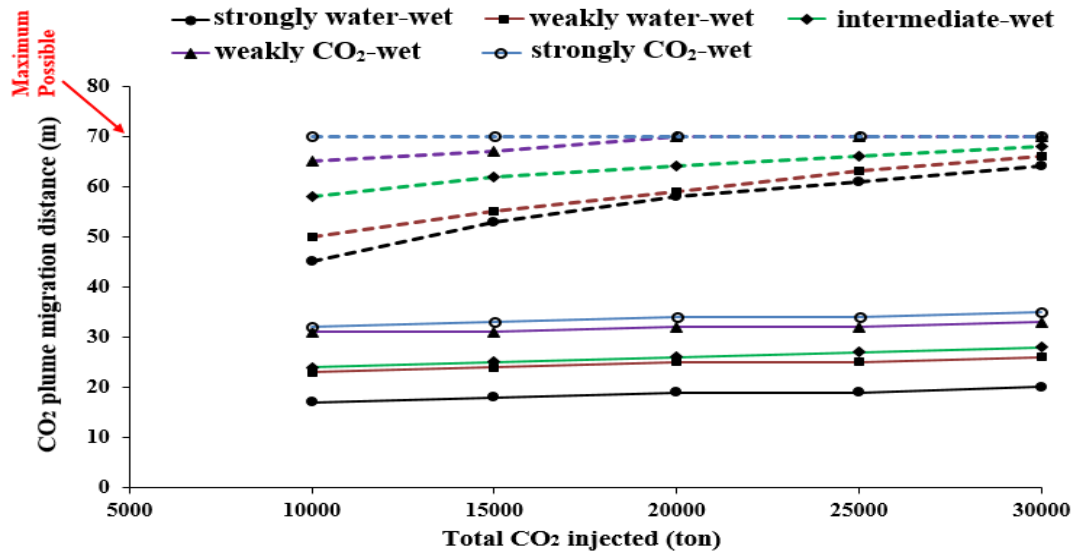


Figure 4-12 CO₂ plume vertical migration distance with different amount of injected CO₂, for the homogeneous (dashed line) and heterogeneous (solid line) reservoir scenarios, at the end of storage period (10 years) and for the five investigated wettability conditions.

4.3.2.2 Impact of the reservoir heterogeneity on CO₂ trapping capacity

Although the general impact of wettability on the trapping mechanisms is the same for both homogeneous and heterogeneous reservoir (i.e. more mobile, more solubility trapped CO₂ and less residually trapped CO₂ in the more CO₂-wet reservoirs), there is a significant difference between the heterogeneous and homogeneous reservoir both in time and in space, as illustrated in Figure 4-13 and Figure 4-14 and Table 4-4. Our simulation results show that more mobile CO₂ (the majority of this mobility following a horizontal movement not a vertical migration) is computed for the heterogeneous reservoir than for the homogeneous reservoir model, for all wettability conditions. We indeed mentioned in the previous sections that, although the heterogeneity inhibited the vertical migration of CO₂ plume, it enhanced the lateral CO₂ flow.

Moreover, our results show that reservoir heterogeneity decreases the percentage of residually trapped CO₂, which is consistent with Han et al.'s [2010] work; these authors concluded that increasing permeability heterogeneities resulted in decreasing the amount of residually trapped CO₂. Our simulations show that heterogeneity also decreases the percentage of dissolved CO₂ (Figure 4-13 and Figure 4-14 and Table 4-

4). It is important to mention here that the amount of dissolved CO₂ is essentially a function of CO₂ plume migration (Metz et al., 2005; Kumar et al., 2005; Doughty, 2010) due to the fact that permeability heterogeneity (or permeability barrier) reduces the vertical plume migration: we observe that the percentage of dissolved CO₂ decreases for all wettability scenarios (Table 4-4).

We thus conclude that even though the general trends of the effect of wettability on CO₂ storage capacities (i.e. that the more water-wet the reservoir, the less CO₂ is mobile, the less CO₂ is solubility trapped and the more CO₂ is residually trapped) is the same for both, the homogeneous and the heterogeneous scenarios, the absolute values are quite different. Hence, not taking into account heterogeneities in the porosity and permeability distributions can lead to erroneous estimates of CO₂ storage capacities.

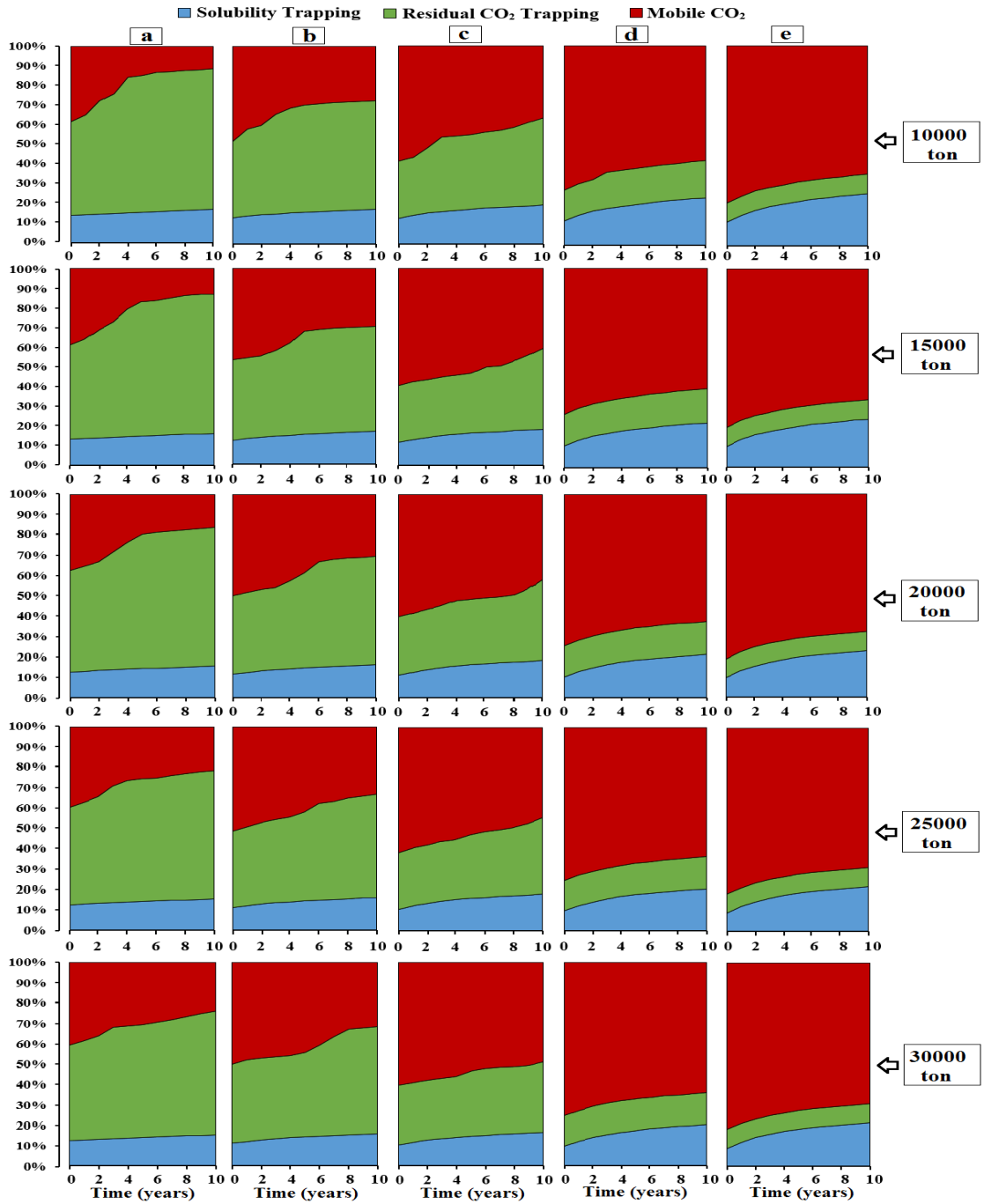


Figure 4-13 Percentage of trapped and free CO₂ (dissolution trapping is in blue, residual trapping in green and mobile CO₂ is in red) as a function of amount of injected CO₂ and storage time, for the case of the heterogeneous aquifer and for the five different rock wettabilities investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet.

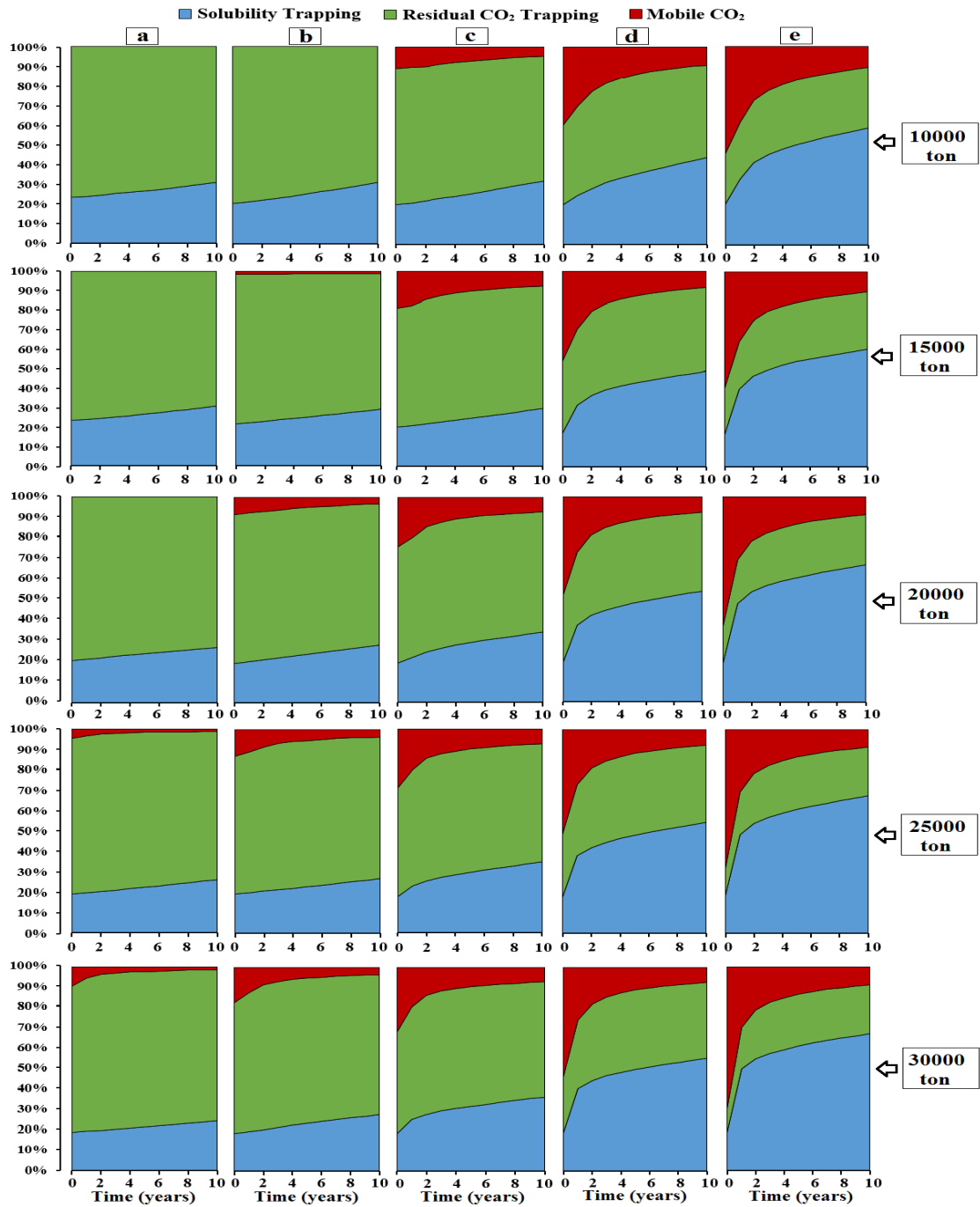


Figure 4-14 Percentage of trapped and free CO₂ (dissolution trapping is in blue, residual trapping in green and mobile CO₂ is in red) as a function of amount of injected CO₂ and storage time, for the case of the homogeneous aquifer and for the five different rock wettabilities investigated: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet.

Table 4-4 Amount of mobile and trapped CO₂ computed at the end of the storage period (10 years) with different amount of CO₂ injected during the injection period, in both heterogeneous and homogenous reservoir models and for the five different wettabilities investigated.

Wettability	Trapping mechanism (ton)	Total injected CO ₂ in heterogeneous reservoir (ton)					Total injected CO ₂ in homogeneous reservoir (ton)				
		10000	15000	20000	25000	30000	10000	15000	20000	25000	30000
Strongly water-wet	solubility trapped CO ₂	1702	2437	3108	3823	4520	2685	3627	5358	6481	7328
	residually trapped CO ₂	7154	10595	13654	15770	18238	7315	11373	14642	18207	22165
	mobile CO ₂	1144	1968	3238	5407	7242	0	0	0	312	507
Weakly water-wet	solubility trapped CO ₂	1775	2517	3254	4040	4742	3117	4354	5600	6744	8362
	residually trapped CO ₂	5483	8054	10666	12703	15819	6883	10472	13714	17387	20608
	mobile CO ₂	2742	4429	6080	8257	9439	0	174	686	869	1030
Intermediate-wet	solubility trapped CO ₂	1971	2745	3655	4492	4884	3224	4426	6844	8712	10940
	residually trapped CO ₂	4410	6231	8006	9429	10520	6328	9434	11738	14532	17005
	mobile CO ₂	3619	6024	8339	11079	14596	448	1140	1418	1756	2055
Weakly CO ₂ -wet	solubility trapped CO ₂	2360	3372	4248	5133	6009	4389	7345	10787	13634	16608
	residually trapped CO ₂	1879	2584	3273	3995	4779	4702	6415	7720	9472	11222
	mobile CO ₂	5761	9044	12479	15872	19212	909	1240	1493	1894	2170
Strongly CO ₂ -wet	solubility trapped CO ₂	2594	3645	4603	5528	6418	5913	9058	13341	16911	20155
	residually trapped CO ₂	981	1479	1901	2360	2821	3028	4402	4896	5948	7239
	mobile CO ₂	6425	9876	13496	17112	20761	1059	1540	1763	2141	2606

4.4 Conclusions

Reservoir wettability can vary widely from strongly water-wet (0° contact angle) to strongly CO₂-wet (170° contact angle) (Dickson et al., 2006; Chiquet et al., 2007; Yang et al., 2008b; Espinoza and Santamarina, 2010; Bikkina, 2011; Kaveh et al., 2011; Wang et al., 2012; Iglauer et al., 2012; Iglauer et al., 2014; Iglauer et al., 2015a; Iglauer et al., 2015b; Chaudhary et al., 2015; Chen et al., 2015a; Sarmadivaleh et al., 2015; Arif et al., 2016a; Arif et al., 2016b). However, the effect of rock wettability on CO₂ storage capacity and containment security has received only little attention (Iglauer et al. 2015a), and no data is available for the combined effect of wettability and reservoir permeability heterogeneity.

In this study, we simulated the CO₂ plume migration and associated CO₂ trapping mechanisms for both a homogeneous and a heterogeneous reservoir. The effect of the amount of CO₂ injected was also investigated. Our results show that a strongly water-wet reservoir has the lowest CO₂ plume migration, with the lowest percentage of mobile CO₂, lowest percentage of CO₂ dissolved in water, and the highest percentage of residually trapped CO₂, independently of the injection rate of CO₂. This observation is valid for both the homogenous and heterogeneous cases but the magnitude of these observations varies both in time and in space between the two cases. Importantly, our results of the impact of wettability on residual trapping (same as capillary trapping) is in line with previous simulation studies (e.g. (Krevor et al., 2015; Juanes et al., 2006)) which concluded that mixed-wet rocks have a lower residual trapping capacity when compared to water-wet rocks. The CO₂ plume spreads more laterally in the heterogeneous reservoir while it has more vertical migration in the homogeneous reservoir. This is due to the fact that the CO₂ plume is prevented to move upwards driven by buoyancy forces because of the presence of lower permeability zones that decrease vertical CO₂ flow velocities. This leads to more mobile CO₂ and less solubility and residually trapped CO₂, consistent with literature data from previous simulation studies which investigated the effect of reservoir heterogeneity, without incorporating the effect of reservoir wettability, (i.e. (Doughty, 2010; Doughty and Myer, 2009; Hovorka et al., 2004; Obi and Blunt, 2006; Bryant et al., 2006; Flett et al., 2007; Zhou et al., 2010; Hesse and Woods, 2010; Han et al., 2010)).

In summary, we conclude that reservoir wettability has a critical impact on the CO₂ plume migration and trapping capacities, with water-wet reservoirs being preferable

CO₂ sinks. Furthermore, simulations, classically performed with homogeneous reservoir properties (e.g. porosity and permeability) may give correct general behaviour of the CO₂ plume in the subsurface (for example that the CO₂ plume vertical migration distance increases with increasing the CO₂ wettability (i.e. when the reservoir becomes more CO₂-wet) for both of homogeneous and heterogeneous models) but the magnitude of these observations may be inaccurate both in time and in space for the real reservoir, for which rock properties are heterogeneous. We conclude that both reservoir wettability and rock properties heterogeneities such as porosity and permeability need to be accurately reflected in CCS reservoir simulations in order to reliably assess CO₂ storage capacities and containment security.

Chapter 5 Influence of Injection Well Configuration and Rock Wettability on CO₂ Plume Behaviour and CO₂ Trapping Capacity in Heterogeneous Reservoirs

5.1 Introduction

The main reason of Earth's climate changes has been identified as due to greenhouse gas emission and carbon dioxide (CO₂) is the major component of this emission (Houghton et al., 2001; Zhang, 2016). The geological sequestration of CO₂ is considered one of the most effective methods to mitigate greenhouse gas emission by collecting it from power plants, refineries, and other industrial resources and injecting it into a deep underground geological formation (Pruess et al., 2003). The CO₂ can be stored underground in different trapping sites (i.e. unminable coal beds, saline aquifers, and depleted hydrocarbon reservoirs (Metz et al., 2005; Xie and Economides, 2009; Zhang et al., 2016b)). The injected CO₂ migrates upward towards the surface due to the density differences between the brine and CO₂ (Dai et al., 2014; Flett et al., 2007; Hassanzadeh et al., 2009). This vertical CO₂ migration (i.e. CO₂ leakage) is considered the most significant risk involved in CO₂ injection (Deel et al., 2007; Li and Liu, 2016). Such CO₂ leakage can result in various risks (i.e. humans and animals health risks, ecosystems damage, underground sources of drinking water contamination, and safety and environmental risks (Humez et al., 2013; Smyth et al., 2009; Kharaka et al., 2009; Kharaka et al., 2013; Oldenburg, 2008)). This CO₂ leakage risk can be reduced by four different trapping mechanisms. In the first one, a significant ratio of the CO₂ will be trapped in the high porosity and permeability zones of the target reservoir and prevented to move further by seal beds of the cap rocks or by low permeability reservoir layers: this trapping mechanism is called the structural trapping (Hesse et al., 2008; Naylor et al., 2011; Iglauer et al., 2015b; Gershenson et al., 2015). The second trapping mechanism is related to the hysteresis of capillary pressure and relative permeability curves and is called residual trapping mechanism (Kumar et al., 2005; Pentland et al., 2011; Iglauer et al., 2011). The third trapping mechanism deals with the amount of CO₂ that will be trapped in the form of dissolved CO₂ in the aqueous phase: it is called solubility trapping (Lindeberg and Wessel-Berg,

1997; Spycher et al., 2003; Iglauer, 2011) and is highly dependent on the CO₂-water interface area (Metz et al., 2005; Kumar et al., 2005; Doughty, 2010). The fourth trapping mechanism, and that can take place over the longest time scale (could reach for thousands of years) is the mineral trapping where the CO₂ can be permanently trapped in the aquifer as carbonate minerals, as a result of CO₂/brine/mineral interactions (Bachu et al., 1994b; Xu et al., 2004; Gaus, 2010; Flett et al., 2007; Metz et al., 2005).

The storage efficiency of saline aquifers can be enhanced either by the accurate evaluation of the geological factors affecting the CO₂ storage process (e.g. caprock properties, porosity and permeability heterogeneity, permeability anisotropy (K_v/K_h)), or by optimising the CO₂ injection process (i.e. the type of injection scenario that includes continuous CO₂ injection, water alternate gas (WAG) and, possibly, as we will explain in this study, injection well configuration (vertical or horizontal wells)). In this context, previous studies show that there are various geological factors that have been identified as significant in affecting the CO₂ storage processes including properties of cap rocks (Iglauer et al., 2015a) and vertical to horizontal permeability ratio (K_v/K_h) (Kumar et al., 2005; Basbug et al., 2005; Hassanzadeh et al., 2009). Furthermore, some initial and boundary conditions of the injection sites have an important impact on the safety of CO₂ storage mechanism which are: injection site temperature (Kumar et al., 2005; Bennion and Bachu, 2006; Ofori and Engler, 2011), and injection depth (Zeidouni et al., 2015). In addition, CO₂ injection scenario, such as using a WAG scenario compared to a traditional pure CO₂ injection, is also considered as an important factor that enhances the storage capacity by increasing the efficiency of residual (Juanes et al., 2006) and dissolution (Hassanzadeh et al., 2009; Leonenko and Keith, 2008; Vivek and Kumar, 2016)) CO₂ trappings. Moreover, the geological heterogeneity (i.e. porosity and permeability heterogeneity) had been shown previously as an important effecting factor on the CO₂ storage efficiency (Hovorka et al., 2004; Flett et al., 2007; Hesse and Woods, 2010; Green and Ennis-King, 2010; Song et al., 2014a; Gershenson et al., 2015; Gershenson et al., 2016). In addition, the top-surface morphology has been well investigated in literature as an affecting factor on CO₂ storage capacity (Nilsen et al., 2012; Shariatipour et al., 2016; Syversveen et al., 2012). Furthermore and importantly, reservoir wettability, from laboratory investigations (mm to cm scale), has been found to be an important rock parameter that has a significant impact on residual trapping (cp. Iglauer et al., 2011 and Andrew

et al., 2013 versus Chaudhary et al., 2013 and Rahman et al., 2016) and structural trapping (cp. Naylor et al., 2011 versus Iglauer et al., 2015). Furthermore, a recent 3D multi-phase flow simulation study by Al-Khdheawi et al., (2017c) showed that wettability affects the CO₂ storage capacity. The impact of CO₂ injection well type has been well investigated for Enhanced Oil Recovery (EOR) and it has been found that the use of horizontal wells leads to improve the productivity and the injectivity because a horizontal well provide a larger flow area than a vertical well, for the same reservoir volume (Kossack et al., 1987; Reiss, 1987; Macdonald, 1988; Joshi, 1988; Joshi, 2003; Babu and Odeh, 1989; Hardman, 1989; Murphy, 1990; Economides et al., 1991; Thomas, 2008). However, the impact of injection well configuration on the CO₂ plume behaviour and on the efficiency of CO₂ trapping mechanisms has not been addressed in detail previously.

In this chapter, we have assessed how to reduce the CO₂ leakage risk by investigating the impact of the CO₂ injection well configuration (vertical versus horizontal) and rock wettability on the CO₂ plume behaviour, its vertical migration and on CO₂ storage capacities (mobile CO₂, residually trapped CO₂ and solubility trapped CO₂). For that purpose, we developed 3D multiphase flow simulations in a heterogeneous reservoir using five different rock wettability conditions and four injection well scenarios: one vertical well, two vertical wells, four vertical wells and one horizontal well.

5.2 Methodology

5.2.1 Numerical simulation models

Four sets of simulation models for a heterogeneous saline aquifer have been developed in this study in order to investigate the influence of the CO₂ injection well configuration and rock wettability on the CO₂ plume movements, shape and vertical migration and on the capacity of mobile and trapped CO₂. These four sets are performed in the same heterogeneous geological model: three use different vertical wells scenarios (i.e. one well (scenario V1), two wells (scenario V2), and 4 wells (scenario V4)) and the fourth one uses one horizontal well (scenario H). These four sets of simulation models have each five scenarios representing five different rock wettability conditions, which are: strongly water-wet, weakly water-wet, intermediate-

wet, weakly CO₂-wet and strongly CO₂-wet. The nonisothermal multicomponent multiphase flow simulator TOUGH2 (Pruess et al., 1999) with the tabular equation of state (EOS) ECO2M which has been used to simulate the thermodynamic and thermophysical properties of the H₂O-NaCl-CO₂ mixtures including the super- and sub-critical conditions, along with phase changes between liquid and gaseous CO₂ (Pruess, 2011). ECO2M EOS uses the Altunin's correlations (1975) to compute the molar volumes of CO₂ (i.e. supercritical CO₂, gaseous CO₂, and CO₂ dissolved in water as liquid). The model dimensions were 1600 m × 1200 m with a thickness of 700 m. The model has 37 × 33 × 80 cells (97680 cells in total). The pressure at the top of the aquifer (800 m) is 8 MPa and at the bottom of the aquifer (1500 m) is 15 MPa following the hydrostatic pressure gradient (10 MPa/km) (Dake, 2007), (Figure 5-1). The top of the reservoir was sealed by reducing the vertical to horizontal permeability ratio (k_v/k_h) from 0.1 in the storage reservoir to 10⁻⁴ in the top layer of the reservoir, so that a barrier prevents the CO₂ from leaking to the surface. Isothermal conditions have been applied for the reservoir model and temperature was set at 333 K. The volume of the outer boundary cells of the model has been multiplied by a large volume factor of 10⁸ to simulate a constant pressure boundary condition (Mo et al., 2005; Nghiem et al., 2009). Initially the reservoir was fully saturated with brine ($S_{wi} = 100\%$) with a salinity of 15% (by weight). The model characteristics, initial and boundary conditions described above are the same for all sets of simulations, dealing with horizontal and vertical configurations of the injection well. Generally, horizontal wells are different from vertical wells by the length of the well-reservoir contact area: horizontal wells provide a larger well-reservoir contact area (Joshi, 1991), which in our model example was around 600 m (Figure 5-2). Thus, we have used different scenarios of vertical well configurations (i.e. one, two, and four vertical injection wells) in order to increase the well-reservoir contact area in the vertical well configurations. These vertical wells have been distributed in the center of the model and separated by a distance of 300 m for the two vertical wells scenario and 200 m for the four vertical wells scenario. We used geometric scaling to implement the porosity and permeability data to the heterogeneous aquifer model (97680 cells) of the tenth SPE comparative solution project (1.122 million cells; Christie and Blunt 2001), which was initially built for the PUNQ project (Floris et al., 1999), (Figure 5-3). The geometric scaling has been used to distribute the porosity and permeability. The used 10th SPE comparative solution project is considered a highly complex geological

model consisting of two formations. The top 35 layers represent the Tarbert formation, which is a prograding near shore environment with a smooth variation in the permeability. The bottom 50 layers represent the Upper-Ness formation which, is a highly heterogeneous fluvial with many channel structures. The porosity was strongly correlated to the permeability during this very complex geological model (Christie and Blunt 2001).

CO₂ was injected into the reservoir at a constant injection rate of 1,000,000 tons CO₂/yr for both the horizontal and vertical modelled cases; this rate is similar to the maximum CO₂ injection rate that was used in the Sleipner CO₂ storage project in Norway (Leung et al., 2014; Torp and Gale, 2004), and in the Quest CCS project in Canada (Bourne et al., 2014). CO₂ was injected at a depth of 1373 m and at the center of the model (Y = 600), for both the horizontal and vertical wells and for all wettability conditions; this injection lasted a period of 2 years (i.e. a total of 2 Mton of CO₂ are injected). This 2 years injection period was followed by a 200 years shut-off injection period to simulate the storage period in the reservoir. The plume behaviour was simulated during the storage period for all injection well configurations and for the 5 wettability conditions. In addition, the amount of mobile CO₂, residually trapped CO₂ and solubility trapped CO₂ were computed as a function of time for both horizontal and vertical injection wells configuration and for all the wettability scenarios. The results obtained from the horizontal well and vertical well models were compared in order to investigate the impact of injection well configuration and rock wettability on CO₂ storage efficiency in a heterogeneous reservoir.

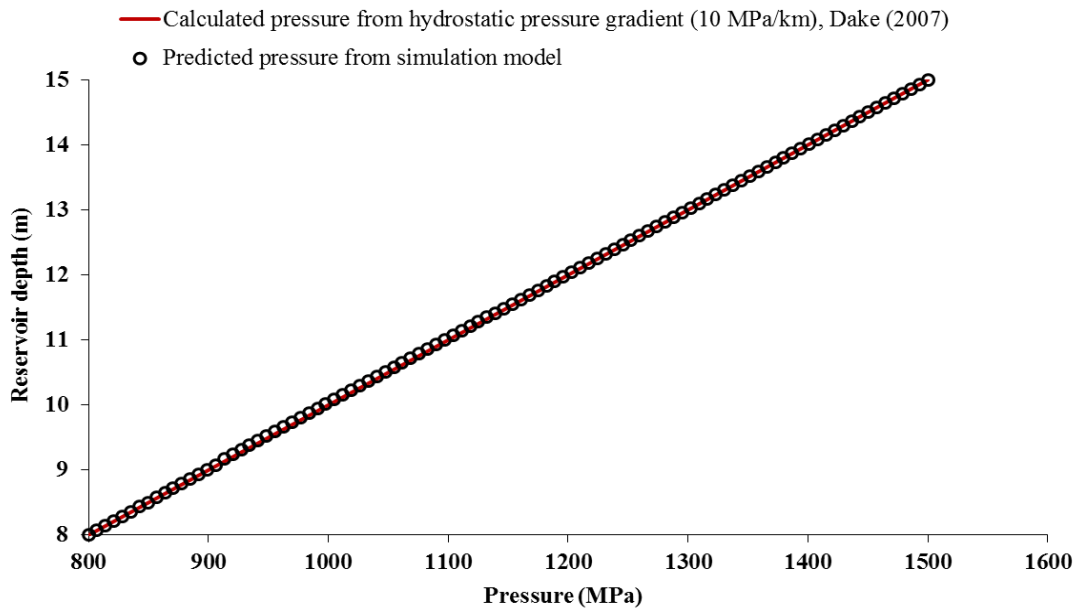


Figure 5-1 Representation of model validation by comparing the predicted and calculated initial reservoir pressure.

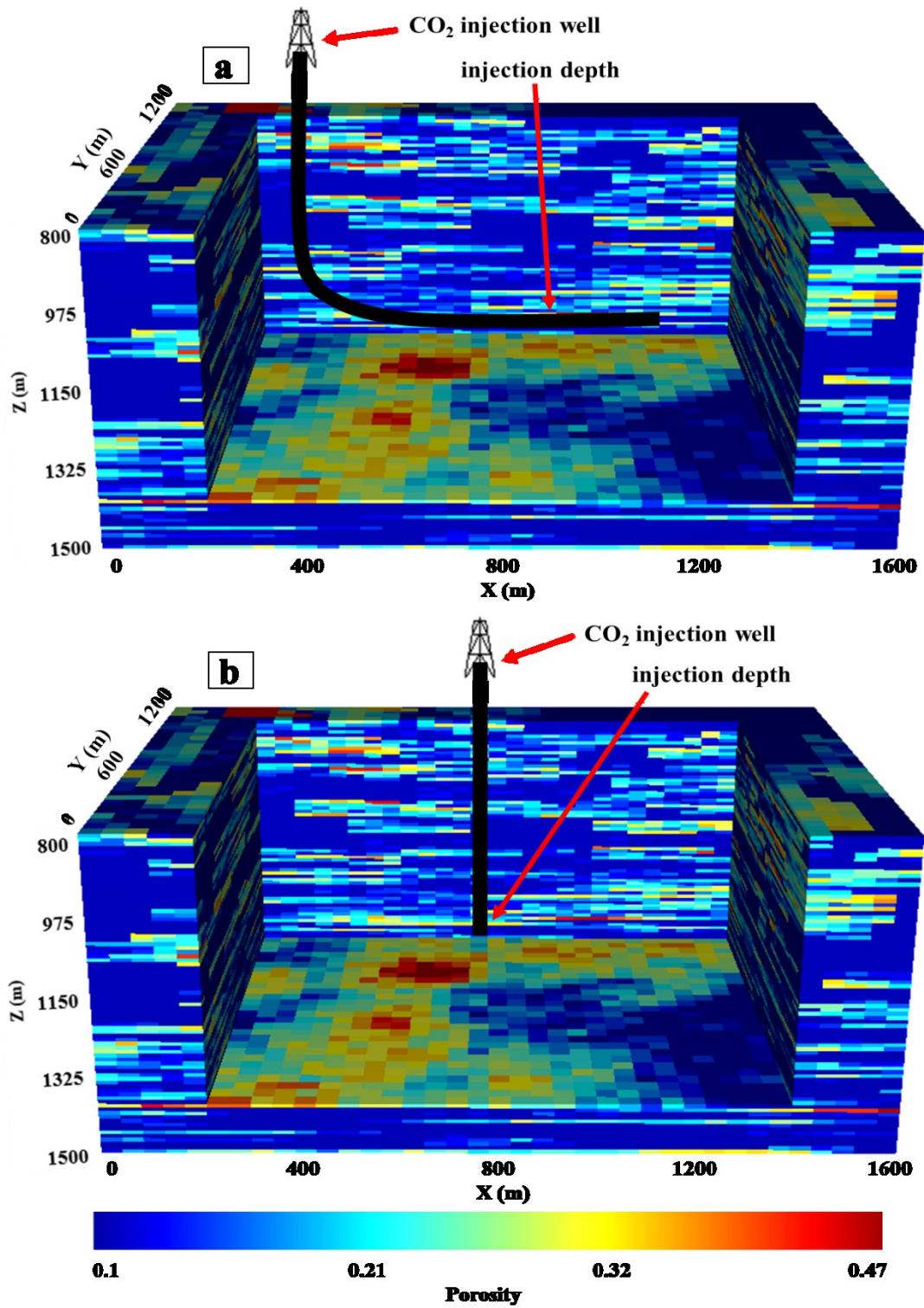


Figure 5-2 Representation of the 3-D model showing injection wells locations with porosity distribution for the reservoir and model dimensions: A) horizontal well; B) single vertical well.

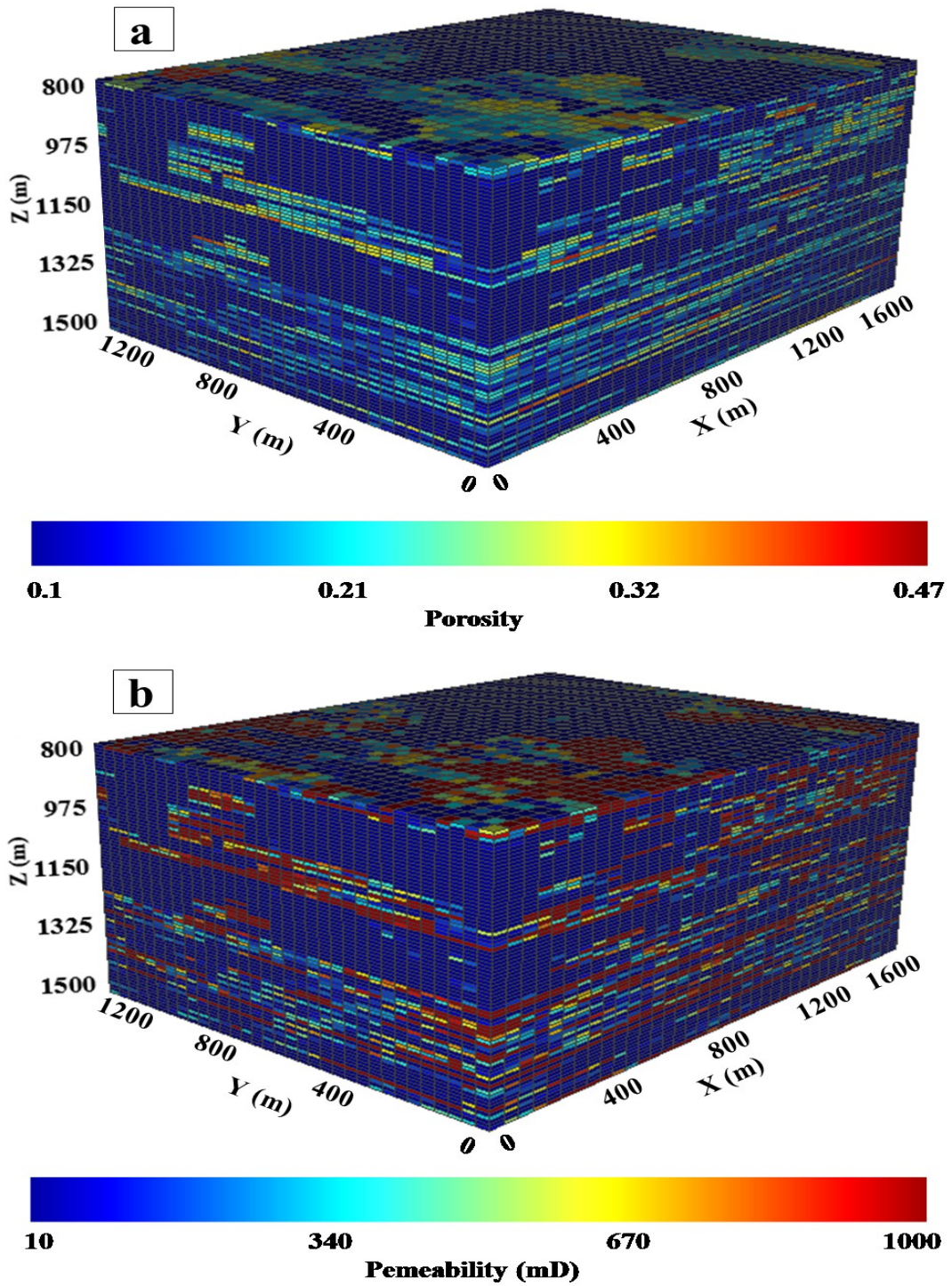


Figure 5-3 Representation of the 3-D model showing the reservoir porosity and permeability heterogeneity with model dimensions: A) porosity distribution; B) permeability distribution.

5.2.2 Simulation of different rock wettability scenarios

We have used 5 different rock wettability conditions for all well configuration models (i.e. vertical wells and horizontal well models): strongly water-wet, weakly water-wet, intermediate-wet, weakly CO₂-wet and strongly CO₂-wet. These different wettabilities have been simulated by using five different pairs of relative permeability and capillary curves for all models of horizontal and vertical injection wells, as previously performed in Al-Khdheawi et al., (2017). McCaffery and Bennison's (1974) relative permeability curves have been used with an accurate modification according to Craig's rules of thumb (Craig, 1993) and comparison with previous studies which showed the effect of wettability on the relative permeability curves (i.e. Owens and Archer, 1971; Heiba et al. 1983; Anderson, 1987b; Krevor et al., 2012; Levine et al., 2014) to fit the various wettability conditions (Figure 5-4). These adjusted curves have been implemented into the developed reservoir simulation models by using the Van Genuchten-Mualem model (Van Genuchten, 1980; Mualem, 1976) summarized by the following set of correlations:

$$k_{rw} = \sqrt{S^*} \left\{ 1 - \left(1 - [S^*]^{1/\lambda} \right)^\lambda \right\}^2 \quad \text{if } S_w < S_{ws} \quad (5.1)$$

$$k_{rw} = 1 \quad \text{if } S_w \geq S_{ws} \quad (5.2)$$

$$k_{rg} = 1 - k_{rw} \quad \text{if } S_{gr} = 0 \quad (5.3)$$

$$k_{rg} = (1 - \hat{S})^2 (1 - \hat{S}^2) \quad \text{if } S_{gr} > 0 \quad (5.4)$$

$$S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}), \quad (5.5)$$

$$\hat{S} = (S_w - S_{wr}) / (1 - S_{wr} - S_{gr})$$

where:

k_{rg} = gas relative permeability, k_{rw} = water relative permeability,

S_{gr} = residual saturation for gas, S_w = water saturation,

S_{ws} = maximum water saturation (= 1), S_{wr} = residual saturation for water,

λ = pore size distribution index (fitting parameter).

In addition, Pini's et al., (2012, 2013) capillary pressure curves for strongly water-wet Berea sandstone have been used and adjusted for the other wettability conditions by using the criteria of (Anderson, 1987a; Batycky et al., 1981; Heiba et al., 1983; Melrose, 1965; Morrow, 1976) that explains the impact of wettability on capillary

pressure curves (Figure 5-5). These adjusted curves have been implemented into the simulation models by using Van Genuchten-Mualem capillary pressure model (Van Genuchten, 1980; Mualem, 1976) which can be summarized by the following correlations:

$$(P_{cap}) = P_0 ([S^*]^{-1/\lambda} - 1)^{1-\lambda} \quad (5.6)$$

with the restriction $0 \leq P_{cap} \leq P_{max}$

$$\text{and, } S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}) \quad (5.7)$$

where:

P_{cap} = rock capillary pressure (Pa), P_0 = pressure scaling parameter (Pa),

S_{ws} = maximum water saturation, S_{wr} = residual saturation for water,

λ = pore size distribution index.

Importantly, the impact of porosity and permeability heterogeneity, for each grid block, on the capillary pressure has been implemented by using the scaling Leverett J-function (Leverett, 1941):

$$J(S_w) = \frac{P_c}{\sigma \cos \Theta} \sqrt{\frac{k}{\phi}} \quad (5.8)$$

where:

s_w = water saturation, P_c = capillary pressure (Pa), k = permeability (m^2 , mD),

ϕ = porosity, σ = surface tension (N/m), Θ = contact angle ($^\circ$).

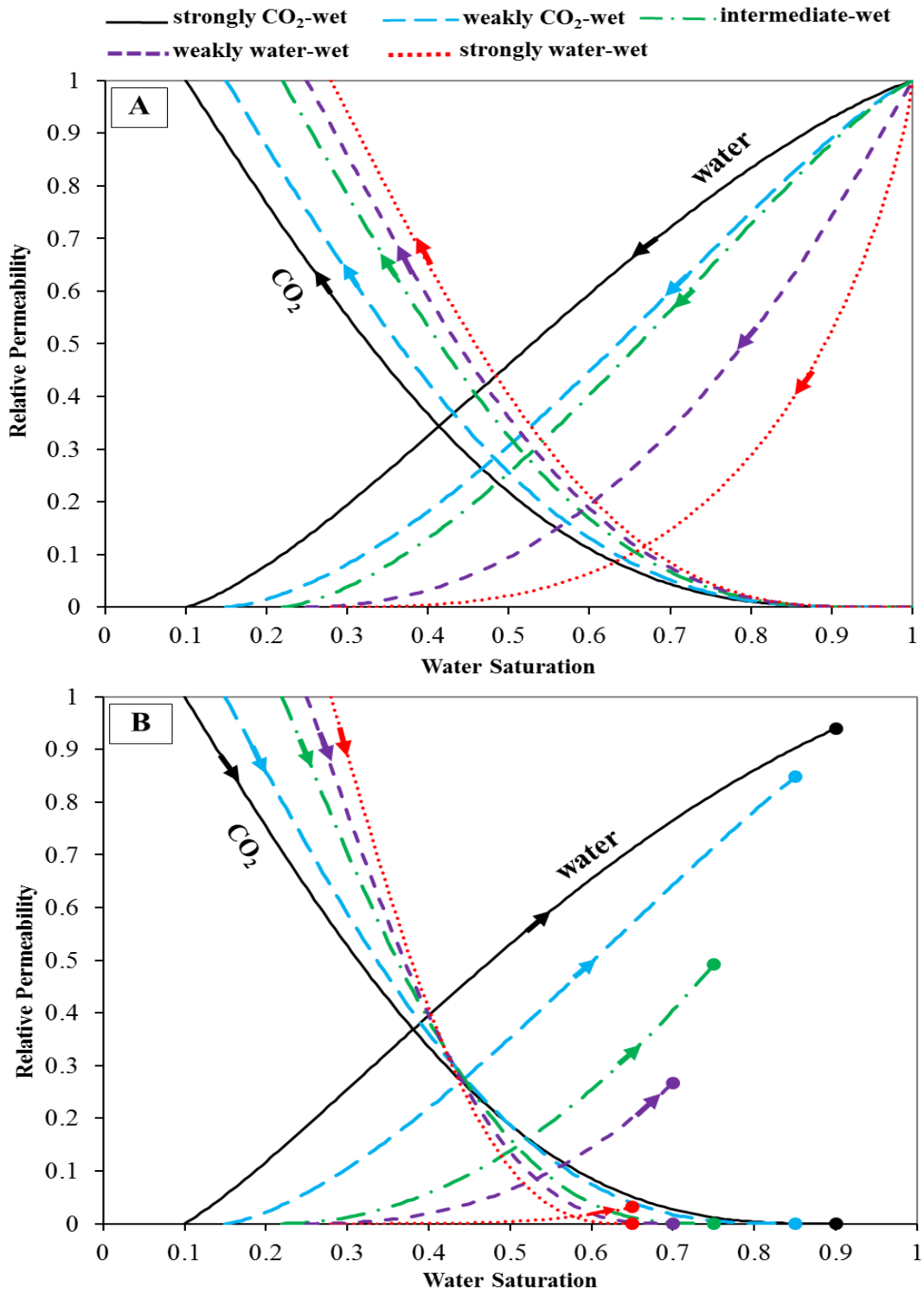


Figure 5-4 Relative permeability curves used for both injection well configurations (vertical and horizontal well) for 5 different rock wettability scenarios: A) CO₂ injection process (drainage); B) CO₂ storage process (imbibition).

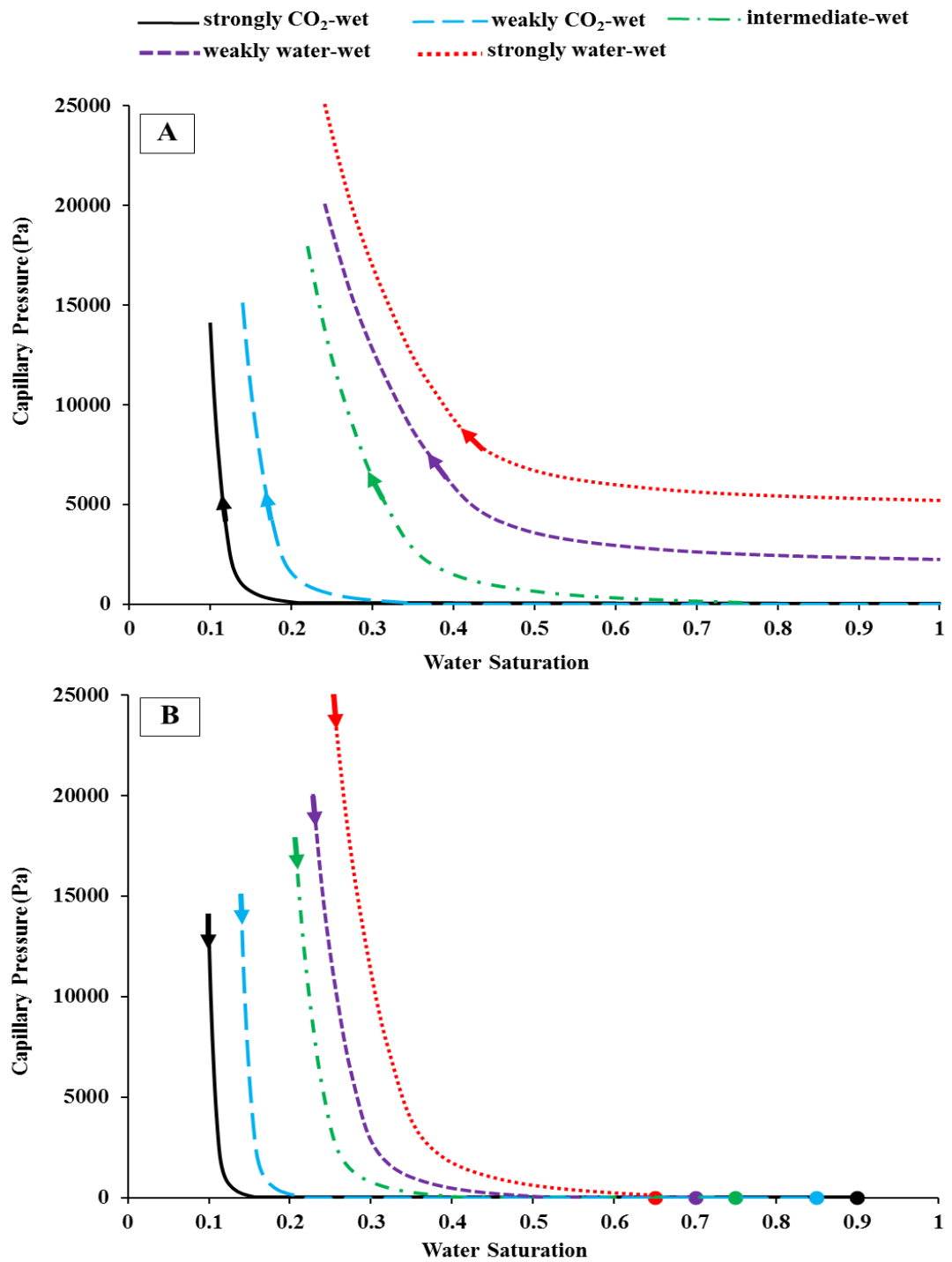


Figure 5-5 Capillary pressure curves used for both injection well configurations (vertical and horizontal well) models for 5 different rock wettability scenarios: A) CO₂ injection process (drainage); B) CO₂ storage process (imbibition).

Initially, at the beginning of the injection period, the reservoir is fully saturated with water ($S_w = 100\%$, $S_g = 0$) so at the start of the CO_2 injection process $k_{rw} = 1$ and $k_{rg} = 0$ (Figure 5-4A, top right). Then during the progress of the CO_2 injection, k_{rw} decreases gradually until it reaches its minimum value $k_{rw} = 0$ at the residual water saturation (S_{wr}) (as represented by the arrows on the 5 curves related to water relative permeability, Figure 5-4A). Simultaneously, k_{rg} increases from zero at the residual gas saturation (S_{gr}) to one at the residual water saturation (S_{wr}). For the storage period, (Figure 5-4B), k_{rg} decreases from one at S_{wr} to zero at S_{gr} and k_{rw} increases from zero at S_{gr} until it reaches its maximum value (according to the specified wettability condition at S_{wr} Craig (1993)). Here it is important to mention that both S_{wr} and S_{gr} are a function of wettability and that strongly water-wet rocks have the highest S_{wr} and S_{gr} values (0.26 and 0.35, respectively), while strongly CO_2 -wet rocks have the lowest S_{wr} and S_{gr} values (0.1 and 0.1, respectively) (Anderson 1987a, b; Craig, 1993; Iglauer et al., 2011; Pentland et al., 2011; Krevor et al., 2012; Chaudhary et al., 2013; Akbarabadi and Piri, 2013; Ruprecht et al., 2014; Rahman et al., 2016; Al-Menhali et al., 2016; Al-Khdheawi et al., 2017). Furthermore, the water saturation (S_w) at which k_{rw} and k_{rg} are equal should be less than 0.5 in the water-wet rock curves, and more than 0.5 in the CO_2 -wet rock curves, according to Craig (1993). For capillary pressures (Figure 5-5), the capillary pressure curves also start from the lowest capillary pressure values at the beginning of the injection period until reaching the maximum capillary pressure values, for each wettability case, at S_{wr} (Figure 5-5A). Then, during the storage period, capillary pressure values reduces again until they reach zero at S_{gr} , for all wettability conditions (Figure 5-5B) (Anderson, 1987a; Pentland et al., 2011; Pini et al., 2012, 2013; Krevor et al., 2015, 2016).

5.3 Results and discussion

5.3.1 Influence of CO_2 injection well configuration and rock wettability on CO_2 plume behaviour

By comparing the shape of the plume, at a given time after the end of the injection period and for a given rock wettability, for the three vertical well scenarios (i.e. one well, two wells, and 4 wells) (Figure 5-6 to Figure 5-8), respectively, and for the horizontal one (Figure 5-9), it is clear that the injection well configuration and rock

wettability have a significant impact on the CO₂ plume behaviour (i.e. its shape, its vertical and horizontal migration). For all rocks wettability scenarios, the CO₂ plume for all vertical well configurations (scenarios V1, V2 and V4), reached a shallower depth (i.e. closer to the top of the reservoir) than for the horizontal well configuration (scenario H); also, for all injection well configurations, the more CO₂-wet rock has the shallower CO₂ plume depth than the more water-wet rock, as reported in Table 5-1 and Figure 5-10 (e.g. V1 = 1217 m, V2 = 1221 m, V4 = 1226 m and H = 1246 m for the strongly water-wet reservoir, while scenarios V1, V2, and V3 = 800 m (i.e. the top depth of the model) and H = 955 m for the strongly CO₂-wet reservoir, at the end of the storage period of 200 years). Moreover, the total migration distance of the CO₂ plume (the distance between the lowest depth and the highest depth reached by the CO₂ plume) in all vertical well models was higher than in the horizontal well model, for all different rock wettability reservoir cases; also, for all injection well configurations, the total CO₂ plume migration distance is higher in the more CO₂-wet rock than in the more water-wet rock, as reported in Table 5-1 and Figure 5-11 (e.g. V1 = 156 m, V2 = 152 m, V4 = 147 m and H = 127 m for the strongly water-wet, while scenarios V1, V2, and V3 = 573 m and H = 418 m for the strongly CO₂-wet, at the end of storage period). Importantly, the CO₂ plume has a higher vertical migration distance in the vertical well configurations than in the horizontal well configuration, because the horizontal well has a larger contact area than the vertical well. Thus, for the horizontal well, the CO₂ lateral movement will be higher than its vertical migration, which is beneficial for controlling CO₂ upward migration. Furthermore, CO₂ plume migration is highly affected by the residual gas saturation (S_{gr}) and increasing S_{gr} leads to a reduction in vertical CO₂ migration (Metz et al., 2005; Doughty, 2010). S_{gr} is in turn a function of reservoir wettability (i.e. S_{gr} reduced from 0.35 in strongly water-wet rocks to 0.1 in strongly CO₂-wet rocks, Figure 5-4) (Anderson 1987a, b; Craig, 1993; Iglauer et al., 2011; Pentland et al., 2011; Krevor et al., 2012; Chaudhary et al., 2013; Akbarabadi and Piri, 2013; Ruprecht et al., 2014; Rahman et al., 2016; Al-Menhali et al., 2016; Al-Khdheawi et al., 2017).

In summary, the results demonstrate that injection well configuration and rock wettability have a significant impact on the CO₂ plume behaviour. Our results indicate that the use of horizontal well technology, for all rock wettability conditions, or the

more water-wet reservoirs, for both well configurations, will reduce the vertical CO₂ plume migration, which implies an improved CO₂ storage process.

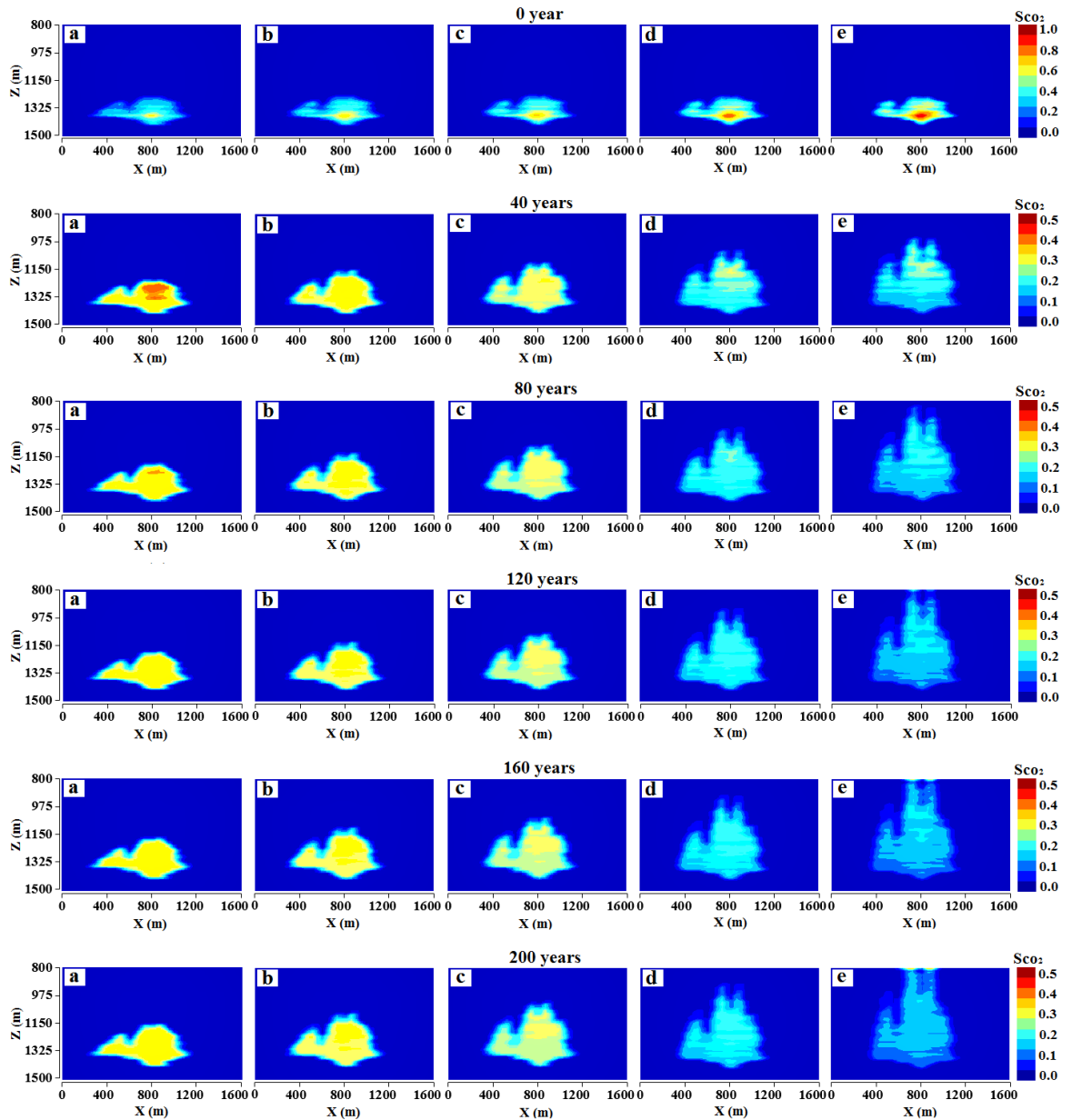


Figure 5-6 2D vertical cross-sections through the centre of the storage reservoir. CO₂ is injected at a depth of (1373 m) through a one vertical well. The CO₂ plume shape and height in the reservoir are shown for 5 different rock wettability types: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. 2 Mt of CO₂ was injected in total.

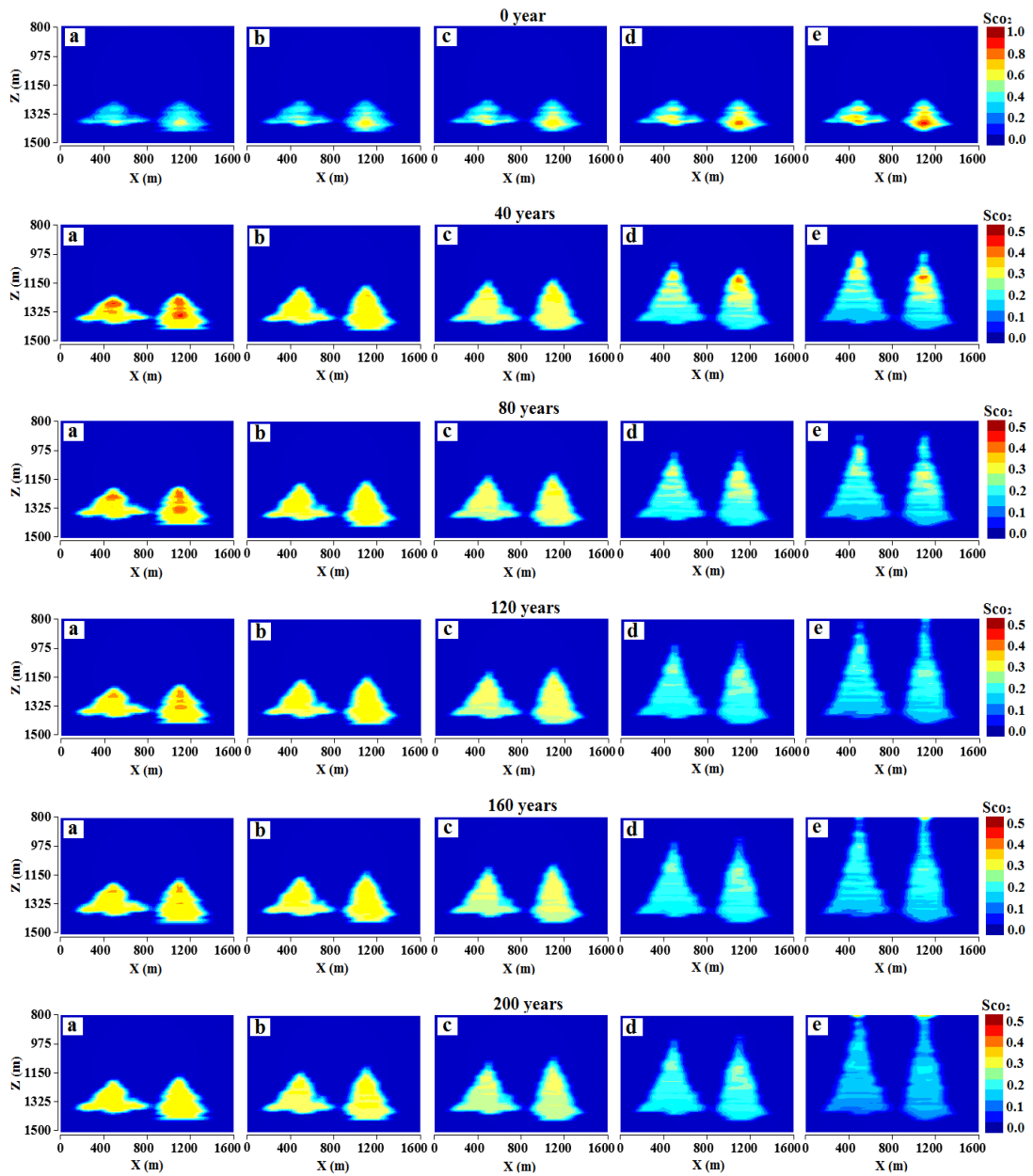


Figure 5-7 2D vertical cross-sections through the centre of the storage reservoir. CO₂ is injected at a depth of (1373 m) through the a two vertical wells. The CO₂ plume shape and height in the reservoir are shown for 5 different rock wettability types: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. 2 Mt of CO₂ was injected in total.

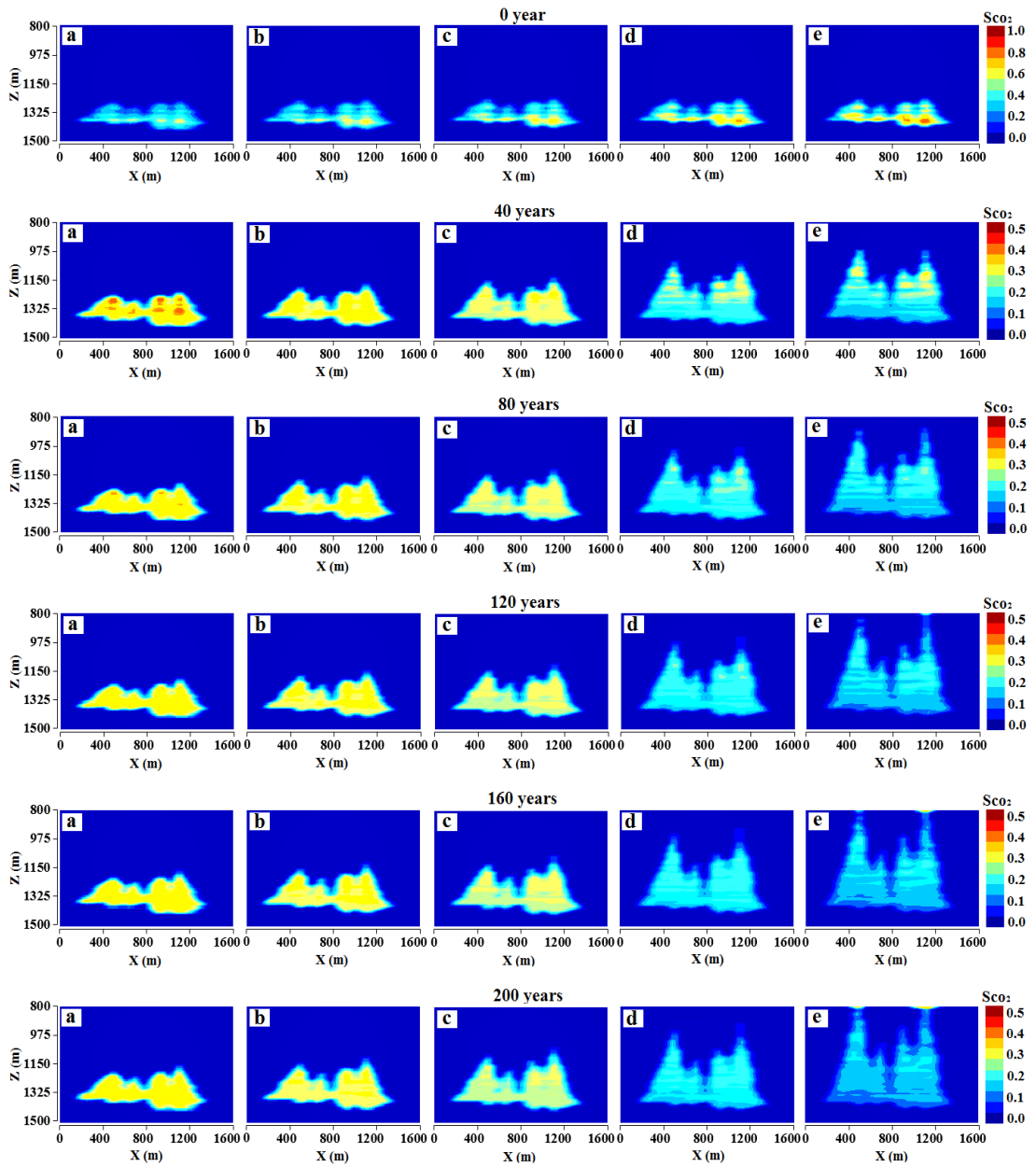


Figure 5-8 2D vertical cross-sections through the centre of the storage reservoir. CO₂ is injected at a depth of (1373 m) through a four vertical wells. The CO₂ plume shape and height in the reservoir are shown for 5 different rock wettability types: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. 2 Mt of CO₂ was injected in total.

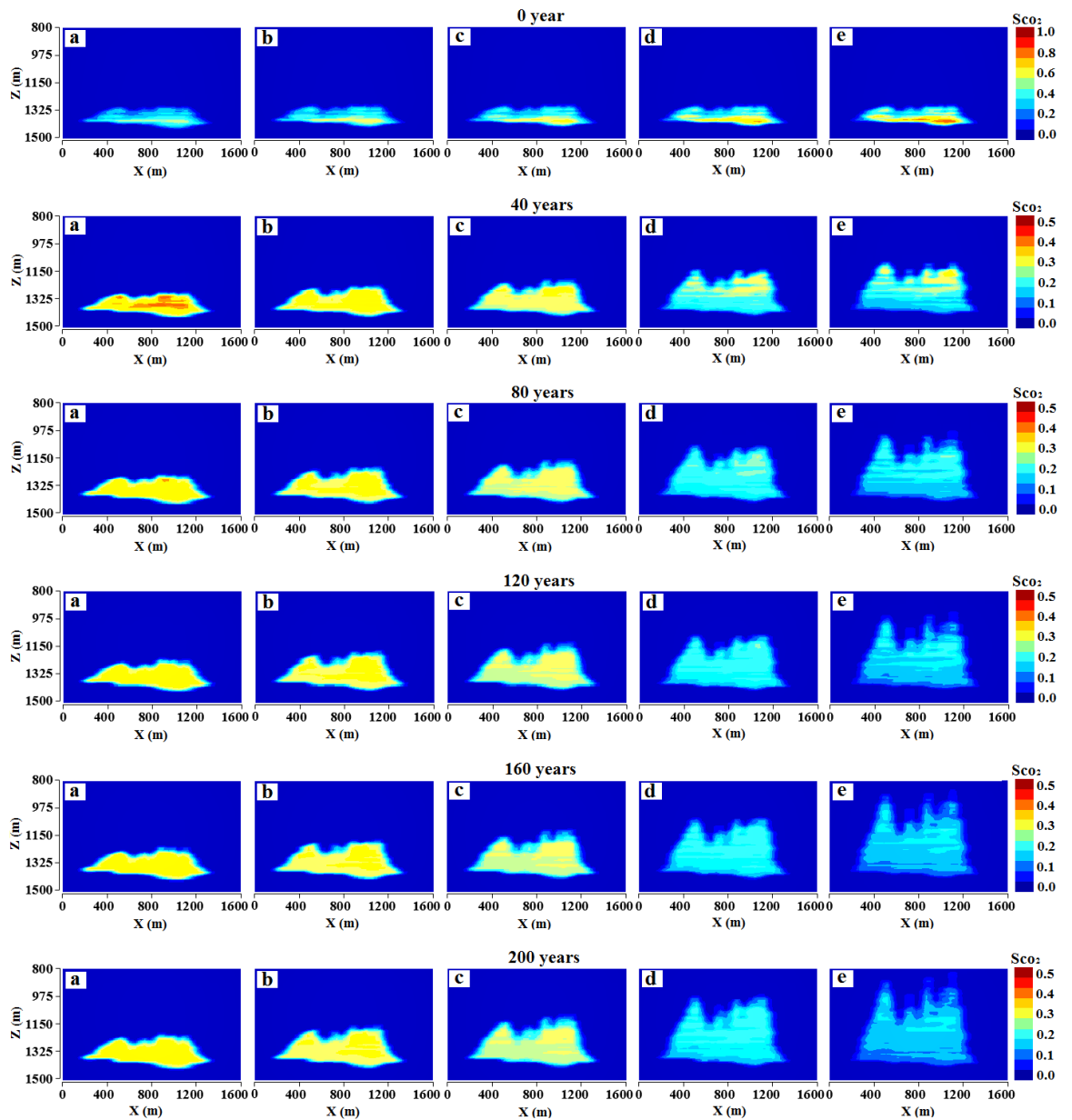


Figure 5-9 2D vertical cross-sections through the centre of the storage reservoir. CO₂ is injected at a depth of (1373 m) through a horizontal well. The CO₂ plume shape and height in the reservoir are shown for 5 different rock wettability types: a) strongly water-wet; b) weakly water-wet; c) intermediate-wet; d) weakly CO₂-wet; e) strongly CO₂-wet. 2 Mt of CO₂ was injected in total.

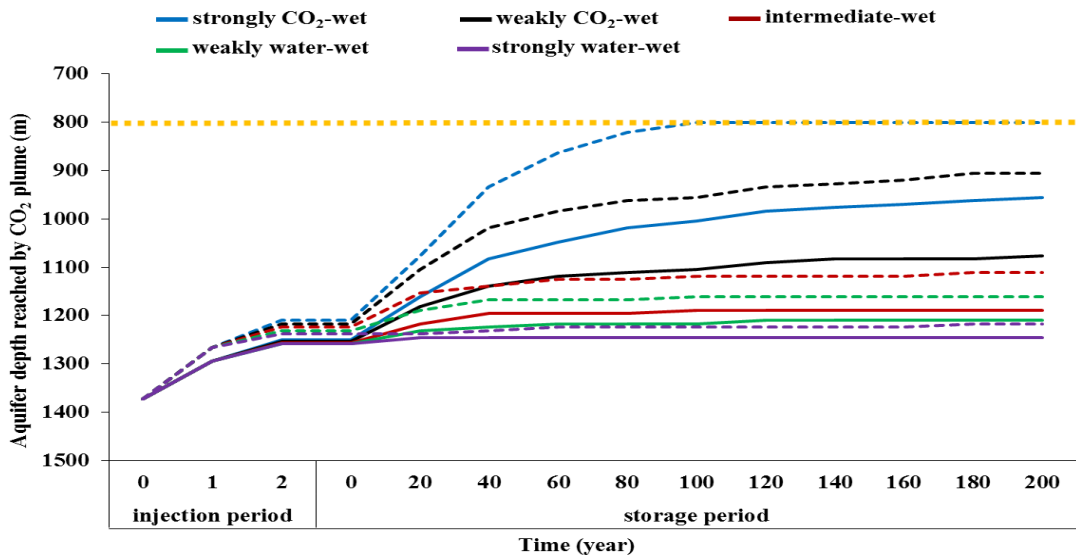


Figure 5-10 Aquifer (reservoir) depth reached by CO₂ plume as a function of injection well configuration (dashed line represents vertical well (i.e. one vertical well) and solid line represents horizontal well) and injection and storage time for different rock wettability reservoirs. The square dot yellow (horizontal) line represents the caprock seal depth at the top of the reservoir. For the two and four vertical well scenarios the CO₂ plume depths reached are very similar. Thus, here we have presented only the one vertical well scenario for simplicity.

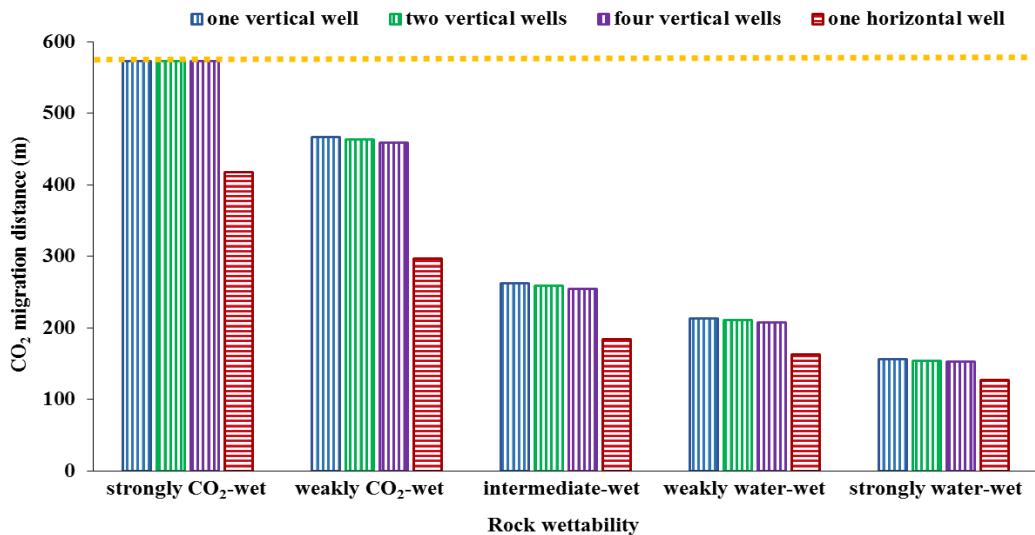


Figure 5-11 CO₂ plume migration distance as a function of injection well configuration and rock wettability at the end of storage period (200 years). The dashed yellow (horizontal) line represents the maximum possible migration distance (i.e. CO₂ reaches the caprock seal at the top of the reservoir).

Table 5-1 Depth^a reached and vertical migration distance of CO₂ plume after the end of the storage period (200 years) for the vertical wells and horizontal well in different rock wettability scenarios

CO ₂ plume depth and migration distance	well configuration	strongly CO ₂ -wet ^b	weakly CO ₂ -wet	Intermediate-wet	weakly water-wet	strongly water-wet
Depth reached by CO ₂ plume (m)	one vertical well	800	906	1111	1160	1217
	two vertical wells	800	915	1120	1167	1221
	four vertical wells	800	927	1132	1175	1226
	one horizontal well	955	1076	1189	1210	1246
Vertical CO ₂ plume migration distance (m)	one vertical well	573	467	262	213	156
	two vertical wells	573	458	253	206	152
	four vertical wells	573	446	241	198	147
	one horizontal well	418	297	184	163	127
Percentage of vertical migration distance to the maximum possible migration distance (%)	one vertical well	100	82	46	37	27
	two vertical wells	100	81	45	37	27
	four vertical wells	100	80	45	36	27
	one horizontal well	73	52	32	28	22

^aCO₂ injection depth is at (1373 m).

^bIn the scenario of the vertical well in the strongly CO₂-wet rock, CO₂ plume reached the caprock seal at the top of the reservoir (800 m) after only 100 years of storage period; CO₂ then flowed laterally beneath the caprock.

5.3.2 Influence of CO₂ injection well configuration and rock wettability on trapping mechanisms

For each wettability case, Figure 5-12 to Figure 5-14 and Table 5-2 report and compare the amount of mobile, solubility trapped and residually trapped CO₂, for each of the two configurations of the wells. Firstly, the results show that use of vertical wells will lead to a greater amount of free CO₂ (mobile CO₂) at the end of the storage period than that when a horizontal well was used, for all rock wettability conditions; also the results show that the more CO₂-wet rock has more CO₂ mobility, for both vertical and horizontal well configurations (e.g., the total amount of mobile CO₂ at the end of the storage period (after 200 years), V1 = 129 kton, V2 = 127 kton, V4 = 126 kton, and H = 94 kton for the strongly water-wet reservoir, while V1 = 809 kton, V2 = 801 kton, V4 = 791 kton, and H = 724 kton for the strongly CO₂-wet reservoir (Figure 5-12)). In addition, the results demonstrate that the use of vertical wells will result in more solubility trapped CO₂ than that obtained with a horizontal injection well; also the more CO₂-wet rock has more solubility trapped CO₂ than that in the more water-wet rock (e.g., the total amount of solubility trapped CO₂ at the end of the storage period (after 200 years), V1 = 271 kton, V2 = 269 kton, V4 = 268 kton, and H = 253 kton for the strongly water-wet reservoir, while V1 = 558 kton, V2 = 556 kton, V4 = 555 kton, and H = 513 kton for the strongly CO₂-wet reservoir (Figure 5-13)). Importantly, the amount of trapped CO₂ increases with increasing plume migration as a result of increasing the brine-CO₂ contact area (Doughty, 2010). Thus, both vertical wells configurations combined with the more CO₂-wet rocks have the highest CO₂ plume migration distance, and consequently higher solubility trapping capacity.

Finally, the results show that the use of horizontal well is associated with more residually trapped CO₂ than in the case of the vertical injection wells, for all wettability conditions; also the results show that the residually trapped CO₂ reduces with increasing the CO₂ wettability of the rock, for both of the injection well configurations (e.g., the total amount of residually trapped CO₂ at the end of the storage period (after 200 years), V1 = 1600 kton, V2 = 1604 kton, V4 = 1606 kton, and H = 1653 kton for the strongly water-wet reservoir, while V1 = 633 kton, V2 = 643 kton, V4 = 654 kton, and H = 763 kton for the strongly CO₂-wet reservoir; Figure 5-14). Importantly, residual trapping is a function of S_{gr} and increasing S_{gr} leads to improve the residual

trapping. Recall, strongly water-wet rocks have the highest S_{gr} value (0.35), while strongly CO_2 -wet rocks have the lowest S_{gr} value (0.1) (Anderson 1987a, b; Craig, 1993; Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016), Figures Figure 5-4 and Figure 5-5. Thus, residual trapping increases with increasing water-wettability. These results, related to the influence of wettability on residual trapping of CO_2 , are in line with previous experimental studies (cp. Iglauer et al., 2011 and Andrew et al., 2013 versus Chaudhary et al., 2013 and Rahman et al., 2016). In summary, our simulation results conclude that the injection well configuration and rock wettability has an important impact on the amounts of mobile and trapped CO_2 and the use of horizontal well and the more water-wet rock will enhance the CO_2 storage process.

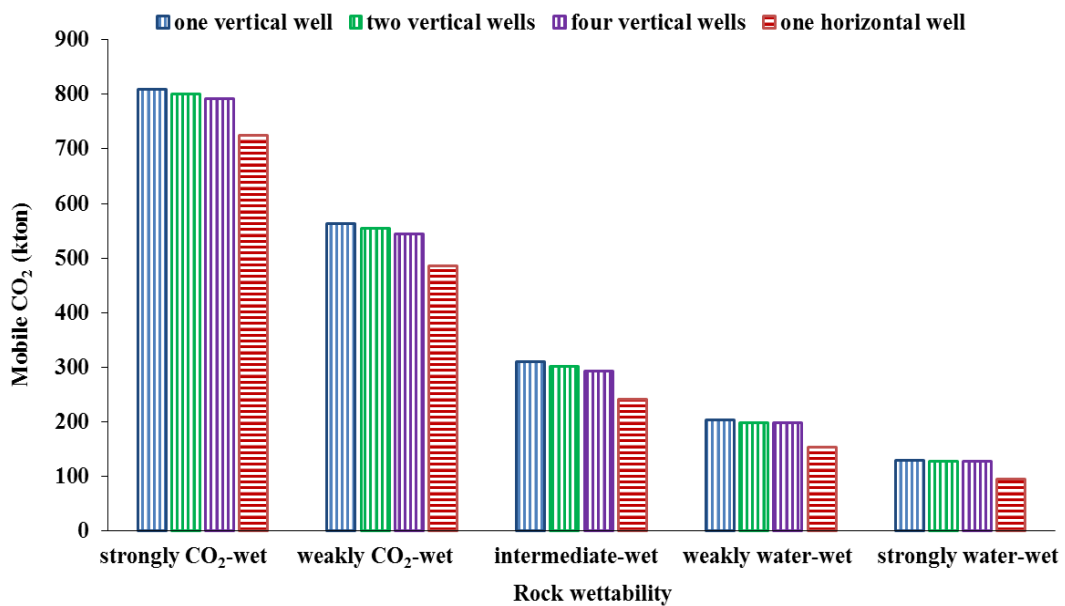


Figure 5-12 Total amount of mobile CO_2 (in thousand tons) as a function of injection well configuration and rock wettability at the end of storage period (200 years). 2 Mt of CO_2 was injected in total.

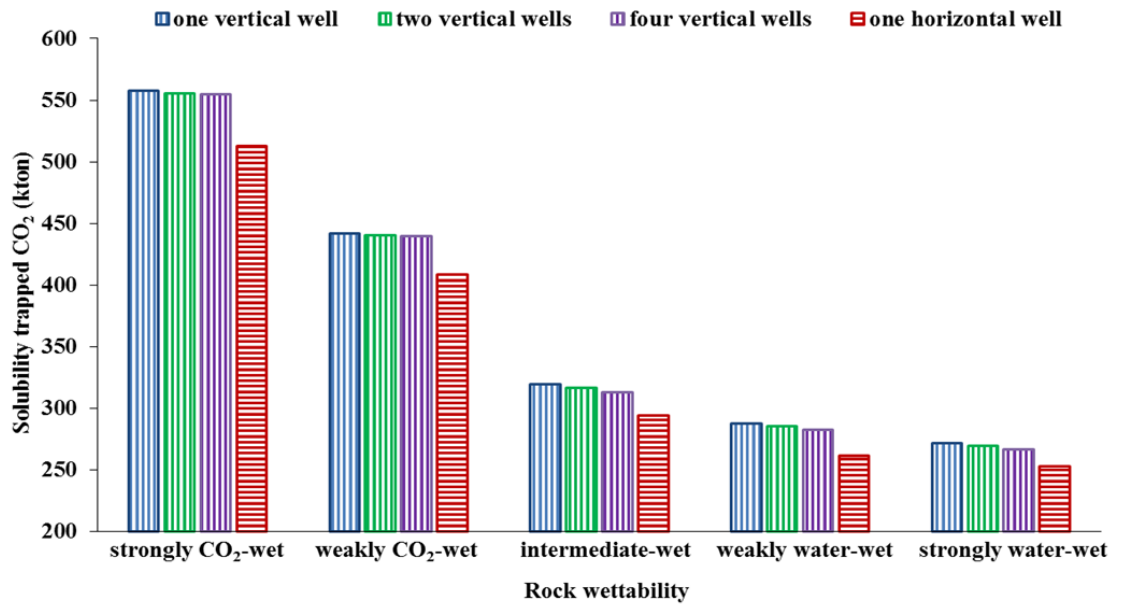


Figure 5-13 Total amount of solubility trapped CO₂ (in thousand tons) as a function of injection well configuration and rock wettability at the end of storage period (200 years). 2 Mt of CO₂ was injected in total.

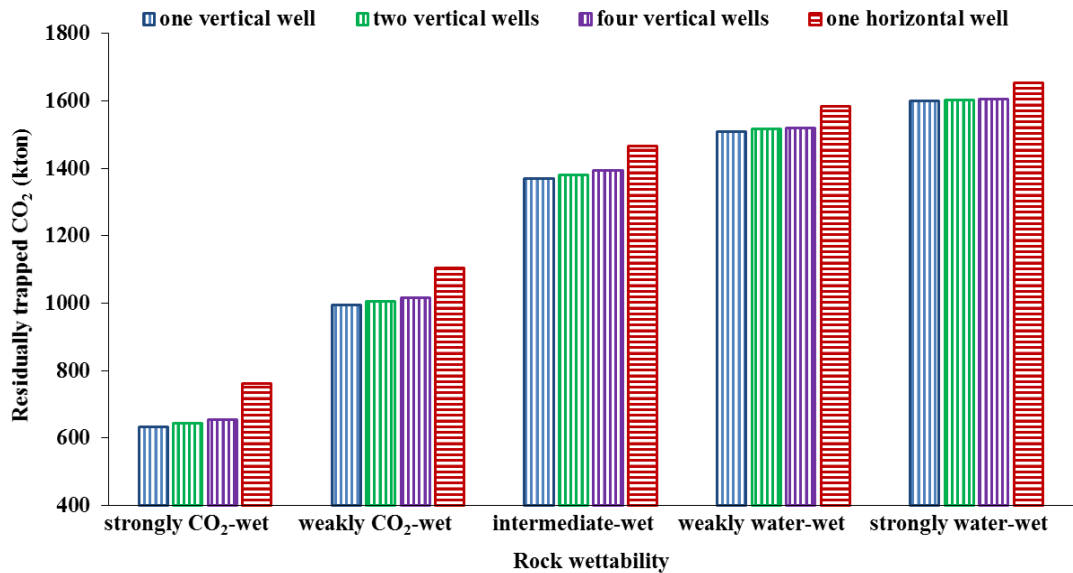


Figure 5-14 Total amount of residually trapped CO₂ (in thousand tons) as a function of injection well configuration and rock wettability at the end of storage period (200 years). 2 Mt of CO₂ was injected in total.

Table 5-2 CO₂ mobility and capacity of trapping mechanisms for horizontal and vertical wells in different rock wettability scenarios at the end of storage period (200 years). For all wettability scenarios and in all injection well configurations (vertical and horizontal) 2 Mt of CO₂ was injected in total

Amount and percentage of mobile and trapped CO ₂		well configuration	strongly CO ₂ -wet	weakly CO ₂ -wet	Intermediate-wet	weakly water-wet	strongly water-wet	
Total amount of mobile and trapped CO ₂ (Thousand tons)	Mobile	one vertical	809	563	310	203	129	
		two vertical	801	554	301	199	127	
		four vertical	791	543	293	198	127	
		one horizontal	724	486	240	153	94	
	Solubility trapped	one vertical	558	441	320	287	271	
		two vertical	556	441	317	285	269	
		four vertical	555	440	312	282	266	
		one horizontal	513	408	294	262	253	
	Residually trapped	one vertical	633	996	1370	1510	1600	
		two vertical	643	1005	1382	1516	1604	
		four vertical	654	1017	1395	1520	1607	
		one horizontal	763	1106	1466	1585	1653	
	Percentage of mobile and trapped CO ₂ to the total injected CO ₂ (%)	Mobile	one vertical	40	28	15	10	6
			two vertical	40	28	15	10	6
			four vertical	40	27	15	10	6
			one horizontal	36	24	12	8	5
Solubility trapped		one vertical	28	22	16	14	14	
		two vertical	28	22	16	14	13	
		four vertical	28	22	16	14	13	
		one horizontal	26	21	15	13	12	
Residually trapped		one vertical	32	50	69	76	80	
		two vertical	32	50	69	76	81	
		four vertical	32	51	69	76	81	
		one horizontal	38	55	73	79	83	

5.4 Conclusions

CO₂ capture and storage (CCS) is highly affected by various factors (e.g. caprock properties, permeability anisotropy, aquifer temperature, aquifer depth, CO₂ injection scenarios, porosity and permeability heterogeneity (cp. the reference for each factor in section 5.1 of this chapter). Other possible affecting factor, which has not received a sufficient attention, is the injection well configuration. Even though the use of horizontal well type has been addressed as an important factor to enhance the oil productivity from the oil reservoirs (Kossack et al., 1987; Reiss, 1987; Macdonald, 1988; Joshi, 1988, 2003; Babu and Odeh, 1989; Hardman, 1989; Murphy, 1990; Economides et al., 1991; Thomas, 2008), its impact on the efficiency of CO₂ storage process in saline aquifers has not been addressed.

Thus, for the first time, we have developed two sets of 3D heterogeneous reservoir simulation models to demonstrate the impact of the injection well configuration, horizontal versus three different vertical injection well scenarios (i.e. one, two and four vertical wells), on the CO₂ plume behaviour and CO₂ storage capacities for a wide range of rock wettabilities.

Our simulation results clearly indicate that, at the end of the 200 years storage period, the CO₂ injection well configuration has a significant impact on the CO₂ plume behaviour and storage capacities. A horizontal well, for all wettability conditions, reduced the CO₂ plume vertical mobility and the amount of the solubility trapped CO₂, while it increased the amount of residually trapped CO₂. Furthermore, our simulation results show that water-wet rocks improved the CO₂ storage capacities for both well configurations. Importantly, our results about the impact of rock wettability on the CO₂ storage capacities are consistent with our previous simulation study in a homogeneous reservoir (Al-Khdheawi et al., 2017) and previous experimental studies (cp. Iglauer et al., 2011; Andrew et al., 2013; Chaudhary et al., 2013; Rahman et al., 2016).

We conclude that technically the use of a horizontal well and geologically the more water-wet rock will significantly improve CO₂ storage capacity and reduce vertical CO₂ plume migration thus de-risking CO₂-projects.

This study has also important implications for reducing the cost associated with of CO₂ geo-sequestration. Joshi (2003) presented a study about the economic benefits of horizontal wells, which showed that the cost of drilling a horizontal well is 1.5 to 2.5

times higher than drilling one vertical well. However, we have found here that using one horizontal well leads to higher CO₂ storage capacities when compared to even 4 vertical wells.

Chapter 6 Effect of Wettability Heterogeneity and Reservoir Temperature on CO₂ Storage Efficiency in Deep Saline Aquifers

6.1 Introduction

CO₂ emissions, from large stationary points and different industrial activities, represent the majority of greenhouse gas emissions (Houghton et al., 2001; Zhang, 2016). To mitigate these emissions, Carbon Capture and Storage (CCS) is the most effective strategy by capturing and injecting the CO₂ into deep geological formations for long-term storage (Pruess et al., 2003; Lebedev et al., 2017; Oldenburg and Unger, 2003). These geological formations include depleted oil and gas reservoirs (Honari et al., 2016), unmineable coal seams (Shi and Durucan, 2005; Zhang et al., 2016b), and deep saline aquifers (DePaolo et al., 2007). Out of all these possible storage formations, deep saline aquifers are the most preferable as they have the highest storage capacity (Lackner, 2003). However, due to the density contrast between aquifer water and the injected CO₂, the low-density CO₂ migrates upwards and can potentially leak through wells, cap rock, and geological fault systems back towards the surface (Li and Liu, 2016; Vialle et al., 2016). There are various physical and chemical trapping mechanisms that can reduce or prevent this CO₂ migration; including structural trapping (Iglauer et al., 2015a; Naylor et al., 2011; Nilsen et al., 2012), residual trapping (Baz et al., 2016; Ide et al., 2007; Iglauer et al., 2011; Krevor et al., 2015; Pentland et al., 2011; Ruprecht et al., 2014; Suekane et al., 2008), dissolution trapping (Bachu and Adams, 2003; Emami-Meybodi et al., 2015; Iglauer, 2011; Lindeberg and Wessel-Berg, 1997; Mito et al., 2008; Pruess and Garcia, 2002; Spycher et al., 2003; Suekane et al., 2008) and mineral trapping (Bachu et al., 1994b; Gaus, 2010; Metz et al., 2005; Mito et al., 2008; Xu et al., 2003; Xu et al., 2004; Xu et al., 2005).

Various site characteristics can affect the efficiency of these trapping mechanisms, including injection well configuration (Al-Khdheawi et al., 2017d), caprock properties (Iglauer et al., 2015a), aquifer brine salinity (Al-Khdheawi et al., 2017a; Al-Khdheawi et al., 2018) or porosity and permeability heterogeneity (Al-Khdheawi

et al., 2017b; Ambrose et al., 2008; Doughty and Pruess, 2004; Doughty and Myer, 2009; Flett et al., 2007; Gershenson et al., 2015; Gershenson et al., 2016; Gershenson et al., 2017; Green and Ennis-King, 2010; Han et al., 2010; Hesse and Woods, 2010; Hovorka et al., 2004; Obi and Blunt, 2006; Saadatpoor et al., 2009; Zhou et al., 2010). In addition, it has been established that reservoir wettability has a significant impact on CO₂ plume migration and trapping mechanisms, and that water-wet rocks are preferable CO₂ sinks due to their slower vertical CO₂ migration and enhanced residual trapping capacity (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017e; Al-Khdheawi et al., 2017b; Iglauer et al., 2015a; Iglauer et al., 2015b; Iglauer, 2017; Krevor et al., 2015; Rahman et al., 2016).

However, in previous studies homogeneous wettability was assumed; and most studies assumed strongly water-wet conditions although this is a very unlikely scenario (Iglauer, 2017). Furthermore, it is well established that reservoir wettability is distributed heterogeneously at various length scales (from the microscale to the macroscale; (Aspenes et al., 2003; Chaouche et al., 1994; Graue et al., 2002; Masalmeh, 2002; Morrow et al., 1986; Spinler et al., 2002; Standnes and Austad, 2000; Vizika and Duquerroix, 1997).

Multiple factors are responsible for this, ranging from adsorption of organics, pore-surface roughness heterogeneity, grain size heterogeneity, chemical heterogeneity of the mineral surface, differences in surface chemistry, reservoir permeability heterogeneities and associated reservoir fluid movements, or depth in the reservoir (Crocker and Marchin, 1988; Drelich and Miller, 1994; Gaydos and Neumann, 1987; Iglauer et al., 2015a; Iglauer et al., 2015b; Iglauer, 2017; Jafari and Jung, 2016; Laroche et al., 1999; Li, 1996; Lin et al., 1993; Morrow et al., 1986; Saghafi et al., 2014; Van Lingen et al., 1996; Vizika and Duquerroix, 1997). It is also well established that wettability heterogeneity has a significant influence on the fluid displacement mechanisms and phase distributions in gas injection processes (Laroche et al., 1999; Anderson, 1986c; Anderson, 1987a; Anderson, 1987b; Chang et al., 1997; Morrow, 1990; Bertin et al., 1998; Blunt, 1997; Kiriakidis et al., 1993).

There is thus a serious gap in knowledge with respect to how wettability heterogeneity influences the CO₂ storage efficiency. Furthermore, the impact of reservoir temperature on CO₂ storage efficiency has not been examined systematically.

Here we thus investigate the effect of wettability heterogeneity (at isothermal and non-isothermal conditions) and reservoir temperature on vertical CO₂ plume migration and dissolution and residual trapping storage capacities.

6.2 Methodology

6.2.1 Numerical model: underlying theory

We developed a 3D heterogeneous porosity and permeability reservoir model using a nonisothermal multicomponent multiphase flow simulator (TOUGH2; Pruess et al., 1999) and the ECO2M equation of state (Pruess, 2011) to predict the thermodynamic and thermophysical properties of the H₂O-NaCl-CO₂ mixtures including super- and sub-critical CO₂ phases, and CO₂ phase changes. The mass and energy balances describing the fluid and heat flow in these (multiphase and multicomponent) systems are given by:

$$\frac{d}{dt} \int_{V_n} M^k dV_n = \int_{\Gamma_n} F^k \cdot \mathbf{n} d\Gamma_n + \int_{V_n} Q^k dV_n \quad (6.1)$$

where:

V_n is an arbitrary sub-domain which is bounded by the closed surface Γ_n , M is the mass or energy per volume, k ($= 1$ to n) is a labeling factor representing different mass components (e.g. water, CO₂, air, solutes...etc.), F is the mass or heat flux, Q is the sink (source) term, \mathbf{n} is the normal vector on surface element $d\Gamma_n$, pointing inward into V_n .

In addition, mass accumulation can be computed in TOUGH2 using the following formula:

$$M^k = \phi \sum_{\psi} S_{\psi} \rho_{\psi} X_{\psi}^k \quad (6.2)$$

where:

ϕ is rock porosity, ψ identifies fluid phase, S_{ψ} is the saturation of phase ψ , ρ_{ψ} is the density of phase ψ , and X_{ψ}^k is the mass fraction of component k in phase ψ .

The individual phase mass flux, which can be used to compute the cumulative mass flux, can be predicted via the multiphase version of Darcy's law:

$$F_{\psi} = \rho_{\psi} u_{\psi} = -k \frac{k_{r\psi} \rho_{\psi}}{\mu_{\psi}} (\nabla P_{\psi} - \rho_{\psi} \mathbf{g}) \quad (6.3)$$

where:

F_{ψ} = individual phase mass flux, u_{ψ} = Darcy velocity of phase ψ ,

k = absolute permeability, $k_{r\psi}$ = relative permeability of phase ψ ,

μ_{ψ} = viscosity of phase ψ , P_{ψ} = pressure of phase ψ .

6.2.2 Model characteristics

The reservoir model had the dimensions: 1600 m length, 1200 m width and 700 m height (Figure 6-1); these were split in $37 \times 33 \times 80$ cells (=97680 cells). For simulating a highly heterogeneous reservoir, the heterogeneous porosity and permeability data of the SPE comparative solution project (Christie and Blunt, 2001) were used (Figure 6-1). The ratio of vertical to horizontal permeability (k_v/k_h) for the storage reservoir was 0.1 and it was $10E-6$ (10^{-6}) for the caprock (i.e. the top boundary of the model at 800 m depth) to prevent vertical flow at the top of the model (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b). The model was initially fully saturated with water (water saturation = 100%) with an initial salinity of 15 wt% NaCl. The initial pressure at the top of the model at 800 m depth was 8 MPa. This initial pressure increased to 15 MPa at the bottom of the reservoir model at 1500 m depth following a pressure gradient of 10 MPa/km (Dake, 2007). Constant pressure boundary conditions were prescribed by expanding the volumes of the lateral outer boundary cells with a large volume modifier of $10E8$ (10^8) for each outer boundary cell (Nghiem et al., 2010).

Two different reservoir temperature conditions were considered (i.e. non-isothermal and isothermal conditions) for both heterogeneous and homogeneous wettability scenarios. For the non-isothermal conditions, a geothermal gradient was applied using a vertical temperature gradient of $30 \text{ }^{\circ}\text{C}/\text{km}$ (Turcotte and Schubert, 2014; Grant and Bixley, 2011; Doughty et al., 2008) with a surface temperature of $25 \text{ }^{\circ}\text{C}$. For the isothermal conditions, and in order to cover most temperatures of interest for subsurface CO_2 storage, a wide range of reservoir and CO_2 injection temperatures was tested (i.e. 303 K, 313 K, 323 K, 333 K, 343 K and 353 K) for both wettability

distribution scenarios. The dynamic pressure-temperature effects (Joule-Thompson) have been included in the reservoir model (Katz and Lee, 1990; Pruess, 2005; Oldenburg, 2007).

CO₂ was then injected at a depth of 1373 m at an injection rate of 1 Mt/year for a period of 10 years (i.e. the total amount of injected CO₂ was 10 Mton) with an injection temperature of 338 K for the non-isothermal scenario. Note that this injection rate is similar to that used in the Sleipner (Leung et al., 2014; Torp and Gale, 2004) and Quest (Bourne et al., 2014) CCS projects. A 500 year post-injection (storage) periods were then simulated, where the migration and behaviour of the CO₂ plume was examined, and the vertical CO₂ migration distance and dissolution and residual trapping capacities were predicted.

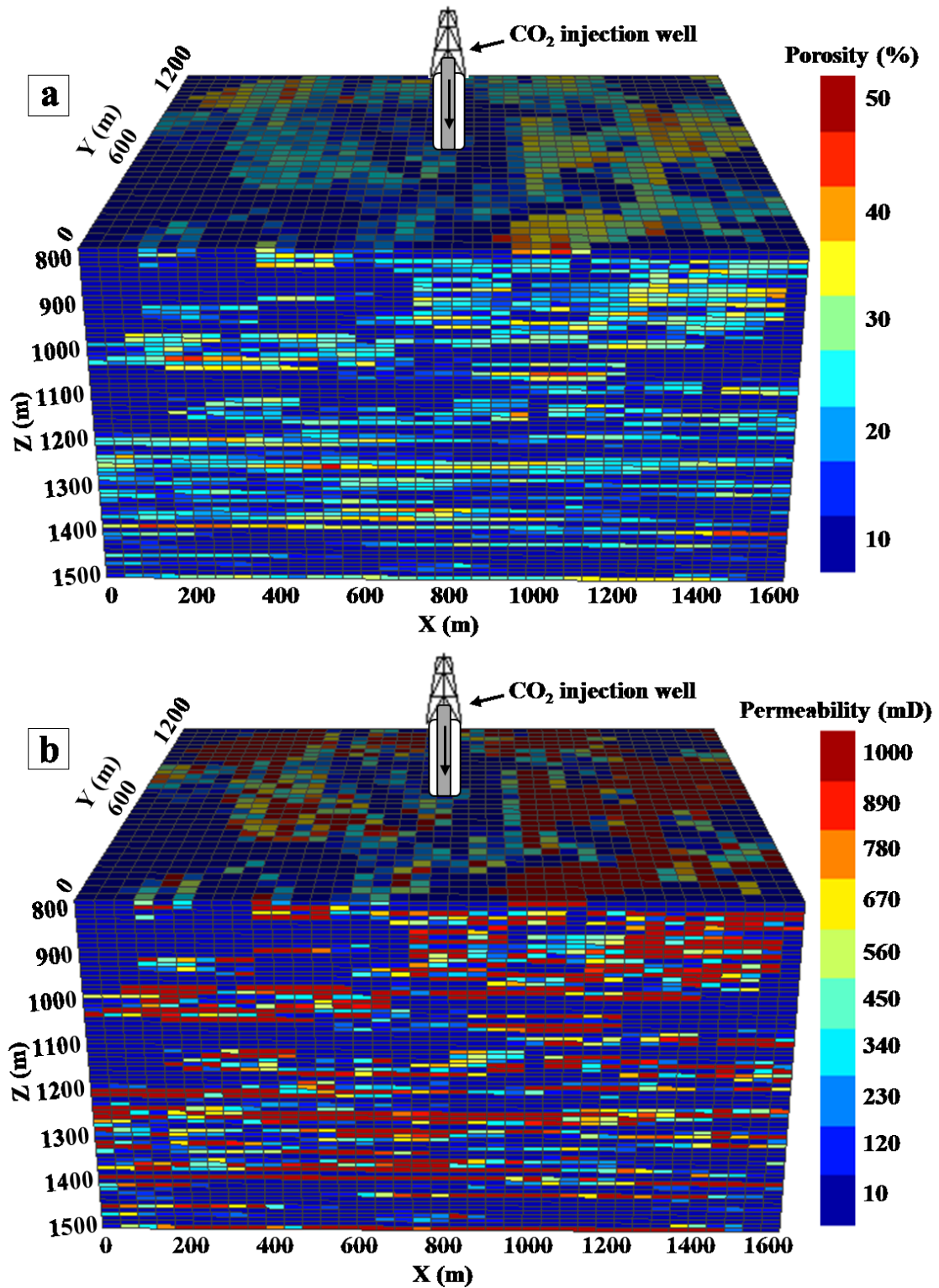


Figure 6-1 3D views of the heterogeneous reservoir model showing the location of the CO₂ injection well, model dimensions and a) heterogeneous porosity field; b) heterogeneous permeability field.

6.2.3 Simulation of homogeneous and heterogeneous wettability

It is clear that reservoir wettability can vary widely (Iglauer, 2017), and that wettability strongly influences relative permeabilities (Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017d; Al-Khdheewi et al., 2017b; Anderson, 1987b; Krevor et al., 2012; Levine et al., 2014; McCaffery and Bennion, 1974; Owens and Archer, 1971), capillary pressures (Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017d; Al-Khdheewi et al., 2017b; Anderson, 1987a; Batycky et al., 1981; Melrose, 1965; Morrow, 1976; Pini et al., 2013; Pini et al., 2012) and the associated residual gas and water saturations (Akbarabadi and Piri, 2013; Al-Menhali et al., 2016; Chaudhary et al., 2013; Craig, 1993; Iglauer et al., 2011; Pentland et al., 2011; Rahman et al., 2016; Ruprecht et al., 2014).

Wettability therefore strongly influences reservoir-scale fluid dynamics (Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017e; Al-Khdheewi et al., 2017b). We thus used our recently developed relative permeability and capillary pressure curves to simulate specific wettability scenarios (Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017d; Al-Khdheewi et al., 2017b). Note that our previous developed relative permeability and capillary pressure curves were built based on various experimental studies (e.g. Melrose, 1965; Owens and Archer, 1971; McCaffery and Bennion, 1974; Morrow, 1976; Batycky et al. 1981; Anderson, 1987a, 1987b; Craig, 1993; Pentland et al., 2011; Krevor et al., 2012; Pini et al., 2012; Pini et al., 2013; Levine et al., 2014). To mimic heterogeneous wettability, five wettabilities (i.e. strongly water-wet, weakly water-wet, intermediate-wet, weakly CO₂-wet and strongly CO₂-wet) were randomly assigned to different cells using a discrete probability distribution. To represent the deep saline aquifer, 5% of the rock was assigned strongly water-wet, 30% weakly water-wet, 45% intermediate-wet, 15% weakly CO₂-wet and 5 % strongly CO₂-wet (Figure 6-2). These distributions were chosen for two reasons: a) we have followed the more likely percentages in an actual saline aquifer (Iglauer et al. 2015a, b, 2017) and b) the weighted averages of the residual gas saturation and residual water saturation for this nonuniform wettability distribution are equal to that used in the homogenous wettability model (intermediate-wet), where $S_{gr} = 0.25$ and $S_{wr} = 0.22$ (Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017d; Al-Khdheewi et al., 2017b). Without any further studies available in literature, as regard any correlation between wettability distribution and other petrophysical or geological parameters (i.e. permeability, porosity and mineralogy) we

have used here a random distribution. Indeed, many previous studies, which discuss the wettability heterogeneity in oil reservoirs, used random distribution to represent the wettability heterogeneity in the reservoir (e.g. (Aspenes et al., 2003; Chaouche et al., 1994; Graue et al., 2002; Masalmeh, 2002; Morrow et al., 1986; Spinler et al., 2002; Standnes and Austad, 2000; Vizika and Duquerroix, 1997)), this is thus the established approach which we follow here.

The weighted averages of residual gas saturation (\bar{S}_{gr}) and residual water saturation (\bar{S}_{wr}) in the heterogeneous wettability model were calculated using the above percentages of each wettability condition with our recently used values of residual gas and residual water saturation for the different wettability scenarios (Table 6-1; (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d; Al-Khdheawi et al., 2017b)).

Thus the weighted average of the residual gas saturation (\bar{S}_{gr}) and residual water saturation (\bar{S}_{wr}) for the heterogeneous wettability scenario have been calculated as follows:

$$\bar{S}_{gr} = \sum_i P_i S_{gri}/100$$

$$\bar{S}_{gr} = \left(\begin{array}{l} (P \times S_{gr})_{\text{strongly water-wet}} + (P \times S_{gr})_{\text{weakly water-wet}} \\ + (P \times S_{gr})_{\text{intermediate-wet}} + (P \times S_{gr})_{\text{weakly CO}_2\text{-wet}} \\ + (P \times S_{gr})_{\text{strongly CO}_2\text{-wet}} \end{array} \right) /100 \quad (6.4)$$

And,

$$\bar{S}_{wr} = \sum_i P_i S_{wri}/100$$

$$\bar{S}_{wr} = \left(\begin{array}{l} (P \times S_{wr})_{\text{strongly water-wet}} + (P \times S_{wr})_{\text{weakly water-wet}} \\ + (P \times S_{wr})_{\text{intermediate-wet}} + (P \times S_{wr})_{\text{weakly CO}_2\text{-wet}} \\ + (P \times S_{wr})_{\text{strongly CO}_2\text{-wet}} \end{array} \right) /100 \quad (6.5)$$

Where:

\bar{S}_{gr} is the weighted average of residual gas saturation for the heterogeneous wettability model, P_i is the wettability percentage of wettability scenario i in the heterogeneous model, i = different wettability scenarios from strongly water-wet to strongly CO₂-wet (5 cases), S_{gri} is the residual gas saturation for wettability scenario i , \bar{S}_{wr} is the weighted average of residual water saturation for the heterogeneous wettability scenario, and S_{wri} is the residual water saturation for wettability scenario i .

Here, we are representing hysteresis by using different values for the parameters (i.e. λ , S_{gr} and S_{wr}) in equations (6-11) to simulate the relative permeability and capillary pressure curves for the CO₂ injection period (dashed lines) and post-injection period (solid lines; Figures Figure 6-3 and Figure 6-4) for the different wettability scenarios. These parameters (Tables 6-1 and 6-2) were imported into the TOUGH2 code using the Genuchten-Mualem model (Mualem, 1976; Van Genuchten, 1980; Equations 6-11 below) to implement the developed relative permeability and capillary pressure curves in the reservoir simulations. Note that the used approach for treating the hysteresis is not as rigorous as using a true hysteretic formulation (Doughty, 2007), but it is able to represent the CO₂ injection and post-injection periods for the different wettability scenarios adequately (Al-Khdheawi et al., 2017b).

$$k_{rw} = \sqrt{S^*} \left\{ 1 - \left(1 - [S^*]^{1/\lambda} \right)^\lambda \right\}^2 \quad \text{if } S_w < S_{ws} \quad (6.6)$$

$$k_{rw} = 1 \quad \text{if } S_w \geq S_{ws} \quad (6.7)$$

$$k_{rg} = 1 - k_{rw} \quad \text{if } S_{gr} = 0 \quad (6.8)$$

$$k_{rg} = (1 - \hat{S})^2 (1 - \hat{S}^2) \quad \text{if } S_{gr} > 0 \quad (6.9)$$

$$(P_{cap}) = P_0 ([S^*]^{-1/\lambda} - 1)^{1-\lambda} \quad (6.10)$$

$$S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}), \quad (6.11)$$

$$\hat{S} = (S_w - S_{wr}) / (1 - S_{wr} - S_{gr})$$

where:

k_{rg} = gas relative permeability,

k_{rw} = water relative permeability,

S_{gr} = residual saturation for gas,

S_w = water saturation,

S_{ws} = saturated water saturation (= 1),

S_{wr} = residual saturation for water

P_c = CO₂-water capillary pressure,

P_o = capillary pressure scaling factor.

λ = pore size distribution index (fitting parameter),

In order to implement the effect of porosity and permeability heterogeneity on the capillary pressure curves for each gridblock we used the Leverett J-function (Leverett, 1941):

$$J(S_w) = \frac{P_c}{\sigma \cos \theta} \sqrt{\frac{k}{\phi}} \quad (6.12)$$

where:

J = dimensionless capillary pressure, k = reservoir permeability,
 ϕ = reservoir porosity, σ = CO₂-brine surface tension,
 θ = CO₂-brine-rock contact angle.

Table 6-1 The values imported into the TOUGH2 code to implement the relative permeability curves for the different wettability scenarios (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d; Al-Khdheawi et al., 2017b)

Wettability scenario	CO ₂ injection period			CO ₂ post-injection period		
	S _{gr}	S _{wr}	λ	S _{gr}	S _{wr}	λ
Strongly water-wet	0	0.26	0.78	0.35	0.26	0.58
Weakly water-wet	0	0.25	1.05	0.30	0.25	0.95
Intermediate-wet	0	0.22	1.22	0.25	0.22	1.17
Weakly CO ₂ -wet	0	0.15	1.41	0.15	0.15	1.51
Strongly CO ₂ -wet	0	0.1	1.7	0.1	0.1	1.9

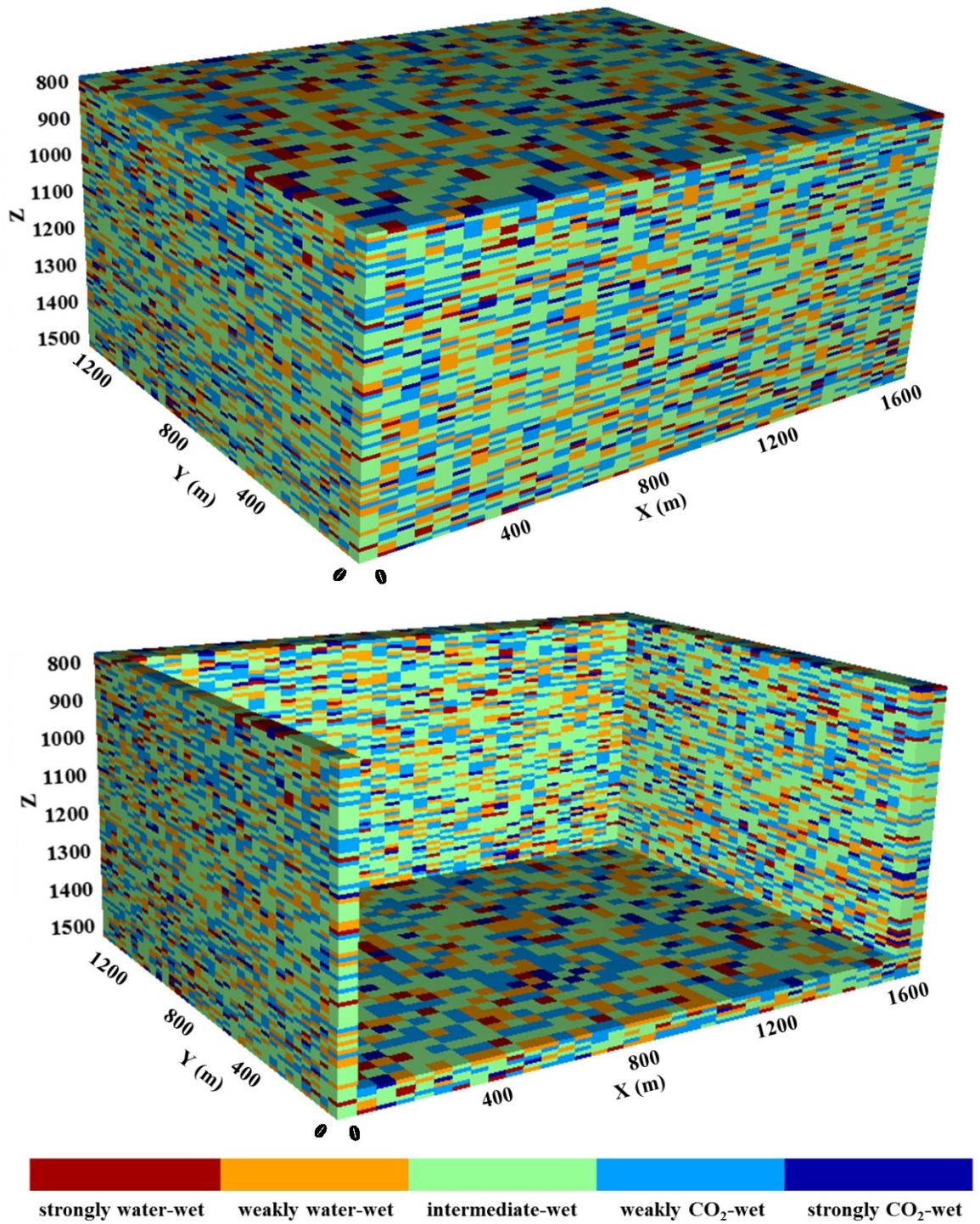


Figure 6-2 3-D views of the heterogeneous wettability model: 5% strongly water-wet; 30% weakly water-wet; 45% intermediate-wet; 15% weakly CO₂-wet; and 5 % strongly CO₂-wet was assigned.

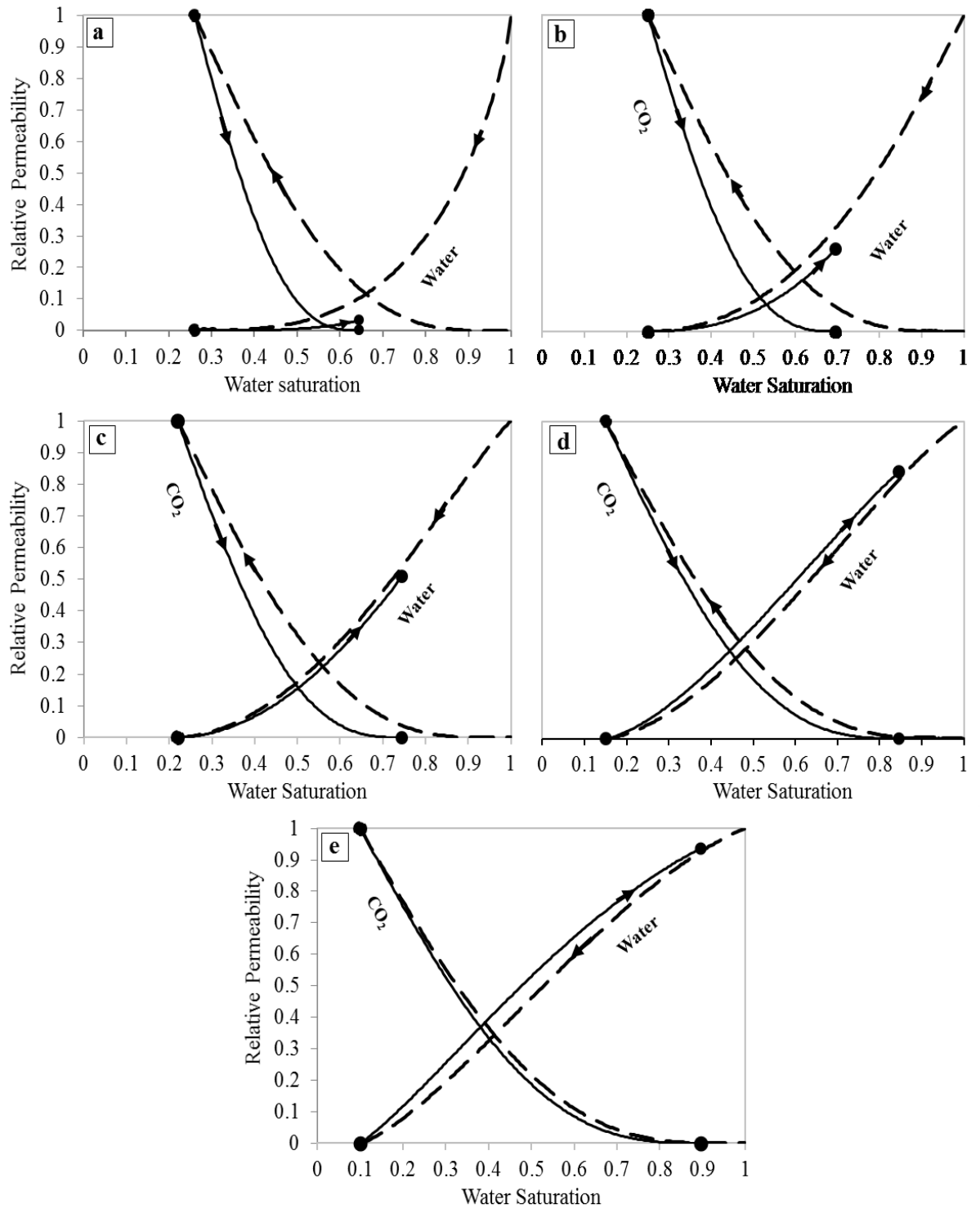


Figure 6-3 Relative permeability curves used for the five rock wettability conditions: (a) strongly water-wet; (b) weakly water-wet; (c) intermediate-wet; (d) weakly CO₂-wet; (e) strongly CO₂-wet (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d; Al-Khdheawi et al., 2017b). Dashed lines represent CO₂ injection period and solid lines represent post-injection (storage) period.

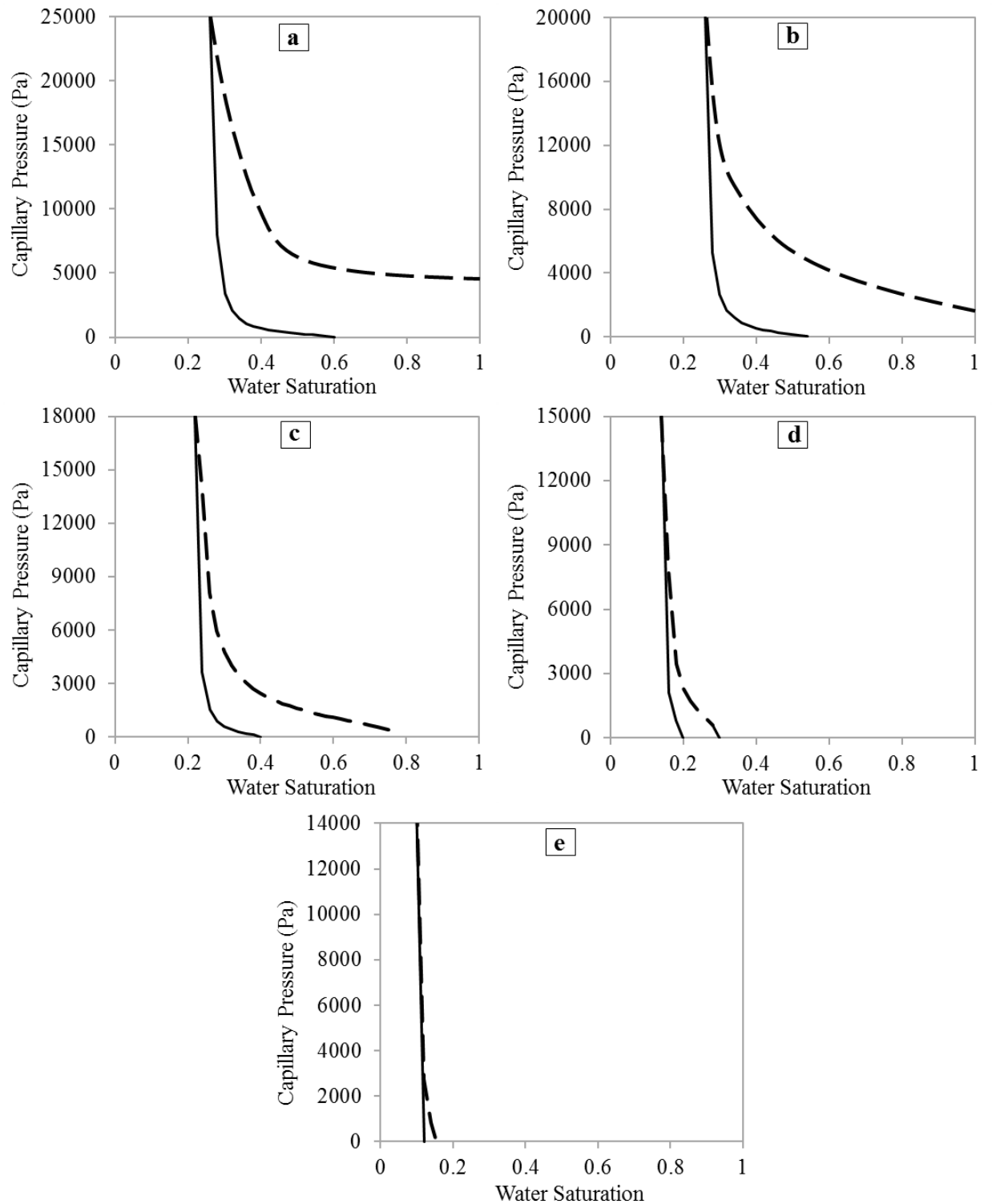


Figure 6-4 Capillary pressure curves used for the five rock wettability scenarios: (a) strongly water-wet; (b) weakly water-wet; (c) intermediate-wet; (d) weakly CO₂-wet; (e) strongly CO₂-wet (Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017d; Al-Khdheewi et al., 2017b). Dashed lines represent CO₂ injection period and solid lines represent post-injection (storage) period.

Table 6-2 The values imported into the TOUGH2 code to implement the capillary pressure curves for the different wettability scenarios (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d; Al-Khdheawi et al., 2017b)

Wettability	CO ₂ injection period					CO ₂ post-injection period				
	S _{wr}	S _{ws}	λ	P ₀ [Pa]	P _{max} [Pa]	S _{wr}	S _{ws}	λ	P ₀ [Pa]	P _{max} [Pa]
Strongly water-wet	0.259	1	0.7	1500	25000	0.259	0.6	0.51	1000	25000
Weakly water-wet	0.249	1	0.7	1500	20000	0.249	0.55	0.51	1000	20000
Intermediate -wet	0.219	0.8	0.7	1500	18000	0.219	0.4	0.51	1000	18000
Weakly CO ₂ -wet	0.149	0.3	0.7	1500	15000	0.149	0.2	0.51	1000	15000
Strongly CO ₂ -wet	0.099	0.15	0.7	1500	14000	0.099	0.1	0.51	1000	14000

6.3 Results and discussion

6.3.1 CO₂ migration analysis

6.3.1.1 Effect of wettability heterogeneity on CO₂ migration

It was previously shown that reservoir wettability has a significant impact on the vertical CO₂ plume migration and that CO₂ migration is highest in CO₂-wet rocks (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d; Al-Khdheawi et al., 2017b). Importantly, in these previous studies, only homogeneous wettability scenarios were investigated. However, it is well established that wettability is heterogeneously distributed in reservoirs (Aspenes et al., 2003; Chaouche et al., 1994; Graue et al., 2002; Masalmeh, 2002; Morrow et al., 1986; Spinler et al., 2002; Standnes and Austad, 2000; Vizika and Duquerroix, 1997). Hence, we investigated this effect and show that wettability heterogeneity leads to a significant increase in vertical CO₂ plume migration, which substantially affects residual and dissolution trapping as shown in

Figure 6-5 (for the non-isothermal scenario), and FiguresFigure 6-6 and Figure 6-7 (heterogeneous and homogeneous wettability cases, respectively; for the isothermal scenario). For both, non-isothermal and isothermal conditions tested, a heterogeneously wetted reservoir had fastest and highest vertical CO₂ plume migration (FiguresFigure 6-5 to Figure 6-8, and Table 6-3). For the non-isothermal scenario, the total vertical CO₂ migration distance at the end of the CO₂ post-injection period (500 years) was 573 m in case of the heterogeneously wetted reservoir, while it was only 464 m in the homogeneous wettability scenario (Figure 6-5 and Table 6-3). Also, for all isothermal temperatures tested (i.e. 303 K, 313 K, 323 K, 333 K, 343 K and 353 K), the heterogeneous wettability scenario had the fastest and highest vertical CO₂ migration (e.g. the total vertical CO₂ migration distance for the 303 K, isothermal temperature model was 375 m in case of heterogeneous wettability, while it was only 304 m in the homogeneous wettability scenario, at the end of the CO₂ post-injection period; Figure 6-8 and Table 6-3). We also noticed that, for non-isothermal conditions, the difference in the CO₂ upward migration rates (when comparing homogenous and heterogeneous wettability scenarios) is larger for higher temperatures.

In summary, our results indicate that spatial heterogeneities in wettability distribution lead to accelerated CO₂ upwards migration for both isothermal and non-isothermal conditions. Hence, simulations made with a homogeneous wettability distribution predicted a lower vertical migration distance than those assuming a heterogeneous wettability distribution.

Table 6-3 CO₂ plume statistics during the post-injection period (0-500 years) for the different reservoir temperature conditions (isothermal and non-isothermal) in both heterogeneous (HT) and homogeneous (HM) wettability scenarios

Reservoir temperature condition		Plume depth at the end of the injection period (m)		Shallowest depth reached by the CO ₂ plume (m)		Time required to reach the shallowest plume depth (years)		Total migration distance (m)*	
		HT	HM	HT	HM	HT	HM	HT	HM
Non-Isothermal****		1049	1049	800	909	35***	500**	573	464
Isothermal	303 K	1196	1196	998	1069	500**	500**	375	304
	313 K	1168	1168	934	1033	500**	500**	439	340
	323 K	1118	1118	807	941	500**	500**	566	432
	333 K	1040	1040	800	821	100** *	500**	573	552
	343 K	941	941	800	800	30***	150***	573	573
	353 K	864	864	800	800	10***	50***	573	573

*The maximum possible vertical migration distance is 573 m, which represents the vertical distance between the CO₂ injection depth (1373 m) and the top seal of the reservoir (800 m).

** For the heterogeneous wettability at 303 K, 313 K and 323 K of the isothermal conditions, and for the homogeneous wettability at non- isothermal condition and 303 K, 313 K, 323 K and 333 K of the isothermal conditions the CO₂ plume did not reach the caprock depth (800 m) during the whole post injection period (0 to 500 years). Thus, the shallowest CO₂ plume depth is recorded at the end of the post-injection period (after 500 years).

*** For the heterogeneous wettability at non- isothermal condition and at 333K, 343 K and 353 K of the isothermal conditions, and for homogeneous wettability at 343 K and 353 K of the isothermal conditions the CO₂ plume reached the caprock (800 m depth) at various times during the post injection period.

**** The geothermal gradient: $T (^{\circ}\text{C}) = 25 + 0.03 \times \text{depth (m)}$.

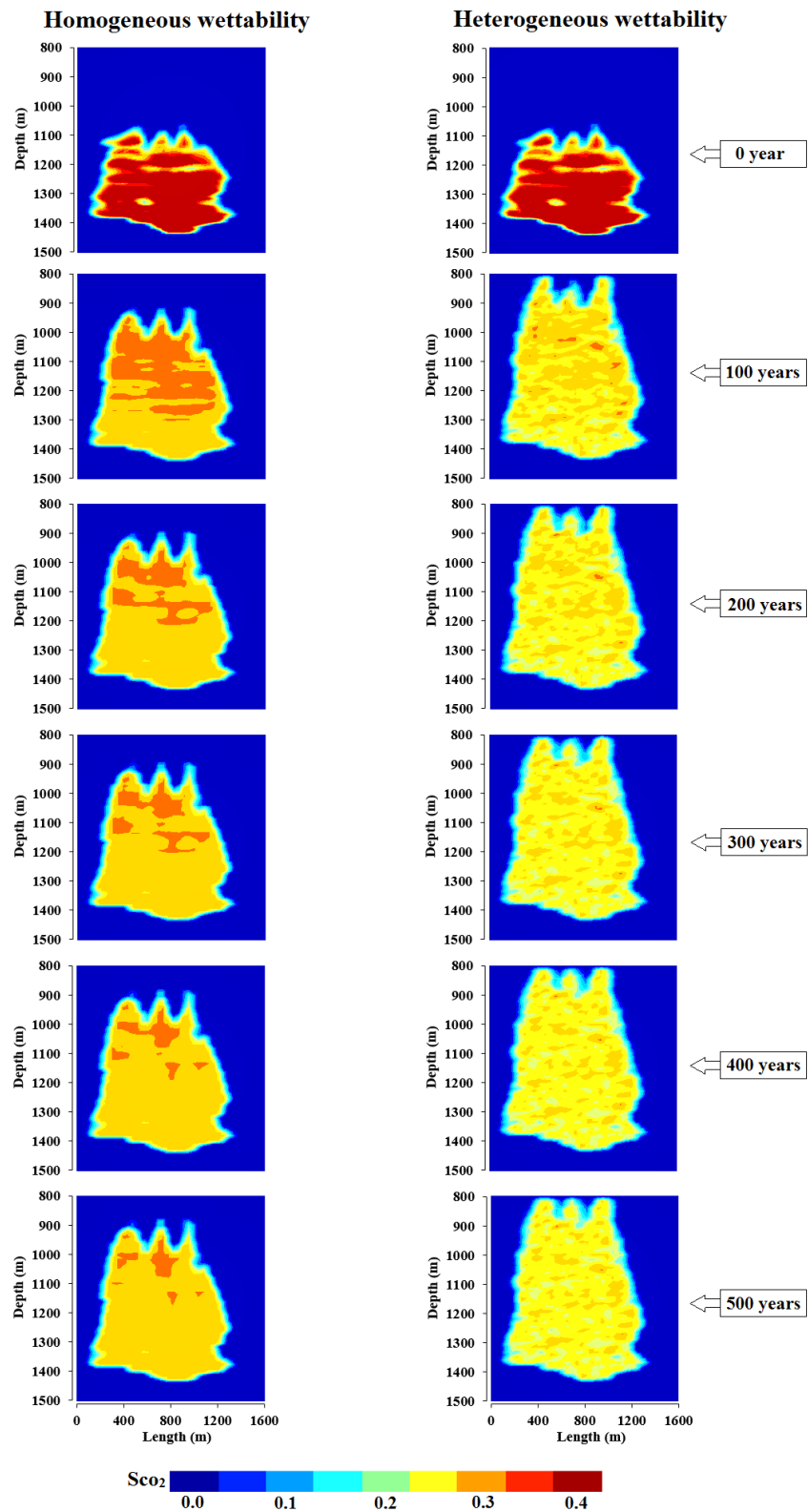


Figure 6-5 2D views through the center of the reservoir showing the CO₂ plume for the model with a vertical geothermal gradient (the non-isothermal conditions) as a function of wettability heterogeneity. Time = 0 year represents the end of the injection period.

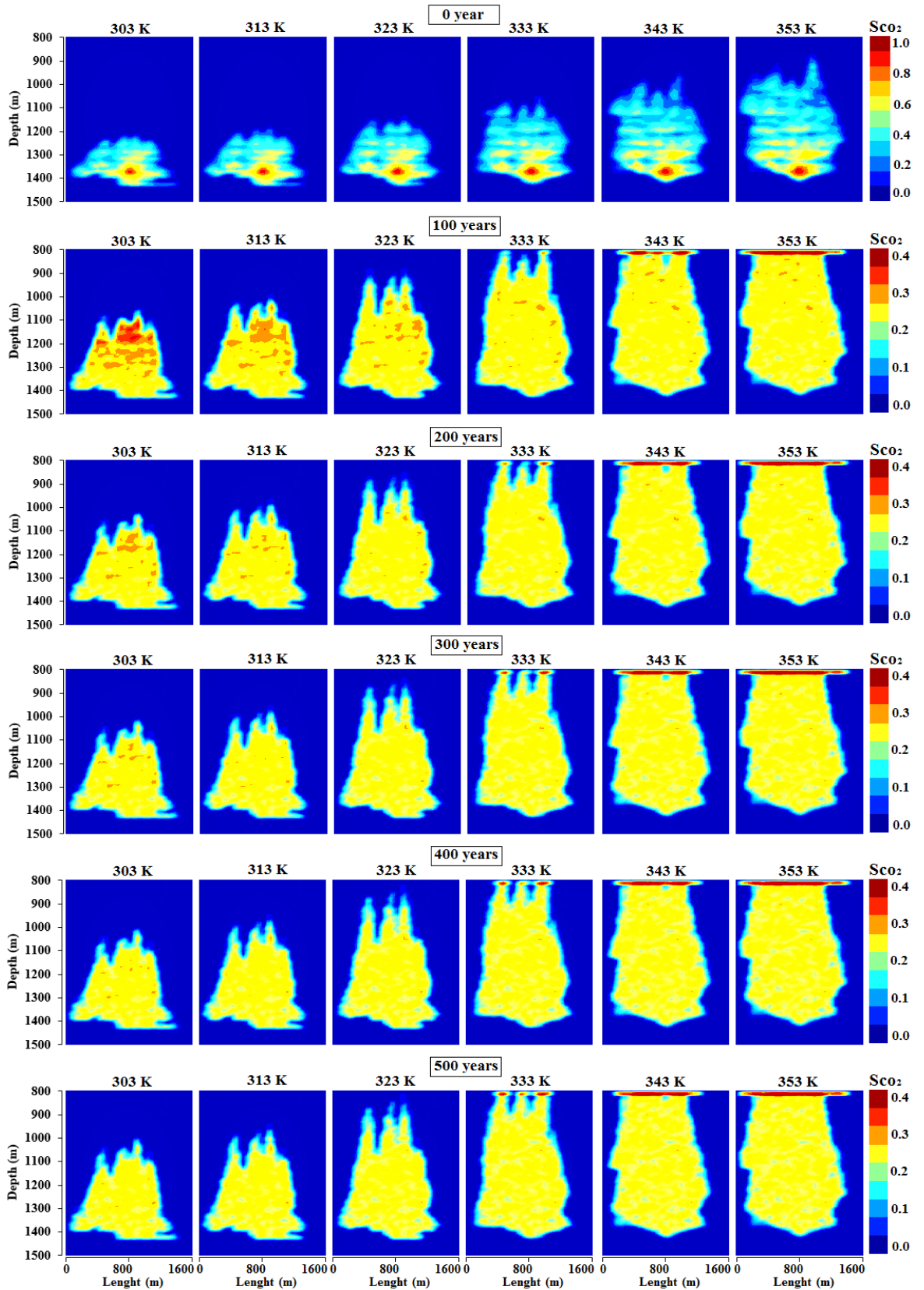


Figure 6-6 2D views through the center of the reservoir showing the CO₂ plume as a function of reservoir temperature and storage time (heterogeneous wettability). Time = 0 year represents the end of the injection period.

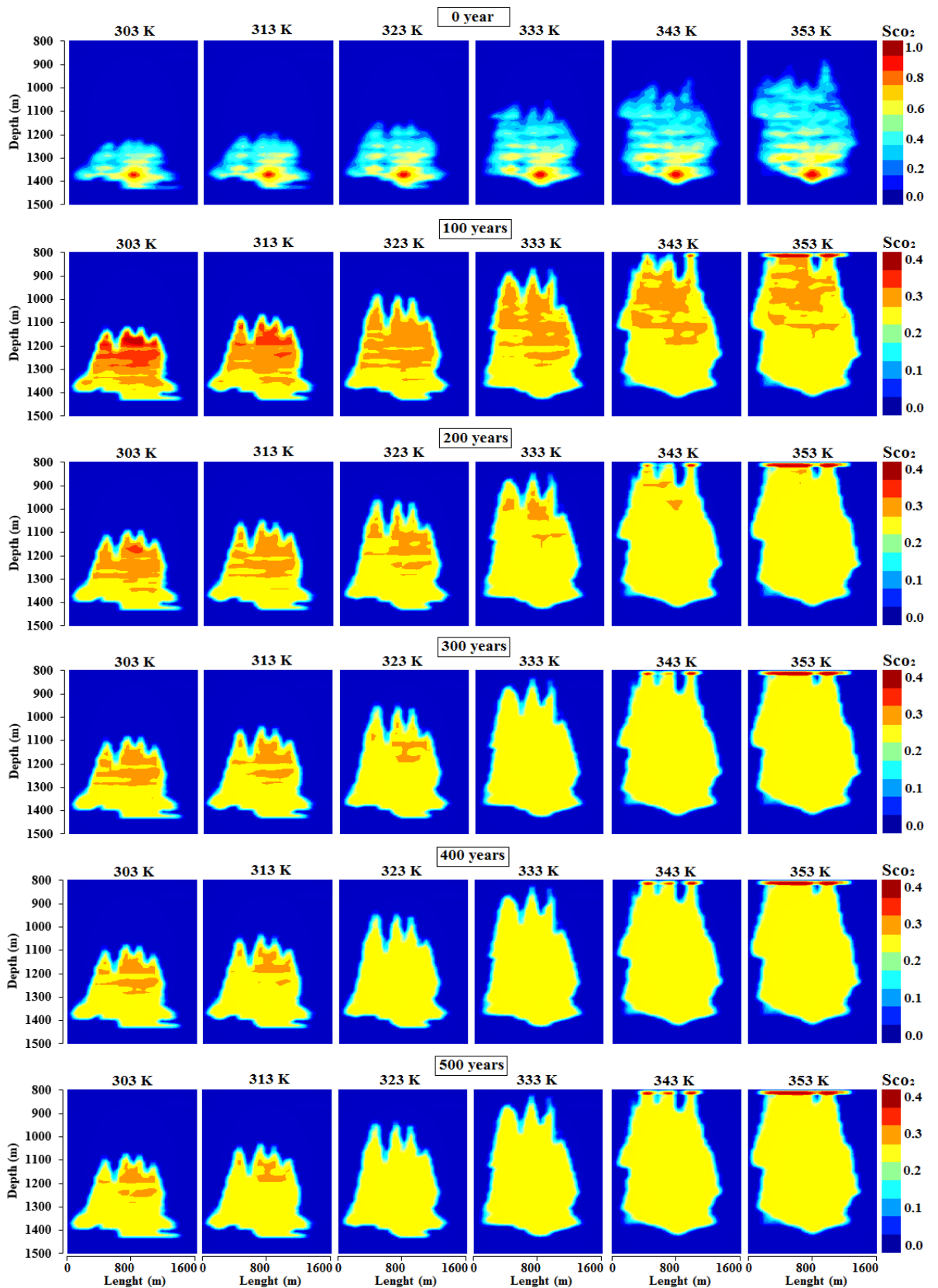


Figure 6-7 2D views through the center of the reservoir showing the CO₂ plume as a function of reservoir temperature and storage time (homogeneous wettability). Time = 0 year represents the end of the injection period.

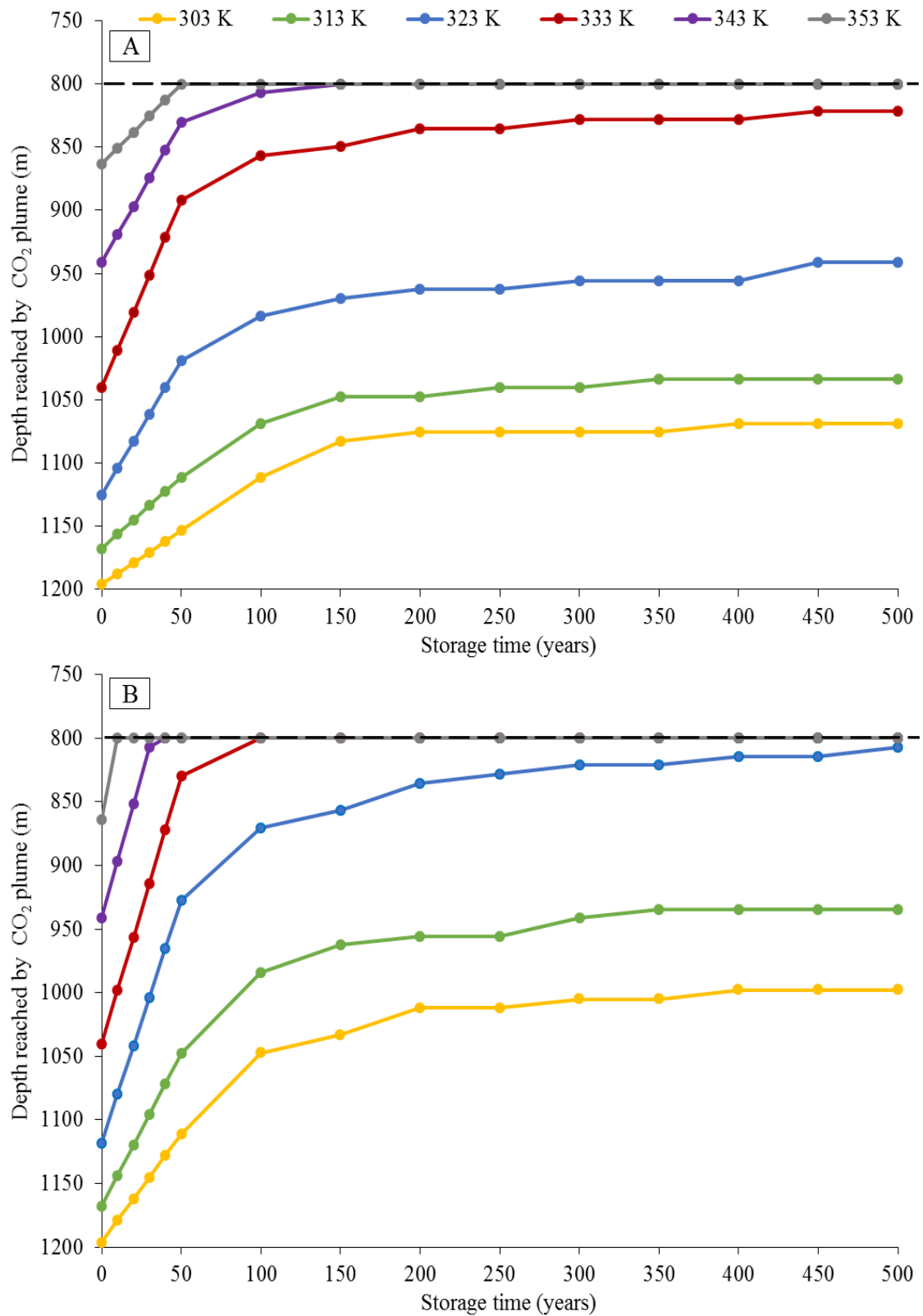


Figure 6-8 Depth reached by the CO₂ plume as a function of storage time and isothermal reservoir temperature; A) homogeneous wettability, B) heterogeneous wettability. The horizontal dashed (black) line shows the caprock seal.

6.3.1.2 Effect of reservoir temperature on CO₂ migration

Clearly, reservoir temperature has a significant impact on the vertical CO₂ migration (Figure 6-8). For both (homogeneous and heterogeneous) wettability scenarios, higher reservoir temperatures drastically accelerated the vertical CO₂ migration.

Specifically, the total post-injection time required for the CO₂ plume to reach the caprock seal depth (800 m) (i.e. the CO₂ plume migrates its maximum possible migration distance; 573 m) was 30 years for the heterogeneous wettability and 150 years for the homogeneous wettability case at 343 K, while it was only 10 years for the heterogeneous wettability and only 50 years for the homogeneous wettability at 353 K (Figure 6-8). Importantly, for all other reservoir temperatures, for both homogeneous and heterogeneous wettability scenarios, a higher reservoir temperatures accelerated the CO₂ plume migration (Figure 6-8 and Table 6-3).

We conclude that higher reservoir temperatures lead to accelerated CO₂ upwards movement.

6.3.2 Trapping capacity analysis

6.3.2.1 Effect of wettability heterogeneity on trapping capacity

Residual and dissolution trapping capacities, and the amount of mobile CO₂ were predicted for both wettability scenarios and for both reservoir temperature conditions investigated (non-isothermal, Figure 6-9, and isothermal, Figure 6-10). Heterogeneous wettability increased the amount of mobile CO₂, while it reduced residual trapping, but improved solubility trapping (Figure 6-9 for the non-isothermal conditions scenario, Figure 6-10 for the isothermal conditions scenario and Table 6-4). Note that the improved solubility trapping is caused by the increased CO₂ plume migration distance, which leads to more dissolved CO₂ as a result of the increased brine-CO₂ contact area (Doughty, 2010).

In summary, our results show that wettability heterogeneity reduces residual trapping, and increases CO₂ mobility and solubility trapping.

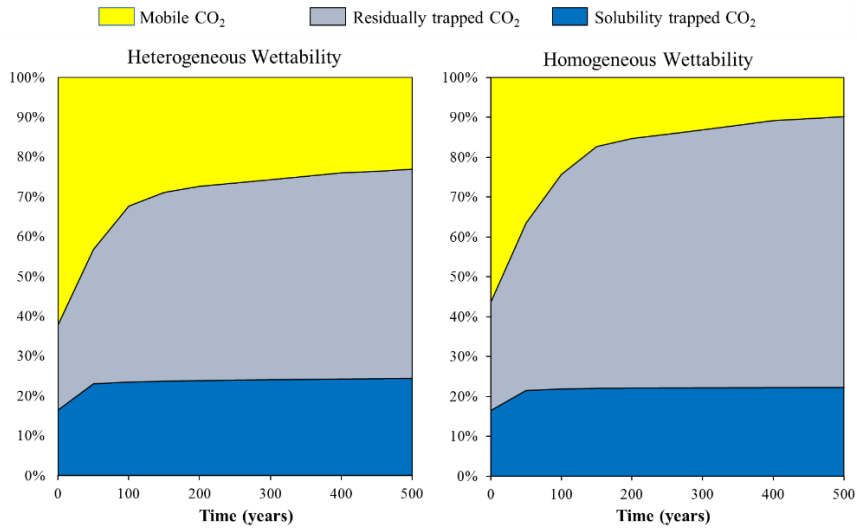


Figure 6-9 Percentages of mobile, residually trapped and solubility trapped CO₂ for the model with vertical geothermal gradient (the non-isothermal conditions) as a function of storage time and wettability heterogeneity. Clearly, wettability heterogeneity leads to increased CO₂ mobility and solubility trapping; while it leads to reduced residual trapping.

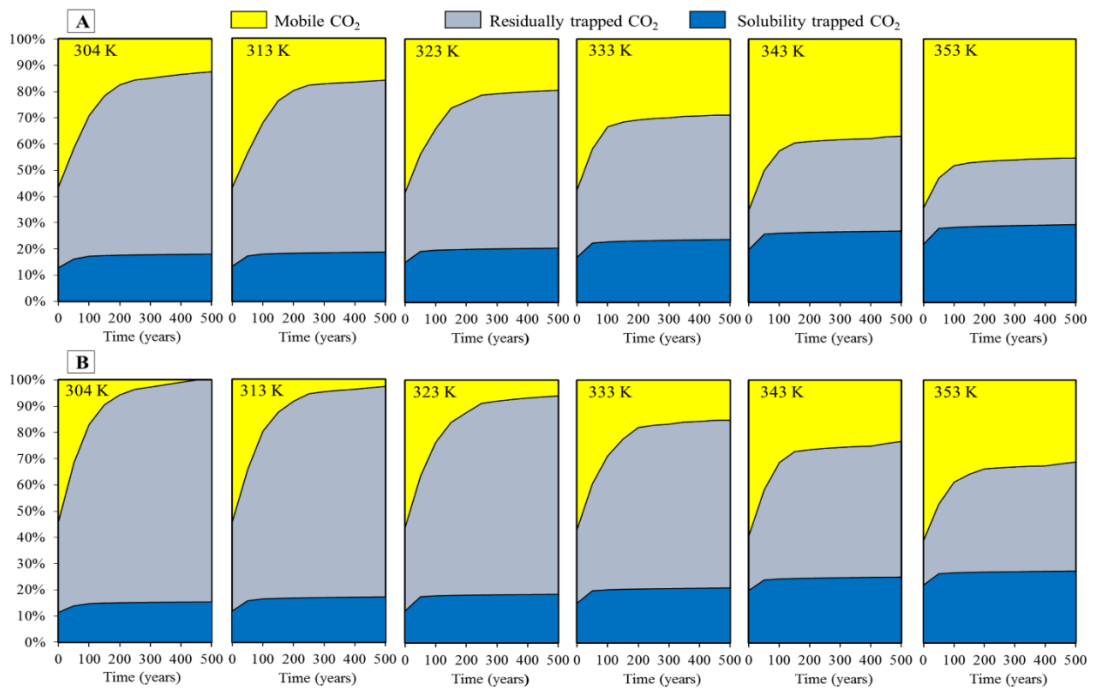


Figure 6-10 Percentages of mobile, residually trapped and solubility trapped CO₂ as a function of storage time, reservoir temperature and wettability heterogeneity; A) heterogeneous wettability scenario and B) homogeneous wettability scenario. Clearly, increasing wettability heterogeneity and increasing reservoir temperature lead to increasing CO₂ mobility and solubility trapping; while it reduces residual trapping.

Table 6-4 Ratio of CO₂ mobility and trapping capacity to the amount of total injected CO₂ (10 Mton) for the different reservoir temperature conditions in both heterogeneous (HT) and homogeneous (HM) wettability scenarios at the end of the post-injection period (500 years)

Ratio of CO ₂ mobility and trapping capacities to the total injected CO ₂ (10 Mton)	Wettability scenario	Isothermal Reservoir temperature (K)						Non-isothermal conditions
		303	313	323	333	343	353	
Residual trapping (%)	HT	70	65	60	47	36	25	53
	HM	85	80	75	64	52	42	68
dissolution trapping (%)	HT	18	19	21	24	27	30	24
	HM	15	17	19	21	25	27	22
Mobile (free) CO ₂ (%)	HT	12	16	19	29	37	45	23
	HM	0	3	6	15	23	31	10

6.3.2.2 Effect of reservoir temperature on trapping capacity

Clearly, reservoir temperature highly affects CO₂ mobility and residual and dissolution trapping (Figure 6-10). Increasing reservoir temperature increased CO₂ mobility (for both, homogeneous and heterogeneous wettability; Figure 6-10 and Table 6-4). This is coupled with reduced residual trapping (e.g. the amount of residually trapped CO₂ decreased from 70% to 25% when the reservoir temperature was increased from 303 K to 353 K; in the heterogeneous-wet case model, after 500 years; but improved solubility trapping (e.g. the amount of solubility trapped CO₂ increased from 18% to 30% when the reservoir temperature increased from 303 K to 353 K; in the heterogeneous-wet model, after 500 years; Figure 6-10 and Table 6-4). These results are consistent with previous studies (e.g. Kumar et al., 2005; Ofori and Engler, 2011) which demonstrated that increasing reservoir temperature leads to increased CO₂ solubility trapping. It is important to note that the solubility of CO₂ in water decreases with increasing temperature (e.g. the mass fraction of CO₂ in the

aqueous phase decreases from 0.053 to 0.032 when the temperature increases from 303K to 353K at 80 bars; Doughty, 2010). However, the greater CO₂ plume migration distance (i.e. the larger CO₂-brine contact area) has also a large effect on CO₂ solubility (Doughty, 2010); so CO₂ solubility trapping increases with increasing reservoir temperature. Recall that our results show that higher reservoir temperatures leads to higher CO₂ plume migration. So this overcompensates the thermodynamic effect.

In summary, we conclude that the reservoir temperature has an important impact on the amount of mobile and trapped CO₂, and lower temperatures are preferred.

6.4 Conclusions

It is well known that reservoir wettability is distributed heterogeneously at various length scales in a reservoir (Aspenes et al., 2003; Chaouche et al., 1994; Graue et al., 2002; Masalmeh, 2002; Morrow et al., 1986; Spinler et al., 2002; Standnes and Austad, 2000; Vizika and Duquerroix, 1997). Furthermore, it has been shown previously that such wettability heterogeneity has a significant impact on the displacement mechanisms and phase distributions in gas injection processes for oil recovery (Laroche et al., 1999; Anderson, 1986, 1987a, b; Chang et al., 1997; Morrow, 1990; Bertin et al., 1998; Blunt, 1997; Kiriakidis et al., 1993). However, the effect of wettability heterogeneity on CO₂ storage efficiency at isothermal and non-isothermal reservoir conditions has not been addressed. Moreover, reservoir temperature is another factor which has not received sufficient attention in this context.

Thus, we studied the effect of wettability heterogeneity and reservoir temperature on CO₂ plume migration and residual and solubility trapping. Our results clearly demonstrate that both increased wettability heterogeneity and increased reservoir temperature significantly increase the vertical CO₂ plume migration distance, CO₂ mobility and CO₂ dissolution trapping, but decrease residual trapping.

We thus conclude that wettability heterogeneity needs to be accurately modelled in reservoir-scale simulations in order to obtain reliable CO₂ storage predictions. Furthermore, we conclude that lower reservoir temperatures are preferential because they improve trapping capacity.

Chapter 7 Impact of Salinity on CO₂ Containment Security in Highly Heterogeneous Reservoirs

7.1 Introduction

Geological Carbon Storage is one of the most promising technologies to mitigate anthropogenic carbon emissions by capturing the CO₂ from large stationary sources and injecting it deep into geological formations. (Pruess et al., 2003). The brine-CO₂ density difference will then cause a vertical CO₂ migration towards the surface. This migration can be reduced and CO₂ can be inhibited from leaking to the atmosphere by four main trapping mechanisms, namely structural trapping, (i.e. caprock seal beds or permeability barrier units),(Hesse et al., 2008; Naylor et al., 2011; Iglauer et al., 2015a) residual trapping which depends on the capillary forces,(Pentland et al., 2011; Iglauer et al., 2011; Krevor et al., 2015; Rahman et al., 2016; Iglauer and Wüiling, 2016) dissolution trapping (Lindeberg and Wessel-Berg, 1997; Spycher et al., 2003; Iglauer, 2011) which is affected by the CO₂-brine interfacial area,(Metz et al., 2005; Kumar et al., 2005; Doughty, 2010; Pentland et al., 2012) and mineral trapping,(Bachu et al., 1994b; Xu et al., 2004; Gaus, 2010) which depends on the chemical reactions between reservoir rock minerals and fluids and the injected CO₂ (Xu et al., 2003; Xu et al., 2005).

Many factors affect the CO₂ storage efficiency, e.g. reservoir heterogeneity,(Doughty and Pruess, 2004; Hovorka et al., 2004; Obi and Blunt, 2006; Bryant et al., 2006; Flett et al., 2007; Ide et al., 2007; Saadatpoor et al., 2009; Zhou et al., 2010; Ambrose et al., 2008; Han et al., 2010; Hesse and Woods, 2010; Green and Ennis-King, 2010; Gershenzon et al., 2015; Gershenzon et al., 2016) rock wettability, (Iglauer et al., 2015a; Iglauer et al., 2015b; Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b) injection well configuration,(Al-Khdheawi et al., 2017d) the ratio of vertical to horizontal permeability,(Basbug et al., 2005) stratum dip angle,(Wang et al., 2016a; Wang et al., 2016b) key model parameters (i.e. outer boundary condition, reservoir size, and CO₂ effective permeability; Zhang et al., 2016a) cap rock properties, (Iglauer et al., 2015a) or aquifer depth and leakage-pathway (Zeidouni et al., 2015). However, although it is well established that brine salinity can vary widely between prospective

storage reservoirs (Morton and Land, 1987; Bachu and Bennion, 2008; Chalbaud et al., 2010; Al-Khdheawi et al., 2017a), there is a knowledge gap in terms of how this can influence storage efficiency. Importantly, experimental data showed that water salinity has a significant impact on the water-CO₂-rock contact angle θ (i.e. rock wettability; Iglauer, 2017). For example, Arif *et al.* (Arif et al., 2016b) reported that increasing water salinity from 0 wt% (DI water) to 20 wt% leads to an increase in advancing contact angle by 16° (from 59° to 75°) and an increase in receding contact angle by 12° (from 54° to 66°) for a mica surface at 323 K and 15 MPa. Similarly, significant increases were measured for quartz (Al-Yaseri et al., 2016; Espinoza and Santamarina, 2010; Farokhpour et al., 2013; Jung and Wan, 2012; Saraji et al., 2014; Wang et al., 2012; Broseta et al., 2012; Sarmadivaleh et al., 2015). This effect is caused by a better surface charge screening at higher salinities. Mechanistically, cations move to the negative surfaces charges and shield them, thus reducing the surface potential and surface polarity, which leads to a dewetting of the surface (Iglauer, 2017). Furthermore, salinity has a slight effect on water viscosity (e.g. Denny (Denny, 1993) reported that water viscosity slightly increased from 0.00065 Pa.s for freshwater to 0.00071 Pa.s for seawater, measured at 313 K). Moreover, several studies have shown that brine salinity has a significant impact on CO₂ solubility in the brine,(Rumpf et al., 1994; Duan and Sun, 2003; Spycher and Pruess, 2004; Iglauer, 2011; El-Maghraby et al., 2012) while its impact on CO₂ plume migration, CO₂ mobility and residually trapped CO₂ has not been addressed.

Thus, in this chapter, the effect of salinity (3-20 wt%) on CO₂ plume migration, CO₂ mobility and dissolution and residual trapping in a highly heterogeneous 3D reservoir are computationally examined.

7.2 Methodology

7.2.1 Modelling theory

To study the effect of brine salinity on the CO₂ geo-storage efficiency, a 3D heterogeneous reservoir scale model has been developed using the TOUGH2 simulator (i.e. a nonisothermal multicomponent multiphase flow simulator; Pruess et al., 1999) with the tabular equation of state ECO2M (Pruess, 2011). TOUGH2 and the ECO2M module describe thermodynamic and thermophysical properties (e.g.

viscosity, density and specific enthalpy) of H₂O–NaCl–CO₂ mixtures, as a function of temperature, pressure and salinity (i.e. composition), including super- and sub-critical CO₂ conditions, and CO₂ phase changes; Pruess, 2011). Furthermore, TOUGH2 solves the mass and energy balance equations which describe fluid and heat flow in multiphase and multicomponent systems, for $k = 1$ to n mass components, by the formula:

$$\frac{d}{dt} \int_{V_n} M^k dV_n = \int_{\Gamma_n} F^k \cdot n d\Gamma_n + \int_{V_n} Q^k dV_n \quad (7.1)$$

where:

V_n is an arbitrary sub-domain which is bounded by the closed surface Γ_n , M is the mass or energy per volume, k is the mass component labelling factor which represents different mass components (e.g. for water, CO₂, air, solutes...etc.), F is the mass or heat flux, Q is the sink and source term, and n is the surface element $d\Gamma_n$, a normal vector, pointing inward into V_n . Furthermore, TOUGH2 predicts the accumulation of mass based on the following formula:

$$M^k = \phi \sum_{\beta} S_{\psi} \rho_{\psi} X_{\psi}^k \quad (7.2)$$

where:

ψ is the number of fluid phases (i.e. liquid, gas, non-aqueous phase liquid), ϕ is the system porosity, S_{ψ} is the saturation of phase ψ , ρ_{ψ} is the density of phase ψ , and X_{ψ}^k is the mass fraction of component k in phase ψ . The cumulative mass and heat fluxes are calculated by summing up all individual fluxes, while the multiphase version of Darcy's law is used to compute these individual phase fluxes:

$$F_{\psi} = \rho_{\psi} u_{\psi} = -k \frac{k_{r\psi} \rho_{\psi}}{\mu_{\psi}} (\nabla P_{\psi} - \rho_{\psi} \mathbf{g}) \quad (7.3)$$

where:

F_{ψ} is the individual mass or heat flux, u_{ψ} is the Darcy velocity of phase ψ , k is the absolute permeability, $k_{r\psi}$ is the phase ψ relative permeability, μ_{ψ} is the viscosity of phase ψ , and P_{ψ} is the pressure of phase ψ .

7.2.2 Model description and initialization

The dimensions of the heterogeneous reservoir model were 1600 m × 1200 m with a thickness of 700 m equivalent to a 37 × 33 × 80 grid (97680 cells in total) as shown in Figure 7-1. The initial water saturation in the reservoir was 100% at isothermal conditions (323 K), and initial pressure was 8 MPa at 800m depth. Pressure at 1500 m depth was 15 MPa assuming a pressure gradient of 10 MPa/km (Dake, 2007). The reservoir outer boundary cells were extended by multiplying their volume by 10^8 to simulate constant pressure boundary conditions (i.e. Dirichlet boundary conditions). (Mo et al., 2005; Nghiem et al., 2009). Porosity and permeability heterogeneity of the SPE comparative solution project (Christie and Blunt, 2001) were used for each grid block to simulate a representative highly heterogeneous reservoir (Figure 7-1). Specifically, porosity ranged from 10% to 47% and permeability ranged from 10 mD to 1000 mD. This comparative solution project was initially developed for the PUNQ project (Floris et al., 1999). The 10th SPE model consists of two parts: The top part is prograding near shore environment, representing the Tarbert formation; and the bottom part is a fluvial reservoir with channel structures representing the Upper-Ness formation (Christie and Blunt, 2001). The vertical to horizontal permeability ratio (k_v/k_h) was reduced from 0.1 in the reservoir to (10^{-4}) at the top boundary of the reservoir to simulate a barrier preventing the CO₂ from leaving the reservoir.

For all investigated brine salinities, (3 wt%, 6 wt%, 10 wt% and 20 wt% NaCl brine), at a reservoir depth of 1373 m, 10 Mton of CO₂ were injected through a one injection well during a 10 years injection period at a rate of 1 Mt/year, which represents the highest Sleipner (Leung et al., 2014; Torp and Gale, 2004) and Quest (Bourne et al., 2014) CO₂ injection rate. This 10 years injection period was followed by a 200 year post-injection period to monitor CO₂ plume development following injection well shut-in. CO₂ plume migration, mobility and residual and dissolution trapping capacities were then computed for this period.

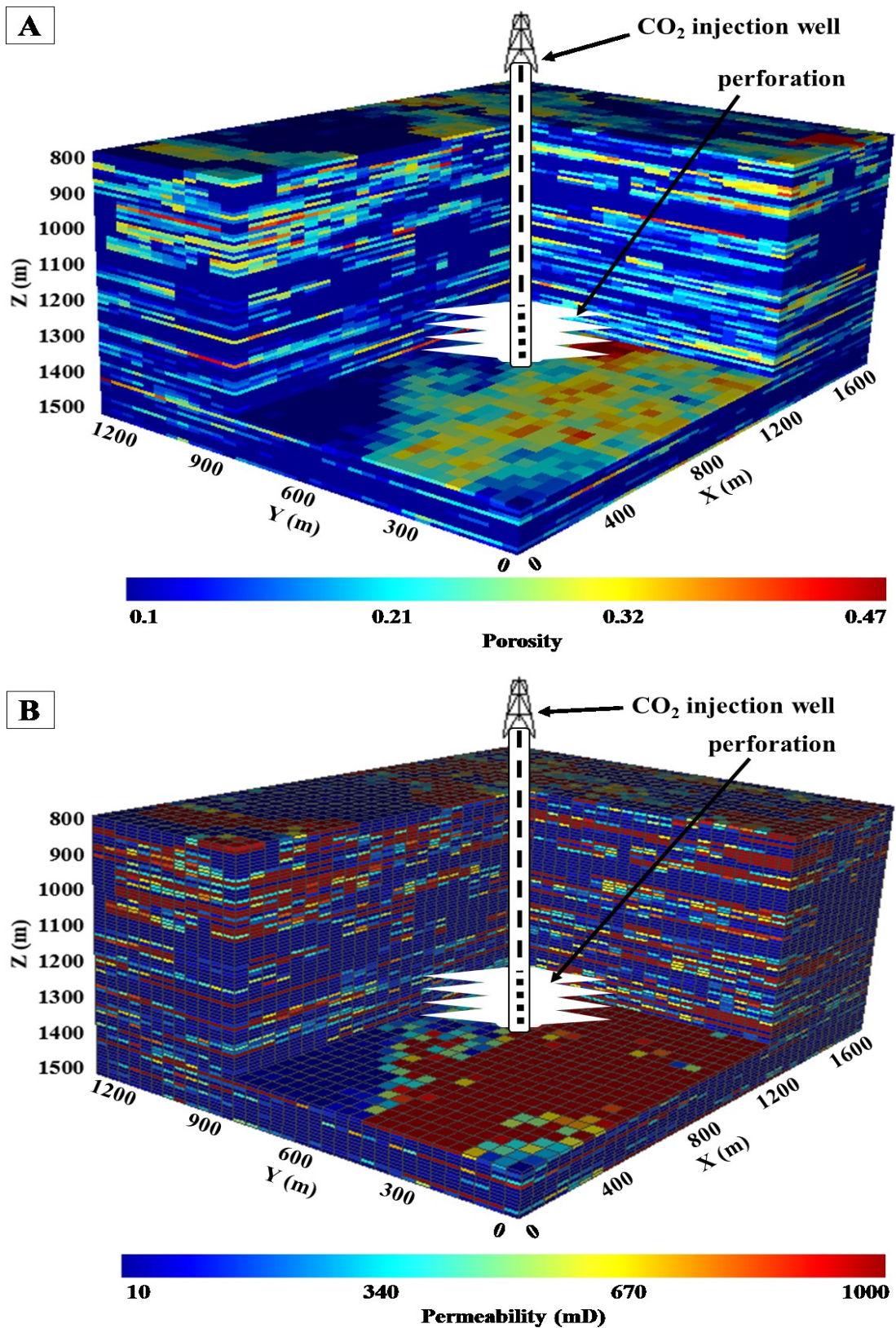


Figure 7-1 3-D reservoir model showing CO₂ injection well, perforation depth and model dimensions with: A) porosity distribution ; B) permeability distribution.

7.2.3 Implementation of different brine salinity scenarios

Variation in salinity has two effects: i) it influences the characteristics curves (i.e. capillary pressure and relative permeability), which have a significant influence on CO₂ migration (Doughty, 2007); and ii) it influences CO₂ solubility in the brine; higher solubilities can be achieved in less saline waters. (Rumpf et al., 1994; Duan and Sun, 2003; Spycher and Pruess, 2004; Iglauer, 2011; El-Maghraby et al., 2012) In terms of point i), Previous studies showed that relative permeability and capillary pressure curves are a function of the rock wettability (i.e. contact angle); while it is clear that brine salinity (and composition) changes the CO₂-rock wettability (increasing salinity reduces water wettability). (Al-Yaseri et al., 2016; Arif et al., 2016b; Chiquet et al., 2007; Espinoza and Santamarina, 2010; Farokhpoor et al., 2013; Jung and Wan, 2012; Saraji et al., 2014; Wang et al., 2012; Broseta et al., 2012; Sarmadivaleh et al., 2015) This shift in wettability (i.e. variation in the contact angle value) is thus the reason for the changes in relative permeability, (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b; Levine et al., 2014; Krevor et al., 2012; Anderson, 1987b; Heiba et al., 1983; McCaffery and Bennion, 1974; Owens and Archer, 1971) capillary pressures, (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b; Pini et al., 2012; Pini et al., 2013; Heiba et al., 1983; Batycky et al., 1981; Morrow, 1976; Melrose, 1965; Anderson, 1987a) and residual saturations. (Rahman et al., 2016; Al-Menhali et al., 2016; Ruprecht et al., 2014; Chaudhary et al., 2013; Akbarabadi and Piri, 2013; Iglauer et al., 2011; Pentland et al., 2011; Craig, 1993) Here, we implemented the impact of salinity via a pair of relative permeability and capillary pressure curves for each brine salinity. Thus, we modified here our recently developed relative permeability and capillary pressure curves for a water-wet system (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b) by adjusting them for different CO₂-water wettabilities (i.e. for different salinities) using the Craig's rules of thumb (Craig, 1993) to consider different brine salinities.

We then modified these curves as follows to consider different brine salinities. (Craig, 1993)

Initially, for all salinity scenarios, water relative permeability (k_{rw}) was 1 and gas relative permeability (k_{rg}) was 0 to simulate a fully brine saturated aquifer (Figure 7-2; A). During CO₂ injection (drainage), k_{rw} decreased from 1 to 0 and k_{rg} increased from 0 to 1 at the endpoints (i.e. residual water and residual CO₂ saturations). During water

influx (imbibition) k_{rg} decreased again from 1 to 0 (at S_{gr}) and k_{rw} increased from 0 until it reached its maximum value at S_{gr} (Figure 7-2; B). Importantly, S_{gr} decreases with increasing brine salinity, because higher salinity increases CO₂-wettability (Al-Yaseri et al., 2016; Arif et al., 2016b; Chiquet et al., 2007; Espinoza and Santamarina, 2010; Farokhpour et al., 2013; Jung and Wan, 2012; Saraji et al., 2014; Wang et al., 2012; Broseta et al., 2012; Sarmadivaleh et al., 2015) and S_{gr} is a function of wettability (i.e. increasing CO₂ wettability reduces S_{gr}). (Rahman et al., 2016; Al-Menhali et al., 2016; Ruprecht et al., 2014; Chaudhary et al., 2013; Akbarabadi and Piri, 2013; Iglauer et al., 2011; Pentland et al., 2011; Craig, 1993) Furthermore, lower salinity (i.e. lower CO₂-wettability) shifts the k_{rw} curves downwards, and the k_{rw} - k_{rg} intercept point towards the right (i.e. to a higher residual water saturation, (Craig, 1993) Figure 7-2).

The curves were subsequently implemented in the Van Genuchten-Mualem model, (Van Genuchten, 1980; Mualem, 1976) cp. also Figure 7-2:

$$k_{rw} = \sqrt{S^*} \left\{ 1 - \left(1 - [S^*]^{1/\lambda} \right)^\lambda \right\}^2 \quad \text{if } S_w < S_{ws} \quad (7.4)$$

$$k_{rw} = 1 \quad \text{if } S_w \geq S_{ws} \quad (7.5)$$

$$k_{rg} = 1 - k_{rw} \quad \text{if } S_{gr} = 0 \quad (7.6)$$

$$k_{rg} = (1 - \hat{S})^2 (1 - \hat{S}^2) \quad \text{if } S_{gr} > 0 \quad (7.7)$$

$$(P_{cap}) = P_0 ([S^*]^{-1/\lambda} - 1)^{1-\lambda} \quad (7.8)$$

$$S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}), \quad (7.9)$$

$$\hat{S} = (S_w - S_{wr}) / (1 - S_{wr} - S_{gr})$$

where:

k_{rg} = gas relative permeability,

k_{rw} = water relative permeability,

S_{gr} = residual saturation for gas,

S_w = water saturation,

S_{ws} = saturated water saturation (= 1),

S_{wr} = residual saturation for water

P_c = CO₂-water capillary pressure,

P_0 = capillary pressure scaling factor.

λ = pore size distribution index (fitting parameter),

The influence of reservoir heterogeneity (i.e. porosity and permeability heterogeneity) on the capillary pressure curves for each grid cell was implemented via the Leverett J-function, (Leverett, 1941) cp. Figure 7-3:

$$J(S_w) = \frac{P_c}{\sigma \cos \theta} \sqrt{\frac{k}{\phi}} \quad (7.10)$$

where

J = dimensionless capillary pressure,

k = reservoir permeability,

ϕ = reservoir porosity,

σ = CO₂-brine surface tension,

θ = CO₂-brine-rock contact angle.

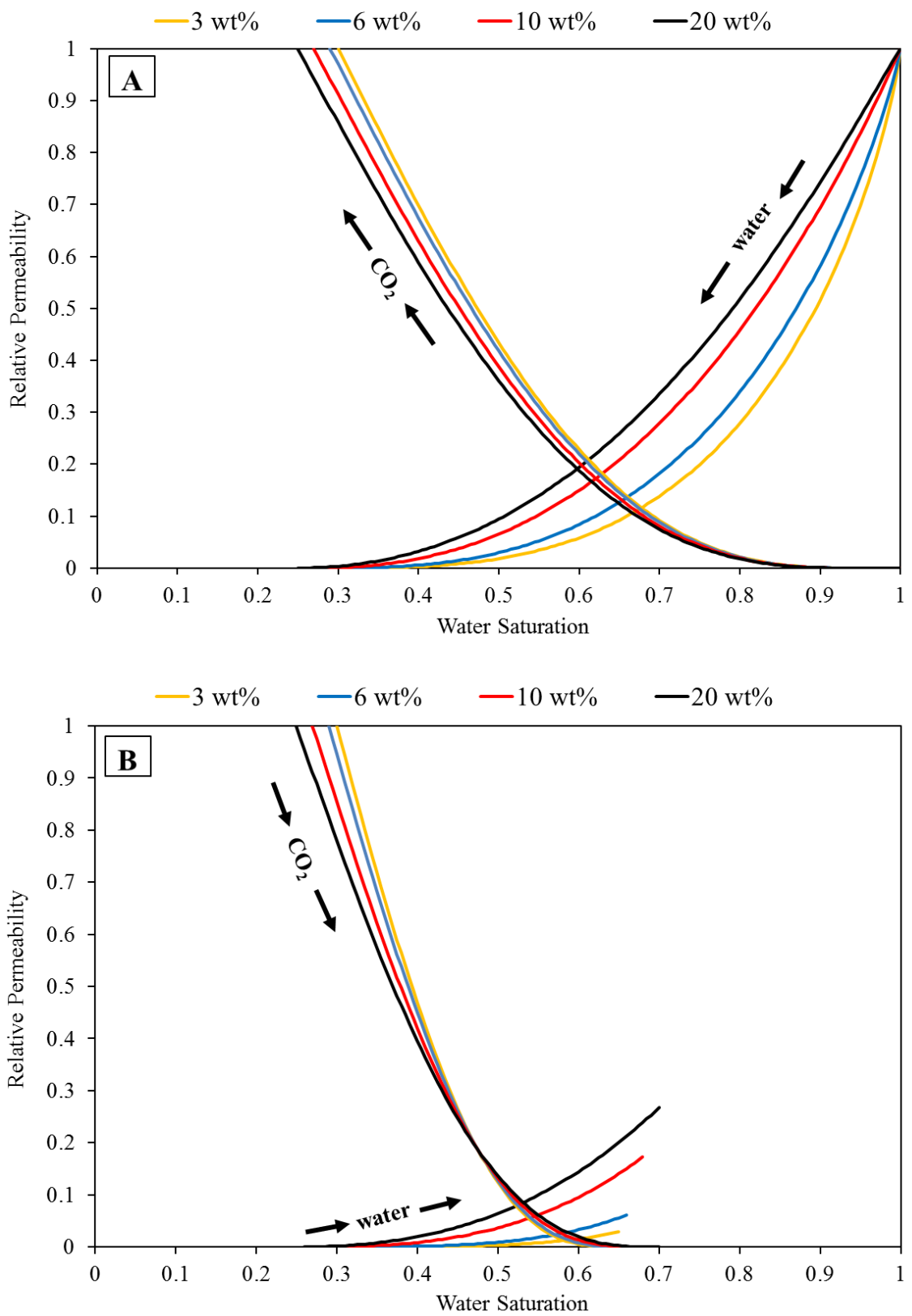


Figure 7-2 Relative permeability curves used for the different salinity scenarios: A) CO₂ injection (drainage); B) CO₂ storage (imbibition).

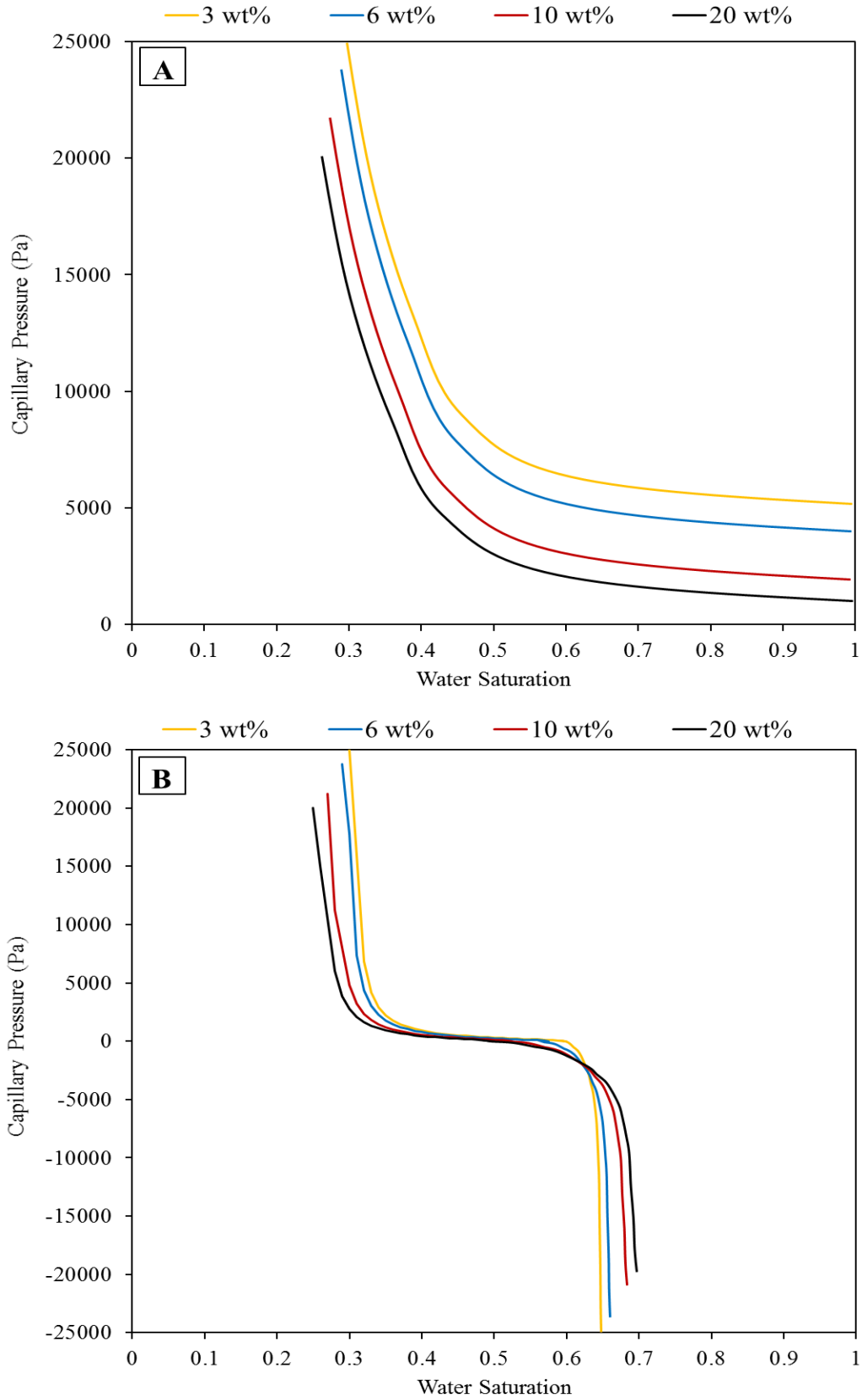


Figure 7-3 Capillary pressure curves used for the 4 different salinity scenarios: A) CO₂ injection (drainage); B) CO₂ storage (imbibition).

7.3 Results and discussion

7.3.1 Impact of salinity on CO₂-plume migration

Clearly, brine salinity has a significant impact on the plume shape and its migration distance (Figure 7-4). Specifically, lower brine salinities reduced vertical CO₂ plume migration distance during the whole storage period (0-200 years; for instance the highest CO₂ plume depth reached was 1090 m for 3 wt% salinity, 1069 m for 6 wt% salinity, 1012 m for 10 wt% salinity and 906 m for 20 wt% salinity; after 200 years, (Figure 7-5). This is equivalent to a vertical CO₂ plume migration distance of 283 m for 3 wt% salinity, 304 m for 6 wt% salinity, 361 m for 10 wt% salinity and 467 m for 20 wt% salinity (after 200 years storage time, (Figure 7-6).

This increase in migration distance with increasing salinity is caused by the shift in the characteristics curves (i.e. relative permeability and capillary pressure); recall that salinity affects the CO₂-rock wettability, and increasing brine salinity leads to reduced water wettability. (Al-Yaseri et al., 2016; Arif et al., 2016b; Chiquet et al., 2007; Espinoza and Santamarina, 2010; Farokhpour et al., 2013; Jung and Wan, 2012; Saraji et al., 2014; Wang et al., 2012; Broseta et al., 2012; Sarmadivaleh et al., 2015)

This wettability shift causes the change in relative permeabilities (Figure 7-2) and capillary pressures (Figure 7-3), see discussion above. (Craig, 1993; Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b) Importantly, these characteristics curves have a significant influence on CO₂ migration.(Doughty, 2007) Furthermore, CO₂ migration increases with decreasing S_{gr} (Doughty, 2010) and thus with decreasing water-wettability (note that S_{gr} decreases when water wettability is reduced.(Rahman et al., 2016; Al-Menhali et al., 2016; Chaudhary et al., 2013)

In summary, our results indicate that brine salinity has a significant impact on CO₂ plume behaviour and its vertical migration distance; we conclude that lower brine salinity is preferable as it reduces CO₂ migration implying higher containment security.

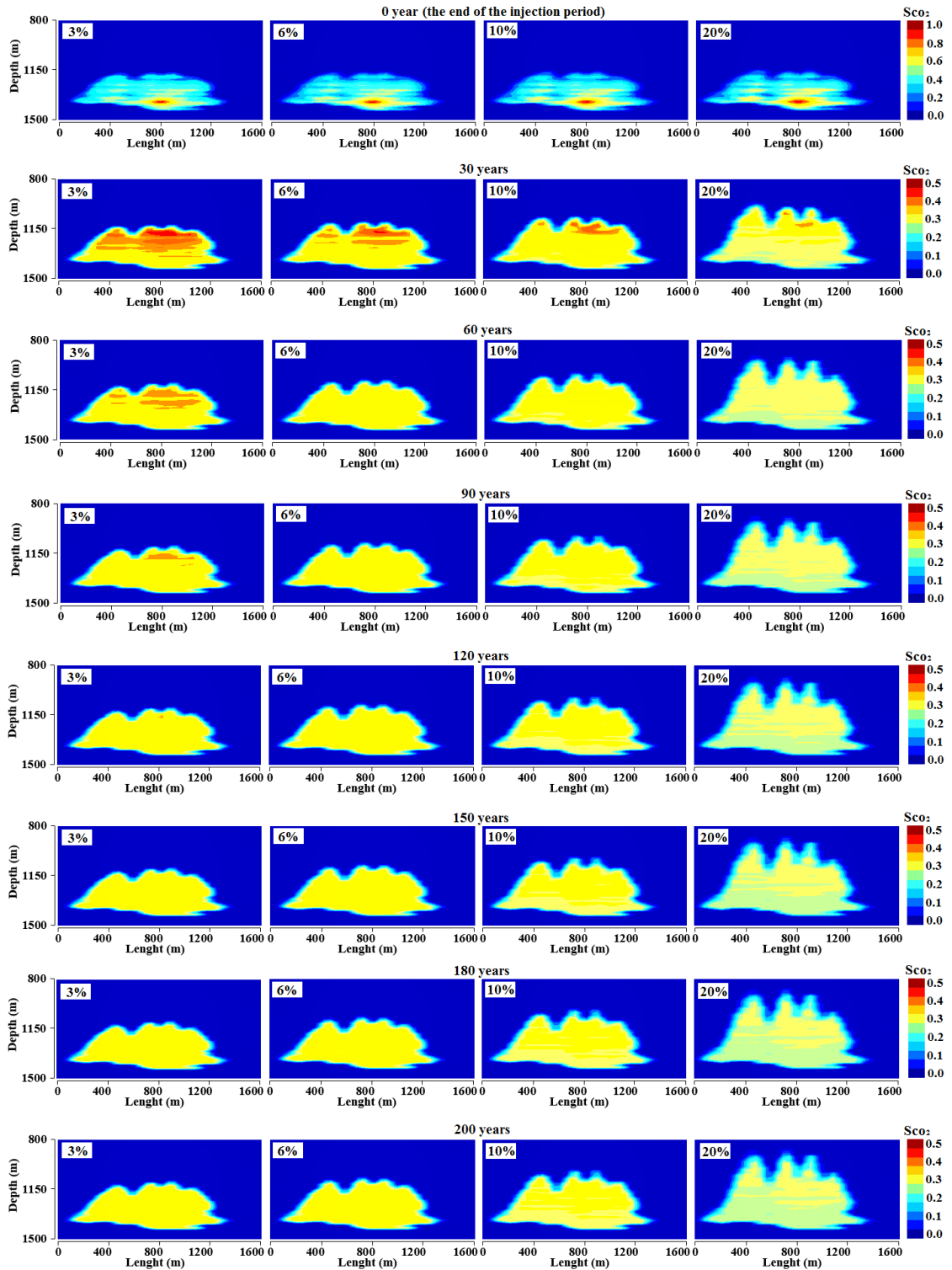


Figure 7-4 2D cross-sections through the middle of the storage reservoir. The CO₂ injection point is at a depth of 1373 m and a length of 800 m. CO₂ plume shape and migration are shown as a function of storage time and brine salinity; clearly salinity has a significant influence. 10 Mton of CO₂ were injected.

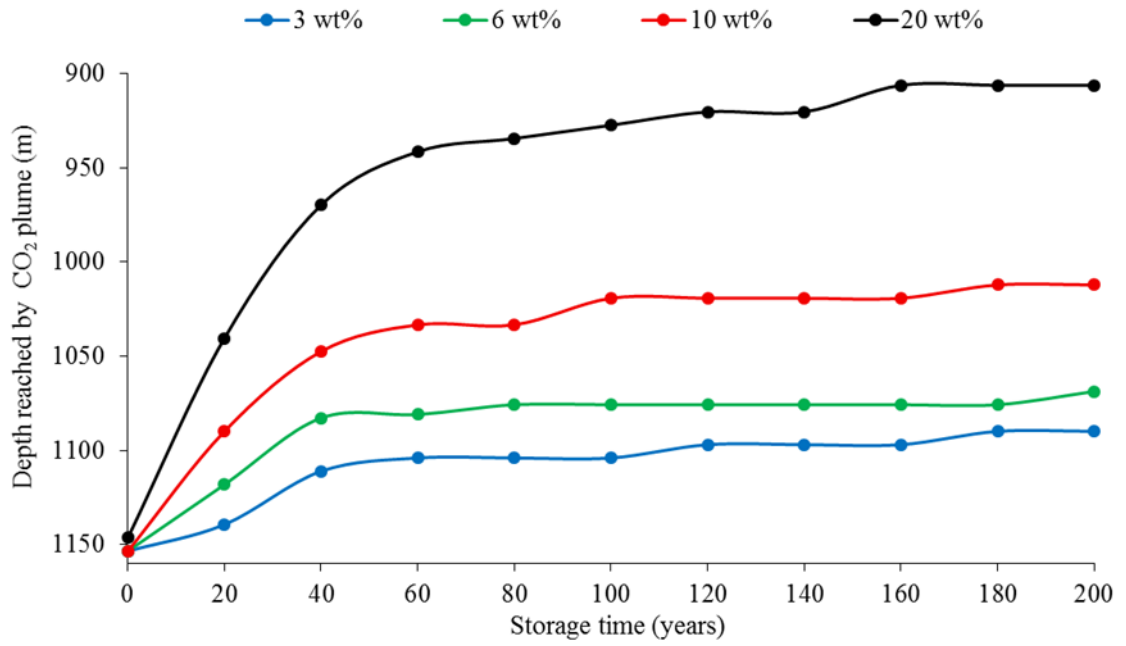


Figure 7-5 Reservoir depth reached by CO₂ plume as a function of storage time and salinity.

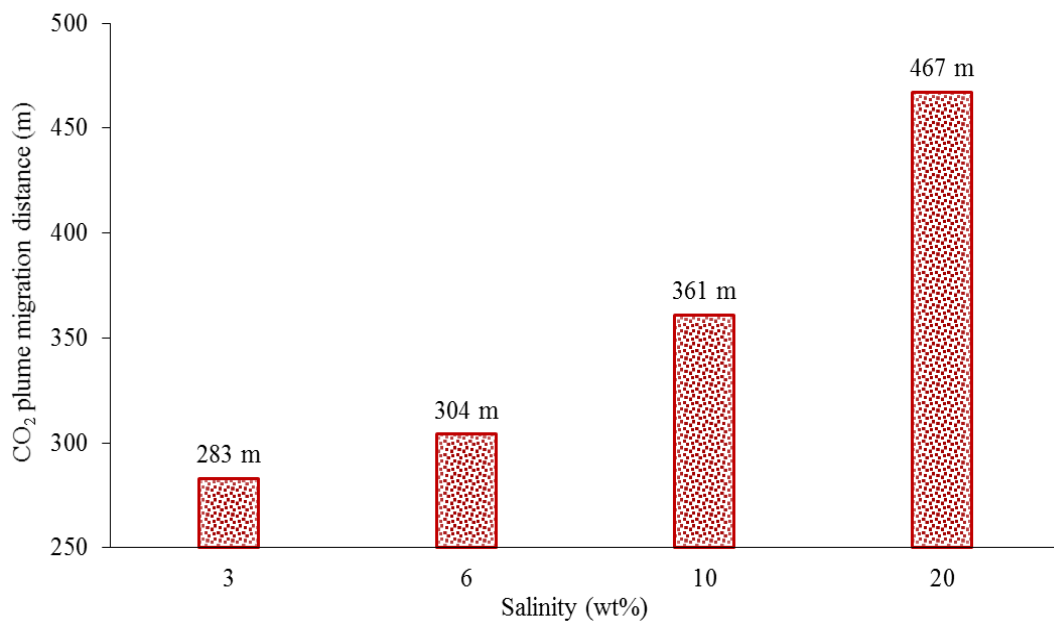


Figure 7-6. Total vertical CO₂ plume migration distance for the four salinity scenarios at the end of the storage period (after 200 years). The numbers indicate the exact distances.

7.3.2 Impact of salinity on CO₂ trapping capacity

Clearly, brine salinity has a considerable impact on solubility and residual (=capillary) trapping. Firstly, higher salinities lead to higher CO₂ mobility (following mobile CO₂ quantities were predicted after a 200 years storage period: 414 kton for 3 wt% salinity, 721 kton for 6 wt% salinity, 1135 kton for 10 wt% salinity, and 1760 kton for 20 wt% salinity; Figure 7-7 and Table 7-1), mainly due to the shifts in characteristics curves (see Figure 7-2 and discussion above). Related to this, capillary trapping capacities (Iglauer et al., 2011) are reduced in more saline brines (e.g. residual trapping capacity was 6445 kton for 20 wt% salinity, 6889 kton for 10 wt% salinity, 7219 kton for 6 wt% salinity and 7434 kton for 3 wt% salinity; at the end of the storage period after 200 years, Figure 7-8 and Table 7-1). This is consistent with our previous prediction that increased water wettability reduces the amount of mobile CO₂ and enhances residual trapping. (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b)

Moreover, it is also clear that lower brine salinity enhances solubility trapping (e.g. solubility trapping capacity was 1795 kton for 20 wt% salinity, 1976 kton for 10 wt% salinity, 2060 kton for 6 wt% salinity and 2152 kton for 3 wt% salinity; at the end of the storage period after 200 years, Figure 7-9 and Table 7-1). This is mainly driven by the higher thermodynamic CO₂ solubility in lower salinity brine. (Rumpf et al., 1994; Duan and Sun, 2003; Spycher and Pruess, 2004; Iglauer, 2011; El-Maghraby et al., 2012)

Our results thus indicate that brine salinity has a major impact on CO₂ storage efficiency, and we conclude that lower salinities improve CO₂ containment security.

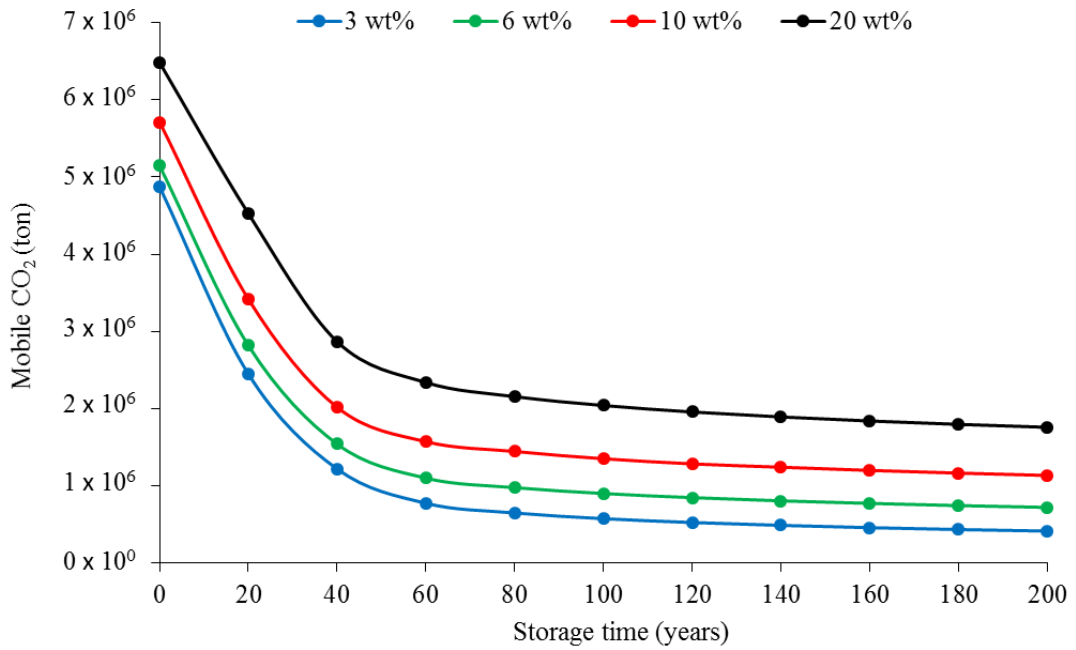


Figure 7-7 Amount of mobile CO₂ as a function of storage time and brine salinity. 10 Mton of CO₂ were injected.

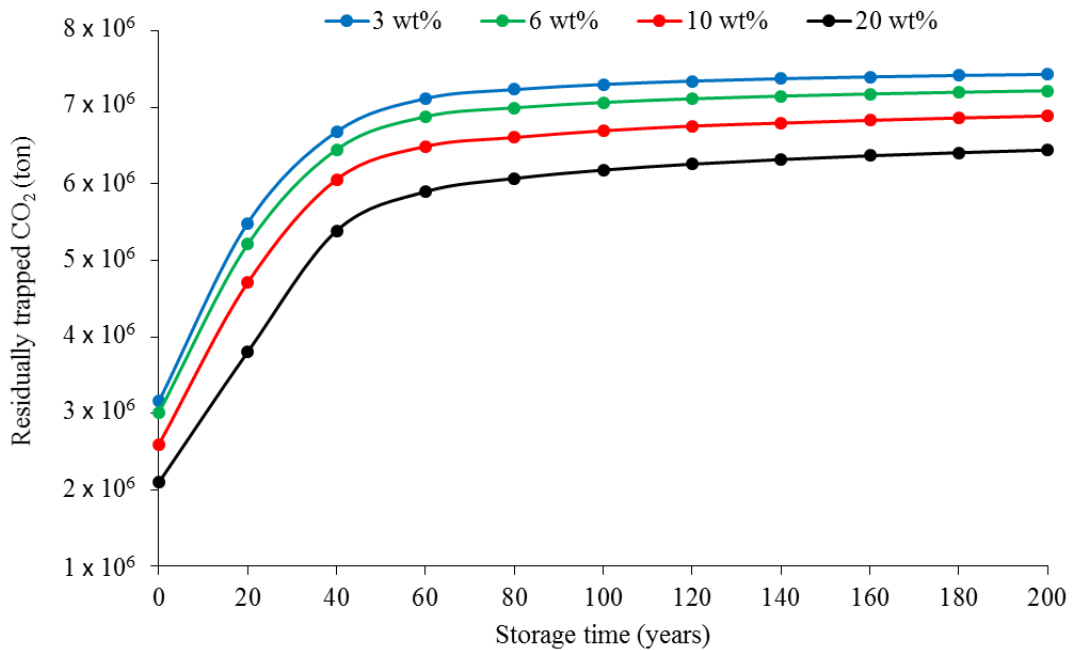


Figure 7-8 Amount of residually trapped CO₂ as a function of CO₂ storage time and brine salinity. 10 Mton of CO₂ were injected.

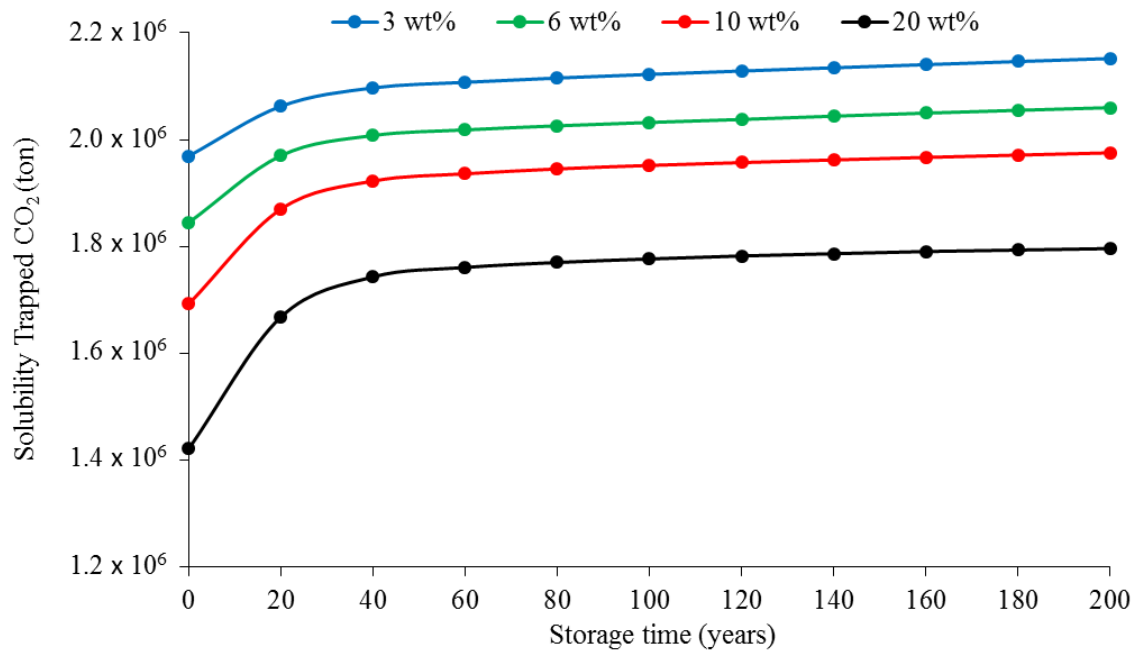


Figure 7-9 Amount of solubility trapped CO₂ as a function of storage time and brine salinity. 10 Mton of CO₂ were injected.

Table 7-1 Amounts of mobile and trapped CO₂ for different salinities after 200 years (at the end of the storage period). In all cases 10 Mt of CO₂ were injected

Salinity wt%	Solubility trapped CO ₂ (kton)	Residually trapped CO ₂ (kton)	Mobile CO ₂ (kton)	Total CO ₂ (kton)
3	2152	7434	414	10000
6	2060	7219	721	10000
10	1976	6889	1135	10000
20	1795	6445	1760	10000

7.4 Conclusions

Many factors influence CO₂ geo-storage efficiency, including permeability anisotropy,(Basbug et al., 2005) caprock properties,(Iglauer et al., 2015a) reservoir depth,(Zeidouni et al., 2015) injection well configuration,(Al-Khdheawi et al., 2017d) stratum dip angle,(Wang et al., 2016a; Wang et al., 2016b) key model

parameters,(Zhang et al., 2016a) reservoir wettability, (Iglauer et al., 2015a; Iglauer et al., 2015b; Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b) and the reservoir porosity and permeability heterogeneity. (Doughty and Pruess, 2004; Hovorka et al., 2004; Obi and Blunt, 2006; Bryant et al., 2006; Flett et al., 2007; Ide et al., 2007; Saadatpoor et al., 2009; Zhou et al., 2010; Ambrose et al., 2008; Han et al., 2010; Hesse and Woods, 2010; Green and Ennis-King, 2010; Gershenzon et al., 2015; Gershenzon et al., 2016) One factor, which, however, has not received sufficient attention, is brine salinity, even though it is well established that brine compositions vary widely in prospective storage reservoirs. (Morton and Land, 1987; Bachu and Bennion, 2008; Chalbaud et al., 2010; Al-Khdheawi et al., 2017a)

We thus computationally investigated the impact of brine salinity on CO₂ storage efficiency. To this end we developed a 3D highly heterogeneous reservoir model and predicted CO₂ migration distance, and the amounts of mobile, residually trapped and solubility trapped CO₂ for four salinity scenarios (i.e. 3 wt%, 6 wt%, 10 wt% and 20 wt% NaCl brines).

Our results indicate that brine salinity has a critical influence on the CO₂ plume migration distance, CO₂ mobility, and trapping capacities. Clearly, lower salinities lead to a reduction in a) the vertical CO₂ plume migration distance, and b) the amount of mobile CO₂; and lower salinities increase residual and solubility trapping capacities (by 10% and 4% over a 200 year storage period). This is consistent with previous predictions that lower water contact angles (i.e. more CO₂-wet rocks) lead to higher CO₂ mobility and less residual trapping (Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017b).

We thus conclude that, from a trapping capacity and containment security perspective, low salinity reservoirs are preferential CO₂ geo-storage sinks.

Chapter 8 Enhancement of CO₂ Trapping Efficiency in Heterogeneous Reservoirs by Water Alternating Gas Injection

8.1 Introduction

Underground storage of carbon dioxide into geological formations is currently one proposed strategy to mitigate CO₂ emissions that result from various industrial activities (Benson and Cole, 2008; Bachu and Adams, 2003; Pruess et al., 2003). However, because of the density difference between the injected CO₂ (the lighter phase, even though CO₂ is in a supercritical state at the injection depth) and the formation water (the heavier phase), CO₂ moves upward with a risk of leakage through pathways such as faults, micro fractures or existing wells (Vialle et al., 2016). One way to mitigate this CO₂ leakage risk is to reduce the vertical CO₂ migration (which is induced by the above mentioned buoyancy forces) by increasing the efficiency of the different CO₂ trapping mechanisms. These mechanisms include structural trapping (caprock barrier) (Iglauer et al., 2015a; Hesse and Woods, 2010), capillary trapping (Iglauer et al., 2011; Iglauer and Wüiling, 2016; Krevor et al., 2015; Pentland et al., 2011; Ruprecht et al., 2014; Suekane et al., 2008; Rahman et al., 2016), solubility trapping (Emami-Meybodi et al., 2015; Iglauer, 2011; Spycher et al., 2003) and mineral trapping (Bachu et al., 1994b; Gaus, 2010; Metz et al., 2005; Xu et al., 2003; Xu et al., 2004; Xu et al., 2005).

The effectiveness of the trapping mechanisms depends on various geological and hydraulic parameters, including geological reservoir heterogeneities (Al-Khdheawi et al., 2017b; Ambrose et al., 2008; Doughty and Pruess, 2004; Flett et al., 2007; Gershenson et al., 2015; Gershenson et al., 2016; Gershenson et al., 2017; Green and Ennis-King, 2010; Hovorka et al., 2004; Saadatpoor et al., 2009), caprock properties (Iglauer et al., 2015a; Naylor et al., 2011), CO₂-rock wettability (Al-Khdheawi et al., 2017b; Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d; Iglauer et al., 2015b; Iglauer, 2017; Al-Menhali et al., 2015; Al-Khdheawi et al., 2017e; Chaudhary et al., 2013), reservoir temperature (Al-Khdheawi et al., 2018a), wettability heterogeneity (Al-Khdheawi et al., 2018a) and brine salinity (Al-Khdheawi et al., 2018; Al-Khdheawi et al., 2017a). In addition, while many

geological parameters cannot be changed for a particular site, it has been shown that intelligent selection of CO₂ injection technology (e.g. the well configuration) can significantly improve storage capacity and containment security (Al-Khdheawi et al., 2017d; Anchliya and Ehlig-Economides, 2009).

Here, we investigate and discuss a process that can be controlled, namely the CO₂ injection scheme. Specifically, we compare the efficiency of a Water-Alternating-Gas (WAG) injection with the traditionally implemented continuous CO₂ injection, used in most of the current CCS projects in the world such as the Sleipner project in Norway (Adam, 2001), and the Quest project in Canada (Bourne et al., 2014). WAG injection could indeed affect the CO₂ storage efficiency because of its well-known role in improving both the microscopic and macroscopic sweep efficiencies in oil reservoirs (Rogers and Grigg, 2001; Zheng and Yang, 2013). Even though WAG technology has been widely used in various industrial applications such as enhanced oil recovery (EOR) (Christensen et al., 2001; Dang et al., 2016; Fatemi and Sohrabi, 2013; Kulkarni and Rao, 2005; Laochamroonvorapongse et al., 2014; Sohrabi et al., 2004; Song et al., 2014b; Teklu et al., 2016), its effect on CO₂ trapping efficiency has not been addressed.

In this chapter, we therefore compare three different CO₂ injection scenarios, namely continuous CO₂ injection, intermittent CO₂ injection, and water alternating gas injection, using numerical multiphase flow simulations in a highly heterogeneous, hectometre-sized storage formation. For each injection scenario, we compute the vertical CO₂ plume migration distance, solubility trapping and capillary trapping capacities over a period of 100 years after the CO₂ injection has ceased. The results of this study show that WAG injection leads to significantly improved storage capacity and containment security: this is an important result for the commercial development and public acceptance of this technology.

8.2 Methodology

To test the efficiency of the different injection scenarios, (i.e. continuous, intermittent, and water alternating gas injection), hectometer scale reservoir simulations were conducted, and the predicted trapping parameters were quantified and compared. Note

that, for all tested injection scenarios, one geological scenario (i.e. heterogeneous reservoir) has been used.

8.2.1 Model characteristics

A heterogeneous reservoir with dimensions 1600 m × 1200 m × 700 m (with a regular Cartesian grid containing $37 \times 33 \times 80 = 97680$ grid blocks) was employed. Geological reservoir heterogeneity parameters (i.e. porosity and permeability distribution) were taken from the SPE comparative solution project (Figure 8-1) (Christie and Blunt, 2001). This comparative solution project was initially developed for the PUNQ project (Floris et al., 1999). The model parameters are tabulated in Table 8-1.

Fluid flow was simulated using the nonisothermal multicomponent multiphase flow simulator TOUGH2 (Pruess et al., 1999) combined with the ECO2M module (Pruess, 2011). ECO2M computes the thermodynamic and thermophysical properties (e.g. viscosity, density and specific enthalpy) of H₂O, NaCl and CO₂ mixtures, including phase changes and compositions (Pruess, 2011).

In all injection scenarios, 5 Megatonnes (Mt) of CO₂ were injected at a depth of 1373 m over a 10 year period. For the continuous CO₂ injection, a continuous injection rate of 0.5 Mt/year was used. For the intermittent injection, 5 CO₂ injection cycles, at a rate of 1 Mt/year each was employed; each injection was followed by a 1 year shut-off period. In the WAG injection, 5 CO₂ injection cycles at a rate of 1 Mt/year were also carried out, but now each injection was followed by a 1 year water injection period at a rate of 3 Mt/year. All injections were followed by a 100 year observation period (post-injection period) for which the vertical CO₂ plume migration distance, solubility and capillary trapping capacities were computed and subsequently analysed.

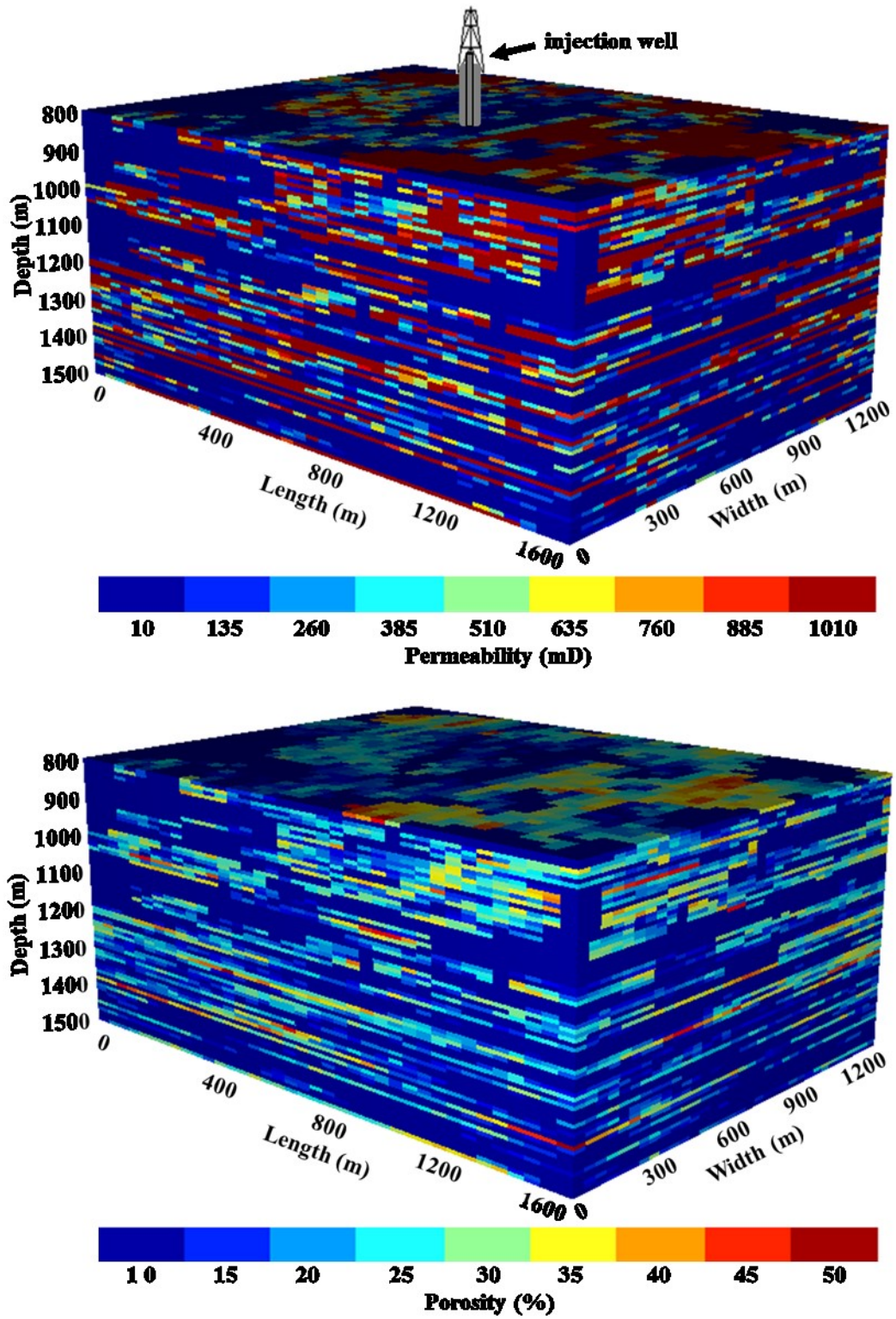


Figure 8-1 3D views of the heterogeneous model showing the location of the CO₂ and water injection well, model dimensions and permeability heterogeneity (on top) and porosity heterogeneity (on bottom).

Table 8-1 Reservoir model parameters

Property	Value
Length	1600 m
Width	1200 m
Thickness	700 m
Depth of the top of the reservoir	800 m
Depth of the bottom of the reservoir	1500 m
Grid block number	$37 \times 33 \times 80 = 97680$ grid blocks
k_v/k_h in the storage reservoir	0.1
k_v/k_h at the top layer of the reservoir ^a	10^{-6}
Boundary cells volume multiplier ^b	10^8
Brine salinity	60000 ppm
Initial pore pressure at 1200 m depth	12 MPa
Pressure gradient ^c	0.01 MPa/m
Reservoir temperature (isothermal)	333 K
Initial water saturation	100%
Injection depth	1373 m
Dip of the strata	0° (i.e. horizontal reservoir)

^aThe vertical to horizontal permeability ratio (k_v/k_h) is reduced to 10^{-6} at the top layer cells of the reservoir to simulate a barrier preventing the CO₂ from leaking out of the model (Al-Khdheawi et al., 2017b; Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d; Hesse and Woods, 2010; Birkholzer et al., 2009; Zhou et al., 2008).

^bThe volumes of the bottom and side boundary cells have been multiplied by a factor of 10E8 to simulate constant pressure conditions at the reservoir outer boundaries (Nghiem et al., 2010).

^cA pressure gradient of 0.01 MPa/m (Dake, 2007) was applied to simulate the hydrostatic pressure gradient in the reservoir.

8.2.2 Implementation of characteristics curves

It has been shown that characteristics curves (i.e. relative permeability and capillary pressure curves) have a significant impact on the CO₂-brine fluid dynamics in the reservoir (Woods and Farcas, 2009; Kopp et al., 2009; Hesse et al., 2008; Doughty, 2007). Thus, for simulating the continuous and intermittent CO₂ injections, we used previously developed relative permeability and capillary pressure curves for an intermediate-wet condition (Al-Khdheewi et al., 2017b; Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017d). These curves were built using experimental data (Anderson, 1987b; Anderson, 1987a; Craig, 1993; Batycky et al., 1981; McCaffery and Bennion, 1974; Melrose, 1965; Morrow, 1976; Owens and Archer, 1971). Intermediate-wet conditions were chosen as this is the most likely wettability scenario in CO₂-brine deep saline aquifer systems (Iglauer et al., 2014; Iglauer et al., 2015a; Iglauer et al., 2015b; Iglauer, 2017). In addition, rock wettability has a significant effect on the residual water and gas saturations (Anderson, 1987a; Anderson, 1987b; Craig, 1993; Iglauer et al., 2011). Specifically, for the construction of the relative permeability curves, we used McCaffery and Bennion's data (1974). We used Didger software (Golden Software Inc, 2013, Colorado) to digitize these curves and obtain data for the relative permeability values as a function of water saturation; we adjusted these curves using Craig's rules of thumb (Craig, 1993). Hence, to simulate the intermediate-wet wettability condition, we used the values of 0.22 for S_{wr} and 0.25 for S_{co_2r} as in our recent publications (Al-Khdheewi et al., 2017b; Al-Khdheewi et al., 2017c; Al-Khdheewi et al., 2017d). In addition, the water saturation (S_w) at which the CO₂ relative permeability of CO₂ (k_{rco_2}) and that of water (k_{rw}) are equal was selected to be 0.5 according to Craig's rule of thumb (Craig, 1993).⁷⁴ For all injection scenarios, initially k_{rw} is set to 1 and k_{rco_2} to 0, which corresponds to full (100%) water saturation. During the first CO₂ injection process k_{rw} reduces gradually, while k_{rco_2} increases until it reaches a maximum at the irreducible water saturation (S_{wr}).

Furthermore, and importantly, these characteristics curves, and the associated residual gas and water saturations (Herring et al., 2013; Herring et al., 2015; Herring et al., 2016; Skauge and Larsen, 1994; Akbarabadi and Piri, 2013; Benson et al., 2013; Krevor et al., 2011; Krevor et al., 2012; Pini et al., 2012; Reynolds and Krevor, 2015; Tokunaga et al., 2013) were modified based on Herring et al.'s data (Herring et al., 2016) for the WAG scCO₂-brine floods (Figure 8-2 and Figure 8-3). For the first WAG

cycle we used the values of the S_{CO_2r} (residual CO_2 saturation) and S_{CO_2i} (initial CO_2 saturation) of the intermediate-wet rock. Then for the cycles 2-5 the S_{CO_2r} has been enhanced based on the experimental studies (Herring et al., 2013; Herring et al., 2015; Herring et al., 2016; Skauge and Larsen, 1994; Tokunaga et al., 2013). For all five WAG cycles the relationship of the S_{CO_2r} versus S_{CO_2i} followed Pentland's et al. measurements (Pentland et al., 2011). The Genuchten-Mualem model (Van Genuchten, 1980; Mualem, 1976) was then used to import the characteristics curves into the TOUGH2 code:

$$k_{rw} = \sqrt{S^*} \left\{ 1 - \left(1 - [S^*]^{1/\lambda} \right)^\lambda \right\}^2 \quad \text{if } S_w < S_{ws} \quad (8.1)$$

$$k_{rw} = 1 \quad \text{if } S_w \geq S_{ws} \quad (8.2)$$

$$k_{rcO_2} = 1 - k_{rw} \quad \text{if } S_{CO_2r} = 0 \quad (8.3)$$

$$k_{rcO_2} = (1 - \hat{S})^2 (1 - \hat{S}^2) \quad \text{if } S_{CO_2r} > 0 \quad (8.4)$$

$$(P_{cap}) = P_0 ([S^*]^{-1/\lambda} - 1)^{1-\lambda} \quad (8.5)$$

$$S^* = (S_w - S_{wr}) / (S_{ws} - S_{wr}), \quad (8.6)$$

$$\hat{S} = (S_w - S_{wr}) / (1 - S_{wr} - S_{CO_2r})$$

where:

k_{rcO_2} = CO_2 relative permeability,

k_{rw} = water relative permeability,

S_{CO_2r} = CO_2 residual saturation,

S_w = water saturation,

S_{ws} = saturated water saturation,

S_{wr} = water residual saturation,

P_{cap} = CO_2 -water capillary pressure,

P_0 = capillary pressure scaling factor.

λ = pore size distribution index,

Moreover, the Leverett J-function (Leverett, 1941) was used to implement the effect of porosity and permeability heterogeneity on the capillary pressure curves for each grid cell:

$$J(S_w) = \frac{P_c}{\sigma \cos \Theta} \sqrt{\frac{k}{\phi}} \quad (8.7)$$

where:

$J(S_w)$ = dimensionless capillary pressure,

k = rock permeability,

ϕ = rock porosity,

σ = CO₂-water interfacial tension,

θ = CO₂-water-rock contact angle.

P_c = CO₂-water capillary pressure

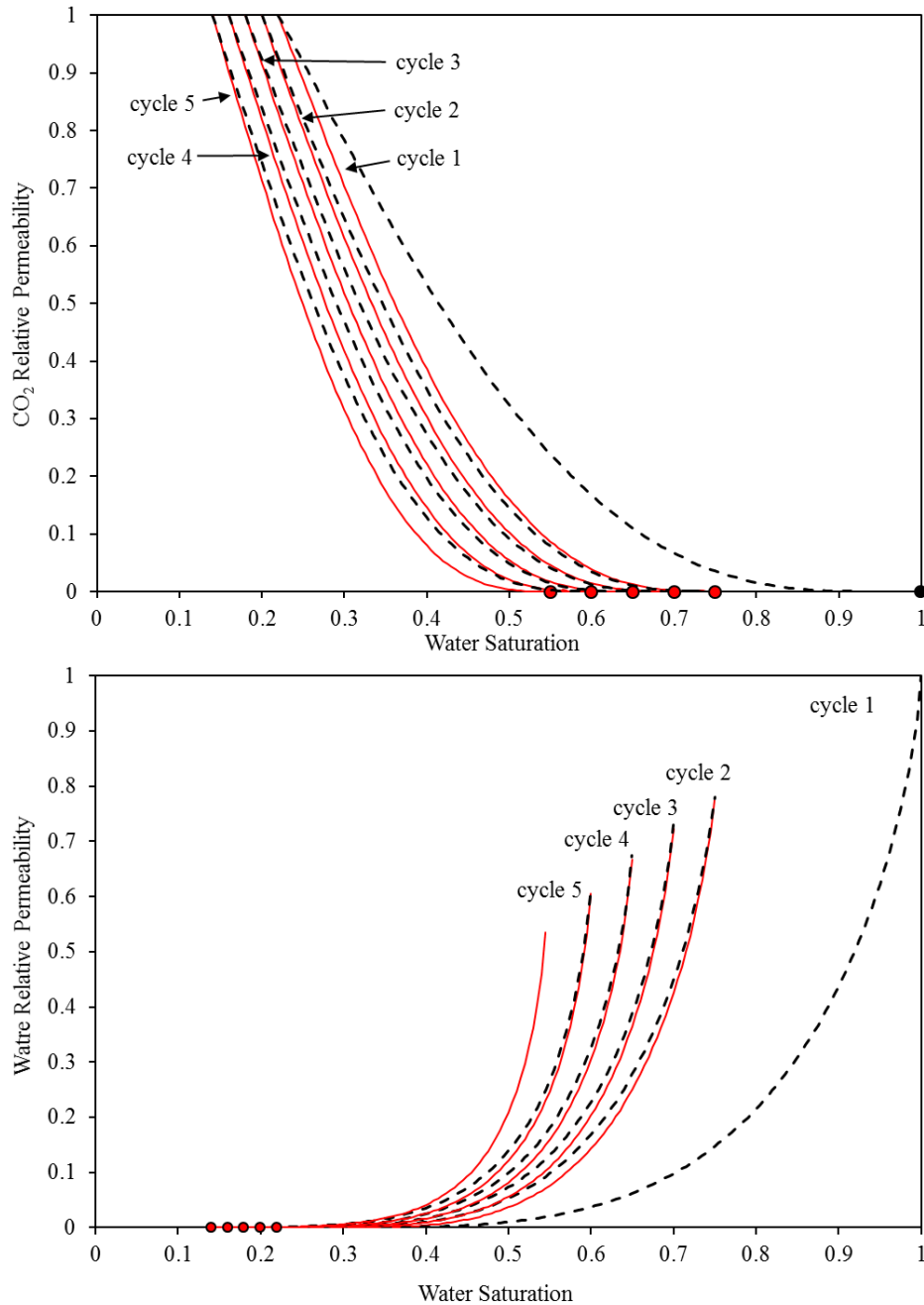


Figure 8-2 CO₂ and water relative permeability curves used for the five water-alternating gas injection cycles. The dashed black lines represent the CO₂ injection process and continuous red lines represent the water injection process. Note that cycle 1 represents the normal drainage and imbibition curves of the intermediate-wet rocks that have been used for the continuous and intermittent injection scenarios. Derived

from Al-Khdheawi et al. (Al-Khdheawi et al., 2017b; Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d) and extended for different WAG cycles based on the experimental studies (Herring et al., 2013; Herring et al., 2015; Herring et al., 2016; Skauge and Larsen, 1994; Tokunaga et al., 2013).

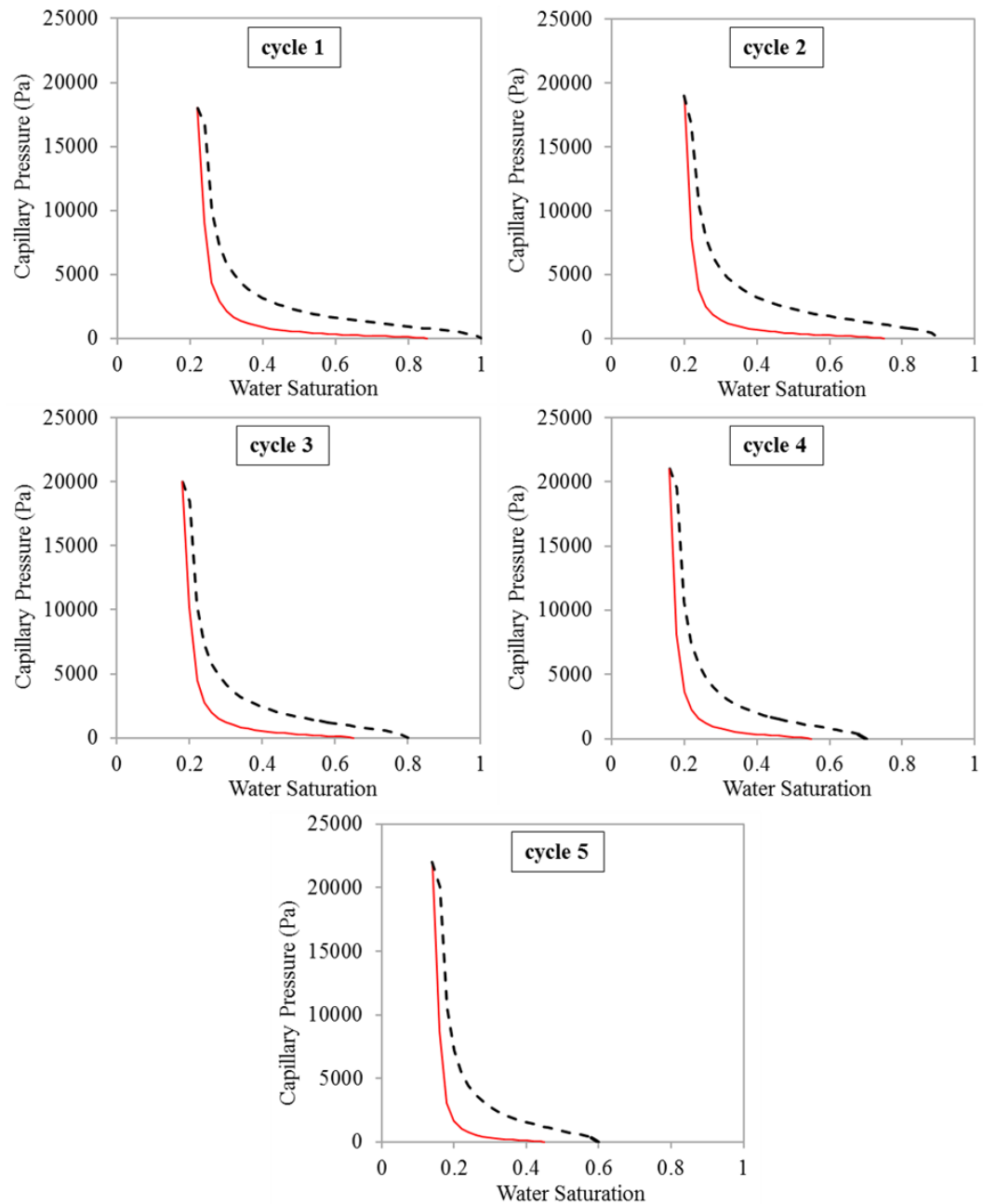


Figure 8-3 Capillary pressure curves used for the five water-alternating gas injection cycles. The dashed black lines represent the CO₂ injection process and continuous red lines represent the water injection process. Note: cycle 1 represents the intermediate-wet rock that underwent normal drainage and imbibition (used for continuous and intermittent injection). Derived from Al-Khdheawi et al. (Al-Khdheawi et al.,

2017b; Al-Khdheawi et al., 2017c; Al-Khdheawi et al., 2017d) and extended for different WAG cycles based on the experimental studies (Herring et al., 2013; Herring et al., 2015; Herring et al., 2016; Skauge and Larsen, 1994; Tokunaga et al., 2013).

8.3 Results and discussion

8.3.1 Storage capacity enhancement via WAG injection

Generally, in CO₂ geo-sequestration projects, higher vertical CO₂ migration is unwanted, because it increases CO₂ migration risk from the reservoir to the surface, while storage capacities need to be maximized. Here, we thus study the effect of continuous, intermittent and WAG injection on these storage parameters. Our results show that WAG injection had the lowest vertical CO₂ plume migration, less than half of the migration distance compared to continuous or intermittent injections, over the whole post-injection period of 100 years (

Figure 8-4 and Figure 8-5). For example, the results show that the total vertical CO₂ plume migration distance was 432 m for the intermittent injection, 453 m for the continuous injection, and only 198 m in case of WAG, after a 100 year storage period.

Mechanistically, this reduction in vertical CO₂ plume migration was caused by the relatively high residual CO₂ saturation (S_{co_2r}) in WAG (Herring et al., 2016), which diminished vertical CO₂ flow. Importantly, previous studies showed that vertical CO₂ plume migration distance is a function of the residual CO₂ saturation and that lower S_{co_2r} leads to an increase in CO₂ plume migration distance (Metz et al., 2005; Hesse et al., 2008; Doughty, 2010; Al-Khdheawi et al., 2017b).

We thus conclude that WAG injection leads to a significant reduction in CO₂ leakage risk, implying superior containment security.

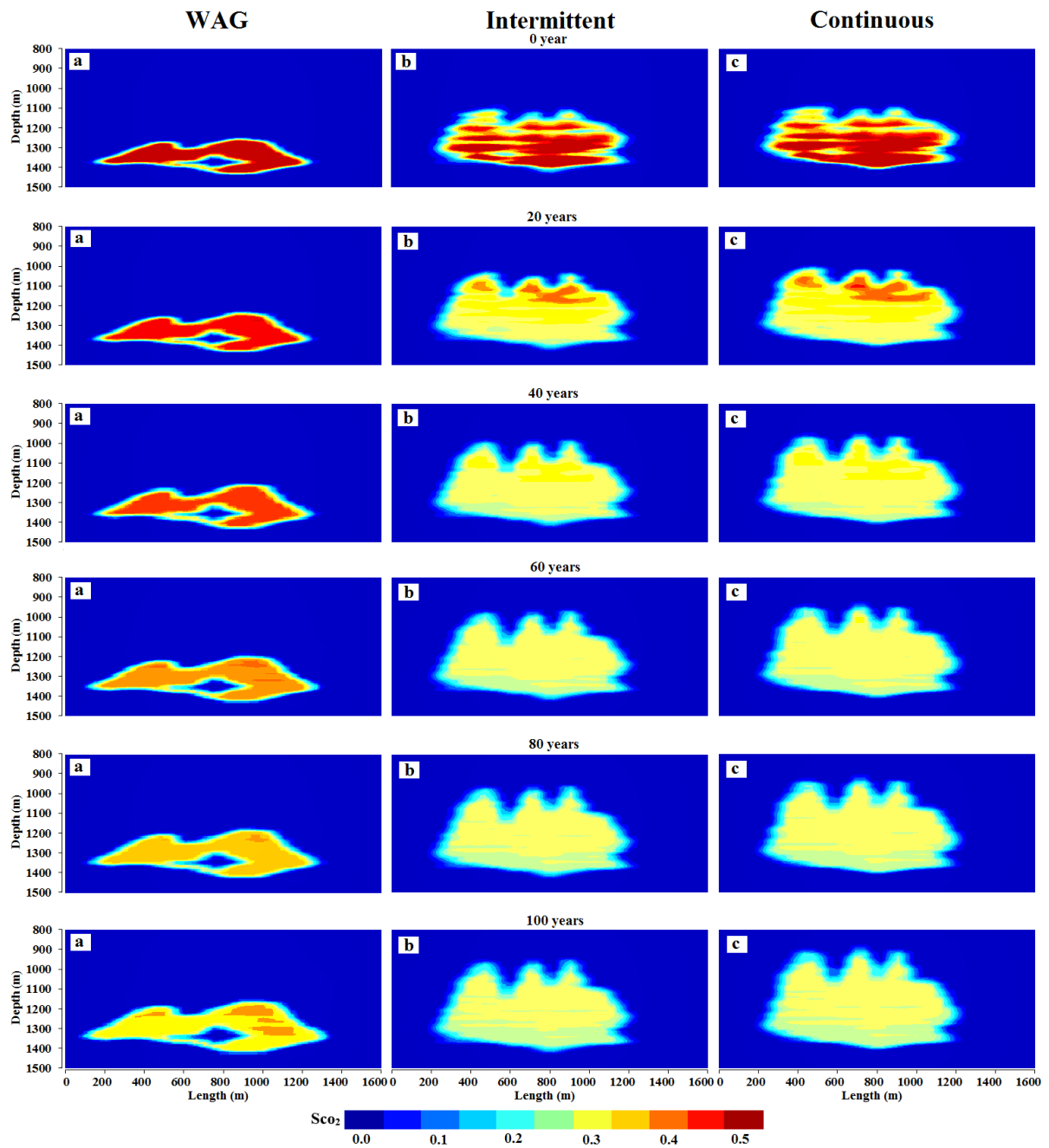


Figure 8-4 2D (X-Z) views through the center of the reservoir showing the CO₂ plume as a function of storage (post-injection) time and CO₂ injection scenario: a) WAG injection, b) intermittent injection, c) continuous injection. S_{co₂} represents the supercritical CO₂ saturation.

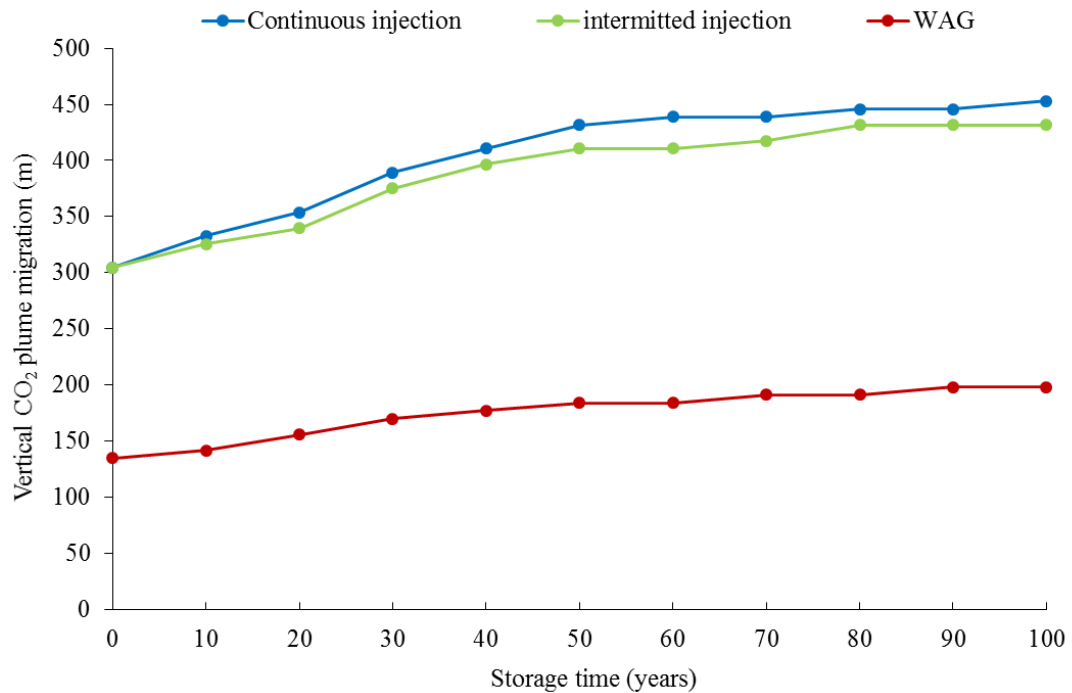


Figure 8-5 Vertical CO₂ plume migration distance as a function of storage time and injection scenario.

8.3.2 Maximising CO₂ trapping

Clearly, WAG injection significantly improves CO₂ trapping efficiency (Figure 8-6). Firstly, WAG maximises residual (capillary) trapping by increasing the residual CO₂ saturation for the imbibition cycle. For instance, after a 100 year storage period, 67% of the CO₂ was residually trapped in case of WAG, while only 52% and 49% (intermittent and continuous injection, respectively) were trapped otherwise. Recall, WAG significantly increases the residual CO₂ saturations (Figure 8-2 and Figure 8-3) (Herring et al., 2016).

Secondly, WAG injection also enhanced solubility trapping. This is of key importance as only CO₂ dissolved in water is truly safely stored and cannot leak back to the surface. As an example, after a 100 year storage period, 30% of the CO₂ was solubility trapped in WAG, while only 21% and 19% (intermittent and continuous injection, respectively) was solubility trapped in the other scenarios. In addition, the maximum mass fraction of CO₂ dissolved in the aqueous phase (X_{CO_2aq}) after the end of the post-injection period (100 years) was ~0.05 for WAG scenario, compared to only ~0.025 for the intermittent and ~0.02 for the continuous CO₂ injection scenarios

(Figure 8-7). The reason for this is the larger lateral CO₂ spreading in cyclic WAG injection, which results in larger CO₂-water contact surface in WAG scenario (Doughty, 2010). Consequently, overall, WAG injection significantly reduces (unwanted) CO₂ mobility (i.e. the amount of free CO₂). In statistical terms, after 100 years the ratio of mobile to total injected CO₂ was only 3% for WAG while it was 27% for intermittent and 32% for continuous CO₂ injection (Figure 8-6).

Furthermore, our simulation results show that WAG injection has a higher overall CO₂ storage efficiency (via both residual and solubility trapping) than simultaneous water and gas injection (SWAG). For instance, after a 100 year storage period, the overall storage efficiency for WAG injection was 97% (i.e. only 3% of CO₂ is free) compared to only 75% (i.e. 25% of CO₂ is free) for the best SWAG injection scenario (Baz et al., 2016). Importantly, our results in terms of improving the storage efficiency by WAG injection are consistent with our previous study performed in homogeneous reservoirs (Al-Khdheawi et al., 2018b), which concluded that, after a 200 years storage period, water alternating CO₂ (WACO₂) injection technology can enhance residual trapping by 13% and dissolution trapping by 14% when compared with the standard continuous CO₂ injection. It is important to mention that changing the boundary conditions may quantitatively affect the estimated trapping efficiencies. However, this change in boundary conditions will not change the basic difference between the different injection scenarios (i.e. the highest trapping efficiency will be obtained in the WAG scenario compared to the other two injection scenarios).

We thus conclude that WAG strongly enhances solubility and residual trapping and minimizes (detrimental) vertical CO₂ mobility, implying much better CO₂ geo-sequestration efficiency. WAG is therefore the preferred injection scheme.

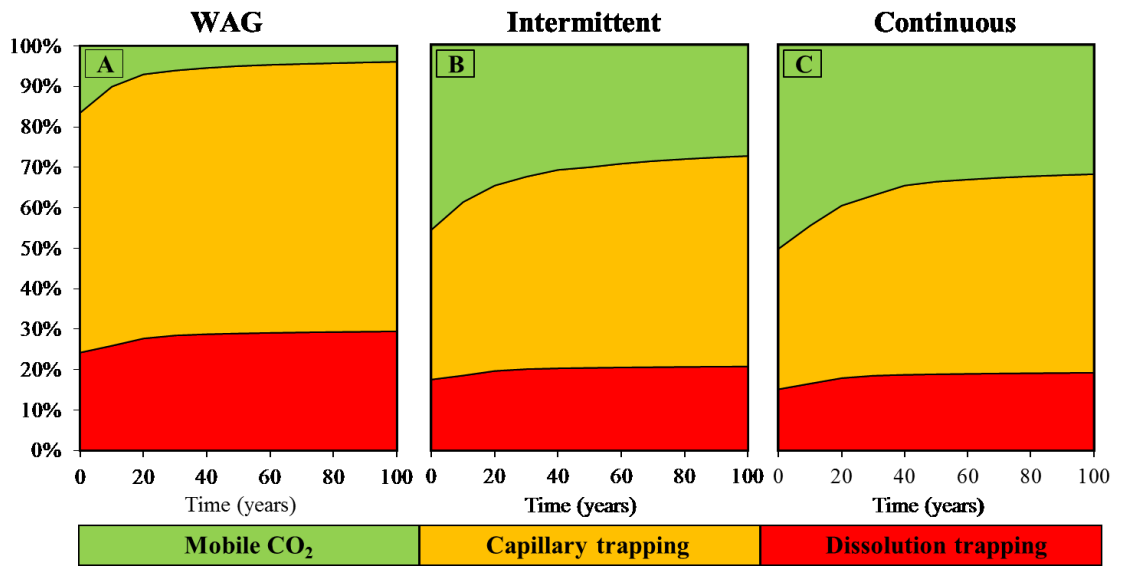


Figure 8-6 Percentage of mobile CO₂ (in green), capillary trapped CO₂ (in yellow) and solubility trapped CO₂ (in red) as a function of post-injection time and CO₂ injection scenario; A) WAG injection, B) intermittent CO₂ injection, C) continuous CO₂ injection. Clearly, WAG is the optimal CO₂ disposal scheme.

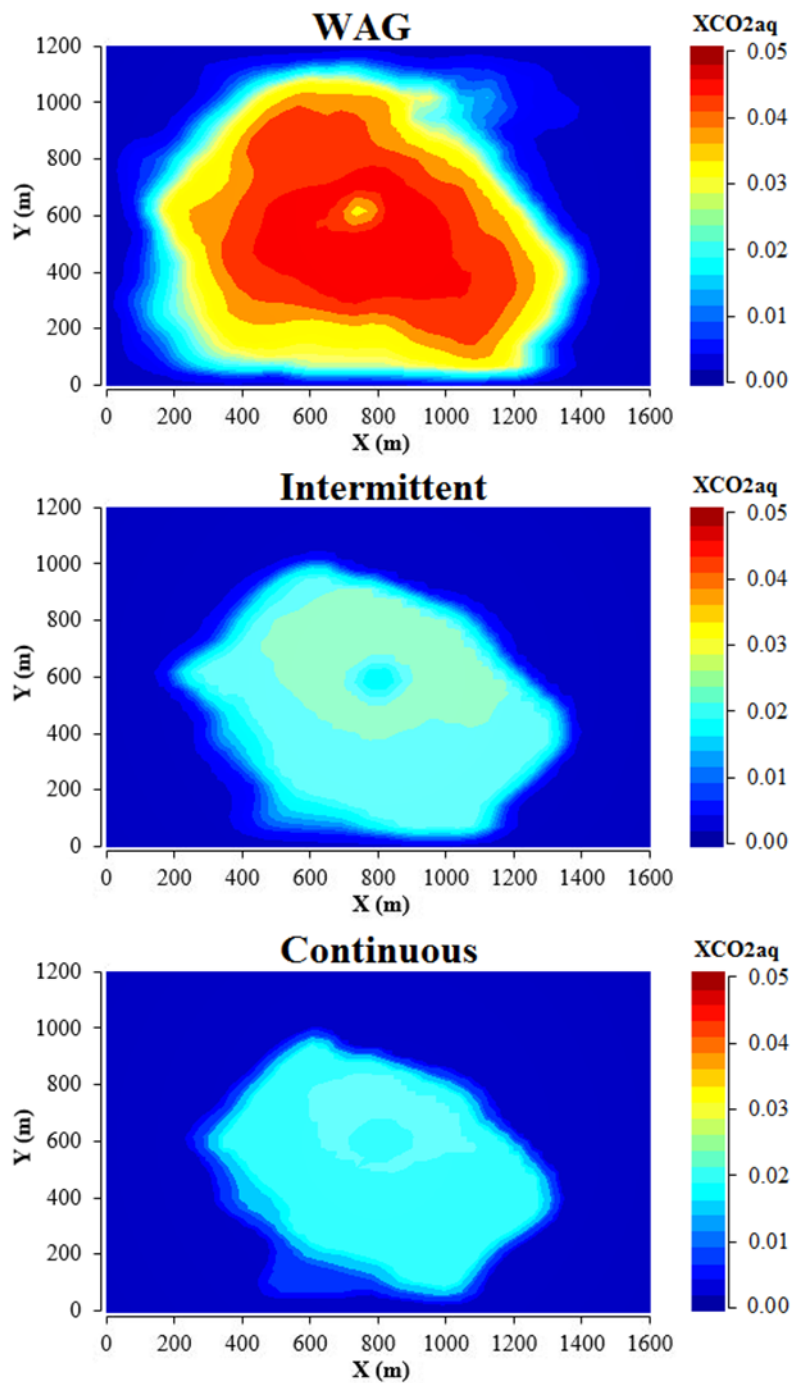


Figure 8-7 2D (X-Y) views through the reservoir at the CO₂ injection point at a depth of 1373 m showing the spreading of the dissolved CO₂ after the post-injection period (100 years) for the different injection scenarios: WAG injection (top), intermittent injection (middle), and continuous injection (bottom). Clearly, WAG has the largest lateral CO₂ spreading area and the highest mass fraction of dissolved CO₂ in aqueous phase (XCO_{2aq}).

8.4 Conclusions

Carbon dioxide storage in deep saline aquifers is currently considered as the most effective technology to mitigate climate change by reducing CO₂ emissions (Benson and Cole, 2008; Bachu and Adams, 2003; Pruess et al., 2003). The efficiency of CO₂ storage in saline aquifers can be improved either by the accurate evaluation of the various factors affecting CO₂ trapping mechanisms (e.g. reservoir heterogeneity, caprock properties, rock wettability and brine salinity) or by optimising the CO₂ injection technology (Al-Khdheawi et al., 2017d; Anchliya and Ehlig-Economides, 2009). Although WAG technology has been widely used as an important method to improve the efficiency of enhance oil recovery process (EOR) (Christensen et al., 2001; Dang et al., 2016; Fatemi and Sohrabi, 2013; Kulkarni and Rao, 2005; Laochamroonvorapongse et al., 2014; Sohrabi et al., 2004; Song et al., 2014b; Teklu et al., 2016) and it affects residual gas and water saturations (Herring et al., 2013; Herring et al., 2015; Herring et al., 2016; Skauge and Larsen, 1994; Tokunaga et al., 2013), its effect on the CO₂ trapping efficiency is not completely understood.

Thus, here we investigated how WAG influences CO₂ storage efficiency in one heterogeneous reservoir scenario. Recall that the impact of the WAG injection on the relative permeability and capillary pressure curves, and the associated residual gas and water saturations were simulated based on previous experimental studies (Herring et al., 2013; Herring et al., 2015; Herring et al., 2016; Skauge and Larsen, 1994; Akbarabadi and Piri, 2013; Benson et al., 2013; Krevor et al., 2011; Krevor et al., 2012; Pini et al., 2012; Reynolds and Krevor, 2015; Tokunaga et al., 2013). Our results clearly showed that WAG injection reduced the total vertical CO₂ plume migration (by more than 50%), and that it also reduced CO₂ mobility (by > 24%) when compared with other injection scenarios (i.e. continuous and intermittent injections). In addition, WAG enhanced residual trapping (by ~ 17%) and solubility trapping (by ~ 9%) when compared with the conventional CO₂ injection process (i.e. continuous CO₂ injection).

Thus, we conclude that WAG injection can significantly enhance CO₂ storage efficiency and containment security, and WAG is thus the preferred injection scheme.

Our results have significant implications for enhancing the CO₂-geosequestration efficiency, especially in CO₂ sequestration projects which rely on residual and

dissolution trapping (e.g. the South West Hub carbon capture and storage project in Western Australia) (Stalker et al., 2013).

Chapter 9 Conclusions, Recommendations and Outlook for Future Work

9.1 Conclusions

In the context of CO₂ geo-sequestration, the flow of CO₂ through porous media, as well as its storage capacity, are highly affected by different parameters. One key parameter, which has received little attention, is the rock wettability. This thesis, for the first time, investigated the influence of the rock wettability and wettability heterogeneity on CO₂ geo-sequestration efficiency and on CO₂ movements through porous media by performing various multiphase flow reservoir simulation models at a hectometre-scale. Importantly, in this thesis, and for the first time, five cases of CO₂-water-rock wettability have been implemented, from strongly water-wet to strongly CO₂-wet. To do so, a pair of relative permeability and capillary pressure curves for each wettability condition has been developed to simulate both CO₂ injection and storage processes. In addition, this thesis also included a sensitivity analysis of the effect of some other reservoir parameters, such as reservoir temperature, formation water salinity and permeability and porosity heterogeneity on CO₂ geo-sequestration efficiency. This thesis also covers the effect of engineering parameters on CO₂ geo-sequestration. Thus, the influence of using different CO₂ injection technologies such as injection well configurations (i.e. vertical and horizontal wells) and CO₂ injection scheme (i.e. continuous injection, intermittent injection, and water-alternating gas (WAG) injection) on CO₂ geo-sequestration efficiency have been investigated.

The thesis concludes that water-wet formations with homogeneous wettability improve the CO₂ geo-sequestration capacity and containment security. Water-wet formations with more homogeneous porosity and permeability distribution, lower formation water salinities, lower reservoir temperatures, and those instrumented with horizontal wells and WAG injection operations, were shown to be superior geo-sequestration candidates. Generally, the findings of this thesis lead to a better understanding of the important effect of rock wettability and wettability distribution, in addition to the effect other reservoir parameters and the used CO₂ injection technology on CO₂ geo-sequestration efficiency and CO₂ movements through porous media. The most significant findings of this thesis are summarized below:

1. Influence of wettability on CO₂ geo-sequestration in homogeneous saline aquifers
 - ❖ CO₂-wettability of rocks can vary tremendously, from strongly water-wet to strongly CO₂-wet
 - ❖ CO₂ is best retained in water-wet rock, while CO₂-wet reservoirs are relatively much more permeable to CO₂.
 - ❖ the shape of the CO₂ plume is also strongly affected by wettability, the plume is much more compact in case of water-wet rock, while it is vertically elongated in CO₂-wet rock.
 - ❖ The amount of residually trapped CO₂ is significantly higher in water-wet rock.
 - ❖ Dissolution trapping is more effective in CO₂-wet rock.
 - ❖ wettability significantly changes migration patterns and storage capacities, which is directly relevant to CO₂ geo-storage projects.
 - ❖ Strongly water-wet reservoirs are preferable CO₂ sinks due to their higher storage capacities and superior containment security.
 - ❖ This result has important implications for designing geoengineering solutions aiming at increasing CO₂ storage especially in situations where an efficient and continuous seal is absent.

2. Impact of reservoir wettability on CO₂ geo-sequestration in heterogeneous reservoirs
 - ❖ It has been confirmed that the effect of wettability on CO₂ geo-sequestration is valid for both the homogenous and heterogeneous reservoirs but the magnitude of these observations varies both in time and in space between the two cases.
 - ❖ CO₂-wet reservoirs have the highest CO₂ vertical migration, while water-wet reservoirs best retain CO₂.
 - ❖ Less residual CO₂ but more dissolved CO₂ is obtained in a CO₂-wet reservoir.
 - ❖ CO₂-wet reservoirs have the highest percentage of of mobile CO₂.

- ❖ Reservoir wettability has a critical impact on the CO₂ plume migration and trapping capacities, with water-wet reservoirs being preferable CO₂ sinks.
 - ❖ Reservoir wettability needs to be accurately reflected in CCS reservoir simulations in order to reliably assess CO₂ storage capacities and containment security.
3. Influence of injection well configuration on CO₂ plume behaviour and CO₂ trapping capacity in heterogeneous reservoirs
- ❖ The impact of the injection well configuration, horizontal versus three different vertical injection well scenarios (i.e. one, two and four vertical wells), on the CO₂ plume behaviour and CO₂ storage capacities for a wide range of rock wettabilities has been investigated, for the first time.
 - ❖ The injection well configuration has a major influence on CO₂ plume migration and on the amount of mobile, residual and dissolved CO₂.
 - ❖ Horizontal wells, for all wettability conditions, reduce CO₂ plume migration, CO₂ mobility and CO₂ solubility trapping.
 - ❖ Horizontal wells, for all wettability conditions, improve CO₂ residual trapping.
 - ❖ From a technical perspective horizontal injection wells are preferable as they increase storage capacity and containment security.
 - ❖ Using horizontal wells to inject the CO₂ leads to reduced costs associated with CO₂ geo-sequestration.
4. Effect of wettability heterogeneity on CO₂ storage efficiency in deep saline aquifers
- ❖ The effect of heterogeneous wettability distribution (at isothermal and non-isothermal conditions) on the vertical CO₂ plume migration, and capillary and dissolution trapping capacities has been investigated, for the first time.
 - ❖ Spatial heterogeneities in wettability distribution lead to accelerated CO₂ upwards migration for both isothermal and non-isothermal conditions.
 - ❖ Simulations made with a homogeneous wettability distribution predicted a lower vertical migration distance than those assuming a heterogeneous wettability distribution.
 - ❖ Heterogeneously distributed wettability significantly increases CO₂ mobility and solubility trapping.
 - ❖ Wettability heterogeneity reduces residual trapping.

- ❖ Wettability heterogeneity is an important factor in the context of CO₂ geo-storage, and heterogeneous wettability reduces storage capacity for both isothermal and non-isothermal conditions.
 - ❖ Wettability heterogeneity needs to be accurately modelled in reservoir-scale simulations in order to obtain reliable CO₂ storage predictions.
5. Impact of porosity and permeability heterogeneity on CO₂-plume migration and trapping capacity
- ❖ The CO₂-plume vertical migration distance is larger in the homogeneous reservoir than in the heterogeneous reservoir, for all different wettability conditions.
 - ❖ For all wettability conditions and for all amounts of CO₂ injected, the CO₂ plume vertical migration distance in the homogeneous model is approximately two times more than that computed for the heterogeneous model.
 - ❖ More mobile CO₂ (the majority of this mobility following a horizontal movement not a vertical migration) is computed for the heterogeneous reservoir than for the homogeneous reservoir model, for all wettability conditions.
 - ❖ Porosity and permeability heterogeneity decreases the percentages of residually trapped and dissolved CO₂.
 - ❖ Porosity and permeability heterogeneity needs to be incorporated into reservoir simulations for accurate predictions of both CO₂ plume behaviour and CO₂ storage capacities.
 - ❖ Not taking into account heterogeneities in the porosity and permeability distributions can lead to erroneous estimates of CO₂ storage capacities.
6. Effect of reservoir temperature on CO₂ storage efficiency in deep saline aquifers
- ❖ Reservoir temperature affects vertical CO₂ migration, and the associated capillary and dissolution trapping mechanisms.
 - ❖ Higher reservoir temperatures significantly accelerated the vertical CO₂ plume movement.
 - ❖ Higher reservoir temperatures increase CO₂ mobility and solubility trapping.
 - ❖ Higher reservoir temperatures reduce residual trapping.

- ❖ Lower temperatures are preferred.
7. Impact of salinity on CO₂ containment security in highly heterogeneous reservoirs
- ❖ A pair of relative permeability and capillary pressure curves for each brine salinity has been developed.
 - ❖ Lower brine salinities reduce CO₂ mobility and vertical CO₂ plume migration distance.
 - ❖ Lower brine salinity enhances capillary trapping and solubility trapping capacities.
 - ❖ Low salinity reservoirs are preferential CO₂ geo-sequestration sinks.
8. Enhancement of CO₂ trapping efficiency in heterogeneous reservoirs by Water Alternating Gas injection
- ❖ The impact of the WAG injection on the relative permeability and capillary pressure curves were simulated based on previous experimental studies.
 - ❖ WAG injection leads to a significant reduction in CO₂ leakage risk.
 - ❖ WAG injection has less than half of the vertical migration distance compared to continuous or intermittent injections.
 - ❖ WAG maximises residual (capillary) trapping by increasing the residual CO₂ saturation for the imbibition cycle.
 - ❖ WAG injection also enhanced solubility trapping.
 - ❖ WAG injection significantly reduces (unwanted) CO₂ mobility
 - ❖ WAG is therefore the preferred injection scheme due to improve CO₂ geo-sequestration efficiency.

9.2 Recommendations and Outlook for future work

Based on the outcomes of this thesis, the following further studies are recommended to be conducted in future work:

- ❖ The influence of rock wettability on mineral reaction and mineral trapping mechanism is an open research area and needs to further research work. Thus, it is recommended to perform reservoir simulation model using TOUGHREACT or any other possible software to test the mineral trapping efficiency for the different wettability conditions.

- ❖ The influence of wettability on CO₂ geo-sequestration efficiency in hydrocarbon reservoirs (oil and gas reservoirs) is not well investigated. Thus, more simulation and experimental works are required. This will need to use three phase relative permeability and capillary pressure curves (oil/water and gas/oil) to simulate the different wettability conditions.
- ❖ The wettability variation during the CO₂ injection process and its effect on the CO₂ storage efficiency in reservoir scale is not well understood. This will need to develop new simulators that are able to predict the wettability variation during the CO₂ injection with time. Then, the new wettability condition should be automatically implemented in the simulation model.
- ❖ In this thesis, to simulate the wettability heterogeneity, and due to the fact that until now there is no particular correlation available in literature studies between wettability distribution and other petrophysical or geological parameters (i.e. permeability, porosity and mineralogy), separate probability distributions have been used to randomly assign the five wettability conditions to different cells. Thus, future studies should focus on investigating the relationship between wettability distribution and the other petrophysical or geological parameters.

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Appendix A: Attribution of Authorship

Paper “Influence of CO₂-wettability on CO₂ migration and trapping capacity in deep saline aquifers” Greenhouse Gases: Science and Technology, 2017, 7 (2), 328-338.

Authors and full affiliations: Emad A. Al-Khdheawi^{a,c,*}, Stephanie Vialle^b, Ahmed Barifcani^a, Mohammad Sarmadivaleh^a, Stefan Iglauer^d

^a Department of Petroleum Engineering, Curtin University, 6151 Kensington, Western Australia, Australia.

^b Department of Exploration Geophysics, Curtin University, 6151 Kensington, Western Australia, Australia.

^c Petroleum Technology Department, University of Technology, Baghdad, Iraq

^d School of Engineering, Petroleum Engineering discipline, Edith Cowan University, Joondalup Western Australia 6027, Australia.

	Conception and design	Acquisition of data and method	Data conditioning and manipulation	Analysis and statistical method	Interpretation and discussion	Final approval
Dr. Stephanie Vialle						✓
I acknowledge that these represent my contribution to the above research output. Signature: [Redacted]						
Dr. Ahmed Barifcani						✓
I acknowledge that these represent my contribution to the above research output. Signature: [Redacted]						
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Dr. Stefan Iglauer						✓
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Paper “**Impact of reservoir wettability and heterogeneity on CO₂-plume migration and trapping capacity**” International Journal of Greenhouse Gas Control, 2017, 58 (2017), 328-338.

Authors and full affiliations: Emad A. Al-Khdheawi^{a,c,*}, Stephanie Vialle^b, Ahmed Barifcani^a, Mohammad Sarmadivaleh^a, Stefan Iglauer^d

^a Department of Petroleum Engineering, Curtin University, 6151 Kensington, Western Australia, Australia.

^b Department of Exploration Geophysics, Curtin University, 6151 Kensington, Western Australia, Australia.

^c Petroleum Technology Department, University of Technology, Baghdad, Iraq

^d School of Engineering, Petroleum Engineering discipline, Edith Cowan University, Joondalup Western Australia 6027, Australia.

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Dr. Stefan Iglauer						✓
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Paper “**Influence of injection well configuration and rock wettability on CO₂ plume behaviour and CO₂ trapping capacity in heterogeneous reservoirs**” Journal of Natural Gas Science and Engineering, 2017, 43 (2017), 190-206.

Authors and full affiliations: Emad A. Al-Khdheawi^{a,c,*}, Stephanie Vialle^b, Ahmed Barifcani^a, Mohammad Sarmadivaleh^a, Stefan Iglauer^d

^a Department of Petroleum Engineering, Curtin University, 6151 Kensington, Western Australia, Australia.

^b Department of Exploration Geophysics, Curtin University, 6151 Kensington, Western Australia, Australia.

^c Petroleum Technology Department, University of Technology, Baghdad, Iraq

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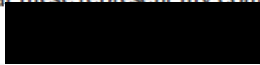
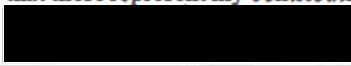


Authors and full affiliations: Emad A. Al-Khdheawi^{a,c,*}, Stephanie Vialle^b, Ahmed Barifcani^a, Mohammad Sarmadivaleh^a, Stefan Iglauer^d

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^b Department of Exploration Geophysics, Curtin University, 6151 Kensington, Western Australia, Australia.

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Paper “**Impact of salinity on CO₂ containment security in highly heterogeneous reservoirs**” Greenhouse Gases: Science and Technology, 2018, 8(1), 93-105.

Authors and full affiliations: Emad A. Al-Khdheawi^{a,c,*}, Stephanie Vialle^b, Ahmed Barifcani^a, Mohammad Sarmadivaleh^a, Yihuai Zhang^a, Stefan Iglauer^d

^a Department of Petroleum Engineering, Curtin University, 6151 Kensington, Western Australia, Australia.

^b Department of Exploration Geophysics, Curtin University, 6151 Kensington, Western Australia, Australia.

^c Petroleum Technology Department, University of Technology, Baghdad, Iraq

^d School of Engineering, Petroleum Engineering discipline, Edith Cowan University, Joondalup Western Australia 6027, Australia.

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Dr. Stefan Iglauer						✓
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Paper “**Enhancement of CO₂ trapping efficiency in heterogeneous reservoirs by Water Alternating Gas injection**” Greenhouse Gases: Science and Technology, 2018, 8(5), 920-931.

Authors and full affiliations: Emad A. Al-Khdheawi^{a,c,*}, Stephanie Vialle^b, Ahmed Barifcani^a, Mohammad Sarmadivaleh^a, Stefan Iglauer^d

^a Department of Petroleum Engineering, Curtin University, 6151 Kensington, Western Australia, Australia.

^b Department of Exploration Geophysics, Curtin University, 6151 Kensington, Western Australia, Australia.

^c Petroleum Technology Department, University of Technology, Baghdad, Iraq

^d School of Engineering, Petroleum Engineering discipline, Edith Cowan University, Joondalup Western Australia 6027, Australia.

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